

ANALIZNA KEMIJA I

UČNI NAČRT PREDMETA/COURSE SYLLABUS

Predmet:	Analizna kemija I
Course title:	Analytical Chemistry I
Članica nosilka/UL Member:	UL FKKT

Študijski programi in stopnja	Študijska smer	Letnik	Semestri	Izbirnost
Kemijska tehnologija, prva stopnja, visokošolski strokovni	Ni členitve (študijski program)	2. letnik	zimski	obvezni

Univerzitetna koda predmeta/University course code:	0072029
Koda učne enote na članici/UL Member course code:	KT115

Predavanja /Lectures	Seminar /Seminar	Vaje /Tutorials	Klinične vaje /Clinical tutorials	Druge oblike študija /Other forms of study	Samostojno delo /Individual student work	ECTS
45	15	15 LV			75	5

Nosilec predmeta/Lecturer: prof. dr. Mitja Kolar

Vrsta predmeta/Course type: obvezni/mandatory

Jeziki/Languages:

Predavanja/Lectures:	Slovenščina
Vaje/Tutorial:	Slovenščina

Pogoji za vključitev v delo oz. za opravljanje študijskih obveznosti:

Študent oz. kandidat mora imeti predmet opredeljen kot študijsko obveznost.

Prerequisites:

The course has to be assigned to the student.

Vsebina:

- Uvod (opredelitev in pomen analizne kemije, delitve na področju analizne kemije in temeljni analizni pojmi, kemik analitik pričakovane lastnosti).
- Stopnje celovitega analiznega procesa (splošno opredeljen problem, analizno oblikovano vprašanje, izbira metode, odvzem vzorca, transport in shranjevanje, priprava vzorca, meritve, vrednotenje rezultatov, oblikovanje sklepov in dokumentacija).
- Rezultat v analizni kemiji in vrednotenje rezultatov (aritmetična sredina, mediana, modus, zaokroževanje in signifikantna mesta, standardni odklon in izpeljanke, merilna negotovost, pravilnost in ponovljivost rezultatov, vrste napak, zagotavljanje sledljivosti rezultatov do osnovnega merskega sistema, zahteva po zanesljivih rezultatih).

Content (Syllabus outline):

- Introduction (definition and role of analytical chemistry, classifications in analytical chemistry, terminology, essential skills and abilities of an analytical chemist).
- The total analytical process (general definition of a problem, specific analytical statement of a problem, selection of a procedure, sampling, sample transport and storage, sample preparation, measurement, evaluation of data, conclusions and report).
- Results in analytical chemistry – evaluation and presentation (arithmetic mean, median, mode, significant figures, standard deviation and derived statistical parameters, measurement uncertainty, accuracy, precision, sources and types of errors, traceability, requirements for reliable results).

<p>- Metode kalibracije (metoda kalibracijske premice, metoda standardnih dodatkov, metoda interne standarda, linearna regresija).</p> <p>- Izbira in vpeljava analizne metode (parametri za vrednotenje metode, vidik ekonomičnosti, pomen validacije metode, zagotavljanje kakovosti, nadzorni grafi).</p> <p>- Razvojne smeri v analizi kemiji (miniatrizacija in on-line senzori, spremljanje meritev na daljavo, avtomatizacija in robotizacija, vezave instrumentov v mreže, resnično inteligentni analizni sistemi, kompleksne metode za reduciranje podatkov).</p> <p>- Gravimetrija (stopnje gravimetričnega postopka, vzroki nečistoč, primeri gravimetričnih določitev).</p> <p>- Titrimetrija (Delitev glede na merjeno količino za nadzor nad dodanim titrirnim sredstvom, kemijsko osnovo in načine ugotavljanje končne točke. Obarjalne titracije: potek titracijske krivulje, načini ugotavljanja končne točke, primeri določitev. Nevtralizacijske titracije: potek titracijske krivulje za različne primere titracij, analizna uporaba in omejitve. Kompleksometrične titracije: titritna sredstva, načini ugotavljanja končne točke, analizne aplikacije. Redoks titracije: potek titracijske krivulje, vrste indikatorjev, analizne aplikacije.)</p> <p>Vaje: Identifikacijske reakcije analiznega pomena.</p>	<p>- Methods of calibration (method of calibration function, method of standard additions, method of internal standard, linear regression).</p> <p>- Selecting and introducing analytical methods into laboratory practice (characteristics of analytical methods, method validation, quality assurance and quality control, control charts, financial aspects).</p> <p>- Development in analytical chemistry (miniaturisation, on-line sensors, remote sensing, automatization, robotization, intelligent analytical systems, networking of analytical instruments, complex methods of data reduction).</p> <p>- Gravimetric analytical methods (gravimetric procedure, sources of impurities, examples of analytical applications)</p> <p>- Titrimetric analytical methods (Subcategories concerning measuring quantity, end point determination and chemical background. Precipitation titrations: titration curve, indication of the end point of titration, analytical applications. Neutralisation titrations: titration curve, applications and limitations. Complexometric titrations: titration reagents, indication of the end point of titration, analytical applications. Reduction-oxidation titrations: titration curve, types of indicators, analytical applications.)</p> <p>Laboratory practical – identification reactions of analytical importance.</p>
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Temeljna literatura in viri/Readings:

- Gary D. Christian, Analytical Chemistry, Wiley, 5th edition or later. 228 pages.

Cilji in kompetence:

Pri predmetu študent osvoji temelje analize kemije, spozna analizni pristop in temelje in uporabo klasičnih analiznih metod ter se usposobi za vrednotenje rezultatov in uporabo metod kalibracije.

Objectives and competences:

Student learns the fundamentals of analytical chemistry, analytical approach and classical analytical methods and develops skills necessary for evaluation of analytical results and calibration.

Predvideni študijski rezultati:

Znanje in razumevanje
Študent razume temelje analize kemije in analizni pristop ter osnovo in uporabo klasičnih analiznih metod.

Uporaba
Študent zna podajati in vrednotiti analizne rezultate ter razlikuje med različnimi metodami kalibracije.

Študent obvlada računske vidike obravnavanih analiznih metod.

Refleksija
Študent ima kritičen odnos do analiznega rezultata in obravnavanih analiznih metod.

Prenosljive spretnosti
Zmožnost statističnega vrednotenja rezultatov, uporaba linearne regresije.

Intended learning outcomes:

Knowledge and Comprehension
Student understands the fundamentals of analytical chemistry, analytical approach and background and applications of classical analytical methods.

Application
Student develops ability of presenting and evaluating analytical results and differentiates calibration methods. Student masters calculation procedures related to the classical analytical methods.

Analysis
Student develops critical attitude towards analytical results and analytical methods.

Skill-transference Ability
Statistical evaluation of results and linear regression.

Metode poučevanja in učenja:

Learning and teaching methods:

Predavanja, vodeni razgovor, sodelovalno učenje, reševanje problemov.	Lectures, guided discussions, cooperative learning, problem solving.
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Načini ocenjevanja:
Delež/Weight
Assessment:

Vaje: uspešno opravljen praktični preizkus iz identifikacije ionov je predpogoj za uspešno opravljene vaje in pristop h kolokviju. Praktični preizkus in kolokvij imata enako težo pri končni oceni vaj.	33,30 %	Tutorial: successfully accomplished assessment in identification of ions is a precondition to sit the written assessment. Both assessments contribute equally to the grade of the tutorial.
Predmet: Pisni izpit. Predpogoj za izpit so uspešno zaključene vaje.	66,70 %	Lectures: Written exam. Successfully accomplished tutorial is a precondition to sit the exam.

Reference nosilca/Lecturer's references:

- JERENEC, Simona, ŠIMIČ, Mario, SAVNIK, Aleš, PODGORNİK, Aleš, **KOLAR, Mitja**, TURNŠEK, Marko, KRAJNC, Peter. Glycidyl methacrylate and ethylhexyl acrylate based polyhipe monoliths : morphological, mechanical and chromatographic properties. *Reactive & functional polymers*, ISSN 1381-5148. [Print ed.], 2014, vol. 78, str. 32-37, doi: 10.1016/j.reactfunctpolym.2014.02.011. [COBISS.SI-ID 17661718]
- BRGLEZ, Polonca, HOLOBAR, Andrej, PIVEC, Aleksandra, **KOLAR, Mitja**. Spin-coating for optical-oxigen-sensor preparation = Uporaba spinskega nanosa pri izdelavi optičnih senzorjev za kisik. *Materiali in tehnologije*, ISSN 1580-2949. [Tiskana izd.], mar.-apr. 2014, letn. 48, št. 2, str. 181-188. [COBISS.SI-ID17746454],
- **KOLAR, Mitja**, DOLIŠKA, Aleš, ŠVEGL, Franc, KALCHER, Kurt. Tungsten - tungsten trioxide electrodes for the long-term monitoring of corrosion processes in highly alkaline media and concrete-based materials. *Acta chimica slovenica*, ISSN 1318-0207. [Tiskana izd.], 2010, vol. 57, no. 4, str. 813-820, graf. prikazi.<http://acta.chem-soc.si/57/57-4-813.pdf>. [COBISS.SI-ID 14701078]

ANALIZNA KEMIJA II

UČNI NAČRT PREDMETA/COURSE SYLLABUS

Predmet:	Analizna kemija II
Course title:	Analytical Chemistry II
Članica nosilka/UL Member:	UL FKKT

Študijski programi in stopnja	Študijska smer	Letnik	Semestri	Izbirnost
Kemijska tehnologija, prva stopnja, visokošolski strokovni	Ni členitve (študijski program)	2. letnik	poletni	obvezni

Univerzitetna koda predmeta/University course code:	0072030
Koda učne enote na članici/UL Member course code:	KT130

Predavanja /Lectures	Seminar /Seminar	Vaje /Tutorials	Klinične vaje /Clinical tutorials	Druge oblike študija /Other forms of study	Samostojno delo /Individual student work	ECTS
30		45 LV			75	5

Nosilec predmeta/Lecturer: prof. dr. Mitja Kolar

Vrsta predmeta/Course type: obvezni/mandatory

Jeziki/Languages:

Predavanja/Lectures:	Slovenščina
Vaje/Tutorial:	Slovenščina

Pogoji za vključitev v delo oz. za opravljanje študijskih obveznosti: **Prerequisites:**

Študent oz. kandidat mora imeti predmet opredeljen kot študijsko obveznost.	The course has to be assigned to the student.
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Vsebina:

Elektrokemijske metode:

- Potencimetrija (značilnosti metode, merilni sistem, referenčne, konvencionalne in ionskoselektivne elektrode ter njihova analizna uporaba, prijemi za potenciometrično določanje koncentracije neke zvrsti, potenciometrične titracije)
- Voltometrija (značilnosti metode, merilni sistem, voltamogram, polarografija, hidrodinamska voltometrija, uporaba v pretočnih in titracijskih sistemih, voltometrični oz. amperometrični senzorji)
- Konduktometrija (značilnosti metode, analizna uporaba)
- Kulometrija (značilnosti metode, analizna uporaba)

Spektroskopske metode:

- Molekularna absorpcijska spektrometrija (značilnosti metode, različne gradnje spektrometrov in preverjanje

Content (Syllabus outline):

Electrochemical methods:

- Potentiometry (method characteristics, measuring system, types of electrodes, approaches for determining concentration, potentiometric titrations)
- Voltametry (method characteristics, measuring system, voltamogram, polarography, hydrodynamic voltametry, applications in flow systems and titrimetry, voltametric/amperometric sensors)
- Conductometry (method characteristics, analytical applications)
- Coulometry (method characteristics, analytical applications)

Spectroscopic methods:

- Molecular absorption spectrometry (method characteristics, different types of spectrometers, testing performance of a spectrometer, Beer-

<p>spektrometra, Beer-Lambertov zakon in njegova veljavnost, uporaba spektrometrije v UV-VIS in IR področju, diferenčna spektrometrija)</p> <ul style="list-style-type: none"> - Molekularna fluorescenčna spektrometrija (temelj metode, instrument, uporaba) - Plamenska emisijska spektrometrija (temelj metode, plamenski spektrometer, značilnosti metode in uporaba) - Atomska emisijska spektrometrija (spektrografska analiza, spektroskopske tehnike na osnovi induktivno sklopljene plazme) - Atomska absorpcijska spektrometrija (temelj metode, atomski absorpcijski spektrometer s plamensko oz. elektrotermično atomizacijo, značilnosti metode in uporaba) - Pregled drugih pogostih spektroskopskih metod <p>Vaje osvetljuje praktične vidike analiznih vsebin obravnavanih v tem in predhodnem semestru. Pri vajah se študent usposobi za uporabo klasičnih in instrumentalnih analiznih metod ter uporabi že pridobljena znanja s področja kalibracije in vrednotenja rezultatov.</p>	<p>Lambert's Law and its limitations, UV-VIS and IR spectrometry, measurements at two or more wavelengths)</p> <ul style="list-style-type: none"> - Molecular fluorescence spectrometry (fundamentals, instrumentation, applications) - Flame emission spectrometry (fundamentals, instrumentation, applications, characteristics of the method) - Atomic emission spectrometry (spectrographic analyses, spectroscopic techniques based on the inductively coupled plasma) - Atomic absorption spectrometry (fundamentals, spectrometer with flame and electrothermal atomisation, characteristics of the method, applications) - Other spectroscopic methods – an overview. <p>Practical: Classical and instrumental analytical methods. Calibration and evaluation of analytical results.</p>
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Temeljna literatura in viri/Readings:

- Francis, Rouessac, Annick Rouessac, Chemical Analysis - Modern Instrumental Methods, Wiley, 2000, Chichester. Or later editions. Chapters 11 to 16 and 18 and 19. 153 pages.

Cilji in kompetence:

Študent pridobi znanja o elektrokemijskih in spektroskopskih analiznih metodah ter njihovi uporabi. Pri vajah se usposobi za praktično izvedbo klasičnih in instrumentalnih analiznih metod.

Objectives and competences:

Students are acquainted with electrochemical and spectroscopic analytical methods and their applications and develop laboratory skills in classical and instrumental analytical methods.

Predvideni študijski rezultati:

Znanje in razumevanje
Študent razume temelje in uporabo obravnavanih analiznih metod.
Uporaba
Študent zna izvesti analizne postopke in meritve vezane na obravnavane analizne metode ter ovrednotiti dobljene rezultate.
Refleksija
Študent je kritičen do dobljenih rezultatov in se zaveda omejitev analiznih metod.
Prenosljive spretnosti
Laboratorijske spretnosti, statistično vrednotenje in pravilno podajanje rezultatov.

Intended learning outcomes:

Knowledge and Comprehension
Student fosters understanding of fundamentals and applications of analytical methods.
Application
Student is able to perform analyses and evaluate analytical results.
Analysis
Student develops a critical attitude towards analytical result and is aware of the limitation of analytical methods.
Skill-transference Ability
Laboratory skills, statistical methods, expression of analytical results.

Metode poučevanja in učenja:

Predavanja, vodeni razgovor, sodelovalno učenje, reševanje problemov.

Learning and teaching methods:

Lectures, guided discussions, cooperative learning, problem solving.

Načini ocenjevanja:

Delež/Weight Assessment:

<p>Vaje: kakovost analiznih rezultatov (33,3 %) in kolokvij (66,7 %). Uspešno zaključen praktični del je predpogoj za kolokvij. Predmet: Pisni izpit. Predpogoj za izpit so uspešno zaključene vaje.</p>		<p>Tutorial: quality of analytical results (33.3 %) written assessment (66.7 %). Successfully accomplished laboratory work is a precondition to sit the written assessment. Lectures: Written exam. Successfully accomplished tutorial is a precondition to sit the exam.</p>
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Reference nosilca/Lecturer's references:

1. HUŠ, Sebastjan, KOLAR, Mitja, KRAJNC, Peter. Separation of heavy metals from water by functionalized glycidylmethacrylate poly (high internal phase emulsions). *Journal of chromatography. A*, ISSN 0021-9673, 2016, vol. 1437, str. 168-175.
2. JERENEC, Simona, ŠIMIĆ, Mario, SAVNIK, Aleš, PODGORNIK, Aleš, KOLAR, Mitja, TURNŠEK, Marko, KRAJNC, Peter. Glycidyl methacrylate and ethylhexyl acrylate based polyhipe monoliths : morphological, mechanical and chromatographic properties. *Reactive & functional polymers*, ISSN 1381-5148. [Print ed.], 2014, vol. 78, str. 32-37.
3. KOLAR, Mitja, DOBČNIK, Danilo, RADIĆ, Njegomir. Chemically treated silver electrodes for the determination of cysteine. *Mikrochimica acta*, ISSN 1436-5073. [Online ed.], 2002, vol. 138, no 1-2, str. 23-27 15.

ANALIZNA KEMIJA III

UČNI NAČRT PREDMETA/COURSE SYLLABUS

Predmet:	Analizna kemija III
Course title:	Analytical Chemistry III
Članica nosilka/UL Member:	UL FKKT

Študijski programi in stopnja	Študijska smer	Letnik	Semestri	Izbirnost
Kemijska tehnologija, prva stopnja, visokošolski strokovni	Ni členitve (študijski program)	3. letnik	zimski	obvezni

Univerzitetna koda predmeta/University course code:	0072045
Koda učne enote na članici/UL Member course code:	KT131

Predavanja /Lectures	Seminar /Seminar	Vaje /Tutorials	Klinične vaje /Clinical tutorials	Druge oblike študija /Other forms of study	Samostojno delo /Individual student work	ECTS
30	30	15 LV			75	5

Nosilec predmeta/Lecturer: doc. dr. Nataša Gros

Vrsta predmeta/Course type: obvezni/mandatory

Jeziki/Languages:

Predavanja/Lectures:	Slovenščina
Vaje/Tutorial:	Slovenščina

Pogoji za vključitev v delo oz. za opravljanje študijskih obveznosti:

Študent oz. kandidat mora imeti predmet opredeljen kot študijsko obveznost.

Prerequisites:

The course has to be assigned to the student.

Vsebina:

Separacijske metode:

- Temelji kromatografije in delitev kromatografskih tehnik
- Tekočinska kromatografija visoke ločljivosti (enote HPLC sistema, vrste polnitev kolon, normalno-fazna, reverzno-fazna, velikostno izključitvena kromatografija, vrste detektorjev, analizne aplikacije)
- Ionska kromatografija (delitve ionske kromatografije, enote ionskega kromatografa, polnitve kolon, analizne aplikacije)
- Tenkoplastna kromatografija (temeljni pojmi in izvedbe, identifikacija in kvantifikacija ločenih komponent, analizne aplikacije)
- Plinska kromatografija (enote plinskega kromatografa, različne izvedbe injektorjev, vrste kolon, detektorji, analizne aplikacije)

Content (Syllabus outline):

Separation methods:

- Fundamentals of chromatography, different chromatographic methods – an overview
- High performance liquid chromatography (HPLC system, chromatographic columns, different techniques in HPLC, detection, analytical applications)
- Ion chromatography (classification in ion chromatography, ion chromatographic systems, chromatographic columns, analytical applications)
- Thin layer chromatography (terminology, designs, identification and quantification, analytical applications)
- Gas chromatography (instrumentation, columns, detectors, analytical applications)

<p>- Kapilarna elektroforeza (elektroforezne tehnike, elektro-osmotski tok, elektroferogram, analize aplikacije)</p> <p>- Sklopitve kromatografskih tehnik z masnim spektrometrom</p> <p>Prikaz, vrednotenje in interpretacija analiznih rezultatov.</p> <p>- Box and Whisker graf, histogramski in poligonski prikazi, test normalne porazdelitve</p> <p>- Interval zaupanja in statistični testi (F-test, različne izvedbe t-testa, Q-test)</p> <p>- Uporaba linearne regresije za primerjavo rezultatov dveh metod.</p> <p>Vaje nadgradijo in razširijo nabor instrumentalnih analiznih metod, ki so jih študentje uporabljali pri vajah v drugem letniku in vključijo dodatne vidike vrednotenja in interpretacije rezultatov.</p>	<p>- Capillary electrophoresis (techniques, electro-osmotic flow, electroferogram, analytical applications)</p> <p>- Hyphenation of chromatographic techniques with mass spectrometry</p> <p>Presentation, evaluation and interpretation of analytical results</p> <p>- Box and Whisker graph, histogram, polygon, test of normal distribution</p> <p>- Confidence interval, statistical tests (F-test, t-test, Q-test)</p> <p>- Use of linear regression for a comparison of the results of two analytical methods.</p> <p>Laboratory class: Instrumental analytical methods and evaluation and interpretation of results.</p>
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Temeljna literatura in viri/Readings:

<p>- Francis, Rouessac, Annick Rouessac, Chemical Analysis - Modern Instrumental Methods, Wiley, 2000, Chichester. Or later editions. Chapters 1 to 5 and 8. 106 pages.</p> <p>- James N. Miller, Jane C. Miller, Statistics and Chemometrics for Analytical Chemistry, 6th Edition, Pearson Education, 2010, Harlow. Chapters 2, 3, 5.10, 6.2. 65 pages.</p>

Cilji in kompetence:

<p>Študent pridobi znanja o separacijskih analiznih metodah. Pri vajah se usposobi za praktično izvedbo dodatnega nabora instrumentalnih analiznih metod. Zmožnost predstavitve, vrednotenja in interpretacije rezultatov osvojene v drugem letniku študent razširi z dodatnimi statističnimi prijemi.</p>	<h3>Objectives and competences:</h3> <p>Understanding and knowledge of separation methods. Experimental skills and ability of using instrumental analytical methods. Upgraded knowledge in interpretation and evaluation of analytical results.</p>
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Predvideni študijski rezultati:

<p>Znanje in razumevanje</p> <p>Študent razume temelje in uporabo obravnavanih analiznih metod. Študent osvoji dodatna znanja za predstavitve, vrednotenje in interpretacijo analiznih rezultatov.</p> <p>Uporaba</p> <p>Študent zna izvesti analizne postopke in meritve vezane na obravnavane analizne metode ter ovrednotiti dobljene rezultate.</p> <p>Refleksija</p> <p>Študent je kritičen do dobljenih rezultatov in se zaveda omejitev analiznih metod.</p> <p>Prenosljive spretnosti</p> <p>Laboratorijske spretnosti, statistično vrednotenje in interpretacija rezultatov.</p>	<h3>Intended learning outcomes:</h3> <p>Knowledge and Comprehension</p> <p>Student fosters understanding of fundamentals and applications of analytical methods. Ability of evaluating, presenting and interpreting analytical results.</p> <p>Application</p> <p>Student is able to perform analyses and evaluate analytical results.</p> <p>Analysis</p> <p>Student develops a critical attitude towards analytical result and is aware of the limitation of analytical methods.</p> <p>Skill-transference Ability</p> <p>Laboratory skills, statistical methods, interpretation of analytical results.</p>
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Metode poučevanja in učenja:

<p>Predavanja, vodeni razgovor, sodelovalno učenje, reševanje problemov.</p>	<h3>Learning and teaching methods:</h3> <p>Lectures, guided discussions, cooperative learning, problem solving.</p>
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Načini ocenjevanja:

Delež/Weight Assessment:

Vaje: Esej: »Ovrednotenje kakovosti naravne vode«.	33,30 %	Tutorial: Essay: "Evaluation of a natural water quality".
Predmet: Pisni izpit. Predpogoj za izpit so uspešno zaključene vaje.	66,70 %	Lectures: Written exam. Successfully accomplished tutorial is a precondition to sit the exam.

Reference nosilca/Lecturer's references:

- **GROS, Nataša.** Microdiffusion-based UV-LED spectrometric setup for determining low levels of ethanol in fruit juice. *Talanta*, ISSN 0039-9140. [Print ed.], 2011, vol. 87, no. 1, str. 174-179.
- **GROS, Nataša, CAMÕES, Maria Filomena, OLIVEIRA, Cristina, SILVA, M. C. R.** Ionic composition of seawaters and derived saline solutions determined by ion chromatography and its relation to other water quality parameters. *Journal of chromatography. A*, ISSN 0021-9673, 2008, vol. 1210, no. 1, str. 92-98.
- **GROS, Nataša, VRTAČNIK, Margareta.** A small-scale low-cost gas chromatograph. *Journal of chemical education*, ISSN 0021-9584, 2005, vol. 82, no. 2, str. 291-293 + supplemental material. [COBISS.SI-ID 1246044]

ANORGANSKA KEMIJA

UČNI NAČRT PREDMETA/COURSE SYLLABUS

Predmet:	Anorganska kemija
Course title:	Inorganic Chemistry
Članica nosilka/UL Member:	UL FKKT

Študijski programi in stopnja	Študijska smer	Letnik	Semestri	Izbirnost
Kemijska tehnologija, prva stopnja, visokošolski strokovni	Ni členitve (študijski program)	1. letnik	poletni	obvezni

Univerzitetna koda predmeta/University course code:	0072017
Koda učne enote na članici/UL Member course code:	KT107S

Predavanja /Lectures	Seminar /Seminar	Vaje /Tutorials	Klinične vaje /Clinical tutorials	Druge oblike študija /Other forms of study	Samostojno delo /Individual student work	ECTS
60	15				75	5

Nosilec predmeta/Lecturer: doc. dr. Saša Petriček

Vrsta predmeta/Course type: obvezni/mandatory

Jeziki/Languages:

Predavanja/Lectures:	Slovenščina
Vaje/Tutorial:	Slovenščina

Pogoji za vključitev v delo oz. za opravljanje študijskih obveznosti: **Prerequisites:**

Študent oz. kandidat mora imeti predmet opredeljen kot študijsko obveznost.	The course has to be assigned to the student.
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Vsebina:

Periodni sistem kot osnova sistematike elementov in anorganskih spojin.

Vodik in kisik ter njune spojine

Vodik. Kisik. Voda. Vodikov peroksid.

VII. skupina periodnega sistema

Elementi VII. skupine. Spojine elementov VII. skupine z vodikom. Spojine s kisikom, oksokisljine in oksosoli.

VI. skupina periodnega sistema

Elementi VI. skupine. Spojine elementov VI. skupine z vodikom. Oksidi in oksospojine žvepla. Spojine s halogeni.

V. skupina periodnega sistema

Elementi V. skupine periodnega sistema. Spojine elementov V. skupine z vodikom. Oksidi in oksospojine.

Content (Syllabus outline):

Introduction: Periodic table of the elements. Hydrogen and oxygen and their compounds. Group 17 elements, hydrogen halides and their salts, oxoacids and their salts. Group 16 elements, compounds of sulphur. Group 15 elements, compounds of nitrogen and phosphorus. Group 14 elements, compounds of carbon, silicon, tin and lead. Group 13 elements, compounds of boron and aluminium. Group 1 and 2 elements and their compounds. Compounds of d-elements, coordination and organometallic compounds.

<p>IV. skupina periodnega sistema Elementi IV. skupine periodnega sistema. Spojine elementov IV. skupine z vodikom. Oksidi, oksospojine in soli.</p> <p>III. skupina periodnega sistema Elementi III. skupine periodnega sistema. Bor in njegove spojine. Aluminij in njegove spojine.</p> <p>II. in I. skupina periodnega sistema Elementi I. in II. skupine periodnega sistema. Kemizem zemeljskoalkalijskih kovin. Kemizem alkalijskih kovin.</p> <p>Pregled kemije prehodnih elementov d-orbitale in njihova vloga v kemiji prehodnih elementov. Sistematika prehodnih elementov. Elementi in njihove kemijske lastnosti. Oksidi, hidroksidi in oksokislina prehodnih elementov. Koordinacijske spojine.</p>	
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Temeljna literatura in viri/Readings:

Temeljna literatura in viri / Readings:

F. LAZARINI in J. V. BRENČIČ, *Splošna in anorganska kemija*, UL FKKT, Ljubljana, 2004, 557 str. (30%)

B. Čeh, *Anorganska kemija*, UL FKKT, 2019, 347 strani (40%)

Cilji in kompetence:

Študenti pri predmetu spoznajo lastnosti elementov periodnega sistema in njihovih spojin. Periodni sistem elementov je študentu osnova razumevanja in ureditev velikega števila kemijskih zakonitosti in dejstev v obvladljivo shemo. Specifično, znanje anorganske kemije je kot osnovno kemijsko znanje potrebno za večino predmetov nadaljnega študija. To znanje študentu omogoča strokoven pristop k analizi in reševanju problemov, s katerimi se bo kot kemik srečal pri delu.

Objectives and competences:

Objectives: Understanding the basic principles of the properties and reactivity of inorganic compounds.

Competences: Ability to use periodic table of the elements for the systemization of properties of substances and their reactivity. The basic knowledge of inorganic chemistry is needed for most of the later courses and allows the students analysis and solving of chemical problems.

Predvideni študijski rezultati:

Znanje in razumevanje

Študent dobi osnovno znanje iz anorganske kemije, ki se nadgradi na predznanje iz predmeta Splošna kemija. Z uporabo zakonitosti splošne kemije je sposoben razumeti in v nekaterih primerih predvideti lastnosti anorganskih spojin.

Uporaba

Znanje anorganske kemije je študent sposoben uporabiti pri drugih kemijskih predmetih. To znanje potrebuje pri reševanju problemov, za komunikacijo o kemijskih temah in za splošno strokovno izobrazbo.

Refleksija

Študent je sposoben oceniti pomen teoretskega znanja za razlago eksperimentalnih dejstev in izkušenj. Pereča kemijska vprašanja bo znal ovrednotiti in najti potrebne dodatne podatke in znanja.

Prenosljive spretnosti

Intended learning outcomes:

Knowledge and Comprehension

To the basic knowledge from General Chemistry course, the students add the understanding of the basic principles of the properties and reactivity of inorganic compounds.

Application

The students are able to use the knowledge in other chemical courses, to solve chemical problems and to get broad chemical education.

Analysis

The student understand the importance of theoretical knowledge and experiments. They will be able to find additional data and knowledge to solve their tasks.

Skill-transference Ability

The students learned that vast amount of knowledge on inorganic compounds was gained by experimental work and that similar is valid for most scientific disciplines. He will be able to find relevant data on his topic of interest. He will be able to design

Študent spozna, da je obstoječe znanje anorganske kemije nastalo z raziskovanjem snovi in da je raziskovalni način nastajanja znanja skupen večini področij. Znal bo ovrednotiti poplavo podatkov na spletu in najti relevantne vire podatkov. Sposoben bo zasnovati poskuse, znal bo izmeriti veličine in jih dokumentirati. Rezultate bo sposoben oceniti in razložiti.	experiments, document the measurements and evaluate the results.
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Metode poučevanja in učenja:

Predavanja in seminarji.

Learning and teaching methods:

Lectures and seminars.

Načini ocenjevanja:

Pisni izpit. Če študent na 2 testih med semestrom zbere vsaj 50 % možnih točk, je oproščen pisnega izpita. Ocenjevalna lestvica v skladu z enotno lestvico ECTS na Univerzi v Ljubljani: 6 – 10 opravil izpit 1 – 5 ni opravil izpita

Delež/Weight

Assessment:

Written exam. Written exam could be done by 2 tests with at least 50% in average. Notes: 6-10 (pass the exam) and 1-5 (does not pass the exam).

Reference nosilca/Lecturer's references:

- DEMŠAR, Alojz, KOŠMRLJ, Janez, **PETRIČEK, Saša**. Variable-temperature nuclear magnetic resonance spectroscopy allows direct observation of carboxylate shift in zinc carboxylate complexes. *Journal of the American Chemical Society*, 2002, vol. 124, no. 15, str. 3951-3958, ISSN 0002-7863.
- **PETRIČEK, Saša**. Syntheses of lanthanide bromide complexes from oxides and the crystal structures of $[\text{LnBr}_3(\text{DME})_2]$ (Ln = Pr, Nd, Sm, Eu), $[\text{LnBr}_3(\text{THF})_4]$ (Ln = Pr, Sm) and $[\text{EuBr}_2(\text{THF})_5][\text{EuBr}_4(\text{THF})_2]$. *Polyhedron*, 2004, vol. 23, no. 14, str. 2293-2301, ISSN 0277-5387.
- **PETRIČEK, Saša**, DEMŠAR, Alojz. Syntheses and crystal structures of manganese, nickel and zinc chloride complexes with dimethoxyethane and di(2-methoxyethyl) ether. *Polyhedron*, 2010, vol. 29, no. 18, str. 3329-3334, ISSN 0277-5387.

BIOTEHNOLOŠKI PROCESI IN NAPRAVE

UČNI NAČRT PREDMETA/COURSE SYLLABUS

Predmet:	Biotehnološki procesi in naprave
Course title:	Processes and Equipment in Biotechnology
Članica nosilka/UL Member:	UL FKKT

Študijski programi in stopnja	Študijska smer	Letnik	Semestri	Izbirnost
Kemijska tehnologija, prva stopnja, visokošolski strokovni	Ni členitve (študijski program)	3. letnik	poletni	obvezni

Univerzitetna koda predmeta/University course code:	0100214
Koda učne enote na članici/UL Member course code:	KTSI5

Predavanja /Lectures	Seminar /Seminar	Vaje /Tutorials	Klinične vaje /Clinical tutorials	Druge oblike študija /Other forms of study	Samostojno delo /Individual student work	ECTS
45	15	15 LV			75	5

Nosilec predmeta/Lecturer: prof. dr. Andreja Žgajnar Gotvajn, prof. dr. Polona Žnidaršič Plazl

Vrsta predmeta/Course type: izbirni strokovni/Elective Professional

Jeziki/Languages:	Predavanja/Lectures:	Slovenščina
	Vaje/Tutorial:	Slovenščina

Pogoji za vključitev v delo oz. za opravljanje študijskih obveznosti:

Študent oz. kandidat mora imeti predmet opredeljen kot študijsko obveznost.

Prerequisites:

The course has to be assigned to the student.

Vsebina:

Osnove načrtovanja, vodenja in analize bioprosesov. Snovne bilance, bioreakcijska kinetika in prenos kisika v bioprosesih. Izbrani primeri industrijskih procesov: proizvodnja antibiotikov, organskih kislin in aminokislin, kortikosteroidov, biofarmacevtikov. Vloga biotehnologije pri valorizaciji lignocelulozne biomase in v krožnem gospodarstvu. Vpliv farmacevtske industrije na okolje. Problematika farmacevtskih učinkovin in mikropolutantov v okolju. Napredne tehnologije za čiščenje farmacevtskih odpadnih vod. Uporaba biotehnološki procesov za odstranjevanje onesnaženja iz odpadnih vod. Biološke in rastlinske čistilne naprave. Odstranjevanje odpadkov s kompostiranjem in anaerobno razgradnjo. Koncept biorafinerij in trajnostne biotehnoške proizvodnje;

Content (Syllabus outline):

Basic principles of bioprocess design, operation, and analysis. Mass balances, bioreaction kinetics and oxygen transfer in bioprocesses. Selected industrial bioprocesses: antibiotics, organic acids, amino acids, corticosteroids, biopharmaceuticals production. The role of biotechnology in lignocellulose biomass valorization and in circular economy. Impacts of pharmaceutical industry on the environment. Pharmaceuticals and micropollutants in the environment. Advanced processes for wastewater treatment. Biotechnological processes for pollutant removal from wastewaters. Biological wastewater treatment plants and constructed wetlands. Degradation of solid waste with composting and anaerobic degradation.

načrtovanje in optimizacija tehnologiji za čiščenje farmacevtskih odpadnih vod.	The concept of biorefineries and sustainable biotechnological production; Design and optimisation of advanced processes for wastewater treatment.
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Temeljna literatura in viri/Readings:

<ul style="list-style-type: none"> - A. Žgajnar Gotvajn, I. Boševski, M. Čehovin, Napredni oksidacijski in hibridni procesi v postopkih čiščenja odpadnih vod, Fakulteta za kemijo in kemijsko tehnologijo UL, Ljubljana, 2023 (10%) - Ratlidge, C., Kristiansen, B. Basic biotechnology, 3rd Ed. Cambridge University Press, Cambridge, 2006 (10%) - Žnidaršič Plazl, P., Podgornik, H. Vaje iz biotehnologije, 2. izdaja. Fakulteta za kemijo in kemijsko tehnologijo UL, Ljubljana, 2011 (15%) - Žnidaršič Plazl, P., Pavko, A. Praktikum iz biokemijskega inženirstva. Fakulteta za kemijo in kemijsko tehnologijo, Ljubljana UL. 2005 (15%) - M. Roš, Sodobni postopki čiščenja odpadnih vod, Fit media, Velenje 2015 (30%)

Cilji in kompetence:

<p>Cilj predmeta je da študent med študijem pridobljeno znanje uporabi pri analizi tipičnega biotehnoškega procesa in procesa ravnanja z nastalimi odpadki.</p> <p>Predmetno specifične kompetence:</p> <ul style="list-style-type: none"> - študent spozna integralno vlogo osnovnih operacij v izbranem kemijsko tehnološkem procesu - obvlada procesno shemo, sestavljeno iz osnovnih operacij oziroma aparatov za izbrani tehnološki proces. - spozna metodologijo ravnanja z nastalimi trdnimi, tekočimi in plinastimi odpadki
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Objectives and competences:

<p>The objective is that the student uses the acquired knowledge to analyse a typical biotechnological as well as waste management process. Specific competences are:</p> <ul style="list-style-type: none"> - student recognizes the integral role of unit operations in a selected biotechnological process, - student acquaints a process scheme, composed of unit operations and equipment for a selected biotechnological process, - student acquaints a methodology of solid, liquid and gas waste management.

Predvideni študijski rezultati:

<p>Znanje in razumevanje Študent zna integrirati kemijsko inženirska znanja pri vodenju tehnoloških procesov in reševanju problemov ki pri tem nastanejo.</p> <p>Uporaba Pridobljena znanja je sposoben uporabiti pri reševanju posameznih praktičnih primerov in problemov v biotehnoški proizvodnji in industrijskih procesih ravnanja z odpadki.</p> <p>Refleksija Uporaba splošnih znanj in osnovnih principov kemijskega inženirstva, analiza in kritično ovrednotenje tehnološkega procesa oziroma posameznega postopka in naprave v laboratorijskem in industrijskem merilu.</p> <p>Prenosljive spretnosti Razvita sposobnost identifikacije in reševanja problemov, kritičnega razmišljanja in logičnega sklepanja. Sposobnost uporabe literature, zbiranja in interpretacije podatkov in njihove kritične evalvacije.</p>
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Intended learning outcomes:

<p>Knowledge and Comprehension Student is able to integrate chemical engineering knowledge during the control of a biotechnological process and solve the arisen problems.</p> <p>Application Student is able to use the acquired knowledge to solve the particular practical problems and cases in a biotechnological and waste treatment process.</p> <p>Analysis Use of general knowledge and basic principles of chemical engineering, analysis and critical evaluation of a biotechnological process as well as particular operation and equipment on a laboratory and industrial scale.</p> <p>Skill-transference Ability Developed skill to identify and solve problem, critical thinking and making logical conclusions. Ability of literature data using, data collection and interpretation as well as their critical evaluation.</p>

Metode poučevanja in učenja:

Predavanja, laboratorijske vaje in seminarji.

Learning and teaching methods:

Lectures, seminars, laboratory exercises.

Načini ocenjevanja:

Delež/Weight Assessment:

Pisni izpit	70,00 %	Written exam
Laboratorijske vaje	15,00 %	Laboratory exercises
Seminarske naloge Ocene: 6-10 (pozitivno), 1-5 (negativno)	15,00 %	Written seminar project Marks: 6-10(positive), 1-5 (negative)

Reference nosilca/Lecturer's references:

- M. Tišma, **P. Žnidaršič-Plazl**, G. Šelo, I. Tolj, M. Šperanda, A. Bucić-Kojić, M. Planinić. *Trametes versicolor* in lignocellulose-based bioeconomy: state of the art, challenges and opportunities. *Bioresour. Technol.*, 2021, **330**, 124997, doi: [10.1016/j.biortech.2021.124997](https://doi.org/10.1016/j.biortech.2021.124997).
- F. A. Vicente, I. Plazl, S.P.M. Ventura, **P. Žnidaršič-Plazl**. Separation and purification of biomacromolecules based on microfluidics. *Green Chem.*, 2020, **22**, 4391-4410, doi: [10.1039/c9gc04362d](https://doi.org/10.1039/c9gc04362d).
- **P. Žnidaršič-Plazl**. The promises and the challenges of biotransformations in micro-flow. *Biotechnol. J.*, 2019, 14, 1800580, doi: [10.1002/biot.201800580](https://doi.org/10.1002/biot.201800580).
- KORICA, Predrag, POŽGAJ, Đurđica, CIRMAN, Andreja, **ŽGAJNAR GOTVAJN, Andreja**. Decomposition analyses of the municipal waste generation and management in Croatian and Slovenian regions. *Journal of material cycles and waste management*, ISSN 1438-4957, 2016.
- **ŽGAJNAR GOTVAJN, Andreja**, KALČIKOVÁ, Gabriela. Delamination of plastic-coated waste paper by enzymes of the white rot fungus *Dichomitus squalens*. *Journal of environmental management*. Dec. 2018, vol. 228, str. 165-168.
- KALČIKOVÁ, Gabriela, TRATAR-PIRC, Elizabeta, **ŽGAJNAR GOTVAJN, Andreja**. Aerobic and anaerobic biodegradation potential of leachate from old active landfill. *Desalination and water treatment*. [Print ed.]. 2016, vol. 57, iss. 19, str. 8619-8625.

DIPLOMSKO DELO ZA SPLOŠNI MODUL

UČNI NAČRT PREDMETA/COURSE SYLLABUS

Predmet:	Diplomsko delo za Splošni modul
Course title:	Diploma Work for General Module
Članica nosilka/UL	UL FKKT
Member:	

Študijski programi in stopnja	Študijska smer	Letnik	Semestri	Izbirnost
Kemijska tehnologija, prva stopnja, visokošolski strokovni (od študijskega leta 2024/2025 dalje)	Ni členitve (študijski program)	3. letnik	poletni	izbirni

Univerzitetna koda predmeta/University course code: 0643672

Predavanja /Lectures	Seminar /Seminar	Vaje /Tutorials	Klinične vaje /Clinical tutorials	Druge oblike študija /Other forms of study	Samostojno delo /Individual student work	ECTS
				150	150	10

Nosilec predmeta/Lecturer:

Vrsta predmeta/Course type: Obvezni/Mandatory

Jeziki/Languages:

Predavanja/Lectures:	
Vaje/Tutorial:	

Pogoji za vključitev v delo oz. za opravljanje študijskih obveznosti:

Odobrena tema diplomskega dela.

Prerequisites:

Approved topic.

Vsebina:

Diplomsko delo se opravlja s področja kemijske tehnologije. Vsebina in naslov se določata v soglasju z izbranim mentorjem. Mentor je lahko učitelj na UL FKKT [t.j. zaposleni na fakulteti na učiteljskem delovnem mestu ali zaposleni na fakulteti na delovnem mestu asistenta, ki ima učiteljski naziv ali nosilec predmeta na študijskem programu 1. ali 2. stopnje UL FKKT, ki ni zaposlen na fakulteti]. Mentor je praviloma učitelj na programu, ki ga je študent vpisal.

Content (Syllabus outline):

Diploma thesis is in the field of chemical. The content and title are defined by agreement with the mentor. Mentor can be a teacher from UL FKKT [i.e. employed at the Faculty as teacher or as assistant with teaching habilitation or course leader on UL FKKT 1st or 2nd cycle study programme who is not employed at the Faculty]. Mentor should be one of the course lecturers in the student's programme.

Temeljna literatura in viri/Readings:

Knjige in članki, ki so povezani z vsebino diplomskega dela. / Books and articles that are related to the content of the thesis.

Cilji in kompetence:

Objectives and competences:

<p>Cilji: Dokončno oblikovanje pričakovanega lika diplomanta. Študent bo ob izdelavi diplomske naloge pokazal sposobnosti iskanja in zaznavanja problemov in znal poiskati rešitev za tak problem. Pri delu bodo pokazal, da je pridobil večino kompetenc, navedenih v programu študija.</p>	<p>The completion of the expected characteristics of the graduate. The student will in their diploma thesis show the ability to search for and detect problems. They will be able to find a solution for such a problem. At work they will show that the majority of competencies listed in the study programme were achieved.</p>
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Predvideni študijski rezultati:

<p>Znanje in razumevanje Pri izdelavi diplomskega dela bo slušatelj pridobil naslednje kompetence:</p> <ul style="list-style-type: none"> - usposobljenost za uporabo teoretičnega znanja in njegov prenos in uporabo v praksi; - sposobnost eksperimentiranja, zbiranje relevantnih podatkov o eksperimentu ali procesu in njihovega vrednotenja; - razumevanje meja zanesljivosti svojih eksperimentalnih podatkov; - sposobnost izvedbe manjšega projekta, za katerega ni nujno, da njegov rezultat zadošča kriterijem za objavo; - razvita profesionalna etična in okoljska odgovornost; - poznavanje in sposobnost uporabe različnih postopkov analize in karakterizacije snovi od enostavnejših analiz do kompleksnih inštrumentalnih metod; - usposobljenost za kvalitetno in varno delo v laboratoriju s poznavanjem laboratorijske opreme in ustreznih laboratorijskih tehnik; - sposobnost za delo z najzahtevnejšo laboratorijsko opremo, inštrumenti in aparaturami. <p>Uporaba Znanje in pridobljene veščine bo diplomant lahko uporabil pri opravljanju poklica.</p> <p>Refleksija Povezovanje vseh pridobljenih teoretskih znanj z reševanjem problemov na področju kemijske tehnologije ter kritični pogled na uporabnost teh znanj.</p> <p>Prenosljive spretnosti Pri delu bo diplomant pridobil znanja o metodah reševanja problemov, o načinu prezentacije teh znanj v pisani in govorjeni obliki povezani z ostalimi metodami posredovanja raziskav, ugotovitev itd.</p>	<p>Intended learning outcomes:</p> <p>Knowledge and understanding During thesis preparation, the student will obtain the following competencies:</p> <ul style="list-style-type: none"> - Ability to apply theoretical knowledge and transfer it to the practical application; - Ability to conduct experiments, collect the relevant data from the experiment or process and to evaluate them; - Understanding of the reliability of their experimental data; - Ability to conduct a project on a smaller scale that is not necessarily publishable; - Developed professional ethical and environmental responsibility; - Knowledge and ability to use different analytical and characterization procedures from simple analyses to more complex instrumental methods; - Ability to conduct quality and safe laboratory work by knowing laboratory equipment and laboratory techniques; - Ability to work with the most sophisticated laboratory equipment, instruments and apparatuses. <p>Application The graduate will use the acquired knowledge and skills in their occupational work.</p> <p>Reflection Integration of acquired theoretical knowledge to solving problems from the area of chemical technology and a critical view of the usefulness of these skills.</p> <p>Skill-transference Ability The graduate will gain knowledge of the methods of problem solving at work, the method of presentation of these skills in written and spoken form associated with other methods of transferring the research findings, etc.</p>
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Metode poučevanja in učenja:

<p>Študijsko in raziskovalno delo pod vodstvom mentorja.</p>	<p>Learning and teaching methods: Study and research under supervision of mentor.</p>
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Načini ocenjevanja:

Načini ocenjevanja:	Delež/Weight	Assessment:
<p>Ocenjuje se diplomsko delo in zagovor diplomskega dela pred komisijo, ki jo</p>	<p>100,00 %</p>	<p>Diploma thesis and its oral defense are assessed before a panel consisting of mentor</p>

sestavljata mentor in član. Ocene so v skladu s Statutom UL (5 negativno, 6-10 pozitivno).

and one member. Grades for assessment are from 5 to 10. Grade 5 is negative, grades 6 to 10 are positive.

Reference nosilca/Lecturer's references:

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UL
ELEKTR

DIPLOMSKO DELO ZA VOJAŠKI MODUL

UČNI NAČRT PREDMETA/COURSE SYLLABUS

Predmet:	Diplomsko delo za Vojaški modul
Course title:	Diploma Work for Military Module
Članica nosilka/UL	UL FKKT
Member:	

Študijski programi in stopnja	Študijska smer	Letnik	Semestri	Izbirnost
Kemijska tehnologija, prva stopnja, visokošolski strokovni (od študijskega leta 2024/2025 dalje)	Ni členitve (študijski program)	3. letnik	poletni	izbirni

Univerzitetna koda predmeta/University course code: 0643676

Predavanja /Lectures	Seminar /Seminar	Vaje /Tutorials	Klinične vaje /Clinical tutorials	Druge oblike študija /Other forms of study	Samostojno delo /Individual student work	ECTS
				120	120	8

Nosilec predmeta/Lecturer:

Vrsta predmeta/Course type: Obvezni/Mandatory

Jeziki/Languages:

Predavanja/Lectures:	
Vaje/Tutorial:	

Pogoji za vključitev v delo oz. za opravljanje študijskih obveznosti:

Odobrena tema diplomskega dela.

Prerequisites:

Approved topic.

Vsebina:

Diplomsko delo se opravlja s področja kemijske tehnologije. Vsebina in naslov se določata v soglasju z izbranim mentorjem. Mentor je lahko učitelj na UL FKKT [t.j. zaposleni na fakulteti na učiteljskem delovnem mestu ali zaposleni na fakulteti na delovnem mestu asistenta, ki ima učiteljski naziv ali nosilec predmeta na študijskem programu 1. ali 2. stopnje UL FKKT, ki ni zaposlen na fakulteti]. Mentor je praviloma učitelj na programu, ki ga je študent vpisal.

Content (Syllabus outline):

Diploma thesis is in the field of chemical. The content and title are defined by agreement with the mentor. Mentor can be a teacher from UL FKKT [i.e. employed at the Faculty as teacher or as assistant with teaching habilitation or course leader on UL FKKT 1st or 2nd cycle study programme who is not employed at the Faculty]. Mentor should be one of the course lecturers in the student's programme.

Temeljna literatura in viri/Readings:

Knjige in članki, ki so povezani z vsebino diplomskega dela. / Books and articles that are related to the content of the thesis

Cilji in kompetence:

Objectives and competences:

<p>Cilji: Dokončno oblikovanje pričakovanega lika diplomanta. Študent bo ob izdelavi diplomske naloge pokazal sposobnosti iskanja in zaznavanja problemov in znal poiskati rešitev za tak problem. Pri delu bodo pokazal, da je pridobil večino kompetenc, navedenih v programu študija.</p>	<p>The completion of the expected characteristics of the graduate. The student will in their diploma thesis show the ability to search for and detect problems. They will be able to find a solution for such a problem. At work they will show that the majority of competencies listed in the study programme were achieved.</p>
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Predvideni študijski rezultati:

<p>Znanje in razumevanje Pri izdelavi diplomskega dela bo slušatelj pridobil naslednje kompetence:</p> <ul style="list-style-type: none"> - usposobljenost za uporabo teoretičnega znanja in njegov prenos in uporabo v praksi; - sposobnost eksperimentiranja, zbiranja relevantnih podatkov o eksperimentu ali procesu in njihovega vrednotenja; - razumevanje meja zanesljivosti svojih eksperimentalnih podatkov; - sposobnost izvedbe manjšega projekta, za katerega ni nujno, da njegov rezultat zadošča kriterijem za objavo; - razvita profesionalna etična in okoljska odgovornost; - poznavanje in sposobnost uporabe različnih postopkov analize in karakterizacije snovi od enostavnejših analiz do kompleksnih inštrumentalnih metod; - usposobljenost za kvalitetno in varno delo v laboratoriju s poznavanjem laboratorijske opreme in ustreznih laboratorijskih tehnik; - sposobnost za delo z najzahtevnejšo laboratorijsko opremo, inštrumenti in aparaturami. <p>Uporaba Znanje in pridobljene veščine bo diplomant lahko uporabil pri opravljanju poklica.</p> <p>Refleksija Povezovanje vseh pridobljenih teoretskih znanj z reševanjem problemov na področju kemijske tehnologije ter kritični pogled na uporabnost teh znanj.</p> <p>Prenosljive spretnosti Pri delu bo diplomant pridobil znanja o metodah reševanja problemov, o načinu prezentacije teh znanj v pisani in govorjeni obliki povezani z ostalimi metodami posredovanja raziskav, ugotovitev itd.</p>	<p>Intended learning outcomes:</p> <p>Knowledge and understanding During thesis preparation, the student will obtain the following competencies:</p> <ul style="list-style-type: none"> - Ability to apply theoretical knowledge and transfer it to the practical application; - Ability to conduct experiments, collect the relevant data from the experiment or process and to evaluate them; - Understanding of the reliability of their experimental data; - Ability to conduct a project on a smaller scale that is not necessarily publishable; - Developed professional ethical and environmental responsibility; - Knowledge and ability to use different analytical and characterization procedures from simple analyses to more complex instrumental methods; - Ability to conduct quality and safe laboratory work by knowing laboratory equipment and laboratory techniques; - Ability to work with the most sophisticated laboratory equipment, instruments and apparatuses. <p>Application The graduate will use the acquired knowledge and skills in their occupational work.</p> <p>Reflection Integration of acquired theoretical knowledge to solving problems from the area of chemical technology and a critical view of the usefulness of these skills.</p> <p>Skill-transference Ability The graduate will gain knowledge of the methods of problem solving at work, the method of presentation of these skills in written and spoken form associated with other methods of transferring the research findings, etc.</p>
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Metode poučevanja in učenja:

<p>Študijsko in raziskovalno delo pod vodstvom mentorja.</p>	<p>Learning and teaching methods: Study and research under supervision of mentor.</p>
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Načini ocenjevanja:

Delež/Weight Assessment:

Ocenjuje se diplomsko delo in zagovor diplomskega dela pred komisijo, ki jo sestavljata mentor in član. Član mora biti učitelj, zaposlen na UL FKKT. Ocene so v skladu s Statutom UL (5 negativno, 6-10 pozitivno).	100,00 %	Diploma thesis and its oral defense are assessed before a panel consisting of mentor and one member. Member should be a teacher employed at UL FKKT. Grades for assessment are from 5 to 10. Grade 5 is negative, grades 6 to 10 are positive.
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Reference nosilca/Lecturer's references:

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UL FKKT

FIZIKA

UČNI NAČRT PREDMETA/COURSE SYLLABUS

Predmet:	Fizika
Course title:	Physics I
Članica nosilka/UL Member:	UL FKKT

Študijski programi in stopnja	Študijska smer	Letnik	Semestri	Izbirnost
Kemijska tehnologija, prva stopnja, visokošolski strokovni	Ni členitve (študijski program)	1. letnik	zimski	obvezni

Univerzitetna koda predmeta/University course code:	0072018
Koda učne enote na članici/UL Member course code:	KT102

Predavanja /Lectures	Seminar /Seminar	Vaje /Tutorials	Klinične vaje /Clinical tutorials	Druge oblike študija /Other forms of study	Samostojno delo /Individual student work	ECTS
60		15 SV			75	5

Nosilec predmeta/Lecturer:	Aleš Mohorič
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Vrsta predmeta/Course type:	obvezni/mandatory
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Jeziki/Languages:	Predavanja/Lectures:	Slovenščina
	Vaje/Tutorial:	Slovenščina

Pogoji za vključitev v delo oz. za opravljanje študijskih obveznosti: **Prerequisites:**

Študent oz. kandidat mora imeti predmet opredeljen kot študijsko obveznost.	The course has to be assigned to the student.
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Vsebina:

Kinematika: Premo in krivo gibanje. Kroženje in nihanje. Dinamika delca: Newtonovi zakoni. Zakona o ohranitvi gibalne in vrtilne količine. Gravitacijski zakon. Merjenje sil. Delo in moč. Energija, izrek o ohranitvi energije. Potencialna energija. Togo telo in sistem teles: Težišče. Vrtenje, navor in vztrajnostni moment. Rotacijska energija. Ravnotežje sil in navorov.

Elastomehanika: Hookov zakon, elastične konstante. Mehanika tekočin in plinov: Tlak, stisljivost, vzgon, površinska napetost, viskoznost, Bernoullijeva enačba. Sila upora, Reynoldsovo število.

Valovanje: Hitrost potovanja motnje, zapis valovanja. Interferenca, odboj, lom in uklon valovanja.

Toplota:

Content (Syllabus outline):

Kinematics: linear and general motion. Circular motion, oscillation. Dynamics: Newton's laws. Linear and angular momentum. Gravitational law. Force. Work, power. Energy, conservation of energy. Gravitational potential energy. Rigid body, system of bodies: center of gravity. Rotation, torque, moment of inertia. Rotational kinetic energy. Static: force and torque balance.

Elastomechanics: Hook's law, moduli. Fluid dynamics: pressure, compressibility, buoyancy, surface tension, viscosity, Bernoulli equation. Resistance, the Reynolds number.

Waves: speed, mathematical expression. Interference, reflection, refraction, diffraction.

Heat:

<p>Energijski zakon: Kelvinova temperaturna skala. Temperaturno raztezanje. Toplota kot energija. Specifična in latentne toplote, entalpija. Kalorimetrija. Izotermne, izobarne, izohorne in adiabatne spremembe idealnega plina. Toplotni stroji. Entropija. Prevajanje toplote.</p> <p>Elektrika in magnetizem:</p> <p>Coulombov zakon: Sile med električnimi naboji. Električno polje. Gaussov zakon. Električno delo in električna apetost. Kondenzator, kapaciteta, energija električnega polja. Snov v električnem polju, dielektričnost.</p> <p>Kirchoffovi zakoni. Električni tok in upor. Ohmov zakon.</p> <p>Merjenje toka, napetosti in upora. Viri električne napetosti. Istosmerni in izmenični tok.</p> <p>Magnetno polje: Magnetna sila in navor, magnetni moment.</p> <p>Indukcija: Indukcijski zakon (samo) induktivnost. Energija magnetnega polja. Snov v magnetnem polju: Dia-, para- in feromagnetizem. Elektromotor in generator. Transformator. Histerezna zanka.</p> <p>Elektromagnetna nihanja in valovanja: Nihajni krog. Elektromagnetno valovanje. Spekter elektromagnetnega valovanja. Svetloba.</p> <p>Geometrijska in valovna optika: Odbojni in lomni zakon. Optični instrumenti. Interferenca in uklon. Optična mreža. Polarizacija.</p>	<p>First law of thermodynamics. Absolute temperature. Thermal expansion. Heat as energy. Specific and latent heat, enthalpy. Calorimetry. Isothermal, isobaric, izohoric and adiabatic changes of ideal gas. Heat engines. Entropy. Heat conduction.</p> <p>Electricity and magnetism:</p> <p>Coulomb's law, electrostatic forces. Electric field. Gauss' law. Electric work and voltage. Capacitor, capacity, electric field energy. Electric field in matter, dielectricity.</p> <p>Kirchhoff's laws. Electrical resistance and current. Ohm's law. Measuring current, voltage and resistivity. Electrical sources. AC, DC.</p> <p>Magnetic field: magnetic force and torque, magnetic dipole moment.</p> <p>Induction: induction law, inductance. Magnetic field energy. Magnetic field in matter, Dia, para and fero magnetism. Electric motor and generator.</p> <p>Transformer. Hysteresis. Electromagnetic oscillation and waves: electric oscillator circuit. EMW spectra.</p> <p>Light.</p> <p>Geometric and wave optics: refraction and reflection laws. Optical instruments. Interference, diffraction. Diffraction grating. Polarization.</p>
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Temeljna literatura in viri/Readings:

- J. Strnad, Fizika I in II, DMFA
- R. Kladnik, Visokošolska fizika, DZS
- I. Kuščer, A. Moljk, ..., Fizika za srednje šole I, II in III, DZS
- A. Mohorič, V. Babič, Fizika 1, 2 in 3, MK

Cilji in kompetence:

Cilji: Nadgradnja srednješolskih osnov fizike, razumevanje osnovnih zakonov s področja mehanike, toplote, elektrike in magnetizma, optike in atomike. Obvladovanje kvantitativnih povezav med fizikalnimi količinami v enostavnih primerih, na katere naleti diplomant pri poklicnem delu.

Kompetence: Sposobnost poenostavitve nalog v obliko, kjer je mogoča enostavna računsko obravnava. Sposobnost uporabe osnovnega matematičnega znanja pri reševanju zastavljenih nalog. Sposobnost iskanja dodatnih informacij po strokovni literaturi in na svetovnem spletu.

Objectives and competences:

Objectives: Upgrade on the secondary school physics, comprehension of basic laws in mechanics, thermodynamics, electricity, magnetism, optics and atomics. Understanding of quantitative links between physical quantities in simple cases encountered by the graduate in his/hers profession.

Competences: The ability to simplify the tasks of design as possible to make calculus simpler. Ability to apply mathematical knowledge in solving the set tasks. Ability to search for additional information in the literature and on the internet.

Predvideni študijski rezultati:

Znanje in razumevanje
Poznavanje osnovnih pojmov in zakonov fizike.
Neambiciozen študent, ki se je pripravljen zadovoljiti z nizko oceno, bo obvladal povezave med

Intended learning outcomes:

Knowledge and Comprehension
Knowledge of basic concepts and laws of physics.
Unambitious student who is willing to settle for the low rating, will handle the connection between

<p>posameznimi fizikalnimi količinami z memoriranjem, bolj ambiciozen študent pa bo pridobil na osnovi izpeljav enačb iz osnovnih zakonov globlji vpogled v naravne zakone, kar mu bo omogočilo temeljito razumevanje snovi.</p> <p>Uporaba</p> <p>Uporaba že osvojenega teoretičnega znanja fizike v realnih, praktičnih situacijah.</p> <p>Refleksija</p> <p>Razumevanje tehnoloških rešitev in primerov iz vsakdanjega življenja v luči fizikalnih naravnih zakonov.</p> <p>Prenosljive spretnosti</p> <p>Predmet Fizika je najprimernejše področje, kjer študent lahko preverja in utrjuje svoje matematično znanje.</p>	<p>physical quantities with memorizing, more ambitious student will gain through derivation of equations from basic laws deeper insight into the laws of nature, which will allow a thorough understanding of the syllabus.</p> <p>Application</p> <p>Applying existing theoretical knowledge of physics to real, practical situations.</p> <p>Analysis</p> <p>Understanding of technology solutions and examples from everyday life in the light of the physical laws of nature.</p> <p>Skill-transference Ability</p> <p>The subject of physics is the preferred subject where students can verify and consolidate their mathematical knowledge.</p>
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Metode poučevanja in učenja:

Predavanja, vaje, domače naloge, konzultacije.

Learning and teaching methods:

Lectures, exercises, homework, consultations.

Načini ocenjevanja:

Delež/Weight

Assessment:

- Pisni izpit (nadomestita ga lahko dva pozitivno ocenjena kolokvija)	50,00 %	- Written examination (can be replaced by two positive tests)
- ustni izpit. ocene od 6-10 (pozitivno) oz. 1-5 (negativno).	50,00 %	- oral examination. grades 6-10 (positive) and 1-5 (negative).

Reference nosilca/Lecturer's references:

- STEPIŠNIK, Janez, LAHAJNAR, Gojmir, ZUPANČIČ, Ivan, **MOHORIČ, Aleš**. Study of translational dynamics in molten polymer by variation of gradient pulse-width of PGSE. Journal of magnetic resonance, ISSN 1090-7807, 2013, vol. 236, str. 41-46, doi: 10.1016/j.jmr.2013.08.003. [COBISS.SI-ID 27404327]
- STEPIŠNIK, Janez, FRITZINGER, Bernd, SCHELER, Ulrich, **MOHORIČ, Aleš**. Self-diffusion in nanopores studied by the NMR pulse gradient spin echo. Europhysics letters, ISSN 0295-5075, 2012, vol. 98, no. 5, str. 57009-p1-57009-p4. <http://iopscience.iop.org/0295-5075/98/5/57009>. [COBISS.SI-ID 2434404]
- STEPIŠNIK, Janez, **MOHORIČ, Aleš**, SERŠA, Igor, LAHAJNAR, Gojmir. Analysis of polymer dynamics by NMR modulated gradient spin echo. V: VOLOVŠEK, Vesna (ur.), BISTRJEVA, Lahorija (ur.). Polymer spectroscopy July 2011, (Macromolecular Symposia, ISSN 1022-1360, Special issue, Volume 305, Issue Basel [etc.]: Hüting & Wepf Verlag, 2011, vol. 305, str. 55-62, doi: 10.1002/masy.201000120. [COBISS.SI-ID 2362212]
- VAN DEN DOEL, L. R., **MOHORIČ, Aleš**, VERGELDT, Frank, VAN DUYNHOVEN, John, BLONK, Han, VAN DALEN, Gerard, VAN AS, Henk, VAN VLIET, Lucas J. Mathematical modeling of water uptake through diffusion in 3D inhomogeneous swelling substrate. AIChE journal, ISSN 0001-1541. [Print ed.], 2009, vol. 55, no. 7, str. 1834-1848, doi: 10.1002/aic.11930. [COBISS.SI-ID 2175844]
- **MOHORIČ, Aleš**, VERGELDT, Frank, GERKEMA, Edo, VAN DALEN, Gerard, VAN DEN DOEL, L. R., VAN VLIET, Lucas J, VAN AS, Henk, VAN DUYNHOVEN, John. The effect of rice kernel microstructure on cooking behaviour : a combined [mⁱ]-CT and MRI study. Food chemistry, ISSN 0308-8146. [Print ed.], 2009, vol. 115, str. 1491-1499, doi: 10.1016/j.foodchem.2009.01.089. [COBISS.SI-ID 2157412]
- **MOHORIČ, Aleš**, STEPIŠNIK, Janez. NMR in the Earth's magnetic field. Progress in nuclear magnetic resonance spectroscopy, ISSN 0079-6565. [Print ed.], 2009, vol. 54, str. 166-182, doi: 10.1016/j.pnmrs.2008.07.002. [COBISS.SI-ID 2157156]
- STEPIŠNIK, Janez, LASIČ, Samo, **MOHORIČ, Aleš**, SERŠA, Igor, SEPE, Ana. Velocity autocorrelation spectra of fluid in porous media measured by the CPMG sequence and constant magnetic field gradient. V: Proceedings of the 8th International Bologne Conference on Magnetic Resonance in Porous Media, Bologna, 2006, (Magnetic resonance imaging, ISSN 0730-725X, vol. 25, no. 4). New York: Pergamon Press, 2007, vol. 25, no. 4, str. 517-520. [COBISS.SI-ID 1987428]

FIZIKALNA KEMIJA I

UČNI NAČRT PREDMETA/COURSE SYLLABUS

Predmet:	Fizikalna kemija I
Course title:	Physical chemistry I
Članica nosilka/UL Member:	UL FKKT

Študijski programi in stopnja	Študijska smer	Letnik	Semestri	Izbirnost
Kemijska tehnologija, prva stopnja, visokošolski strokovni	Ni členitve (študijski program)	2. letnik	poletni	obvezni

Univerzitetna koda predmeta/University course code:	0072031
Koda učne enote na članici/UL Member course code:	KT116

Predavanja /Lectures	Seminar /Seminar	Vaje /Tutorials	Klinične vaje /Clinical tutorials	Druge oblike študija /Other forms of study	Samostojno delo /Individual student work	ECTS
40	10	25 LV			75	5

Nosilec predmeta/Lecturer: prof. dr. Ksenija Kogej

Vrsta predmeta/Course type: obvezni/mandatory

Jeziki/Languages:

Predavanja/Lectures:	Slovenščina
Vaje/Tutorial:	Slovenščina

Pogoji za vključitev v delo oz. za opravljanje študijskih obveznosti: **Prerequisites:**

Študent oz. kandidat mora imeti predmet opredeljen kot študijsko obveznost.	The course has to be assigned to the student.
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Vsebina:

Plini. Lastnosti plinov. Enačbe stanja, idealni in realni plini. Kritični pojavi, utekočinjanje plinov. Prvi zakon termodinamike. Delo in toplota. Notranja energija, entalpija. Toplotne kapacitete. Kalorimetrija. Termokemija. Odvisnost entalpije od temperature. Drugi zakon termodinamike. Obrnljivi in neobrnljivi procesi. Entropija. Računanje entropijskih sprememb. Tretji termodinamični zakon. Gibbsova in Helmholtzova prosta energija. Odprti sistemi. Kemijski potencial. Kriterij za snovno ravnotežje. Fazna ravnotežja. Clausius-Clapeyronova enačba. Fazni diagrami. Fazno pravilo. Raztopine. Parcialne molske količine. Idealne in neidealne raztopine. Termodinamika mešanja.

Content (Syllabus outline):

Gasses. Properties of gasses. Equations of state, ideal and real gasses. Critical phenomena, condensation of gasses. The first law of thermodynamics. Work and heat. Internal energy, enthalpy, and heat capacities. Calorimetry. Thermo-chemistry. Dependence of enthalpy on temperature. The second law of thermodynamics. Reversible and irreversible processes. Entropy. Calculation of entropy changes. The third law of thermodynamics. Gibbs and Helmholtz free energy. Open systems. Chemical potential. Criterion for equilibrium. Phase equilibria. The Clausius-Clapeyron equation. Phase diagrams. Phase rule. Solutions. Partial molar quantities. Ideal and non-ideal solutions. Thermodynamics of mixing. Raoult's and Henry's law. Vapour pressure diagrams.

<p>Raoultov in Henryjev zakon. Diagrami parnih tlakov. Vrelni diagrami in frakcionirna destilacija. Koligativne lastnosti.</p> <p>Laboratorijske vaje: 1. Kalorimetrija 2. Parni tlak 3. Vrelni diagram 4. Določanje molske mase s krioskopsko metodo 5. Viskoznost tekočin</p>	<p>Temperature-composition diagrams and fractional distillation. Colligative properties.</p> <p>Laboratory practice. 1. Calorimetry 2. Vapor pressure 3. Boiling point diagram 4. Determination of molar mass by cryoscopy 5. Viscosity</p>
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Temeljna literatura in viri/Readings:

<p>- R. A. ALBERTY, R. J. SILBEY: Physical Chemistry, John-Wiley, New York, 1995; 217 strani (24 %):s slikami, grafičnimi prikazi in računskimi nalogami.</p> <p>- BONČINA, Matjaž, CERAR, Janez, GODEC, Andrej (avtor, urednik), HRIBAR, Barbara, JAMNIK, Andrej, LAH, Jurij, LAJOVIC, Andrej, LUKŠIČ, Miha, PODLIPNIK, Črtomir, PRISLAN, Iztok, REŠČIČ, Jurij, ŠARAC, Bojan, TOMŠIČ, Matija, VESNAVER, Gorazd. Fizikalna kemija - praktikum. 1. izd. Ljubljana: Fakulteta za kemijo in kemijsko tehnologijo, 2012. XXXII, 227 str., ilustr. ISBN 978-961-6756-32-7. [COBISS.SI-ID 261552640]</p>
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Cilji in kompetence:

<p>Fizikalna kemija I, skupaj s Fizikalno kemijo II, je osnovni naravoslovni predmet, pri katerem študenti spoznajo temeljne fizikalno-kemijske zakonitosti. Predmet študentu pomaga pri razumevanju fizikalno-kemijskih procesov v laboratoriju in ga nauči reševanja zelo različnih problemov iz naravoslovja. Pri tem razvije sposobnost logičnega sklepanja in povezovanja znanj.</p>	<p>Objectives and competences:</p> <p>Physical chemistry I, together with Physical chemistry II, is a fundamental natural science course where students learn basic physico-chemical principles. After the completion of the course they understand physical processes and learn to solve various problems in the field of natural sciences. Students develop a critical way of thinking and uniting knowledge.</p>
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Predvideni študijski rezultati:

<p>Znanje in razumevanje Študent pri predmetu pridobi razumevanje osnovnih naravnih zakonitosti sistemov (npr. plinov, tekočin, raztopin neelektrolitov in elektrolitov) in fizikalno kemijskih količin (npr. dela, toplote, notranje energije, entalpije, entropije itd.). Sposoben je ločiti med funkcijami stanja in funkcijami poti. Spozna vse termodinamične zakone, pojme obrnljivi in neobrnljivi procesi in kriterije za spontanost procesov in za ravnotežje. Pridobljeno teoretično znanje je osnova za razumevanje številnih fizikalno kemijskih metod, ki se uporabljajo v analitskem in drugih laboratorijih. Študent svoje razumevanje snovi preveri v laboratoriju z merjenjem osnovnih termodinamskih količin.</p> <p>Uporaba Ker je fizikalna kemija temeljni naravoslovni predmet, nudi učna snov študentu široko razumevanje raznih pojavov v naravoslovju. Študij tega predmeta je nujna podlaga za to, da bo študent razumel principe določenih meritev in metod, ki jih bo uporabljal v laboratoriju.</p> <p>Refleksija Študent pridobi občutek za fizikalno-kemijski način razmišljanja in razvije zmožnost abstraktne predstave o fizikalno-kemijskih količinah in njihove interpretacije v okviru enostavnih modelov ter hipotez. S pridobljenim znanjem zna razne tehnološke probleme analizirati s stališča osnovnih</p>	<p>Intended learning outcomes:</p> <p>Knowledge and Comprehension Students acquire understanding of fundamental natural laws in various systems (e.g. gasses, liquids, nonelectrolyte and electrolyte solutions) and for various properties (e.g. work, heat etc.). They are able to distinguish between state function and path functions. They learn all thermodynamic laws, concepts of reversibility and irreversibility, criteria for spontaneity and equilibrium. The acquired theoretical knowledge is the basis for understanding numerous physicochemical methods used in analytical and other laboratories. The understanding of the subject is verified in laboratory by measuring basic thermodynamic properties.</p> <p>Application As a fundamental natural science course, physical chemistry offers students broad understanding of various phenomena in nature. The obtained knowledge is necessary for understanding principals of measurements.</p> <p>Analysis Students gain the physicochemical way of thinking and develop abstract conception of physical-chemical quantities. They are able to interpret their observation in the framework of simple models and hypothesis and to analyse technological problems with basic laws. Through this they should be able to critically evaluate the results of their work.</p> <p>Skill-transference Ability</p>
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<p>zakonitosti in s tem bi moral biti zmožen kritično vrednotiti rezultate svojega dela.</p> <p>Prenosljive spretnosti</p> <p>Pri študiju fizikalne kemije študenti razvijajo abstrakten, kritičen in analitičen način razmišljanja, kar jim bo koristilo pri prepoznavanju in reševanju problemov v raznih okoljih. Naučijo se iskati in uporabljati domačo in tujo literaturo ter poročati o izsledkih svojega dela. Predmet študenta navaja k povezovanju znanja kemije, fizike, matematike in ostalih tehnoloških ved.</p>	<p>The abstract, critical, and analytical way of thinking helps the student in recognizing and solving problems in various environments. They learn to use literature and data basis and how to report their work. They are able to link knowledge of chemistry, physics, and mathematics with that of other technological fields.</p>
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<p>Metode poučevanja in učenja:</p> <p>Predavanja. Seminarji z računskimi primeri, kjer študenti aktivno sodelujejo in tako preizkusijo, ali pravilno razumejo pridobljeno teoretično znanje. Laboratorijske vaje z vodenjem dnevnika in oddajanjem poročil o opravljenem delu in rezultatih.</p>	<p>Learning and teaching methods:</p> <p>Lectures. Seminars (problem solving) with active participation of students. In this way students check their theoretical knowledge. Laboratory practice, writing laboratory diary and submitting reports of measurements and obtained results.</p>
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Načini ocenjevanja:	Delež/Weight	Assessment:
Predavanja: pisni izpit, ki ga študenti lahko opravijo z dvema delnima kolokvijema tekom predavanj.	60,00 %	Lectures: Written examination that can be passed by two written tests during semester.
Vaje: ocena dnevnika tekom opravljanja vaj in končni pisni kolokvij iz vaj. Ocene: 6-10 (pozitivno), 1-5 (negativno).	40,00 %	Laboratory practice: grade for the laboratory diary and for the final written test. Grades: 6-10 (positive), 1-5 (negative)

<p>Reference nosilca/Lecturer's references:</p> <p>- KOGEJ, Ksenija, FONSECA, Sofia M., ROVISCO, J., AZENHA, M. E., LUÍSA RAMOS, M., SEIXAS DE MELO, J., BURROWS, Hugh. Understanding the interaction between trivalent lanthanide ions and stereoregular polymethacrylates through luminescence, binding isotherms, NMR, and interaction with cetylpyridinium chloride. <i>Langmuir</i>, ISSN 0743-7463, 2013, vol. 29, no. 47, str. 14429-14437, [COBISS.SI-ID 1656879]</p> <p>- ANŽLOVAR, Alojz, CRNJAK OREL, Zorica, KOGEJ, Ksenija, ŽIGON, Majda. Polyol-mediated synthesis of zinc oxide nanorods and nanocomposites with poly(methyl methacrylate). <i>Journal of nanomaterials</i>, ISSN 1687-4110, 2012, vol. 2012, art. no. 760872 (9 str.), [COBISS.SI-ID 36033029]</p> <p>- PRELESNIK, Simona, GODERIS, Bart, HANSSON, Per, KOGEJ, Ksenija. Phase diagram and structures in mixtures of poly(styrenesulfonate anion) and alkyltrimethylammonium cations in water : significance of specific hydrophobic interaction. <i>The journal of physical chemistry. B, Condensed matter, materials, surfaces, interfaces & biophysical</i>, ISSN 1520-6106, 2012, vol. 116, no. 15, str. 4634-4645, [COBISS.SI-ID 36006917]</p>
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FIZIKALNA KEMIJA II

UČNI NAČRT PREDMETA/COURSE SYLLABUS

Predmet:	Fizikalna kemija II
Course title:	Physical Chemistry II
Članica nosilka/UL Member:	UL FKKT

Študijski programi in stopnja	Študijska smer	Letnik	Semestri	Izbirnost
Kemijska tehnologija, prva stopnja, visokošolski strokovni	Ni členitve (študijski program)	3. letnik	zimski	obvezni

Univerzitetna koda predmeta/University course code:	0072047
Koda učne enote na članici/UL Member course code:	KT132

Predavanja /Lectures	Seminar /Seminar	Vaje /Tutorials	Klinične vaje /Clinical tutorials	Druge oblike študija /Other forms of study	Samostojno delo /Individual student work	ECTS
35	10	30 LV			75	5

Nosilec predmeta/Lecturer: prof. dr. Ksenija Kogej

Vrsta predmeta/Course type: obvezni/mandatory

Jeziki/Languages:

Predavanja/Lectures:	Slovenščina
Vaje/Tutorial:	Slovenščina

Pogoji za vključitev v delo oz. za opravljanje študijskih obveznosti:

Študent oz. kandidat mora imeti predmet opredeljen kot študijsko obveznost.

Prerequisites:

The course has to be assigned to the student.

Vsebina:

Kemijsko ravnotežje. Sprememba Gibbsove proste entalpije in termodinamična konstanta kemijskega ravnotežja. Reakcije v plinski fazi. Vpliv temperature, tlaka in inertnega plina na kemijsko ravnotežje. Heterogeno ravnotežje.
Kemijska kinetika. Hitrostni zakon, red reakcije in konstanta reakcijske hitrosti. Razpolovni čas. Vpliv temperature na hitrost kemijske reakcije.
Aktivacijska energija.
Elektrokemija. Galvanski členi: napetost in termodinamika galvanskih členov. Nernstova enačba. Standardni elektrodni potenciali in napetostna vrsta. Izbrana poglavja. Površinska kemija. Adsorpcija. Površinska napetost. Površinsko aktivne snovi. Laboratorijske vaje: 1. Napetost in notranja upornost galvanskega člena; merjenje pH 2. Potenciometrična

Content (Syllabus outline):

Chemical equilibrium. The Gibbs free energy change and the thermodynamic equilibrium constant. Reactions in the gaseous phase. Effect of temperature, pressure and inert gas on equilibrium. Heterogeneous equilibrium.
Chemical kinetics. Rate laws, reaction order and reaction rate constant. The half time. Effect of temperature on reaction rate. Activation energy.
Electrochemistry. Electrochemical cells. Electromotive force (EMF) and thermodynamics of galvanic cells. The Nernst equation. EMF measurements. Standard electrode potentials. Selected topics. Surface chemistry. Adsorption. Surface tension. Surface active compounds.
Laboratory practice. 1. Electromotive force and

titracija 3. Prevodnost elektrolitov 4. Adsorpcija 5. Kemijska kinetika. 6. Površinska napetost.	internal resistance of galvanic cells; pH measurements 2. Potentiometric titration 3. Conductivity of electrolytes 4. Adsorption 5. Chemical kinetics. 6. Surface tension.
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Temeljna literatura in viri/Readings:

- R. A. ALBERTY, R. J. SILBEY: Physical Chemistry, John-Wiley, New York, 1995; 217 strani (24 %):s slikami, grafičnimi prikazi in računskimi nalogami.
- BONČINA, Matjaž, CERAR, Janez, GODEC, Andrej (avtor, urednik), HRIBAR, Barbara, JAMNIK, Andrej, LAH, Jurij, LAJOVIC, Andrej, LUKŠIČ, Miha, PODLIPNIK, Črtomir, PRISLAN, Iztok, REŠČIČ, Jurij, ŠARAC, Bojan, TOMŠIČ, Matija, VESNAVER, Gorazd. Fizikalna kemija - praktikum. 1. izd. Ljubljana: Fakulteta za kemijo in kemijsko tehnologijo, 2012. XXXII, 227 str., ilustr. ISBN 978-961-6756-32-7. [COBISS.SI-ID 261552640]

Cilji in kompetence:

Fizikalna kemija II je nadaljevanje predmeta Fizikalna kemija I. Osnovni cilj učne enote Fizikalna kemija I in II je spoznavanje in razumevanje temeljnih fizikalno-kemijskih zakonitosti. V drugem delu je poseben poudarek na praktikumu, katerega poglobljen cilj je demonstracija osvojenih teoretičnih osnov na raznih fizikalno kemijskih procesih in meritvah. Študent se spozna z načinom določanja fizikalno-kemijskih količin v laboratoriju. Pri tem razvije sposobnost samostojnega eksperimentalnega dela in kritičnega vrednotenja dobljenih rezultatov.

Objectives and competences:

The course in Physical Chemistry II is a continuation of Physical Chemistry I. The main goal of both courses is to provide students the knowledge and understanding of basic physicochemical principles. In Part II, the emphasis is on laboratory practice with the main goal to demonstrate the presented theoretical concepts. Students learn how to determine (measure) physicochemical properties in laboratory. Through this they acquire skills for individual experimental work and critical evaluation of results.

Predvideni študijski rezultati:

Znanje in razumevanje
Študent pri predmetu dopolni svoje znanje o osnovnih naravnih zakonitostih in fizikalno kemijskih količinah, ki ga je dobil pri študiju predmeta Fizikalna kemija I. Spozna razliko med termodinamiko in kinetiko ter nekaj pomembnih fizikalnih lastnosti tekočin ter pojavov na površinah. Študent se seznani s principi merjenja fizikalnih količin. Nauči se pravilnega izvajanja meritev in podajanja rezultatov, identificirati in vrednotiti napake pri meritvah in uporabljati enostavne modele za interpretacijo rezultatov. Pri tem razvija sposobnost kritičnega vrednotenja izmerjenih količin.

Uporaba
Teoretično in praktično znanje fizikalne kemije bo študentu pomagalo pri uporabi analitskih metod ter pri izvajanju raznih operacij za nadziranje in usmerjanje kemijskih procesov. Nepogrešljivo je tudi pri vključevanju v razvojno in proizvodno delo, pri kontroli procesov v raznih panogah industrije in pri merjenju fizikalno-kemijskih količin v analitskem in drugih laboratorijih.

Refleksija
Študent pridobi občutek za fizikalno-kemijski način razmišljanja in razvije zmožnost abstraktne predstave o fizikalno-kemijskih količinah. Nauči se uporabljati teoretično pridobljeno znanje pri eksperimentalnem

Intended learning outcomes:

Knowledge and Comprehension
Students complete their knowledge of basic natural concepts and physicochemical quantities, which they have met already at Physical Chemistry I. They learn the difference between thermodynamics and kinetics and get to know some important physical properties and phenomena at interfaces Students learn the correct and precise way of measuring and presenting measured quantities, the way to report uncertainties in their measurements and to use simple models for interpretation of the data. Through this they learn to critically evaluate measured quantities.

Application
Theoretical and practical knowledge of physical chemistry can be used in applications of analytical methods and in executing various operations in chemical processes. It is indispensable in research and production work, in controlling technological processes in industry and in measurements of physicochemical quantities in analytical and other laboratories.

Analysis
Students develop the physicochemical way of thinking and ability of abstract conception of physicochemical quantities. They learn how to use the theoretical knowledge in experimental work and how to critically evaluate their experimental results and correct mistakes.

<p>delu ter kritično vrednotiti rezultate in popravljati svoje napake.</p> <p>Prenosljive spretnosti</p> <p>Pri študiju fizikalne kemije študenti razvijajo abstrakten, kritičen in analitičen način razmišljanja, ki je osnova za kvalitetno delo na različnih tehnoloških področjih. Naučijo se kritično vrednotiti in interpretirati rezultate meritev. Poleg tega se naučijo iskati in uporabljati domačo in tujo literaturo. Pri vajah osvojijo primeren način podajanja rezultatov in poročil o opravljenem delu. Študenti se privadijo timskega dela, kar jim bo koristilo pri vsakem delu v laboratoriju ali proizvodnem obratu. Spoznajo pravila varnega dela v laboratoriju.</p>	<p>Skill-transference Ability</p> <p>In the course on Physical chemistry students develop abstract, critical and analytical way of thinking that is the basis for competent work in various technological fields. They learn to critically evaluate and interpret results of measurements, to search data bases and literature. In laboratory, students learn to correctly handle with instruments and to collect, treat and interpret results of their measurements. They learn to work in smaller teams, which can be useful for any work in laboratory or production. They also learn safety rules for the work in chemical laboratories.</p>
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Metode poučevanja in učenja:

Predavanja s seminarskimi vajami. Laboratorijske vaje z vodenjem dnevnika in oddajanjem poročil o opravljenem delu in rezultatih.

Learning and teaching methods:

Lectures and seminars (problem solving). Laboratory practice, writing laboratory diary and submitting reports of measurements and obtained results.

Načini ocenjevanja:

Delež/Weight

Assessment:

Predavanja: pisni izpit, ki ga študenti lahko opravijo z dvema delnima kolokvijema tekom predavanj.	60,00 %	Lectures: Written examination that can be passed by two written tests during semester.
Vaje: ocena dnevnika tekom opravljanja vaj in končni pisni kolokvij iz vaj. Ocene: 6-10 (pozitivno), 5 (negativno).	40,00 %	Laboratory practice: grade for the laboratory diary and for the final written test. Grades: 6-10 (positive), 1-5 (negative)

Reference nosilca/Lecturer's references:

- **KOGEJ, Ksenija**, FONSECA, Sofia M., ROVISCO, J., AZENHA, M. E., LUÍSA RAMOS, M., SEIXAS DE MELO, J., BURROWS, Hugh. Understanding the interaction between trivalent lanthanide ions and stereoregular polymethacrylates through luminescence, binding isotherms, NMR, and interaction with cetylpyridinium chloride. *Langmuir*, ISSN 0743-7463, 2013, vol. 29, no. 47, str. 14429-14437, [COBISS.SI-ID 1656879]
- ANŽLOVAR, Alojz, CRNJAK OREL, Zorica, **KOGEJ, Ksenija**, ŽIGON, Majda. Polyol-mediated synthesis of zinc oxide nanorods and nanocomposites with poly(methyl methacrylate). *Journal of nanomaterials*, ISSN 1687-4110, 2012, vol. 2012, art. no. 760872 (9 str.), [COBISS.SI-ID 36033029]
- PRELESNIK, Simona, GODERIS, Bart, HANSSON, Per, **KOGEJ, Ksenija**. Phase diagram and structures in mixtures of poly(styrenesulfonate anion) and alkyltrimethylammonium cations in water : significance of specific hydrophobic interaction. *The journal of physical chemistry. B, Condensed matter, materials, surfaces, interfaces & biophysical*, ISSN 1520-6106, 2012, vol. 116, no. 15, str. 4634-4645, [COBISS.SI-ID 36006917]

INDUSTRIJSKI PROCESI IN TRAJNOSTNI RAZVOJ

UČNI NAČRT PREDMETA/COURSE SYLLABUS

Predmet:	Industrijski procesi in trajnostni razvoj
Course title:	Industrial Processes and Sustainable Development
Članica nosilka/UL Member:	UL FKKT

Študijski programi in stopnja	Študijska smer	Letnik	Semestri	Izbirnost
Kemijska tehnologija, prva stopnja, visokošolski strokovni	Ni členitve (študijski program)	3. letnik	zimski	obvezni

Univerzitetna koda predmeta/University course code:	0072048
Koda učne enote na članici/UL Member course code:	KT134

Predavanja /Lectures	Seminar /Seminar	Vaje /Tutorials	Klinične vaje /Clinical tutorials	Druge oblike študija /Other forms of study	Samostojno delo /Individual student work	ECTS
45	10	20 LV			75	5

Nosilec predmeta/Lecturer: prof. dr. Andreja Žgajnar Gotvajn

Vrsta predmeta/Course type: obvezni/mandatory

Jeziki/Languages:

Predavanja/Lectures:	Slovenščina
Vaje/Tutorial:	Slovenščina

Pogoji za vključitev v delo oz. za opravljanje študijskih obveznosti:

Študent oz. kandidat mora imeti predmet opredeljen kot študijsko obveznost.

Prerequisites:

The course has to be assigned to the student.

Vsebina:

Trajnostni razvoj: definicije, terminologija, vzporednice in razlike med industrijskimi procesi in biološkimi sistemi, vpliv industrijskih sistemov na naravo in človeka, etika industrijske proizvodnje, čistejša proizvodnja za izboljšanje materialnih in energijskih izkoristkov, zaščita naravnih virov, pomen celovitega pristopa k posameznem okoljskem problemu, zakonodaja.
Orodja: vrste, viri in transport onesaženja, vrednotenje vplivov onesaženja na okolje, konvencionalne in sodobne tehnike za preprečevanje onesaženja in remediacijo obstoječega onesaženja, tehnike za doseganje eko-učinkovitosti industrijskega procesa ter čistejše proizvodnje, optimiranje tehnoloških postopkov, energetska in snovna izraba odpadkov, zamenjava/minimizacija surovin,

Content (Syllabus outline):

Sustainable development: definitions, terminology, comparison of industrial and natural processes, the impact of industrial processes to nature and mankind, ethics of industrial production, cleaner production for reduction of material and energy use, preservation of natural resources, the importance of complex approach for solving environmental problems, legislation.
Tools: types, sources and environmental pathways of pollution, assessment of environmental impacts of pollution, conventional and advanced pollution prevention techniques, remediation techniques, industrial eco-efficiency methods, methods for cleaner production, optimization of production processes, material and energy recovery of solid wastes, replacement/minimisation of natural

<p>varčevanje z naravnimi viri in energijo, uporaba sekundarnih surovin, recikliranje, zaprti tehnološki procesi, koncept proizvodnje brez odpadkov, izobraževanje zaposlenih in uporabnikov proizvodov. Viri energije: konvencionalni viri energije, obnovljivi viri energije, možnosti prehoda iz obstoječih na trajnostne vire energije, problematika obnovljivih virov energije (transport, dostopnost, sezonska variacija, globalen vpliv na okolje), pričakovane smernice uporabe.</p> <p>Osnove industrijske ekologije: integrirana strategija preventive pred onesnaženjem, LCA (Life Cycle Assessment) izdelkov, eko-optimizacija produkta in procesa, učinkovit transport, nadzor in izboljševanje sistema, upoštevanje lokalne mikroklima, lokalni, regionalni in globalni pristop.</p> <p>Uporaba koncepta trajnostnega razvoja na primeru, ki bo relevanten s slovenskega prostora.</p> <p>Napredni oksidacijski procesi ter hibridni procesi. Uporaba sekundarnih surovin (bioogljje, zeleni betoni).</p> <p>Ocena življenjskega cikla izdelka - LCA (Life Cycle Assessment) metoda, primerjava različnih računalniških LCA modelov, interpretacija LCA rezultatov.</p>	<p>resources usage, preservation of natural resources, use of secondary raw materials, recycling, implementation of closed loops, zero waste management, education of employees and product users.</p> <p>Energy: conventional and renewable energy sources, possibilities of renewable energy usage, problems related to renewable energy implementation (logistics, availability, variations, environmental impact), expected trends.</p> <p>Basics of industrial ecology: integrated pollution prevention strategies, LCA (Life Cycle Assessment) approaches, eco-optimization of products and processes, effective transportation, monitoring and improvement of existing systems, importance of local micro-conditions, local, regional and global approach.</p> <p>Implementation of concept of industrial ecology: A case study with product or process relevant in time and place.</p> <p>Advanced oxidation processes and hybrid processes. Use of secondary raw materials (biochar, green concrete).</p> <p>LCA (Life Cycle Assessment) approaches, comparison of different computational LCA models, interpretation of LCA results.</p>
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Temeljna literatura in viri/Readings:

OSNOVNA LITERATURA-knjižnica:

- P. P. Rogers, K. F. Jalal, J. A. Boyd: An Introduction to Sustainable Development, Harvard University Press, 2006, 404 strani (25%).
- M. Jeguirim, L. Limousy: Char and Carbon Materials Derived from Biomass; Production, Characterisation and Applications, Elsevier, 2019, 423 strani (5%).
- G. M. Masters: Introduction to Environmental Engineering and Science, Prentice Hall, 1996, 651 strani (15%).
- J. Zagorc-Končan, A. Žgajnar Gotvajn: Zbirka nalog iz ekološkega inženirstva, UL, FKKT, 2022, 45 strani (70%).
- R.U. Ayres, L. Ayres: A Handbook of Industrial Ecology, Edward Elgar, 2002, 680 strani (15%).
- A. Žgajnar Gotvajn, I. Boševski, M. Čehovin, Napredni oksidacijski in hibridni procesi v postopkih čiščenja odpadnih vod, Fakulteta za kemijo in kemijsko tehnologijo UL, Ljubljana, 2023, 67 strani (5%).

Cilji in kompetence:

Cilji:

Poznavanje definicij in pomena trajnostnega razvoja. Spoznanje o lokalnih in globalnih vplivih in vpetosti industrijskega procesa v lokalno in globalno okolje.

Kompetence:

Znanje o pristopih in metodah za doseg trajnostnega razvoja. Razumevanje osnovnih načel za načrtovanje čistejših proizvodnih v obstoječi ali na novo načrtovani proizvodnji v različnih industrijskih branžah. Zavedanje o etični in družbeni odgovornosti in potrebi po nenehnem izpopolnjevanju že postavljenega sistema.

Objectives and competences:

Objectives:

Knowledge on definitions in sustainable development and awareness of its relevance in modern society. Knowledge on local and global impacts of industrial processes and their incorporation into the local and global environment.

Competences:

Knowledge on approaches and methods for implementation of concept of sustainable development. Basics of design and implementation of clean production in existing or/and planned industrial processes. Awareness of ethical and social responsibility and need for constant improvement of existing systems.

Predvideni študijski rezultati:

Znanje in razumevanje
 Razumevanje povezav in odnosov med osvojenimi pojmi. Sposobnost vrednotenja vpliva procesov (lokalno in globalno) na ljudi in okolje. Znanje za vpeljavo koncepta trajnosti v obstoječ industrijski proces
 Uporaba
 Uporaba pridobljenih znanj pri reševanju okoljskih problemov. Sposobnost sinteze in interdisciplinarnega pristopa k reševanju problemov.
 Refleksija
 Razumeti pomen izbire ustreznih tehnoloških postopkov in surovin za ohranjanje naravnih virov.
 Kritično vrednotiti vpliv svojega dela na lokalni in globalni ravni.
 Prenosljive spretnosti
 Spretnost uporabe domače in tuje literature.
 Spretnost identifikacije problema in pristopa k njegovemu učinkovitemu reševanju. Uporaba ustnega in pisnega načina poročanja. Spretnost sinteze na različnih področjih pridobljenih znanj. Spretnost skupinskega dela.

Intended learning outcomes:

Knowledge and Comprehension
 Understanding relationships between different terms. Ability to evaluate the impact of processes to environment and people locally and globally. Ability to implement the concept of sustainability into new or existing industrial process.
 Application
 Ability of applying acquired knowledge for solving environmental problems. Ability of interdisciplinary approach to problem solving.
 Analysis
 Understand the importance of selection of appropriate technologies and raw materials to protect natural resources. Evaluate the work critically on local as well as global basis.
 Skill-transference Ability
 Ability to search, select and apply different types of literature. Ability to independently identify various environmental problems and search for solution. Development of oral and literate skills. Developing the skill of team work.

Metode poučevanja in učenja:

Predavanja
 Laboratorijske vaje
 Projektno delo

Learning and teaching methods:

Lectures
 Lab course
 Project work

Načini ocenjevanja:**Delež/Weight****Assessment:**

Opravljene obveznosti pri vajah.	20,00 %	Accomplished lab course.
Pisni in ustni izpit.	70,00 %	Written and oral exam.
Predstavitev projekta.	10,00 %	Project presentation.

Reference nosilca/Lecturer's references:

ŽGAJNAR GOTVAJN, Andreja, KALČIKOVÁ, Gabriela. Delamination of plastic-coated waste paper by enzymes of the white rot fungus *Dichomitus squalens*. *Journal of environmental management*. Dec. 2018, vol. 228, str. 165-168.
BOŠEVSKI, Igor, ŽGAJNAR GOTVAJN, Andreja. The impact of single step ozonation of antibiotics-contaminated waste sludge to biogas production. *Chemosphere*. 2021, vol. 271, str. 1-8.
KORICA, Predrag, CIRMAN, Andreja, ŽGAJNAR GOTVAJN, Andreja. Comparison of end-of-life vehicles management in 31 European countries : a LMDI analysis. *Waste management & research*. 2022, vol. 40, iss. 8, str. 1156-1166.

KEMIJA OKOLJA

UČNI NAČRT PREDMETA/COURSE SYLLABUS

Predmet:	Kemija okolja
Course title:	Environmental Chemistry
Članica nosilka/UL Member:	UL FKKT

Študijski programi in stopnja	Študijska smer	Letnik	Semestri	Izbirnost
Kemijska tehnologija, prva stopnja, visokošolski strokovni	Ni členitve (študijski program)	2. letnik	poletni	obvezni

Univerzitetna koda predmeta/University course code:	0072032
Koda učne enote na članici/UL Member course code:	KT133

Predavanja /Lectures	Seminar /Seminar	Vaje /Tutorials	Klinične vaje /Clinical tutorials	Druge oblike študija /Other forms of study	Samostojno delo /Individual student work	ECTS
30	20	25 LV			75	5

Nosilec predmeta/Lecturer: prof. dr. Helena Prosen

Vrsta predmeta/Course type: obvezni/mandatory

Jeziki/Languages:

Predavanja/Lectures:	Slovenščina
Vaje/Tutorial:	Slovenščina

Pogoji za vključitev v delo oz. za opravljanje študijskih obveznosti:

Študent oz. kandidat mora imeti predmet opredeljen kot študijsko obveznost.

Prerequisites:

The course has to be assigned to the student.

Vsebina:

- Splošni pojmi, lastnosti troposfere, stratosfere.
- Nastanek, pretvorbe in transport atmosferskih onesnaževal (trdni delci, CO, CO₂, SO₂, NO_x, O₃, ogljikovodiki). Pojav ozonskih lukenj in tople grede. Posledice onesnaževanja atmosfere (kisel dež, pojav mračenja). Ukrepi za zmanjšanje onesnaževanja.
- Površinske in podtalne vode. Kemija in biokemija onesnaževal v hidrosferi. Razgradljiva in nerazgradljiva onesnaževala voda in njihov vpliv na zdravje ljudi. Ukrepi za zmanjševanje onesnaženja voda.
- Zemlja in glavna onesnaževala. Problem nitratov in fosfatov v površinskih vodah in nitratov v podtalnici. Obstojna kemijska onesnaževala (klorirane spojine, policiklični aromati, fitofarmacevtska sredstva, kovine) in njihova usoda v okolju.

Content (Syllabus outline):

- General concepts, properties of troposphere and stratosphere.
- Sources, transformations and transport of atmospheric pollutants (particulate matter, CO, CO₂, SO₂, NO_x, O₃, hydrocarbons). Ozone hole and greenhouse phenomena. Atmospheric pollution consequences (acid rain, dimming). Measures to decrease pollution.
- Surface and ground water. Chemistry and biochemistry of pollutants in hydrosphere. Degradable and non-degradable pollutants of waters, their influence on public health. Measures to decrease water pollution.
- Soil and its principal pollutants. Role of nitrates and phosphates in surface waters and nitrates in ground waters. Stable chemical pollutants

<p>5. Trdni odpadki - viri. Problemi z odlagališči in sežiganjem odpadkov.</p> <p>6. Energija in okolje. Jedrska energija in radioaktivni odpadki.</p> <p>7. Določanje splošnih in specifičnih onesnaževal. Vzorčenje in tehnike priprave okoljskih vzorcev. Hitri testi in senzorji za spremljanje onesnaženja okolja. Analitske tehnike za določanje organskih in anorganskih onesnaževal v atmosferi, v vodah in v zemlji.</p> <p>8. Ukrepi za zmanjševanje onesnaženja okolja.</p> <p>Laboratorijske vaje: določanje onesnaževal v vzorcih zraka, vode in tal z različnimi analiznimi tehnikami.</p>	<p>(chlorinated compounds, polycyclic aromatics, phytopharmaceuticals, metals) and their environmental fate.</p> <p>5. Solid waste - sources. Problematic issues of landfills and waste incinerators.</p> <p>6. Energy and environment. Nuclear energy and radioactive waste.</p> <p>7. Determination of general and specific pollutants. Sampling and sample preparation techniques for environmental samples. Rapid tests and sensors for pollution monitoring. Analytical techniques for organic and inorganic pollutant determination in atmosphere, water and soil.</p> <p>8. Measures to decrease environmental pollution. Laboratory work: pollutant determination in atmospheric, aqueous and soil samples with different analytical techniques.</p>
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Temeljna literatura in viri/Readings:

Temeljna literatura:

- G.W. vanLoon, S.J. Duffy: Environmental Chemistry, 3rd ed., Oxford Univ. Press, Oxford UK, 2011, 545 str.
- G. Fellenberg: The Chemistry of Pollution, Wiley 2000, 192 str. (20%)
- B.B. Kebbekus, S.Mitra: Environmental Chemical analysis, Blackie Academic&Professional, London 1998, 330 str. (30%)

Dopolnilna literatura:

- F.W. Fifield, P.J. Haines (eds.): Environmental Analytical Chemistry, 2nd ed., Blackwell Science, Oxford UK, 2000
- J.E. Girard: Principles of Environmental Chemistry, 2nd ed., Jones and Bartlett Publ., Sudbury, MA, USA, 2010
- znanstveni in strokovni članki / scientific and professional articles

Cilji in kompetence:

Cilji: Predstaviti študentom glavna onesnaževala atmosfere, vod in zemlje, njihove vplive na okolje in njihovo analitiko v okoljskih vzorcih.

Kompetence: Sposobnost razumevanja osnovnih okoljskih dejstev; sposobnost opazovanja različnih pojavov; sposobnost predstavitve določenih okoljskih problemov ustno in v pisni obliki; sposobnost razreševanja konkretnih okoljskih problemov, sposobnost izbire ustrezne tehnike priprave vzorca in analize za različna onesnaževala.

Objectives and competences:

Objectives: To inform the students about the principal pollutants in atmosphere, water and soil; their influence on the environment; analytical determination in environmental samples.

Competences: Ability to understand basic environmental facts; ability to observe diverse phenomena; ability to present selected environmental problems in oral and written form; ability to solve particular environmental problems; ability to select an appropriate sample preparation and analytical technique for different pollutants.

Predvideni študijski rezultati:

Znanje in razumevanje
Študent bo spoznal osnovna okoljska onesnaževala. Iz lastnosti okoljskih onesnaževal, ki jih je že delno spoznal pri drugih predmetih, lahko oceni njihov vpliv na kvaliteto okolja. Iz predstavljenih procesov za zmanjševanje emisij bo znal oceniti mejne vrednosti posameznih onesnaževal v okolju in jih pravilno določiti s primerno analizno tehniko. Uporaba

Intended learning outcomes:

Knowledge and Comprehension
Student will be informed about principal environmental pollutants. They can evaluate their influence on environment quality from their properties, which were in part introduced in other courses. Limit values of certain pollutants in the environment will be evaluated from the presented processes for emission lowering and accurately determined by an appropriate analytical technique.

<p>Študent je sposoben kritično ovrednotiti vpliv posameznega onesnaževala na okolje in oceniti nevarnost, ki jo predstavlja za ljudi.</p> <p>Refleksija Študent bo pridobil tudi določen občutek za kritično oceno kvalitete okolja.</p> <p>Prenosljive spretnosti Študent bo znal uporabljati osnovne analize metode za hitro določanje onesnaževal. Na osnovi teh meritev in njihove kritične ocene bo lahko sklepal o onesnaženosti okolja.</p>	<p>Application Student is able to critically evaluate the influence of particular pollutant on the environment and assess the risk for the population.</p> <p>Analysis Student will gain a certain ability to critically evaluate the environmental quality.</p> <p>Skill-transference Ability Student will be able to apply basic analytical methods for rapid pollutant determination. They will be able to assess the environmental pollution, based on these measurements and their critical evaluation.</p>
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Metode poučevanja in učenja:

Predavanja, seminarji in laboratorijske vaje

Learning and teaching methods:

Lectures, seminars, laboratory work

Načini ocenjevanja:

Delež/Weight

Assessment:

pisni izpit (poz. ocena 6-10)	60,00 %	written exam (pass grade 6-10)
seminarska naloga z ustno predstavitvijo,	30,00 %	seminar work with oral presentation,
laboratorijske vaje	10,00 %	laboratory work

Reference nosilca/Lecturer's references:

KRAŠEVEC, Ida, **PROSEN, Helena**. Determination of polar benzotriazoles in aqueous environmental samples by hollow-fibre microextraction method with LC-MS/MS and its comparison to a conventional solid-phase extraction method. *Microchemical Journal* **2021**, 166, 106191, 9 str.

VEROVŠEK, Taja, JANŽA, Mitja, HEATH, David John, ŠUŠTARIČ, Ariana, **PROSEN, Helena**, HEATH, Ester. Occurrence and sources of residues of drugs of abuse in an urban aquifer: chemical analysis and solute transport modelling. *Science of the Total Environment* **2023**, 892, 164364, 8 str.

KUŠČER, Danjela, REPIČ, Barbara, MAKAROVIC, Kostja, DEKLEVA, Maksimiljan, MAROLT, Gregor, **PROSEN, Helena**, BELAVIČ, Darko. An advanced miniature fluidic system in multilayer ceramic technology with precise temperature and flow control for *in situ* pollution monitoring. *Sensors and Actuators. A, Physical* **2024**, 366, 114946, 9 str.

KEMIJSKA IN PROCESNA VARNOST

UČNI NAČRT PREDMETA/COURSE SYLLABUS

Predmet:	Kemijska in procesna varnost
Course title:	Chemical and Process Safety
Članica nosilka/UL Member:	UL FKKT

Študijski programi in stopnja	Študijska smer	Letnik	Semestri	Izbirnost
Kemijska tehnologija, prva stopnja, visokošolski strokovni	Ni členitve (študijski program)	2. letnik	poletni	obvezni

Univerzitetna koda predmeta/University course code:	0072033
Koda učne enote na članici/UL Member course code:	KT136

Predavanja /Lectures	Seminar /Seminar	Vaje /Tutorials	Klinične vaje /Clinical tutorials	Druge oblike študija /Other forms of study	Samostojno delo /Individual student work	ECTS
45	15	15 LV			75	5

Nosilec predmeta/Lecturer: doc. dr. Barbara Novosel

Vrsta predmeta/Course type: obvezni/mandatory

Jeziki/Languages:

Predavanja/Lectures:	Slovenščina
Vaje/Tutorial:	Slovenščina

Pogoji za vključitev v delo oz. za opravljanje študijskih obveznosti:

Študent oz. kandidat mora imeti predmet opredeljen kot študijsko obveznost.

Prerequisites:

The course has to be assigned to the student.

Vsebina:

Prepoznavanje, razumevanje in obvladovanje nevarnosti zaradi uporabe različnih kemikalij in nevarnih snovi. Nevarne snovi: eksplozivne, vnetljive, oksidativne, strupene, radioaktivne, jedke in okolju nevarne kemikalije ter plini. Prevoz in skladiščenje nevarnih kemikalij. Nov evropski sistem obvladovanja tveganja pri ravnanju s kemikalijami - REACH.

Varnost pri delu v laboratoriju, kemijskih in drugih procesnih industrijah. Kompleksnost delovanja industrijskega procesa, tehnološki režim in standardni proizvodni postopki, ustreznega vzdrževanja posameznih naprav in celotnega sistema.

Verjetnost za nastanek izrednih situacij ter preprečevanje in ukrepanje v takih primerih. Analiza industrijskih procesov in priprava ocen tveganja.

Content (Syllabus outline):

Identifying, understanding and managing risk due to the use of various chemicals and hazardous substances. Hazardous substances: explosive, flammable, oxidizing, toxic, radioactive, corrosive and environmentally hazardous chemicals and gases.

Transportation and storage of hazardous chemicals. A new European system of risk management in handling the chemicals - REACH.

Safety at work in a laboratory, chemical and other process industries. The complexity of the operation of industrial processes, technological regime and the standard manufacturing procedures, proper maintenance of individual devices and the entire system.

Probability of stress and prevent or take action in such cases. Analysis of industrial processes and the

<p>Zajemanje pomembnih in kritičnih parametrov ter pogojev procesa, ki vplivajo na njegovo varnost, možni scenariji izrednih dogodkov. Verjetnost dogodka in analiza odpovedi. Kvalitativna in kvantitativna ocena tveganja. Določevanje ukrepov za zmanjšanje tveganja.</p> <p>Vaje: iskanje varnostnega lista, priprava navodil za delo z nevarno kemikalijo; vnetljive tekočine, določitev plamenišča; vnetljive trdne snovi, razvrščanje; plini, označevanje, določevanje trenutnih koncentracij; prašne eksplozije, določevanje minimalne eksplozijske energije; jedke snovi, razredčevanje, nevtralizacija.</p>	<p>preparation of risk assessments. Capture important and critical process parameters and conditions that affect the security of the possible scenarios of emergencies. Probability of the event and failure analysis. Determination of risk reduction measures.</p> <p>Tutorial: Student must find the safety data sheet, and createsafety card; flammable liquids, flash point determination; flammable solids, clasification; labeling, measuring of current concentrations of gases; dust explosions, minimum ignition energy determination; corrosive substances, dilution, neutralization</p>
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Temeljna literatura in viri/Readings:

- Burke R.: Hazardous materials chemistry for emergency responders, 3rd Ed. CRC Press, Boca Raton, 2013., 527 str., (30%)
- Brauer, R., L.: Safety and health for engineers, 3rd Ed., Wiley, cop. Hoboken (New Jersey), 2016, 765 str. (30%)
- Crowl D.A., Louvar J.F., Chemical Process Safety, 2nd Ed., Prentice Hall PTR, New Jersey 2002, (str.625) (10 %)

Cilji in kompetence:

Pri predmetu se študenti seznanijo z zahtevnostjo in kompleksnostjo zagotavljanja varnosti pri delu z različnimi kemikalijami v laboratoriju in v kemijskih ter procesnih industrijah. Spoznajo potrebo po natančnem poznavanju vseh lastnosti kemikalij, ki jih pri svojem delu uporabljajo. Pridobljeno znanje usposobi študenta za razumevanje in presojanje nevarnosti oziroma stopnje tveganja ter za določitev ukrepov, ki zagotavljajo varno delo z nevarnimi kemikalijami.

Študentje spoznajo, da je za varno delo v industriji osnovni pogoj natančno poznavanje vseh faz procesa in podrobna analiza njegovega delovanja na osnovi katere se izvede ocene tveganja. Spoznajo, da je varnost procesa pogojena z mnogo dejavniki in da je za njegovo varno obratovanje potrebno tako optimalno delovanje posameznih procesnih operacij kot tudi usklajeno delovanje sistema kot celote.

Neustrezno načrtovanje, vodenje ali, vzdrževanje procesa lahko vodi do odstopanj od tehnološkega režima in ogrožanja varnosti. Napake se lahko širijo po celotnem sistemu, potencirajo in lahko vodijo do nastanka izrednih razmer, ogrožanja delovnega in širšega okolja. Študentje se pri predmetu usposobijo za sistematičen pregled kemijskih in drugih sorodnih procesov, zaznavanje potencialnih kritičnih mest, priprave ocene tveganja in ukrepov za zmanjšanje tveganja.

Objectives and competences:

At this subject students get acquainted with pretentiousness and the complexity of ensuring safety at work with various chemicals in the laboratory and in the chemical and process industries. Learn about the need of precise knowledge of the characteristics of chemicals they use at work. The course enables understanding and assessment of hazards and level of risk, and the measures for safe and healthy work. Students learn that basic conditions for safe work in industry are exact knowledge of all phases of the process and detailed analysis of its performance on the basis a risk assessment. Students learn that process safety depends on many factors and for safe operation is necessary so optimize the performance of individual part of the process operation as well as the coordinated operation of the system as a whole. Inadequate planning, management or maintenance process can lead to deviations from the technological regime and security threats. Errors may spread throughout the system, potentiate, and may lead to the formation of an emergency, endangering the work and the wider environment. Students are trained in the course of a systematic review of chemical and other related processes, identify potential critical points, preparation of risk assessment and risk reduction measures.

Predvideni študijski rezultati:

Znanje in razumevanje
Študentje se pri predmetu usposobijo za sistematičen pregled nevarnosti pri uporabi nevarnih kemikalij ter

Intended learning outcomes:

Knowledge and Comprehension
The subject makes students capable of a systematic review of the risks of using dangerous chemicals and

<p>pri delu v kemijskih in sorodnih procesih, za zaznavanje potencialnih kritičnih mest, pripravo ocene tveganja in ukrepov za zmanjšanje tveganja.</p> <p>Uporaba Delo z nevarnimi kemikalijami. Vodenje in nadzor kemijskih procesov. Ocenjevanje tveganja kemijskih procesov. Analiza nezgod in določevanje ukrepov za preprečitev nezgod.</p> <p>Refleksija Študenta se usmeri v podrobnejši pregled lastnosti posamezne kemikalije z namenom, da ugotovi nevarnosti snovi za človeka in okolje. Na osnovi spoznanj mora določiti varnostne ukrepe za zmanjšanje ali celo eliminacijo tveganja pri rabi kemikalije.</p> <p>Glede na veljavno SI zakonodajo so podana znanja osnova za opravljanje izpita za svetovalce za kemikalije v različnih podjetjih, kakor tudi temelji za delo v carinski, komercialni ali inšpektorski službi.</p> <p>Prenosljive spretnosti Sistematičen, analitičen pristop do reševanja problema, več razumevanja in upoštevanja varnostne kulture v širšem kontekstu.</p>	<p>work in the chemical and related processes for detecting potential critical points, risk assessment and risk reduction measures.</p> <p>Application Work with hazardous chemicals. Management and control of chemical processes. Risk assessment of chemical processes. Analysis of accident and determination of measures to prevent accidents.</p> <p>Analysis A student will be directed to a more detailed examination of the properties of each chemical in order to identify the hazards of the substance for humans and the environment. Based on the findings should establish the security measures for the reduction or even elimination of risk in the use of chemicals.</p> <p>According to the current SI legislation are given the knowledge base for the exam for counselors of chemicals in different companies, as well as based on the work of customs, commercial or inspector service.</p> <p>Skill-transference Ability Systematic, analytical approach to problem solving, more understanding and observance of safety culture in a broader context.</p>
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Metode poučevanja in učenja:

Predavanja, seminarji, praktične vaje.

Learning and teaching methods:

Lectures, seminars, practical exercises.

Načini ocenjevanja:

Delež/Weight

Assessment:

Opravljenе vaje, seminar (predstavitel), pisni izpit (računske naloge), ustni izpit (15 min), ocenjevalna lestvica skladna s Statutom UL.		Submitted reports on laboratory practical exercises.
Vaje	20,00 %	Laboratory practice
Seminar	10,00 %	Presentation of the seminar.
pisni izpit	40,00 %	Written exam
ustni izpit	30,00 %	Oral examination.

Reference nosilca/Lecturer's references:

- SLABAJNA, Dominika, **NOVOSEL, Barbara**. Smernica za zagotavljanje varnosti in zdravja v kemijskih laboratorijih : projekt Kemijska varnost 3. Ljubljana: Urad RS za kemikalije: Univ. v Ljubljani, Fak. za kemijo in kemijsko tehnologijo, 2010. 48 str., ilustr. <http://www.fkkt.uni-lj.si/si/?2416>. [COBISS.SI-ID 34765317]

- **NOVOSEL, Barbara**, MARINŠEK, Marjan. Računska obravnava kemijskih procesov : zbirka nalog. V Ljubljani: Fakulteta za kemijo in kemijsko tehnologijo, 2003. 132 str., ilustr. ISBN 961-6286-56-0. [COBISS.SI-ID 125977600]

-NOVOSEL, Barbara. Ugotavljanje kritičnih mest v kemijski industriji in zmanjševanje tveganja nezgod. V: BRVAR, Miran (ur.). Kemijske nesreče na delovnem mestu : zbornik prispevkov. Ljubljana: Slovensko zdravniško društvo, Sekcija za klinično toksikologijo, 2013, str. 78-83. [COBISS.SI-ID 1654319]

- MAČEK, Jadran, **NOVOSEL, Barbara**, MARINŠEK, Marjan. Anorganska kemijska tehnologija, Navodila za vaje za 3. letnik UN ŠP Kemijsko inženirstvo. Ljubljana: Fakulteta za kemijo in kemijsko tehnologijo, Katedra za anorg. kem. tehnologijo in materiale, 2001/2002. III, 61 str., tabele. [COBISS.SI-ID 24156165]

- MISLEJ, Vesna, **NOVOSEL, Barbara**, VUK, Tomaž, GRILC, Viktor, MLAKAR, Ernest. Combustion behaviour and products of dried sewage sludge - prediction by thermogravimetric analysis and monitoring the co-incineration process in a cement factory. V: 20th International Congress of Chemical and Process

ULB FAKULTA

MATEMATIKA IN STATISTIKA

UČNI NAČRT PREDMETA/COURSE SYLLABUS

Predmet:	Matematika in statistika
Course title:	Mathematics and Statistics
Članica nosilka/UL	UL FKKT
Member:	

Študijski programi in stopnja	Študijska smer	Letnik	Semestri	Izbirnost
Kemijska tehnologija, prva stopnja, visokošolski strokovni	Ni členitve (študijski program)	1. letnik	celoletni	obvezni

Univerzitetna koda predmeta/University course code:	0100781
Koda učne enote na članici/UL Member course code:	KT101S

Predavanja /Lectures	Seminar /Seminar	Vaje /Tutorials	Klinične vaje /Clinical tutorials	Druge oblike študija /Other forms of study	Samostojno delo /Individual student work	ECTS
90		60 SV			150	10

Nosilec predmeta/Lecturer:	prof. dr. Jakob Cimprič
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Vrsta predmeta/Course type:	obvezni/mandatory
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Jeziki/Languages:	Predavanja/Lectures:	Slovenščina
	Vaje/Tutorial:	Slovenščina

Pogoji za vključitev v delo oz. za opravljanje študijskih obveznosti:

Študent oz. kandidat mora imeti predmet opredeljen kot študijsko obveznost.

Prerequisites:

The course has to be assigned to the student.

Vsebina:

Limite funkcij: računske operacije s funkcijami (vsota, produkt, kompozitum, inverzna funkcija), zveznost, asimptote, lastnosti zveznih funkcij.
Odvod in njegova uporaba: geometrijski pomen, pravila za odvajanje, odvodi elementarnih funkcij, diferencial in njegova uporaba, višji odvodi, L'Hospitalovo pravilo, ekstremini, konveksnost, konkavnost in prevoji, uporaba odvoda.
Funkcije več spremenljivk: funkcija dveh spremenljivk in njen graf, zveznost, parcialni odvodi, posredno odvajanje, implicitne funkcije, totalni diferencial, gradient, ekstremini, vezani ekstremini.
Nedoločeni integral: osnovne lastnosti, integriranje po delih, vpeljava nove spremenljivke, integrali osnovnih elementarnih funkcij (racionalnih in nekaterih trigonometrijskih).

Content (Syllabus outline):

Limits of functions: computation with functions (sum, product, composition, inverse), continuity, asymptotes, properties of continuous functions.
The derivative and its application: the geometric meaning, rules for differentiation, the derivatives of elementary functions, the differential and its applications, higher derivatives, L'Hospital rule, minima and maxima, convexity and concavity, application of the derivative.
Functions of several variables: functions of two variables and their graphs, continuity, partial derivatives, the chain rule, implicit functions, extrema, constrained extrema.
The indefinite integral: basic properties, integration per partes, change of variables, integration of

<p>Določeni integral: geometrijski pomen in osnovne lastnosti, zveza z nedoločnim integralom, izlimitirani integrali.</p> <p>Uporaba integrala: ploščina, ločna dolžina, prostornina in površina vrtenine, težišče, vztrajnostni moment.</p> <p>Diferencialne enačbe: enačbe prvega reda z ločljivima spremenljivkama, homogene, linearne, linearne diferencialne enačbe drugega reda s konstantnimi koeficienti, sistemi linearnih diferencialnih enačb prvega reda s konstantnimi koeficienti, uporaba v kemiji in drugod.</p> <p>Osnove statistike: predstavitev podatkov, populacija, vzorec, cenilke. Hipoteze, korelacija in linearna regresija, metoda najmanjših kvadratov. Intervali zaupanja za srednjo vrednost in disperzijo ter testiranje hipotez</p>	<p>elementary functions (rational and certain trigonometric ones).</p> <p>The definite integral: the geometric meaning and basic properties, the fundamental theorem of calculus, improper integrals.</p> <p>Application of integration: calculations of areas, arc lengths, volumes and surfaces of revolution, centers of mass, moments of inertia.</p> <p>Differential equations: equations of order 1, separation of variables, homogeneous and linear equations, second-order linear differential equations with constant coefficients, systems of linear differential equations, applications to chemistry and elsewhere.</p> <p>The basics of statistics: data presentation, population, samples, estimators. Hypothesis, correlation and linear regression, the least square method. Interval estimation and hypothesis testing.</p>
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Temeljna literatura in viri/Readings:

R. Jamnik, Matematika, DMFA Slovenije, Ljubljana, 1994.
P. Šemrl, Osnove višje matematike, DMFA Slovenije, Ljubljana, 2009.

Dopolnilna literatura:

- A. Turnšek, Tehniška matematika, FS, Ljubljana, 2007, 306 str.
- P. Mizori – Oblak, Matematika za študente tehnike in naravoslovja, FS, UL Ljubljana, 2001.
- P. Mizori – Oblak, Matematika za študente tehnike in naravoslovja, 2. del, FS UL, Ljubljana, 1997.
- I. Vidav, Višja matematika I, DMFA Slovenije, Ljubljana, 1994, 477 str.
- G. Doggett, B. T. Sutcliffe, Mathematics for Chemistry, Longman, 1995, 286 str.

Cilji in kompetence:

Cilj predmeta: Seznaniti študente z osnovnimi metodami matematične analize in linearne algebre, potrebnimi pri nadaljnem študiju, ki spadajo v temeljno izobrazbo naravoslovca ali tehnika.

Predmetno specifične kompetence: Pridobljeno znanje bo študentu omogočilo boljše razumevanje drugih strokovnih predmetov. Imel bo možnost pridobiti nekaj temeljnih matematičnih pojmov in spretnosti, ki so potrebne za razumevanje strokovne literature in tudi za uspešno opravljanje dela.

Objectives and competences:

To familiarize students with calculus and basic linear algebra necessary for further study. This is a usual part of curriculum for students of science and technology. This enables students to better understand some other areas of their study. It gives them an opportunity to acquire basic mathematical skills needed to follow the literature in their own speciality.

Predvideni študijski rezultati:

Znanje in razumevanje
Razumevanje pojmov funkcijske odvisnosti, limite, odvoda in integrala,
poznavanje metod reševanja nekaterih elementarnih tipov diferencialnih enačb in njihove uporabe v kemiji (in drugod), osnovna analiza funkcij več spremenljivk.
Uporaba
Uporaba zgoraj omenjenih pojmov pri reševanju konkretnih nalog iz matematike, fizike in kemije.
Refleksija

Intended learning outcomes:

Knowledge and Comprehension
Students should understand the concepts of functional dependence, limits, differentiation and integration, and acquire the skill of solving certain types of differential equations and their application to chemistry (and elsewhere), basic analysis of functions of several variables.

Application
Students should be able to apply calculus and linear algebra to problems from physics and chemistry.

Analysis

<p>Gre za poglobitev in bistveno razširitev v srednji šoli pridobljenega znanja matematike, ki je nujno za razumevanje naravoslovnih znanosti.</p> <p>Prenosljive spretnosti</p> <p>Predmet daje tudi osnovo za razumevanje nekaterih računalniških postopkov in metod, ki jih bodo spoznali kasneje pri drugih predmetih in ob delu.</p>	<p>The course gives a considerable extension of the mathematical knowledge that the students acquired in high school, which is essential for the understanding of any natural science and chemistry in particular.</p> <p>Skill-transference Ability</p> <p>The knowledge of calculus is necessary for effective use of computer modeling in science, which the students will meet later in the course of their study.</p>
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Metode poučevanja in učenja:	Learning and teaching methods:
Predavanja, vaje, sodelovalno učenje / poučevanje.	Lectures, exercises, homework, consultations.

Načini ocenjevanja:	Delež/Weight	Assessment:
Pisni izpit iz praktičnega dela (ali štiri kolokviji).	50,00 %	Written practical exam (or four midterm tests).
Teoretični izpit. Od 6-10 (pozitivno), 1-5 (negativno).	50,00 %	Oral exam.

Reference nosilca/Lecturer's references:

- **Cvetko-Vah, Karin**; Leech, Jonathan; Spinks, Matthew. Skew lattices and binary operations on functions, J. Appl. Log. 11 (2013), no. 3, 253--265.
- Bauer, Andrej; **Cvetko-Vah, Karin**. Stone duality for skew Boolean algebras with intersections, Houston J. Math. 39 (2013), no. 1, 73--109.
- **Cvetko-Vah, Karin**; Leech, Jonathan. Rings whose idempotents form a multiplicative set, Comm. Algebra 40 (2012), no. 9, 3288--3307.
- **Cvetko Vah, Karin**; Pisanski, Tomaž. A census of edge-transitive planar tilings, J. Combin. Math. Combin. Comput. 80 (2012), 243--265.
- Carfi, David; **Cvetko-Vah, Karin**. Skew lattice structures on the financial events plane, Appl. Sci. 13 (2011), 9--20.

MEHANSKE OPERACIJE

UČNI NAČRT PREDMETA/COURSE SYLLABUS

Predmet:	Mehanske operacije
Course title:	Mechanical operations
Članica nosilka/UL Member:	UL FKKT

Študijski programi in stopnja	Študijska smer	Letnik	Semestri	Izbirnost
Kemijska tehnologija, prva stopnja, visokošolski strokovni (od študijskega leta 2023/2024 dalje)	Ni členitve (študijski program)	3. letnik	zimski	obvezni

Univerzitetna koda predmeta/University course code:	0072049
Koda učne enote na članici/UL Member course code:	KTSI3

Predavanja /Lectures	Seminar /Seminar	Vaje /Tutorials	Klinične vaje /Clinical tutorials	Druge oblike študija /Other forms of study	Samostojno delo /Individual student work	ECTS
45	15	15 LV			75	5

Nosilec predmeta/Lecturer: prof. dr. Marjan Marinšek

Vrsta predmeta/Course type: obvezni/mandatory

Jeziki/Languages:	Predavanja/Lectures:	Slovenščina
	Vaje/Tutorial:	Slovenščina

Pogoji za vključitev v delo oz. za opravljanje študijskih obveznosti:

Prerequisites:

Študent oz. kandidat mora imeti predmet opredeljen kot študijsko obveznost.	The course has to be assigned to the student.
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Vsebina:

Content (Syllabus outline):

- Priprava zrnatih kolektivov** (fizikalno kemijske metode priprave, mehanske metode priprave)
- Naprave in tehnologije priprave ter procesov **večanja površin**
- Velikost in porazdelitev velikosti**, fizikalni principi merjenja velikosti delcev
- Oblika delcev** (faktor oblike)
- Specifična površina** (določevanje specifične površine iz granulacijske sestave, določevanje specifične površine z adsorpcijo plinov)
- Poroznost** in porazdelitev por
- Gostota** sipkih snovi
- Vzorčenje** sipkih snovi
- Procesi **manjšanja površin**
- Transport** sipkih snovi

- Preparation of dispersed systems:** physical-chemical methods, mechanical methods
- Principles of **grinding and milling**
- Particle size and size distribution:** particle size definition and measurements principles.
- Particle shape:** (shape factor)
- Specific surface of dispersed systems:** basic principles and surface measurement
- Porosity and pore distribution in disperse systems**
- Density of dispersed systems**
- System sampling**
- Agglomeration in dispersed systems**
- Transport of dispersed systems**

<p>11. Izbrani primeri tehnologij ločevanja komponent v različnih disperznih sistemih</p> <p>Seminar: Slušatelji v okviru seminarja rešujejo računske primere, ki ilustrirajo principe priprave in karakterizacije zrnatih kolektivov.</p> <p>Vaje: Pri vajah se slušatelji praktično seznanijo z bistvenimi metodami priprave (mletje) in karakterizacije delcev (vzorčenje, merjenje velikosti, določevanje specifične površine)</p>	<p>11. Selected examples of component separation technologies in various dispersed systems</p> <p>Seminar work: Mechanical operations through mathematical problems</p> <p>Tutorial work: Laboratory work to demonstrate basic principles of mechanical operations.</p>
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Temeljna literatura in viri/Readings:

<ul style="list-style-type: none"> - J. Stražičar, Mehanska procesna tehnika, Univerza v Ljubljani, FNT, Ljubljana 1996, 138 strani - T. Allen, Particle Size Measurement, Vol.1, 5th ed., Chapman & Hall, London 1997, 774 strani - M. Rhodes, Introduction to Particle technology, John Wiley & Sons, 1998, 320 strani - R.H. Perry, Perry's Chemical Engineers Handbook, 7th. ed., McGraw-Hill, New York, 1997, 2581 strani
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Cilji in kompetence:

<p>Namen predmeta je priprava in karakterizacija zrnatih kolektivov.</p> <p>V okviru predmeta študent osvoji osnovna znanja o karakterizaciji delcev, ki vključujejo definicije velikosti delcev, njihove oblike, specifične površine, vzorčenja ter določevanja gostote sipkih snovi.</p> <p>V okviru predmeta bo študent pridobil tudi osnovna znanja o procesih večanja površin ter potrebni tehnološki opremi.</p>	<p>Objectives and competences:</p> <p>Definition and characterization of dispersed systems. Students acquire basic knowledge about dispersed system preparation and characterization including particle size and shape, system specific surface and density, as well as basic knowledge of dispersed system grinding and milling or system agglomeration.</p>
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Predvideni študijski rezultati:

<p>Znanje in razumevanje Študentje spoznajo osnovne principe ter naprave za pripravo zrnatih kolektivov. Osvojijo tudi osnovne metode njihove karakterizacije.</p> <p>Uporaba S pridobljenim znanjem je študent sposoben v laboratorijske kakor tudi v industrijskem merilu, glede na predpisane zahteve, izbrati primeren postopek priprave delcev ter njihove karakterizacije.</p> <p>Refleksija Na osnovi pridobljenega znanja študent pridobi občutek za smotno izbiro najprimernejšega tehnološkega postopka priprave zrnatega kolektiva.</p> <p>Prenosljive spretnosti Laboratorijske spretnosti na področju zrnatih kolektivov, seznanitev in učenje iz domače in tuje literature, pisna in ustna predstavitev izbrane tematike, delov skupina</p>	<p>Intended learning outcomes:</p> <p>Knowledge and Comprehension Basic knowledge about principles and equipment for dispersed systems preparation and characterization.</p> <p>Application Students are capable of choosing equipment for particle tailoring in laboratory or industrial scale.</p> <p>Analysis Students are capable of choosing appropriate equipment for dispersed system preparation in laboratory or industrial scale.</p> <p>Skill-transference Ability Developed laboratory skills in the field of dispersed systems, literature research; literature data collecting, data analysis and interpretation, team work.</p>
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Metode poučevanja in učenja:

<p>Predavanja, seminar, laboratorijsko delo, poučevanje in učenje izbrane vsebine s pomočjo umetniške izkušnje</p>	<p>Learning and teaching methods:</p> <p>Lectures, seminars, tutorial work, teaching and learning of selected topics through artistic experience</p>
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Načini ocenjevanja:

Delež/Weight Assessment:

<p>Preverjanja znanja – izpit - (2/3 skupne ocene) od tega pisni izpit (40% ocene) in ustni izpit (60% ocene). Izpit se v celoti lahko opravi z vmesnimi kolokviji in opravljeno seminarsko nalogo (seminarska naloga 20% ocene). Vaje: Opravljene vaje. Ocena vaj predstavlja 1/3 skupne ocene predmeta. Pozitivna ocena 6-10, negativna ocena 1-5</p>		<p>Written (40%) and oral (60%) exam. The exam can be accomplished also by achieving positive grades at written colloquiums during the semester (80%) and prepared individual seminar work (20%). Pass grades from 6 to 10, fail grades from 1 to 5. Tutorial work must be done before taking an exam.</p>
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Reference nosilca/Lecturer's references:

ŠTUKOVNIK, Petra, BOKAN-BOSILJKOV, Violeta, **MARINŠEK, Marjan**. Alkali-dolomite reaction in air lime mortar : implications for increased strength and water resistance. *Journal of cultural heritage*. 2020, letn. xx, str. 1-9

BITENC, Marko, **MARINŠEK, Marjan**, CRNJAK OREL, Zorica. Preparation and characterization of zinc hydroxide carbonate and porous zinc oxide particles. *Journal of the European Ceramic Society*, ISSN 0955-2219. [Print ed.], 2008, vol. 28, no. 15, str. 2915-2921

MARQUES, Susana, MESTRE, Ana S., MACHUQUEIRO, Miguel, ŽGAJNAR GOTVAJN, Andreja, **MARINŠEK, Marjan**, CARVALHO, Ana Paula. Apple tree branches derived activated carbons for the removal of β -blocker atenolol. *Chemical engineering journal*. Aug. 2018, vol. 345, str. 669-678

MERITVE IN OSNOVE REGULACIJE PROCESOV

UČNI NAČRT PREDMETA/COURSE SYLLABUS

Predmet:	Meritve in osnove regulacije procesov
Course title:	Measurements and fundamentals of process control
Članica nosilka/UL	UL FKKT
Member:	

Študijski programi in stopnja	Študijska smer	Letnik	Semestri	Izbirnost
Kemijska tehnologija, prva stopnja, visokošolski strokovni	Ni členitve (študijski program)	3. letnik	zimski	obvezni

Univerzitetna koda predmeta/University course code:	0072050
Koda učne enote na članici/UL Member course code:	KT139

Predavanja /Lectures	Seminar /Seminar	Vaje /Tutorials	Klinične vaje /Clinical tutorials	Druge oblike študija /Other forms of study	Samostojno delo /Individual student work	ECTS
30	15	30 LV			75	5

Nosilec predmeta/Lecturer: prof. dr. Andrej Jamnik

Vrsta predmeta/Course type: obvezni/mandatory

Jeziki/Languages:

Predavanja/Lectures:	Slovenščina
Vaje/Tutorial:	Slovenščina

Pogoji za vključitev v delo oz. za opravljanje študijskih obveznosti:

Prerequisites:

Študent oz. kandidat mora imeti predmet opredeljen kot študijsko obveznost.

The course has to be assigned to the student.

Vsebina:

Merjenje procesnih spremenljivk
Metrologija. Merski sistem. Enote in standardi. Hierarhija standardov. Sledljivost kalibracijske opreme do nacionalnih in mednarodnih standardov. Statične karakteristike merilnih instrumentov. Merjenje tlaka, nivoja, pretoka, temperature. Merjenje nekaterih ostalih količin.

Avtomatska regulacija procesov
Namen in pomen avtomatske regulacije kemijskih procesov. Osnovni pojmi in terminologija. Instrumentacijski in blokovni diagram, značilni elementi regulacijskega kroga. Odprti in zaprti regulacijski krog. Negativna povratna zanka. Standardni signali za prenos informacije v regulacijski zanki.

Content (Syllabus outline):

Measurements of process variables
Metrology. Measurement system. Units and standards. Hierarchy of standards. Traceability of measurements. Static characteristics of measuring instruments. Pressure, level, flow, temperature measurements. Measurements of some other process variables.

Process control
Incentives for chemical process control. Basic concepts and terminology. Feedback loop. Open and closed control loop. Instrumentation and block diagrams. Standard signals for transmission of information in the control loop. Analog-to-digital and digital-to-analog signal conversion.

Dynamic characteristics of systems
Dynamic response of first and second-order

Analogno-digitalna in digitalno analogna pretvorba signalov.

Dinamične karakteristike sistemov

Časovni odzivi sistemov prvega in drugega reda na različne vhodne spremenljivke.

Izvršilni členi: Avtomatski regulirni ventil. Regulirne črpalke.

Vrste povratnozančnih (feedback) regulatorjev

Dvopoložajna in tripoložajna regulacija. Proporcionalni, integrirni in diferencirni ter kombinirani načini regulacije.

Avtomatska regulacija enostavnih procesov: regulacija temperature, nivoja in pH.

Laboratorijske vaje

Merjenje tlaka, pretoka, temperature in nekaterih drugih procesnih količin. Umerjanje instrumentov.

Dinamične karakteristike sistemov 1., 2. in višjih redov. Laplaceova transformacija. Prenosne funkcije in njihova uporaba v regulacijski tehniki. Regulacija temperature, nivoja in pH. Karakteristike regulirnih ventilov. Dvopoložajna, P, PI in PID regulacija. Avtomatsko zajemanje podatkov.

measuring devices and processes on different input variables.

Final control elements: pneumatic control valve and control pumps.

Types of feedback controllers

Two and three position controllers. Proportional, integral, derivative and other types of feedback controllers.

Examples of simple control systems: control of temperature, level, and pH.

Laboratory practice

Measurement of pressure, flow, temperature and some other process variables. Calibration of some process measuring instruments.

Dynamic characteristics of first, second and higher order systems. Laplace transforms. Transfer functions and the input-output models.

Temperature, liquid level and pH control.

Characteristics of control valves. Characteristics of the ON/OFF, P, PI and PID controllers. Automatic data acquisition.

Temeljna literatura in viri/Readings:

- P.W. Murrill, *Fundamentals of process control theory* (3. izdaja), International Society of Automation (2012), 485 str., 40%.

- C.D. Johnson, *Process Control Instrumentation Technology* (8. izdaja), Pearson Education Limited (2014), 680 str., 25%

- C. Pohar in sodelavci, *Laboratorijske vaje iz meritev in osnov regulacije procesov*, interna skripta Katedre za fizikalno kemijo, Ljubljana (2017).

Dopolnilna literatura:

- R. Karba, *Gradniki sistemov vodenja* (1. izdaja), Ljubljana: Založba FER (1994). 326 str.

- G. Platt, *Process Control – A Primer for the Nonspecialist and the Newcomer* (2. izdaja), Instrument Society of America (1998), 216 str., 40%.

Cilji in kompetence:

Študenti spoznajo osnovne principe regulacijskega kroga s poudarkom na merjenju procesnih spremenljivk. Pridobijo temeljno znanje, ki je potrebno za nadaljnji samostojni študij na tem hitro razvijajočem se področju in za dialog s specialisti iz drugih strok.

Objectives and competences:

The course informs the student with the basic principles of control loop with special emphasis on the measurement of process variables. It offers them the basic knowledge needed for further independent study in this rapidly developing field and make possible to dialogue with specialists from other disciplines.

Predvideni študijski rezultati:

Znanje in razumevanje

Poznavanje različnih načinov merjenja procesnih spremenljivk. Razumevanje delovanja regulacijske zanke, osnov avtomatske regulacije, dinamičnih karakteristik regulacijskih sistemov in merilnih inštrumentov.

Uporaba

Pridobljeno znanje je širše uporabno in ni omejeno zgolj na področje kemijske tehnologije.

Refleksija

Intended learning outcomes:

Knowledge and Comprehension

Knowledge of different ways for measurements of various process variables. Understanding the operation of control loop, the basics of automatic control and dynamic characteristics of control systems and measuring instruments.

Application

Acquired knowledge is widely applicable and not limited only to the field of chemical technology.

Analysis

<p>Na podlagi pridobljenega znanja o meritvah in regulaciji se bo študent sposoben bolje in hitreje vključiti v tehnološko-procesno delovno okolje. Prenosljive spretnosti</p> <p>Študent je sposoben povezati znanje iz več predmetov v celoto ter ga uporabiti pri svojem delu tako v teku študija kot tudi kasneje pri opravljanju svojega poklica.</p>	<p>On the basis of the knowledge about the measurements and process control, the student will be able to better and more quickly incorporate into the technological process working environment.</p> <p>Skill-transference Ability</p> <p>The students are able to complement knowledge from this course with those obtained in other subjects and use it in their work both in the course of the study as well as later in the professional career.</p>
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Metode poučevanja in učenja:	Learning and teaching methods:
Predavanja, seminarji, laboratorijske vaje.	Lectures, seminars, laboratory practice.

Načini ocenjevanja:	Delež/Weight	Assessment:
Pisni izpit po uspešno opravljenih laboratorijskih vajah. Ocenjevanje: 6-10 (pozitivno); 1-5 (negativno).		Written examination after successful completion of practical work. Marks: 6-10 (positive); 1-5 (negative).

Reference nosilca/Lecturer's references:

- **M. Tomšič**, J. Cerar, **A. Jamnik**, *Supramolecular structure vs. rheological properties : 1,4-butanediol at room and elevated temperatures*, Journal of colloid and interface science, 557, 2019, 328-335.
- **M. Tomšič**, **A. Jamnik**, G. Fritz, O. Glatter, L. Vlček, *Structural properties of pure simple alcohols from ethanol, propanol, butanol, pentanol, to hexanol : comparing Monte Carlo simulations with experimental SAXS data*, The journal of physical chemistry B, 111, 2007, 1738-1751.
- I. Dogša, **M. Tomšič**, J. Orehek, E. Benigar, **A. Jamnik**, D. Stopar, *Amorphous supramolecular structure of carboxymethyl cellulose in aqueous solution at different pH values as determined by rheology, small angle X-ray and light scattering*, Carbohydrate polymers, 111, 2014, 492-504.

NAČRTOVANJE TRAJNOSTNIH KEMIJSKIH PRODUKTOV

UČNI NAČRT PREDMETA/COURSE SYLLABUS

Predmet:	Načrtovanje trajnostnih kemijskih produktov
Course title:	Design of Sustainable Chemical Products
Članica nosilka/UL	UL FKKT
Member:	

Študijski programi in stopnja	Študijska smer	Letnik	Semestri	Izbirnost
Kemijska tehnologija, prva stopnja, visokošolski strokovni	Ni členitve (študijski program)	2. letnik, 3. letnik	poletni	izbirni

Univerzitetna koda predmeta/University course code: 0643415

Predavanja /Lectures	Seminar /Seminar	Vaje /Tutorials	Klinične vaje /Clinical tutorials	Druge oblike študija /Other forms of study	Samostojno delo /Individual student work	ECTS
20	55				75	5

Nosilec predmeta/Lecturer: doc. dr. Blaž Zupan, prof. dr. Urška Šebenik

Vrsta predmeta/Course type: izbirni strokovni/Elective Professional

Jeziki/Languages:

Predavanja/Lectures:	Slovenščina
Vaje/Tutorial:	

Pogoji za vključitev v delo oz. za opravljanje študijskih obveznosti:

Študent oz. kandidat mora imeti predmet opredeljen kot študijsko obveznost.

Prerequisites:

The course has to be assigned to the student.

Vsebina:

- Definicija načrtovanja kemijskih produktov.
- Koncept zelenega prehoda in trajnostni pristopi pri načrtovanju produktov.
- Piramida kemijskega produkta. Vplivi na funkcionalnost produkta (materiali in sestava, proces izdelave, struktura, pogoji uporabe, kazalci učinkovitosti in faktorji kakovosti). Funkcije lastnosti, procesa in uporabe.
- Osnovni koraki pri načrtovanju produktov in procesov, ki vključujejo definicijo problema in potencialne tehnične rešitve ob upoštevanju trajnostnih dejavnikov. Potrebe, ideje, izbor ideje, izdelava in ekonomika.
- Okoljski projektni management, funkcije managementa, deležniki organizacije, družbena inovacija, organizacija in vodenje, pregled organizacijskih struktur, kultura organizacije, navade

Content (Syllabus outline):

- Definition of chemical product design.
- Green transition concept and sustainable approaches in product design.
- The chemical product pyramid. Effects on chemical product performance (materials and composition, use, process, structure and performance indices and quality factors). Property, process, and usage functions.
- Basic steps in product and process design with problem definition and possible technical solutions considering sustainability issues.
- Needs, ideas, idea selection, manufacturing and economics.
- Environmental project management, functions of management, organizational stakeholders, social innovation, organization and leadership, overview of organizational structures, organizational culture,

<p>in rutine, trajnostno vodenje</p> <ul style="list-style-type: none"> - Poslovno načrtovanje, izdelava poslovnega modela, ekonomika poslovanja, upravljanje z denarnimi tokovi in poslovnimi tveganji - Primeri načrtovanja trajnostnih kemijskih produktov. Študentje v skupinah rešujejo realen problem in razvijajo trajnosten/zelen/družbeno odgovoren produkt. 	<p>habits and routines, sustainable leadership</p> <ul style="list-style-type: none"> - Business planning, business model development, economics, cash flow management and business risk management - Examples of sustainable chemical product design. Students solve a real-world problem in groups and design a sustainable/green/socially responsible product.
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Temeljna literatura in viri/Readings:

<ul style="list-style-type: none"> • E. L. Cussler in G. D. Moggridge, <i>Chemical Product Design</i>, Cambridge University Press, Cambridge, 2. izdaja, 2011. • J. A. Wesselingh, S. Kuil, M. E. Viglid, <i>Design and Development of Biological, Chemical, Food and Pharmaceutical Products</i>, John Wiley & Sons, Chichester, 2007. • R. Costa, G. D. Moggridge in P. M. Saraiva, <i>Chemical Product Engineering: An Emerging Paradigm Within Chemical Engineering</i>, AIChE, 52 (6) 2016, 1976 - 1986. • M. Marc, N. Ponikvar, M. Tekavčič, <i>Ekonomika projektov</i>, Ljubljana, 2020. • A. Stare, <i>Projektne management: teorija in praksa</i>, Ljubljana: Agencija POTI, 2011. • Maurya, Ash. <i>Delaj vitko: od načrta A do načrta, ki deluje</i>. Pasadena, 2014. • Antončič in ostali: <i>Podjetništvo – glavi dejavnik razvoja</i>, 2022. Dostopno na: http://maks2.ef.uni-lj.si/zaloznistvoslike/499/Podjetni%C5%A1tvo.pdf • Dodatna znanstvena in strokovna literatura za opravljanje seminarske naloge in izdelavo pisnega poročila o seminarski nalogi.

Cilji in kompetence:

<p>Študent osvoji metodološki pristop za razvoj novih produktov, ki vključuje identifikacijo in analizo potreb, s poudarkom na trajnostnih pristopih in zelenem prehodu. Osvoji različne stopnje razvoja kemijskih produktov in splošni pristop k definiranju potencialnih tehničnih rešitev ob upoštevanju ekonomskih, varstvenih in okoljskih dejavnikov. Študent je sposoben uporabiti kemijsko inženirsko znanja za opis in reševanje realnega problema. Osvoji osnove poslovnega načrta, projektne managementa in vodenja projektne skupine.</p>	<h3>Objectives and competences:</h3> <p>The student acquires a methodological approach to new product development that includes identification and analysis of needs, with emphasis on sustainable approaches and a green transition. The student understands the different stages of chemical product development and the general approach to define possible technical solutions by considering economic, safety and environmental factors. Ability to apply chemical engineering knowledge to describe and solve a real-world problem. Understanding the basic of business planning, project management and project team management.</p>
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Predvideni študijski rezultati:

<p>Znanje in razumevanje</p> <p>Študent zna identificirati ključne dejavnike pri razvoju trajnostnih kemijskih produktov in možnosti za zeleni prehod. Razume povezavo med osnovnimi kemijsko inženirskimi znanji in specifičnimi znanji za načrtovanje lastnosti določenega kemijskega produkta. Pozna osnovno metodologijo razvoja produkta in procesa za zadovoljitev potreb trga. Zna upoštevati spoznanja s področja (projektne) managementa, ekonomske in okoljske dejavnike pri načrtovanju procesa za kemijski produkt. Razume pomen zelenega prehoda. Razume pomen interdisciplinarnega sodelovanja pri reševanju kompleksnih izzivov, ki se pojavijo pri razvoju trajnostnih kemijskih produktov.</p> <p>Uporaba</p> <p>Študent je pridobljena znanja sposoben uporabiti pri</p>	<h3>Intended learning outcomes:</h3> <p>Knowledge and Comprehension</p> <p>Ability to identify key factors in the development of sustainable chemical products and opportunities for a green transition. Understanding of the relationship between basic chemical engineering skills and specific skills to design the properties of a given chemical product. Knowledge of basic product and process development methodology to meet market needs. Knowledge of how to consider insights from (project) management, economic and environmental factors when developing a process for a chemical product. Understanding of the importance of the green transition. Understanding of the importance of interdisciplinary collaboration in solving the complex challenges that arise in the development of sustainable chemical products.</p>
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razvojnem in raziskovalnem delu na področjih razvoja in optimizacije procesov in produktov.

Refleksija

Študent je sposoben sintetizirati pridobljena kemijsko inženirska znanja in jih uporabiti pri reševanju realnih problemov. Študent je sposoben identificirati potrebo po specifičnih tehničnih in naravoslovnih znanjih ter osnovnih ekonomsko/poslovnih znanjih za reševanje realnega problema. Pridobljeno znanje mu omogoča aktivno komuniciranje in sodelovanje na interdisciplinarnih področjih.

Prenosljive spretnosti

- razumevanje koncepta zelenega prehoda,
- sposobnost ustvarjalnega razmišljanja in delovanja,
- sposobnost iskanja, interpretacije, kritične analize in povezovanja relevantnih informacij,
- sposobnost analize, sinteze in razumevanja vpliva tehniških rešitev na okoljske in socialne odnose,
- sposobnost komunikacije (razprava s poudarkom na argumentaciji), tudi v angleščini, in uporabe modernih komunikacijskih orodij,
- sposobnost komunikacije s strokovnjaki ekonomsko/poslovnih ved
- sposobnost delovanja in organizacije dela v skupini,
- sposobnost razumevanja poklicne in etične odgovornosti,
- sposobnost samostojnega učenja,
- digitalna pismenost.

Application

The student can apply the acquired knowledge in development and research work in the areas of process and product development and optimization.

Analysis

The student is able to integrate and apply the acquired chemical engineering knowledge in solving problems in practise. The ability to identify the need for specific technical and natural science knowledge as well as fundamental economic/business skills to solve a real-world problem. The knowledge acquired will enable the student to actively communicate and cooperate in interdisciplinary fields.

Skill-transference Ability

- the understanding of the concept of green transition,
- the ability to think and act creatively,
- the ability to find, interpret, critically analyse, and link relevant information,
- the ability to analyse, synthesize and understand the impact of technical solutions on environmental and social conditions,
- the ability to communicate (discussion with emphasis on argumentation), and to use modern communication tools,
- the ability to communicate with experts in the field of economics and business,
- the ability to work in a group and to organize work,
- the ability to understand professional and ethical responsibilities,
- the ability to learn independently,
- digital literacy.

Metode poučevanja in učenja:

Predavanja, seminarji – projektno delo v manjših skupinah
Poleg predavanja v živo v predavalnici se največ do 15 ur pedagoško delo izvede z elektronskimi oblikami dela.

Learning and teaching methods:

Lectures, seminars – project work in small groups
In addition to the live lectures in the lecture hall, a maximum of 15 hours of pedagogical work will be conducted using electronic forms of work.

Načini ocenjevanja:

Pogoj za pristop k pisnem izpitu je izdelana seminarska naloga (izvedba, pisno poročilo in predstavitev).

Delež/Weight

40,00 %

Assessment:

The prerequisite for participation in the written examination is a prepared seminar paper (implementation, written report, and presentation).

Pisni izpit

60,00 %

Written exam

Reference nosilca/Lecturer's references:

- RUČIGAJ, Aleš, ŠTIRN, Žiga, ŠEBENIK, Urška, KRAJNC, Matjaž. Main-chain benzoxazine oligomers: effects of molecular weight on the thermal, mechanical, and viscoelastic properties. Journal of applied polymer science. 2018, 135(35),1-11.
- KAJTNA, Jernej, ŠEBENIK, Urška. Novel acrylic/nanocellulose microsphere with improved adhesive properties. International journal of adhesion and adhesives. 2017, 74, 100-106.
- LAPASIN, Romano, GRASSI, Mario, ABRAMI, Michela, ŠEBENIK, Urška. Structural evolution of salt-free aqueous Laponite dispersions : a study based on low-field NMR relaxometry and rheological investigations. Colloids and surfaces. A, Physicochemical and Engineering Aspects. 2020, 602, 1-11.

- PUSTOVRH, Aleš, JAKLIČ, Marko, BOLE, Domen, **ZUPAN, Blaž**. How to create a successful regional startup ecosystem : a policy-making analysis. Lex localis : revija za lokalno samoupravo. [Tiskana izd.]. Jul. 2019, vol. 17, iss. 3, str. 749-770.

- **ZUPAN, Blaž**, CANKAR, Franc, SETNIKAR-CANKAR, Stanka. The development of an entrepreneurial mindset in primary education. European journal of education. [Print ed.]. Sep.2018, vol. 53, iss. 3, str. 427-439.

- LIKAR, Borut, CANKAR, Franc, **ZUPAN, Blaž**. Educational model for promoting creativity and innovation in primary schools. Systems research and behavioral science : the official journal of the International Federation for Systems Research. Mar./Apr. 2015, vol. 32, iss. 2, str. 205-213.

ULB FERRETTI

OBNOVLJIVI VIRI IN SUROVINE

UČNI NAČRT PREDMETA/COURSE SYLLABUS

Predmet:	Obnovljivi viri in surovine
Course title:	Sustainable Resources and Raw Materials
Članica nosilka/UL	UL FKKT
Member:	

Študijski programi in stopnja	Študijska smer	Letnik	Semestri	Izbirnost
Kemijska tehnologija, prva stopnja, visokošolski strokovni	Ni členitve (študijski program)	2. letnik, 3. letnik	zimski ali poletni	izbirni

Univerzitetna koda predmeta/University course code: 0643419

Predavanja /Lectures	Seminar /Seminar	Vaje /Tutorials	Klinične vaje /Clinical tutorials	Druge oblike študija /Other forms of study	Samostojno delo /Individual student work	ECTS
60	15				75	5

Nosilec predmeta/Lecturer: doc. dr. Mitja Mori, doc. dr. Tina Skalar, izr. prof. dr. Igor Pušnik

Vrsta predmeta/Course type: izbirni strokovni/Elective Professional

Jeziki/Languages:

Predavanja/Lectures:	Slovenščina
Vaje/Tutorial:	

Pogoji za vključitev v delo oz. za opravljanje študijskih obveznosti:

Študent mora imeti predmet opredeljen kot študijsko obveznost.

Prerequisites:

The course has to be assigned to the student.

Vsebina:

- Povezovanje in poznavanje osnovnih pojmov s področja energetike, transporta, kemijske in procesne industrije
- Klasifikacija virov in surovin (obnovljivi, neobnovljivi)
- Vrste obnovljivih virov in njihova dostopnost izrabe (primarna, sekundarna)
- Zgodovinski pregled izkoriščanja virov, vpliv socialnih, ekonomskih in družbenih sprememb na vrsto, način in količino njihove izrabe
- Ocena trenutna izkoriščenost, zaloge in potrebe po virih na svetovni ravni in dejavniki, ki vplivajo na njihove spremembe
- Vpliv preskrbe z energijo na okolje (študije življenjskih ciklov tehnologij in procesov) in družbo
- Osnove procesov pridobivanja končne energije

Content (Syllabus outline):

- Connecting and knowledge of fundamental concepts from energy, transportation, chemistry, and process industries
- Classification of resources and raw materials (renewable, non-renewable)
- Types of renewable resources and their uses (primary, secondary)
- Historical review of resource use, the impact of societal, economic, and social changes on the type, manner, and extent of use
- Assessment of current uses, stocks, and demands of resources on a global scale and factors affecting their changes
- Impact of energy supply on the environment (life cycle studies of technologies and processes) and society

<p>(energija iz fosilnih goriv, vetra, sonca, vode, jedrska energija, geotermalna energija)</p> <ul style="list-style-type: none"> - Učinkovitejša raba energije v kemijski industriji - Skladiščenje in distribucija energije in surovin v kemijski industriji - Trajnostni viri v kemijski industriji in možnosti za zeleni prehod (surovin, električne energije, odpadne toplote) - Pospeševanje zelenega prehoda s termokemijskimi procesi, orientiranimi za prestrazanje sekundarnih surovin in sočasno proizvodnjo energije in energentov - Osnove energetskih sistemov, omrežij, distribucije električne energije, analiza njihove razpoložljivosti in energetske učinkovitosti - Vodik energijski vektor in uporaba vodikovih tehnologij v energetiki 	<ul style="list-style-type: none"> - Fundamentals of final energy production processes (energy from fossil fuels, wind, solar, hydro, nuclear, geothermal) - More efficient use of energy in the chemical industry - Storage and distribution of energy and raw materials in the chemical industry - Sustainable resources in the chemical industry and options for a green transition (raw materials, electricity, waste heat) - Accelerating the green transition through thermochemical processes focused on the recovery of secondary raw materials and the simultaneous production of energy and energy products - Fundamentals of energy systems, grids, electricity distribution, analysis of their availability and energy efficiency - Hydrogen energy vector and application of hydrogen technologies in the energy sector
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Temeljna literatura in viri/Readings:

- F. W. Wellmer, Raw materials for future energy supply, Springer, 2019, 225 strani.
- J. O'Connor, B. Noble, T. Lieuwen, Renewable Fuels, Sources, Conversion, and Utilization, Cambridge University Press, 2023, 566 strani.
- Znanstvena literatura pridobljena za pripravo seminarjev v okviru tega predmeta.

Cilji in kompetence:

- Študent pozna stanje obnovljivih virov in surovin na svetovnem trgu.
- Prepozna potrebe in zmožnosti na področju pridobivanja in zagotavljanja virov.
- Obvlada glavne vire energije in surovin in njihove procese pridobivanja in distribucije.
- Seznanjen je s problematiko vpliva na okolje in družbo ter možnosti optimizacije.
- Kandidat je sposoben identificirati in reševati osnovne probleme zagotavljanja virov z upoštevanjem njihove smotrne rabe v smeri zelenega prehoda.
- Študent je zmožen konstruktivne in strokovne komunikacije z vključevanjem različnih profilov strokovnjakov na tem področju (inženirji elektrotehnike, strojnimi inženirji, ...)

Objectives and competences:

- The student knows the situation of renewable raw materials and resources in the world market.
- Recognizes the needs and opportunities in the area of resource procurement and supply.
- Masters the main sources of energy and raw materials and their extraction and distribution processes. - Knows the impact on the environment and society and the optimization options. - The candidate is able to identify and solve the basic problems of resource provision, taking into account their rational use in terms of the green transition. - The student is able to communicate constructively and professionally, involving different profiles of experts in the field (electrical engineers, mechanical engineers, ...)

Predvideni študijski rezultati:

Znanje in razumevanje

Študent spozna osnovne procese pridobivanja energije, jih med seboj primerjati in predlagati optimizacije. Sposoben je opraviti izračun osnovnih masnih in energetskih bilanc procesa in zna preučiti njihov vpliv na okolje in družbo.

Uporaba

Znanja mu omogočajo konstruktivno interdisciplinarno komunikacijo in reševanje problemov s področja učinkovitega izkoriščanja virov

Intended learning outcomes:

Knowledge and Comprehension

The student learns about the basic processes of energy production, compares and optimizes them with each other. Student is able to calculate the basic mass and energy balances of the process and can examine their impact on the environment and society.

Application

His knowledge enables constructive interdisciplinary communication and problem solving in the field of efficient resource utilization in chemical technology.

<p>v kemijski tehnologiji.</p> <p>Refleksija Študent je zmožen povezati že pridobljena znanja s področja matematike, fizike, mehanskih operacij, procesov v industrijski kemiji in jih nadgraditi na področju zelene izrabe energije in učinkovite rabe drugih virov in surovin.</p> <p>Prenosljive spretnosti Študent je usposobljen implementirati že znane tehnologije za zeleni prehod v tehnologije uporabe obnovljivih virov in jih nadgraditi z do sedaj osvojenim znanjem. Pri študiju razvija kritičen način razmišljanja, kar mu koristi pri reševanju okoljskih problemov, ki zahtevajo upravljanje z različnimi viri.</p>	<p>Analysis Student will be able to link previously acquired knowledge in the fields of mathematics, physics, mechanics, industrial chemistry processes and expand them in the field of green energy use and efficient use of other resources and raw materials.</p> <p>Skill-transference Ability The student is empowered to implement already known technologies for green transition to renewable resource technologies and expand them with the knowledge acquired so far. During studies, student develops a critical thinking skills useful in solving environmental problems that require the management of various resources.</p>
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<p>Metode poučevanja in učenja: Predavanja in seminarji. Predavanja se lahko v obsegu do 15 ur izvedejo v drugih oblikah poučevanja (na daljavo, konzultacije, strokovni obisk industrije).</p>	<p>Learning and teaching methods: Lectures and seminars. Lectures of up to 15 hours may be delivered through other forms of instruction (distance learning, consultations, professional visits to industry).</p>
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Načini ocenjevanja:	Delež/Weight	Assessment:
Pogoj za pristop k pisnemu izpitu je uspešno izdelana seminarska naloga (napisana seminarska naloga in predstavitev).	20,00 %	Prerequisite for participation in the written examination is a successfully completed seminar paper (written seminar paper and presentation).
Pisni izpit	80,00 %	Written exam

Reference nosilca/Lecturer's references:

SKALAR, Tina, ZUPAN, Klementina, MARINŠEK, Marjan. Microstructure tailoring of combustion-derived Ni-GDC and Ni-SDC composites as anode materials for intermediate temperature solid oxide fuel cells. *Journal of the Australian Ceramic Society*. Mar. 2019, vol. 55, iss. 1, str. 123-133, ilustr. ISSN 2510-1579.

SKALAR, Tina, LUBEJ, Martin, MARINŠEK, Marjan. Optimization of operating conditions in a laboratory SOFC testing device. *Materiali in tehnologije*. [Tiskana izd.]. št. 5, 2015, str. 731-738. ISSN 1580-2949.

SKALAR, Tina, MAČEK, Jadran, GOLOBIČ, Amalija. A new simple synthesis and characterization of Sm-doped ceria and its homogeneous precursor Ce_{0.80}Sm_{0.20}(Ac)₃(Gly)_n. *Journal of the European ceramic society*. 2012, vol. 32, no. 10, str. 2333-2339. ISSN 0955-2219.

Univerzitetni, visokošolski ali višješolski učbenik z recenzijo

PUŠNIK, Igor. Preskušanje in energetska učinkovitost. Ljubljana: Fakulteta za elektrotehniko, Laboratorij za metrologijo in kakovost, 2016. 2 zv. (160 str.; 190 str.), ilustr. [COBISS.SI-ID 11821908]

ŠTUKELJ, Marina, HAJDINJAK, Melita, **PUŠNIK, Igor.** Stress-free measurement of body temperature of pigs by using thermal imaging : useful fact or wishful thinking. *Computers and electronics in agriculture*. [Online ed.]. 2022, vol. 193, art. no. 106656, str. 1-14, ilustr. ISSN 1872-7107. <https://www.sciencedirect.com/science/article/pii/S0168169921006736?via%3Dihub>, [COBISS.SI-ID 93839363]

PUŠNIK, Igor, GERŠAK, Gregor. Evaluation of the size-of-source effect in thermal imaging cameras. *Sensors*. Jan.-2 2021, iss. 2, 607, str. 1-22, ilustr. ISSN 1424-8220. <https://www.mdpi.com/1424-8220/21/2/607>, [COBISS.SI-ID 47911683]

MORI, Mitja, GUTIÉRREZ, Manuel, SEKAVČNIK, Mihael, DROBNIČ, Boštjan. Modelling and

environmental assessment of a stand-alone micro-grid system in a mountain hut using renewables. *Energies*. 2022, vol. 15, iss. 1, str. 1-21, ilustr. ISSN 1996-1073. <https://www.mdpi.com/1996-1073/15/1/202>, DOI: 10.3390/en15010202. [COBISS.SI-ID 91685379]

MORI, Mitja, GUTIÉRREZ, Manuel, CASERO, Pedro. Micro-grid design and life-cycle assessment of a mountain hut's stand-alone energy system with hydrogen used for seasonal storage. *International journal of hydrogen energy*. [Print ed.]. Aug. 2021, vol. 46, iss. 57, str. 29706-29723, ilustr. ISSN 0360-3199. <https://www.sciencedirect.com/science/article/pii/S0360319920344013>, DOI: 10.1016/j.ijhydene.2020.11.155. [COBISS.SI-ID 41896707]

ŠTROPNIK, Rok, SEKAVČNIK, Mihael, FERRIZ, Ana María, **MORI, Mitja**. Reducing environmental impacts of the ups system based on PEM fuel cell with circular economy. *Energy*. 2018, vol. 165, part b, str. 824-835, ilustr. ISSN 0360-5442. <https://www.sciencedirect.com/science/article/pii/S0360544218319790?via%3Dihub>, DOI: 10.1016/j.energy.2018.09.201. [COBISS.SI-ID 16276763]

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ORGANSKA ANALITIKA IN SPEKTROSKOPIJA

UČNI NAČRT PREDMETA/COURSE SYLLABUS

Predmet:	Organska analitika in spektroskopija
Course title:	Organic Analysis and Spectroscopy
Članica nosilka/UL Member:	UL FKKT

Študijski programi in stopnja	Študijska smer	Letnik	Semestri	Izbirnost
Kemijska tehnologija, prva stopnja, visokošolski strokovni	Ni členitve (študijski program)	2. letnik, 3. letnik		izbirni

Univerzitetna koda predmeta/University course code:	0099254
Koda učne enote na članici/UL Member course code:	KTSI33

Predavanja /Lectures	Seminar /Seminar	Vaje /Tutorials	Klinične vaje /Clinical tutorials	Druge oblike študija /Other forms of study	Samostojno delo /Individual student work	ECTS
45		30 LV			75	5

Nosilec predmeta/Lecturer: prof. dr. Janez Košmrlj

Vrsta predmeta/Course type: izbirni strokovni/Elective Professional

Jeziki/Languages:

Predavanja/Lectures:	Slovenščina
Vaje/Tutorial:	Slovenščina

Pogoji za vključitev v delo oz. za opravljanje študijskih obveznosti:

Študent oz. kandidat mora imeti predmet opredeljen kot študijsko obveznost.

Prerequisites:

The course has to be assigned to the student.

Vsebina:

Separacijske metode. Ločba zmesi na osnovi razlik v fizikalnih lastnostih (topnost v vodnih raztopinah, raztopinah kislin in baz, organskih topilih; vrelišče). Kromatografske metode (teoretske osnove, vrste kromatografskih metod) in uporaba v analizi organskih spojin (uporaba v preparativne namene, ločba zmesi, ločba enantiomerov).

Identifikacija organskih spojin. Kvalitativna in kvantitativna organska analiza. Testi na funkcionalne skupine in priprava derivatov.

Spektroskopske metode. Ultravijolična in vidna spektroskopija. Infrardeča spektroskopija (karakteristični absorpcijski trakovi za funkcionalne skupine, interpretacija infrardečih spektrov, aplikacija IR na organske strukturne probleme). Jedrska magnetna resonanca (kemijski premik, integral,

Content (Syllabus outline):

Separation techniques. Separation of organic compounds from the mixtures based on different physical properties (distillation, and solubility in water, organic solvents, acid-base extraction). Chromatographic techniques: Theoretical background, types of chromatographic methods (thin-layer-, column-, gas-chromatography). Qualitative and quantitative organic analyses, separation of mixtures, chiral separation. Identification of organic compounds based on chemical methods: Qualitative and quantitative analysis, analysis of functional groups and preparation of their derivatives. Molecular spectroscopy: NMR (chemical shift, integral, coupling patterns), IR (characteristic absorption bands of functional groups, interpretation of IR spectra). Mass spectrometry

<p>sklopitveni vzorci, aplikacija NMR na organske strukturne probleme) s poudarkom na ¹H NMR spektrih. Masna spektrometrija. Fragmentacije in premestitve nekaterih tipov organskih spojin. Določanje molske mase.</p> <p>Vaje Individualni pristop k analizi sestave in strukture kompleksnega vzorca (separacija, čiščenje in identifikacija na osnovi kemijskih metod in spektroskopskih tehnik).</p>	<p>(basic ionization techniques, fragmentations). Organic analysis laboratory course is based on students individual analysis of an organic sample; separation, purification, and identification (based on chemical and spectroscopic methods).</p>
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Temeljna literatura in viri/Readings:

- Shiner, R. L., C. K. F.; Morrill, T. C.; Curtin, D. Y.; Fuson, R. C. The Systematic Identification of Organic Compounds, 8th Edition, J. Wiley & Sons, 2003.
- Laboratory course - Dolenc, D. Vaje iz organske analize : praktikum. Ljubljana: Fakulteta za kemijo in kemijsko tehnologijo, 2007.

Dodatna literatura / Additional literature:

- Pavia, L. D; Lampman, G. M.; Kriz, G. S. Introduction to Spectroscopy. A Guide for Students of Organic Chemistry, 3th Edition, Hartcourt, Inc. 2001.
- Harwood, L. M.; Moody, C. J. Experimental Organic Chemistry. Principles and Practice. Blackwell Scientific Publications, 1994.

Cilji in kompetence:

Študent pridobi sposobnost analitičnega razmišljanja ter povezovanja podatkov dobljenih s posameznimi analiznimi metodami pri reševanju kvalitativnih in kvantitativnih analiznih problemov. Poudarek je na analizi funkcionalnih skupin s kemijskimi in spektroskopskimi metodami.

Objectives and competences:

Understanding the principles of qualitative and quantitative organic analysis using different basic analytical methods and techniques. Ability to design and perform standard experimental techniques for separation, isolation and purification of organic compounds. Analysis and characterization of organic compounds based on typical reactions at functional groups and by using basic spectroscopic methods.

Predvideni študijski rezultati:

Znanje in razumevanje
Pozna osnovne kriterije za ločevanje zmesi organskih spojin. Zna povezati spektroskopske lastnosti spojin s strukturo spojin.
Uporaba
Zna uporabiti teoretično znanje za kvalitativno in kvantitativno analizo vzorcev organskih spojin.
Refleksija
Kritično vrednotenje rezultatov pri vajah na osnovi teoretičnega znanja.
Prenosljive spretnosti
Študent pridobi laboratorijske spretnosti in zna eksperimentalne podatke ustrezno obdelati in primerno interpretirati. Uporaba že pridobljenega znanja iz organske kemije in analizne kemije.

Intended learning outcomes:

Knowledge and Comprehension
Student learns basics of separation, identification and characterization as well as spectroscopic methods and their application in characterization of organic compounds.
Application
Student learns to use theoretical knowledge for qualitative and quantitative analysis of organic compounds in research work.
Analysis
Student learns to critically evaluate results acquired at practical course and connect it to the theoretical knowledge.
Skill-transference Ability
Student gains skills for laboratory work and learns how to assess and interpret the experimental results. Student combines the knowledge from courses in organic and analytical chemistry.
Learning and teaching methods: Lectures and laboratory practical courses.

Metode poučevanja in učenja:

Learning and teaching methods:

Predavanja in laboratorijske vaje.	Lectures and laboratory practical courses.
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Načini ocenjevanja:	Delež/Weight	Assessment:
a) Pri vajah se ocenjuje motivacija, samostojnost in sposobnost študenta pri reševanju praktičnih problemov in pripravi laboratorijskih poročil. b) Pisni izpit. 10 (odlično), 9 in 8 (prav dobro), 7 (dobro), 6 (zadostno), 5-1 (nezadostno)		Written examination (60–100%: student passes the exam, 0–59%: student fails the exam).

Reference nosilca/Lecturer's references:

- A. Demšar, **J. Košmrlj**, S. Petriček: Variable-temperature nuclear magnetic resonance spectroscopy allows direct observation of carboxylate shift in zinc carboxylate complexes. *J. Am. Chem. Soc.* 2002, 124, 3951–3958.
- **J. Košmrlj**, S. Kafka, I. Leban, M. Grad: Formation and Structure Elucidation of Two Novel Spiro[2H-indol]-3(1H)-ones, *Magn. Reson. Chem.* 2007, 45, 700–704.
- D. Urankar, A. Pevec, I. Turel, **J. Košmrlj**: Pyridyl Conjugated 1,2,3-Triazole is a Versatile Coordination Ability Ligand Enabling Supramolecular Associations. *Cryst. Growth Des.* 2010, 10, 4920–4927.
- D. Urankar, B. Pinter, A. Pevec, F. De Proft, I. Turel, **J. Košmrlj**: Click-Triazole N2 Coordination to Transition Metal Ions is Assisted by a Pendant Pyridine Substituent. *Inorg. Chem.* 2010, 49, 4820–4829.
- B. Pinter, A. Demšar, D. Urankar, F. De Proft, **J. Košmrlj**: Conformational Fluxionality in a Palladium(II) Complex of Flexible Click Chelator 4-phenyl-1-(2-picolyl)-1,2,3-triazole. A dynamic NMR and DFT study. *Polyhedron* 2011, 30, 2368–2373.
- B. Pinter, D. Urankar, A. Pevec, F. De Proft, **J. Košmrlj**: Platinum mediated dinitrogen liberation from 2-picolylazide through a putative Pt=N double bond containing intermediate. *Inorg. Chem.* 2013, 4528–4533.

ORGANSKA KEMIJA I

UČNI NAČRT PREDMETA/COURSE SYLLABUS

Predmet:	Organska kemija I
Course title:	Organic Chemistry I
Članica nosilka/UL Member:	UL FKKT

Študijski programi in stopnja	Študijska smer	Letnik	Semestri	Izbirnost
Kemijska tehnologija, prva stopnja, visokošolski strokovni	Ni členitve (študijski program)	1. letnik	poletni	obvezni

Univerzitetna koda predmeta/University course code:	0072020
Koda učne enote na članici/UL Member course code:	KT108

Predavanja /Lectures	Seminar /Seminar	Vaje /Tutorials	Klinične vaje /Clinical tutorials	Druge oblike študija /Other forms of study	Samostojno delo /Individual student work	ECTS
60	15				75	5

Nosilec predmeta/Lecturer: prof. dr. Franc Požgan

Vrsta predmeta/Course type: obvezni/mandatory

Jeziki/Languages:

Predavanja/Lectures:	Slovenščina
Vaje/Tutorial:	Slovenščina

Pogoji za vključitev v delo oz. za opravljanje študijskih obveznosti: **Prerequisites:**

Študent oz. kandidat mora imeti predmet opredeljen kot študijsko obveznost.	The course has to be assigned to the student.
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Vsebina:

Nomenklatura (poimenovanje) organskih spojin. Struktura in vezi v organskih spojinah, kislost in bazičnost. Narava kemijske vezi, vezna entalpija, hibridizacija, ionske in kovalentne vezi; struktura etana, etena in etina; polarnost vezi in elektronegativnost; toksičnost kemikalij in varno delo z njimi.

Narava organskih spojin.

Alkani in cikloalkani: nomenklatura, izomerija, konformacije; *cis-trans* izomerija pri cikloalkanah, ekvatorialni in aksialni substituenti pri cikloheksanu; fizikalne lastnosti alkanov, nafta.

Alkeni in alkini: nomenklatura, elektronska struktura, *cis-trans* izomerija, pravila sekvence in *Z,E*-konfiguracija; tipi organskih reakcij, mehanizmi in reakcijski intermedijati ter prehodna stanja; reakcijska

Content (Syllabus outline):

Nomenclature of organic compounds. Structure and bonding of organic compounds; acids and bases. The nature of chemical bonds, bond (dissociation) enthalpy, hybridization, ionic and covalent bonds; structure of ethane, ethene and ethyne; bond polarity and electronegativity; toxicity of chemicals and risks. The nature of organic compounds. Alkanes and cycloalkanes: nomenclature, isomerism, conformations; *cis-trans* isomerism of cycloalkanes, equatorial and axial substituents in cyclohexane; physical properties of alkanes; naphtha. Alkenes and alkynes: nomenclature, electronic structure, *cis-trans* isomerism, sequence rules and *Z,E*-configuration; kinds of organic reactions; mechanisms, reaction intermediates and transition states; rate of the reaction, kinetics and thermodynamics; synthesis and

<p>hitrost, kinetika in termodinamika; sinteza alkenov in alkinov ter njihove pretvorbe; adicijske reakcije, mehanizmi, polimerizacija, naravni kavčuk.</p> <p>Osnove stereokemije: tetraedrski ogljikov atom in stereokemija; kiralnost, optična aktivnost, absolutna konfiguracija; enantiomeri, diastereoizomeri, mezo spojine; projekcijske formule, stereokemija reakcij; kiralnost v naravi, kiralna zdravila.</p> <p>Alkil halogenidi: nomenklatura, sinteza in pretvorbe alkil halogenidov; SN1 in SN2 reakciji, E1 in E2 reakciji, E1cb; PCB in drugi naravni polutanti.</p> <p>Alkoholi, etri in fenoli: nomenklatura, lastnosti, sinteza in pretvorbe; vodikova vez, kislost alkoholov in fenolov, ciklični etri (epoksidi); etanol kot kemikalija, zdravilo in strup.</p>	<p>transformations of alkenes and alkynes; addition reactions, mechanisms, polymerization; natural rubber. Stereochemistry: tetrahedral carbon atom; chirality, optical activity, absolute configuration; enantiomers, diastereo(iso)mers, meso compounds; presentation of the structure, stereochemistry of reactions; chirality in Nature, chiral drugs. Alkyl halides: nomenclature, synthesis and transformations; SN1 and SN2 reactions, E1 and E2 reactions, E1cb; PCB and other pollutants in Nature. Alcohols, ethers and phenols: nomenclature, properties, synthesis and transformations; hydrogen bonding, acidity; cyclic ethers (epoxides); ethanol as chemical, drug and poison.</p>
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Temeljna literatura in viri/Readings:

- J. McMurry: Fundamentals of Organic Chemistry, Brooks/Cole Publishing Company, 4th Edition, Pacific Grove 1998, 566 pages (50%).

Supplementary reading:

(a) P. W. Atkins, M. J. Frazer, M. J. Clugston, R. A. Y. Jones: Kemija, zakonitosti in uporaba (translated by A. Kornhauser, S. A. Glažar), Tehniška založba Slovenije d. d., Ljubljana, 1997.

(b) M. Tišler: Organska kemija, 3. popravljena in dopolnjena izdaja, Državna založba Slovenije, Ljubljana, 2005.

Cilji in kompetence:

Cilj predmeta je, da študent pridobi temeljno znanje iz organske kemije tj. poznavanje nomenklature organskih spojin, poznavanje posameznih vrst organskih spojin (po funkcionalnih skupinah) in njihovih strukturnih značilnosti ter reaktivnosti, poznavanje osnov organske stereokemije, reakcijskih mehanizmov in intermediatov, poznavanje osnovnih principov organske sinteze.

Objectives and competences:

Objectives of the course are mastering basic knowledge of organic chemistry: nomenclature, structural features, functional groups, reactivity, and typical transformations of organic compounds. Basic knowledge of organic stereochemistry, reaction mechanisms and intermediates. Knowledge of the basic principles of organic chemistry and accessing literature sources.

Competences:

Basic knowledge of organic chemistry: nomenclature, structure, reactivity, toxicity of fundamental organic compounds and basic transformations of organic compounds. Ability to understand structure-reactivity relationship, i.e. to predict chemical properties of a given organic compound from its structure and vice versa.

Predvideni študijski rezultati:

Znanje in razumevanje

Študent se nauči osnov organske kemije: poimenovanja organskih spojin, posameznih vrst organskih spojin (po funkcionalnih skupinah) in njihovih strukturnih značilnosti ter reaktivnosti, osnov organske stereokemije, reakcijskih mehanizmov in intermediatov ter osnovnih principov organske sinteze.

Uporaba

Študent pridobi temeljno znanje, ki je osnova za nadaljnji študij kemije in se navezuje na večino ostalih

Intended learning outcomes:

Knowledge and Comprehension

Student understands and is familiar with:

- Structure of organic compounds and functional groups
- Isomerisation and nomenclature of organic compounds
- Basic transformations of organic compounds
- Typical reagents used for performing basic organic reactions

Application

Mastered knowledge of organic chemistry is basic knowledge needed for studying chemical engineering.

<p>predmetov študija kemije. To znanje je nujno potrebno pri kasnejšem delu v praksi. Refleksija Zavedanje, da je za popolno razjasnitev kemijske reakcije potreben natančen študij vsake reakcijske stopnje. Prenosljive spretnosti Pri predmetu se študenti z reševanjem različnih problemov izurijo v uporabi znanja in analitičnega razmišljanja.</p>	<p>The knowledge is interconnected with majority of other subjects concerning the program. Course is also fundamental for understanding biochemistry subjects and courses concerning organic materials and ecology. Analysis Student is capable of recognising different basic types of organic compounds, predict their physical properties, toxicity, reactivity, etc. Skill-transference Ability The student acquires skills that are required for a basic synthetic laboratory work and for handling with chemicals. The knowledge on organic chemistry enables better understanding of the basic principles of other subjects and courses within the study.</p>
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Metode poučevanja in učenja:

Predavanja; seminarji

Learning and teaching methods:

Lectures, seminar work, training by analytical solving of the theoretical problems.

Načini ocenjevanja:

Pisni izpit (nadomestita ga lahko dva pozitivno ocenjena kolokvija).

Delež/Weight

Assessment:

Written exam. Written exam can be passed by two positive partial exams.

Reference nosilca/Lecturer's references:

1. ŠTEFANE, Bogdan, PERDIH, Franc, VIŠNJEVAC, Aleksander, POŽGAN, Franc. Novel triazole-based ligands and their zinc(II) and nickel(II) complexes with a nitrogen donor environment as potential structural models for mononuclear active sites. *New journal of chemistry*, ISSN 1144-0546, no. 1, 2015, str. 566-575, ilustr. <http://pubs.rsc.org/en/content/articlepdf/2014/nj/c4nj01642d?page=search>, doi: [10.1039/c4nj01642d](https://doi.org/10.1039/c4nj01642d). [COBISS.SI-ID [1536036291](https://www.cobiss.si/id/1536036291)]
2. WAGGER, Jernej, POŽES, Aljaž, POŽGAN, Franc. Synthesis of European pharmacopoeial impurities A, B, C, and D of cabergoline. *RSC advances*, ISSN 2046-2069, 2013, vol. 3, no. 45, str. 23146-23156, ilustr. <http://pubs.rsc.org/en/content/articlepdf/2013/ra/c3ra43417f>, doi: [10.1039/c3ra43417f](https://doi.org/10.1039/c3ra43417f). [COBISS.SI-ID [1646639](https://www.cobiss.si/id/1646639)]
3. POŽGAN, Franc, DIXNEUF, Pierre H. Ruthenium(II) acetate catalyst for direct functionalisation of sp²-C-H bonds with aryl chlorides and access to tris- heterocyclic molecules. *Advanced Synthesis & Catalysis*, ISSN 1615-4150. [Print ed.], 2009, vol. 351, no. 11/12, str. 1737-1743, doi: [10.1002/adsc.200900350](https://doi.org/10.1002/adsc.200900350). [COBISS.SI-ID [30712837](https://www.cobiss.si/id/30712837)]

ORGANSKA KEMIJA II

UČNI NAČRT PREDMETA/COURSE SYLLABUS

Predmet:	Organska kemija II
Course title:	Organic Chemistry II
Članica nosilka/UL Member:	UL FKKT

Študijski programi in stopnja	Študijska smer	Letnik	Semestri	Izbirnost
Kemijska tehnologija, prva stopnja, visokošolski strokovni	Ni členitve (študijski program)	2. letnik	zimski	obvezni

Univerzitetna koda predmeta/University course code:	0072035
Koda učne enote na članici/UL Member course code:	KT113

Predavanja /Lectures	Seminar /Seminar	Vaje /Tutorials	Klinične vaje /Clinical tutorials	Druge oblike študija /Other forms of study	Samostojno delo /Individual student work	ECTS
45	30				75	5

Nosilec predmeta/Lecturer: prof. dr. Franc Požgan

Vrsta predmeta/Course type: obvezni/mandatory

Jeziki/Languages:

Predavanja/Lectures:	Slovenščina
Vaje/Tutorial:	Slovenščina

Pogoji za vključitev v delo oz. za opravljanje študijskih obveznosti: **Prerequisites:**

Študent oz. kandidat mora imeti predmet opredeljen kot študijsko obveznost.	The course has to be assigned to the student.
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Vsebina:

Aromatske spojine: nomenklatura, struktura benzena in njegova stabilnost, aromatičnost; elektrofilne aromatske substitucije; policiklične aromatski ogljikovodiki.

Aldehidi in ketoni: tipi karbonilnih spojin, nomenklatura, sinteza, pretvorbe in lastnosti aldehydov in ketonov; nukleofilne adicije na aldehyde in ketone; kislost alfa vodikovih atomov, keto-enolna tautomerija, kondenzacijske reakcije.

Karboksilne kisline in derivati: nomenklatura, struktura, lastnosti in pretvorbe karboksilnih kislin in njihovih derivatov; najlon in poliestri.

Amini: nomenklatura, struktura in lastnosti aminov; bazičnost aminov, sinteza in pretvorbe.

Biomolekule.

Ogljikovi hidrati: razdelitev, struktura, Fisherjeva

Content (Syllabus outline):

Aromatic compounds: nomenclature, structure of benzene and its stability; electrophilic aromatic substitution reactions; polycyclic aromatic compounds.

Aldehydes and ketones: kinds of carbonyls, nomenclature, synthesis, transformations and properties of aldehydes and ketones; nucleophilic additions; acidity of alpha hydrogen atoms, keto-enol tautomerism, condensation reactions.

Carboxylic acids and derivatives: nomenclature, structure, properties and conversions of carboxylic acids and their derivatives; nylon and polyesters.

Amines: nomenclature, structure and properties of amines; synthesis and conversions. Biomolecules.

Carbohydrates: classification, structure, Fischer projection, D- and L-monosaccharides; conversions

<p>projekcija, D- in L-monosaharidi; pretvorbe monosaharidov; disaharidi in polisaharidi.</p> <p>Aminokislina, peptidi in beljakovine: nomenklatura, struktura in lastnosti aminokislin; peptidi in beljakovine, razdelitev beljakovin in njihova struktura.</p> <p>Lipidi: razdelitev, nomenklatura; trigliceridi, masti in olja, mila in umetni detergenti; steroidi, holesterol.</p> <p>Nukleinske kisline: uvod in sestavine.</p> <p>Nukleinske kisline: uvod in sestavine.</p>	<p>of monosaccharides; disaccharides and polysaccharides. Amino acids, peptides and proteins: nomenclature, structure and properties of amino acids; peptides and proteins, classification of proteins and their structure. Lipids: classification, nomenclature; triglycerides, fats and oils, soaps and synthetic detergents; steroids, cholesterol. Nucleic acids: introduction and components.</p>
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Temeljna literatura in viri/Readings:

- J. McMurry: Fundamentals of Organic Chemistry, Brooks/Cole Publishing Company, 4th Edition, Pacific Grove 1998, 566 pages (50%).

Supplementary reading:

(a) P. W. Atkins, M. J. Frazer, M. J. Clugston, R. A. Y. Jones: Kemija, zakonitosti in uporaba (translated by A. Kornhauser, S. A. Glažar), Tehniška založba Slovenije d. d., Ljubljana, 1997. (b) M. Tišler: Organska kemija, 3. popravljena in dopolnjena izdaja, Državna založba Slovenije, Ljubljana, 2005.

Cilji in kompetence:

Cilj predmeta je, da študent pridobi oz. poglobi temeljno znanje iz organske kemije tj. poznavanje nomenklature organskih spojin, poznavanje posameznih vrst organskih spojin (po funkcionalnih skupinah) in njihovih strukturnih značilnosti ter reaktivnosti, poznavanje osnov organske stereokemije, reakcijskih mehanizmov in intermediatov, poznavanje osnovnih principov organske sinteze, poznavanje literaturnih virov in njihove uporabe.

Objectives and competences:

Understanding of the basic principles of organic chemistry, i.e. knowledge of individual classes of organic compounds (by functional groups), their nomenclature, structural characteristics and reactivity. Knowledge of basis of organic stereochemistry, reaction mechanisms and intermediates, knowledge of basic principles of organic synthesis, knowledge of sources of chemical literature and their use.

Competences:

Ability to interpret structural characteristics of individual classes of organic compounds and to predict their reactivity on the basis of their structures. Ability to explain the fundamental organic reactions.

Predvideni študijski rezultati:

Znanje in razumevanje
Študent se nauči osnov organske kemije iz področja: karboksilnih spojin, aminov in biomolekul (ogljikovih hidratov, aminokislin, peptidov in beljakovin, lipidov in nukleinskih kislin) ter spozna literaturne vire in njihovo uporabo. V obliki seminarjev pa utrjuje tudi področja iz Organske kemije I.

Uporaba

Študent pridobi oz. utrdi temeljno znanje, ki je osnova za nadaljnji študij kemije in se navezuje na večino ostalih predmetov študija kemije. To znanje je nujno potrebno pri kasnejšem delu v praksi.

Refleksija

Zavedanje, da je za popolno razjasnitev kemijske reakcije potreben natančen študij vsake reakcijske stopnje.

Prenosljive spretnosti

Pri predmetu se študenti z reševanjem različnih problemov izurijo v uporabi znanja, analitičnega razmišljanja in uporabe literaturnih virov.

Intended learning outcomes:

Knowledge and Comprehension

Student understands and is familiar with organic chemistry subjects:

Carboxylic compounds

Amines and biomolecules (carbohydrates, amino acids, peptides, lipids, and nucleic acids)

Is familiar with literature sources and its use.

By solving the theoretical problems they review the knowledge of Organic chemistry I

Application

Mastered knowledge of organic chemistry is basic knowledge needed for studying Chemical technology.

The knowledge is interconnected with majority of other subjects concerning the program. Course is also fundamental for understanding analytical chemistry subjects and courses concerning organic materials and ecology.

Analysis

Student is capable of recognising different types of organic compounds, predict their physical properties, toxicity, reactivity, etc.

	Skill-transference Ability The student acquires skills that are required for a basic synthetic laboratory work and for handling with chemicals. The knowledge on organic chemistry enables better understanding of the basic principles of other subjects and courses within the study.
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Metode poučevanja in učenja: Predavanja; seminarji, individualni in skupinski projekti.	Learning and teaching methods: Lectures, seminar work, training by analytical solving of the theoretical problems.
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Načini ocenjevanja: Pisni izpit (nadomestita ga lahko dva pozitivno ocenjena kolokvija).	Delež/Weight	Assessment: Written exam. Written exam can be passed by two positive partial exams
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Reference nosilca/Lecturer's references:

1. ŠTEFANE, Bogdan, PERDIH, Franc, VIŠNJEVAC, Aleksander, POŽGAN, Franc. Novel triazole-based ligands and their zinc(II) and nickel(II) complexes with a nitrogen donor environment as potential structural models for mononuclear active sites. *New journal of chemistry*, ISSN 1144-0546, no. 1, 2015, str. 566-575, ilustr. <http://pubs.rsc.org/en/content/articlepdf/2014/nj/c4nj01642d?page=search>, doi: [10.1039/c4nj01642d](https://doi.org/10.1039/c4nj01642d). [COBISS.SI-ID [1536036291](https://www.cobiss.si/id/1536036291)]
2. WAGGER, Jernej, POŽES, Aljaž, POŽGAN, Franc. Synthesis of European pharmacopoeial impurities A, B,C, and D of cabergoline. *RSC advances*, ISSN 2046-2069, 2013, vol. 3, no. 45, str. 23146-23156, ilustr. <http://pubs.rsc.org/en/content/articlepdf/2013/ra/c3ra43417f>, doi: [10.1039/c3ra43417f](https://doi.org/10.1039/c3ra43417f). [COBISS.SI-ID [1646639](https://www.cobiss.si/id/1646639)]
3. POŽGAN, Franc, DIXNEUF, Pierre H. Ruthenium(II) acetate catalyst for direct functionalisation of sp²-C-H bonds with aryl chlorides and access to tris- heterocyclic molecules. *Advanced Synthesis & Catalysis*, ISSN 1615-4150. [Print ed.], 2009, vol. 351, no. 11/12, str. 1737-1743, doi: [10.1002/adsc.200900350](https://doi.org/10.1002/adsc.200900350). [COBISS.SI-ID [30712837](https://www.cobiss.si/id/30712837)]

OSNOVE BIOKEMIJE Z BIOTEHNOLOGIJO

UČNI NAČRT PREDMETA/COURSE SYLLABUS

Predmet:	Osnove biokemije z biotehnologijo
Course title:	Fundamentals of biochemistry with biotechnology
Članica nosilka/UL Member:	UL FKKT

Študijski programi in stopnja	Študijska smer	Letnik	Semestri	Izbirnost
Kemijska tehnologija, prva stopnja, visokošolski strokovni	Ni členitve (študijski program)	1. letnik	poletni	obvezni

Univerzitetna koda predmeta/University course code:	0072021
Koda učne enote na članici/UL Member course code:	KT109

Predavanja /Lectures	Seminar /Seminar	Vaje /Tutorials	Klinične vaje /Clinical tutorials	Druge oblike študija /Other forms of study	Samostojno delo /Individual student work	ECTS
50	10	15 LV			75	5

Nosilec predmeta/Lecturer: prof. dr. Polona Žnidaršič Plazl

Vrsta predmeta/Course type: obvezni/mandatory

Jeziki/Languages:

Predavanja/Lectures:	Slovenščina
Vaje/Tutorial:	Slovenščina

Pogoji za vključitev v delo oz. za opravljanje študijskih obveznosti:

Študent oz. kandidat mora imeti predmet opredeljen kot študijsko obveznost.

Prerequisites:

The course has to be assigned to the student.

Vsebina:

Temeljna vsebinska področja predmeta so:

1. Molekule in življenje

- Lastnosti bioloških sistemov. Prokariotske in evkariontske celice. Raznolikost živega sveta. Mikrobiološke zbirke.

- Biološke makromolekule. Aminokisliline, peptidi, proteini. Encimi. Ogljikovi hidrati. Lipidi. DNA in RNA.

- Osnove celičnih procesov. Ohranjanje in prenos biološke informacije. Osnove celičnega metabolizma in bioenergetike.

2. Biotehnologija

- Vloga biotehnologije v družbi. Uporaba v industriji (kemijska, farmacevtska, živilska), v medicini, kmetijstvu in okoljevarstvu. Biološka varnost in etična vprašanja sodobne biotehnologije.

Content (Syllabus outline):

The primary subject areas of the course are:

1. Molecules and living systems

- Characteristics of biological systems. Prokaryotic and eukaryotic cells. The diversity of the living world. Microbial collections.

- Biological macromolecules. Amino acids, peptides, proteins. Enzymes. Carbohydrates. Lipids. DNA and RNA.

- Principles of cellular processes. Conservation and transfer of biological information. Basics of cell metabolism and bioenergetics.

2. Biotechnology

- The role of biotechnology in society. Use in industry (chemical, pharmaceutical, food), in medicine, agriculture and environmental engineering.

<ul style="list-style-type: none"> - Interdisciplinarnost načrtovanja, vodenja in analize bioprosesov. Stopnje razvoja bioprosesa. Vloga bioprosesnega inženirja. Kvantitativni pristop. - Industrijska biotehnologija. Biotransformacije. Integrirani procesi. Izbrani primeri industrijskih procesov. - Smernice razvoja sodobne biotehnologije. Zarodne celice. Nanobiotehnologija. Mikroreaktorji in »lab on a chip« sistemi. 	<p>Biological safety and ethical issues of modern biotechnology.</p> <ul style="list-style-type: none"> - Interdisciplinary planning, management and analysis of bioprocesses. The development of bioprocesses. The role of bioprocess engineer. Quantitative approach. - Industrial biotechnology. Biotransformations. Integrated processes. Selected examples of industrial processes. - Guidelines in the development of modern biotechnology. Stem cells. Nanobiotechnology. Microreactors and "lab on a chip" systems.
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Temeljna literatura in viri/Readings:

<ul style="list-style-type: none"> - Boyer, R. Temelji biokemije. Študentska založba, Ljubljana. 2005. 634 p. (30%) - Raspor, P. (ur.) Biotehnologija. Bia, d.o.o., Ljubljana. 1996. 815 p. (20%) - Žnidaršič Plazl, P., Podgornik, H. Vaje iz biotehnologije, 2. Izd. Fakulteta za kemijo in kemijsko tehnologijo, Ljubljana. 2011. 96 p.

Cilji in kompetence:

<p>Cilj predmeta je seznaniti študente z nekaterimi osnovami o molekularni zgradbi in delovanju celic ter o njihovi uporabi v biotehnologiji, razvijati zavest o pomenu biotehnologije v sodobni družbi in razvijati sposobnosti za sodelovanje pri razvijanju in analizi bioprosesov.</p> <p>Študent si pri predmetu pridobi naslednje specifične kompetence:</p> <ul style="list-style-type: none"> - poznavanje nekaterih temeljnih pojmov iz biokemije in biologije celice - poznavanje uporabe sodobne biotehnologije v industrijskih procesih, kmetijstvu in okoljevarstvu - razumevanje specifičnosti bioprosesov z inženirskega vidika - poznavanje nekaterih etičnih vprašanj sodobne biotehnologije - osvajanje nekaterih izbranih laboratorijskih tehnik: aseptično delo z mikroorganizmi, določevanje parametrov encimske kinetike, vodenje in analiza bioprosesa v laboratorijskem bioreaktorju 	<h3>Objectives and competences:</h3> <p>Students obtain the following specific competencies:</p> <ul style="list-style-type: none"> - Knowledge of some fundamental concepts in biochemistry and cell biology - Knowledge of the use of modern biotechnology in industrial processes, agriculture and environmental engineering - Understanding of the specificity of bioprocesses from the engineering point of view - Knowledge of some of the ethical issues of modern biotechnology - Knowledge of some selected laboratory techniques: sterilization, aseptic work with microorganisms, the determination of enzyme kinetics parameters, management and analysis of bioprocesses in a laboratory bioreactor
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Predvideni študijski rezultati:

<p>Znanje in razumevanje Študent pridobi osnovna znanja o sestavi celic, bioloških makromolekulah in specifičnosti vodenja in analize bioprosesov. Razume zakonitosti delovanja encimov, nekatere osnovne metabolne poti in principe ohranjanja in prenosa genetske informacije, ki omogočajo razumevanje njihove uporabe v biotehnologiji.</p> <p>Uporaba Pridobljena znanja je študent sposoben uporabljati pri razvijanju in analizi bioprosesov.</p> <p>Refleksija</p>	<h3>Intended learning outcomes:</h3> <p>Knowledge and Comprehension Understanding of the basics of molecular structure and functioning of cells and their application in biotechnology, to develop awareness of the importance of biotechnology in modern society.</p> <p>Application Student will develop the ability to participate in the development and analysis of bioprocesses.</p> <p>Analysis Student will interpret and analyse the knowledge on selected bioprocesses.</p> <p>Skill-transference Ability</p>
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<p>Študent bo interpretiral ter pred kolegi analiziral lastno razumevanje izbranih bioprocsov. Pri tem bo uporabil pridobljena teoretična znanja ter jih vrednotil s praktičnimi izkušnjami.</p> <p>Prenosljive spretnosti</p> <p>Računalniška obdelava eksperimentalnih podatkov, uporabljanje spletnih virov, pisanje poročil, priprava računalniške predstavitve seminarja, timsko delo.</p>	<p>Analysis of experimental data, the use of internet as a data source, writing of reports, a seminar preparation and oral presentation, team work.</p>
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Metode poučevanja in učenja:

Predavanja, laboratorijske vaje, individualno in skupinsko delo pri pripravi seminarjev. Spletna gradiva za določena poglavja.

Learning and teaching methods:

Lectures, seminars, practical training.

Načini ocenjevanja:

Delež/Weight

Assessment:

Pisni izpit. Ocene: 6-10 pozitivno.	70,00 %	Written exam
Laboratorijske vaje	15,00 %	Laboratory exercises
Seminarska naloga	15,00 %	Seminar work

Reference nosilca/Lecturer's references:

- TIŠMA, Marina, ŽNIDARŠIČ PLAZL, Polona, VASIĆ-RAČKI, Đurđa, ZELIĆ, Bruno. Optimization of laccase production by *Trametes versicolor* cultivated on industrial waste. *Appl. Biochem. Biotechnol.*, 2012, 166, 36-46.
- ŽNIDARŠIČ PLAZL, Polona, PLAZL, Igor. Development of a continuous steroid biotransformation process and product extraction within microchannel system. *Catalysis Today*, 2010, 157, 315-320.
- ŽNIDARŠIČ PLAZL, Polona, RUTAR, Vera, RAVNJAK, David. The effect of enzymatic treatments of pulps on fiber and paper properties. *Chemical and biochemical engineering quarterly*, 2009, 23, 497-506.

OSNOVE INDUSTRIJSKE KEMIJE

UČNI NAČRT PREDMETA/COURSE SYLLABUS

Predmet:	Osnove industrijske kemije
Course title:	Fundamentals of industrial chemistry
Članica nosilka/UL Member:	UL FKKT

Študijski programi in stopnja	Študijska smer	Letnik	Semestri	Izbirnost
Kemijska tehnologija, prva stopnja, visokošolski strokovni	Ni členitve (študijski program)	1. letnik	zimski	obvezni

Univerzitetna koda predmeta/University course code:	0072022
Koda učne enote na članici/UL Member course code:	KT104

Predavanja /Lectures	Seminar /Seminar	Vaje /Tutorials	Klinične vaje /Clinical tutorials	Druge oblike študija /Other forms of study	Samostojno delo /Individual student work	ECTS
45	30				75	5

Nosilec predmeta/Lecturer: izt. prof. dr. Boštjan Genorio

Vrsta predmeta/Course type: obvezni/mandatory

Jeziki/Languages:

Predavanja/Lectures:	Slovenščina
Vaje/Tutorial:	Slovenščina

Pogoji za vključitev v delo oz. za opravljanje študijskih obveznosti:

Študent oz. kandidat mora imeti predmet opredeljen kot študijsko obveznost.

Prerequisites:

The course has to be assigned to the student.

Vsebina:

Osnove industrijske kemije:
Kemija in industrijska kemija, razdelitev in organiziranost kemijske industrije, slovenska kemijska industrija,
Malotonažna in malolitražna industrijska proizvodnja, velikotonažna idnustrijska proizvodnja, predelovalna industrija, visokotehnološka proizvodnja
Surovinska osnova procesov, primarne, sekundarne surovine, racionalna uporaba surovinskih virov, reprovegrige, predpriprava surovin pred predelavo, bogatenje surovin
Energija v kemijski industriji, smotrna uporaba različnih surovinskih virov in odpadne toplote
Kemijski tehnološki procesi, tehnološki režim, tehnološke sheme, šaržni in kontinuirni procesi

Content (Syllabus outline):

Fundamentals of Industrial Chemistry:
Chemistry and industrial chemistry, distribution and structure of the chemical industry, Slovenian chemical industry,
Small scale industrial production, large scale industrial production, high-tech manufacturing.
The raw material base processes, primary, secondary raw materials, rational use of raw materials resources, reproductions chains, pretreatment of raw materials before processing, beneficiation of raw materials
Energy in the chemical industry, rational use of different sources of raw materials and waste heat
Chemical technological processes, technological regime, technological schemes, batch and continuous processes

Infrastrukturni pogoji industrijske proizvodnje, oprema, proces Tehnološki procesi, masne bilance procesov, karakteristike in kvaliteta produktov, izkoristek procesov.	Infrastructure conditions in industrial production, equipment, process Technological processes, mass balance processes, characteristics and quality of products, efficiency of processes.
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Temeljna literatura in viri/Readings:

<ul style="list-style-type: none"> - Ignatowitz, Eckhard, Kemijska tehnika, Jutro, 1996, ISBN 961-6006-30-4 (60%) - Heaton, C.A., Introduction to Industrial Chemistry, Springer 1996, ISBN 978-0-7514-0272-8 (10%); - Brockel, U.; Meier, W.; Wagner, G., Product Design and Engineering, Vol 1 & Vol 2, Wiley-vch Verlag GmbH & Co. KgaA 2007, ISBN 978-3-527-31529-1 (30%)
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Cilji in kompetence:

<p>Slušatelji se v okviru predavanj in seminarja seznanijo z različnimi nivoji in osnovnimi značilnostmi kemijske industrijske proizvodnje, z osnovami kemijskih industrijskih procesov ter njihovimi infrastrukturnimi pogoji, možnostmi za uspešno in varno vodenje procesov.</p>	<p>Objectives and competences: Students in the context of lectures and seminars familiar with the different levels and the basic characteristics of industrial chemical production, with the fundamentals of chemical industrial processes and their infrastructure requirements, the possibilities for successful and safe management processes.</p>
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Predvideni študijski rezultati:

<p>Znanje in razumevanje Študent se nauči razlikovati med pogoji in zahtevami laboratorijskega dela in industrijskim nivojem. Spozna osnovne značilnosti, robne pogoje in zahteve za uspešno in sodobno industrijsko proizvodnjo.</p> <p>Uporaba Analizira probleme, ki se pojavijo pri prenosu v večje merilo pri prenosu procesov v industrijsko prakso.</p> <p>Refleksija Analiza racionalne izvedbe industrijskih kemijskih reakcijskih sistemov in procesov omogoča študentu boljši vpogled in razumevanje osnovnih konceptov kemije in industrijske kemije.</p> <p>Prenosljive spretnosti Teoretične principe dodatno spoznava pri njihovi implementaciji v industrijsko merilo in prakso ter kritično vrednoti skladnost med teoretičnimi načeli in možnostjo za njihovo praktičnim uveljavljanje.</p>	<p>Intended learning outcomes: Knowledge and Comprehension Students learn how to distinguish between the terms and conditions of the laboratory work and industrial level. Learn about the basic features, boundary conditions and requirements for a successful and modern industrial production.</p> <p>Application The student analyzes the problems that occur in the transmission of higher criterion in the transfer processes in industrial practice.</p> <p>Analysis Analysis of the rational performance of industrial chemical reaction systems and processes allows students a better understanding of the basic concepts of chemistry and industrial chemistry.</p> <p>Skill-transference Ability A student gets additionally acquainted with theoretical principles at implementation into industrial scale and practice and to critically evaluate the consistency between theoretical principles and practical option for their enforcement.</p>
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Metode poučevanja in učenja:

Predavanja in seminar.	Learning and teaching methods: Lectures and seminars.
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Načini ocenjevanja:

Načini ocenjevanja:	Delež/Weight	Assessment:
<ul style="list-style-type: none"> • dva neobvezna kolokvija • Pisni izpit (lahko ga nadomestita dva pozitivno ocenjena kolokvija) in • Ustni izpit (pogoj za pristop k ustnem izpitu je pozitivno opravljen Pisni izpit) 		

Reference nosilca/Lecturer's references:

- Genorio B, Strmcnik D, Subbaraman R, Tripkovic D, Karapetrov G, Stamenkovic V R, Pejovnik S and Marković N M 2010 Selective catalysts for the hydrogen oxidation and oxygen reduction reactions by patterning of platinum with calix [4] arene molecules Nat. Mater. 9 998–1003
- Genorio B, Lu W, Dimiev A M, Zhu Y, Raji A-R O, Novosel B, Alemany L B and Tour J M 2012 In Situ Intercalation Replacement and Selective Functionalization of Graphene Nanoribbon Stacks ACS Nano 6 4231–40
- Xiang C, Behabtu N, Liu Y, Chae H G, Young C C, Genorio B, Tsentalovich D E, Zhang C, Kosynkin D V, Lomeda J R, Hwang C-C, Kumar S, Pasquali M and Tour J M 2013 Graphene Nanoribbons as an Advanced Precursor for Making Carbon Fiber ACS Nano 7 1628–37
- Xiang C, Cox P J, Kukovecz A, Genorio B, Hashim D P, Yan Z, Peng Z, Hwang C-C, Ruan G, Samuel E L G, Sudeep P M, Konya Z, Vajtai R, Ajayan P M and Tour J M 2013 Functionalized Low Defect Graphene Nanoribbons and Polyurethane Composite Film for Improved Gas Barrier and Mechanical Performances. ACS Nano 7 10380–6
- Raji A-R O, Varadhachary T, Nan K, Wang T, Lin J, Ji Y, Genorio B, Zhu Y, Kittrell C and Tour J M 2016 Composites of Graphene Nanoribbon Stacks and Epoxy for Joule Heating and Deicing of Surfaces ACS Appl. Mater. Interfaces 8 3551–6

OSNOVE KEMIJSKEGA INŽENIRSTVA

UČNI NAČRT PREDMETA/COURSE SYLLABUS

Predmet:	Osnove kemijskega inženirstva
Course title:	Introduction to chemical engineering
Članica nosilka/UL Member:	UL FKKT

Študijski programi in stopnja	Študijska smer	Letnik	Semestri	Izbirnost
Kemijska tehnologija, prva stopnja, visokošolski strokovni	Ni členitve (študijski program)	2. letnik	zimski	obvezni

Univerzitetna koda predmeta/University course code:	0072036
Koda učne enote na članici/UL Member course code:	KT117

Predavanja /Lectures	Seminar /Seminar	Vaje /Tutorials	Klinične vaje /Clinical tutorials	Druge oblike študija /Other forms of study	Samostojno delo /Individual student work	ECTS
45	30				75	5

Nosilec predmeta/Lecturer: prof. dr. Igor Plazl

Vrsta predmeta/Course type: obvezni/mandatory

Jeziki/Languages:

Predavanja/Lectures:	Slovenščina
Vaje/Tutorial:	Slovenščina

Pogoji za vključitev v delo oz. za opravljanje študijskih obveznosti:

Študent oz. kandidat mora imeti predmet opredeljen kot študijsko obveznost.

Prerequisites:

The course has to be assigned to the student.

Vsebina:

Osnovni koncepti termodinamike (zaprt-odprt sistem, vrste energij, lastnosti sistema, stanje sistema in ravnotežje, proces in cikel). Lastnosti čistih komponent (fazne spremembe čistih komponent, diagrami procesov faznih sprememb, enačbe stanja. Prvi zakon termodinamike (a) zaprt sistem (toplota, delo; notranja energija, entalpija in specifična toplota idealnih plinov, tekočin in trdnih substanc) (b) kontrolni volumen (pretočni stacionarni in nestacionarni sistem, zakon o ohranitvi mase, zakon o ohranitvi energije). Drugi zakon termodinamike. Entropija (definicija entropije, sprememba entropije čistih substanc: idealnih plinov, tekočin in trdnih substanc; izentropski procesi). Mešanica plinov in hlapov ter lastnosti zraka (osnove psihrometrije: definicije vlažnosti zraka, definicije

Content (Syllabus outline):

Basic concepts of thermodynamics (closed-opened system, types of energies, system characteristics, state of the system and equilibrium and process cycle). Properties of pure components (phase transition of pure components, diagrams of phase transitions, equations of state. First law of thermodynamics (a) closed system (heat, work, internal energy of a thermodynamic system, enthalpy, specific heat of ideal gasses, liquids and solids) (b) control volume (stationary and unsteady flow systems, law of mass conservation, law of energy conservation). Second law of thermodynamics. Entropy (definition of entropy, entropy changes of pure components: gasses, liquids and solids, isentropic processes). Mixtures of gases and vapors, properties of air (fundamentals of psychrometry, definitions of: air

temperature adiabatsnega nasičenja, temperatura rosišča in temperature mokrega termometra, Mollierov diagram.)

Inženirske osnove tehnoloških operacij: opredelitev pretočnega sistema, makroskopska snovna in energijska bilanca pretočnega sistema, bilanca energije za diferencialni element in posebne oblike le-te, Bernoullijeva enačba). Tok tekočin (definicija viskoznosti, newtonijske in neneutronijske tekočine, tokovne oblike, mejni sloji, turbulenten tok tekočin v cevi, frikcijski faktor, računanje izgub v sestavljenih cevni sistemih).

Prenos toplote: mehanizmi prenosa toplote: (prevajanje, konvekcija, medfazni prenos, radiacija). Stacionarno prevajanje toplote (za različne geometrije in kombiniran prenos toplote), Nestacionarno prevajanje. Konvekcija toplote (toplotna prestopnost, podobnost med prenosom toplote in gibalne količine, kriterijske enačbe za oceno toplotnih prestopnosti, toplotna prehodnost). Načrtovanje toplotnih menjalnikov. Prenos toplote s sevanjem.

Prenos snovi: mehanizmi prenosa snovi (difuzija, konvekcija, (definicije, difuzija v dvokomponentnih sistemih). Snovna bilanca za diferencialni element v sistemu in posebne oblike le-te. Stacionarna difuzija (difuzija skozi mirujočo komponento, pseudostacionarna difuzija, ekvimolarna protidifuzija). Nestacionarna difuzija. Konvekcijski prenos snovi (snovna prestopnost, podobnost med prenosom snovi in gibalne količine, kriterijske enačbe za oceno snovnih prestopnosti, snovna prehodnost). Načrtovanje naprav za snovni prenos.

Izbrane tehnološke operacije (destilacija, sušenje, mešanje, uparjanje) s poudarkom na varnostnem vidiku. Izbrane mehanske operacije (mletje).

Osnove inženirske kinetike (hitrostna enačba, enostavne reakcije, kompleksne reakcije).

Eksperimentalno določevanje hitrostne enačbe (integralna metoda, diferencialna metoda).

Dimenzioniranje reaktorjev (obratovalna enačba šaržnega reaktorja in mešalnega ter cevne pretočnega reaktorja na osnovi snovnih in energijskih bilanc). Primerjava mešalnega in cevne reaktorja. Primeri zelenih tehnologij v kemijskem inženirstvu: zelena proizvodnja vodikovega peroksida, depolimerizacija odpadne plastike na osnovi najlona.

humidity, temperature of adiabatic saturation, dry-bulb temperature, wet-bulb temperature, dew point temperature, Mollier diagram.)

Engineering fundamentals of technological operations: definition of flow system, macroscopic mass and energy balances of flow system, energy balance of differential element and its special cases, Bernoulli equation). Fluid flow (viscosity definition, Newtonian and non-Newtonian liquids, types of flows, boundary layer, turbulent pipe flow, friction factor, calculation of head losses due to friction in assembled pipe systems).

Heat transfer: mechanisms of heat transfer (conduction, convection, interfacial exchange, radiation). Stationary heat conduction (for different geometries, combined heat transfer). Unsteady heat conduction. Convective heat transfer, transport coefficient, similarity between heat and momentum transfers, criterion equations for prediction of heat transport coefficient, overall heat transfer coefficient). Design of heat exchangers. Heat transfer by radiation.

Mass transfer: mechanisms of mass transfer (diffusivity, convection, diffusivity of two-components systems). Mass balance for differential element of the system and its special cases. Stationary diffusivity, (diffusivity through stagnant component, pseudo-stationary diffusivity, equi-molar opposite diffusivity). Unsteady diffusion. Convective mass transfer (transport coefficient, similarity between mass and momentum transfers, criterion equations for prediction of mass transport coefficient, overall mass transfer coefficient. Design of mass transfer devices.

Selected technological operations (distillation, drying, mixing, evaporation) with a focus on security aspects. Selected mechanical operation (crushing, milling). Basic of engineering kinetics (rate equation, simple and complex chemical reaction). Experimentally determination of rate equation (integral method, differential method). Reactor design (operational equations based on energy balances for batch, mixing and plug flow reactors. Comparison of mixing and plug flow reactors.

Examples of green technologies in chemical engineering: green production of hydrogen peroxide, depolymerization of nylon-based plastic waste.

Temeljna literatura in viri/Readings:

Z. Rant, Termodinamika: knjiga za uk in prakso, Ljubljana, Fakulteta za strojništvo, 2001, 607 str., (30%)
Koloini, Prenos toplote in snovi, FKKT, Ljubljana, 1999, 273 str., (30%)

Cilji in kompetence:

Cilj predmeta je omogočiti globlje spoznavanje in razumevanje baznih znanj termodinamike, transportnih pojavov in kemijske tehnike, skupaj s poznavanjem osnovnih inženirskih konceptov.

Objectives and competences:

Objective of the course is deeper knowledge and understanding fundamentals of thermodynamics, transport phenomena and chemical technology, on basis of chemical engineering concepts.

<p>Predvideni študijski rezultati:</p> <p>Znanje in razumevanje: Varnostni inženir je pri svojem delu soočen s potrebo po osnovnem razumevanju tehnoloških postopkov in raznovrstnih specifičnih procesov, ki ga usposablja za kvalitetno in inovativno delo na svojem področju.</p> <p>Uporaba Tako izobražen profil je sposoben ustrezne strokovne komunikacije z ostalimi tehnično podkovanimi kadri, kar privede do uspešnih rešitev konkretnih problemov. Pomemben vidik predmeta je študenta naučiti kritičnega pogleda na problem in na osnovi sintetiziranih znanj podati hitre in učinkovite rešitve v praksi.</p> <p>Refleksija Kritična primerjava teoretskega znanja s praktično uporabo na področju pojava in širjenja požarov.</p> <p>Prenosljive spretnosti Pri predmetu se študent nauči sintetizirati vsebine znanj, pridobljene z različnih področij tehničnih in naravoslovnih segmentov, ter tako pridobi vzorec za inovativno delo na drugih področjih.</p>	<p>Intended learning outcomes:</p> <p>Knowledge and Comprehensions: Safety Engineer is faced at his/her work with the need to understand the basic techniques and understanding of specific processes, which qualify him/her for excellence and innovative work in his/her field</p> <p>Application Such educated profile student is skilled for professional communication with other technically knowledgeable staffs which leads to successful solutions to concrete problems. Student learns to have critical view on the problem and on the basis of synthesized knowledge to provide quick and effective solutions in practice.</p> <p>Analysis Critical comparison of theoretical knowledge with practical application in candidate's working field.</p> <p>Skill-transference Ability By matter of this course student learns to connect knowledge of different technical and natural science segments and acquires mode for innovative work on different fields.</p>
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<p>Metode poučevanja in učenja:</p> <p>Predavanja, seminarji in vaje.</p>	<p>Learning and teaching methods:</p> <p>Lectures, seminars and laboratory work.</p>
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<p>Načini ocenjevanja:</p> <p>- Pisni in ustni izpit - pisna seminarska naloga. Ocene: 5-10, pri čemer velja, da je pozitivna ocena 6-10</p>	<p>Delež/Weight</p>	<p>Assessment:</p> <p>- written and oral exam; - Written seminar Grading scale: 5-10, positive grade is 6-10</p>
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<p>Reference nosilca/Lecturer's references:</p> <ol style="list-style-type: none"> 1. G.N. Jovanovic, J. E. Atwater, P. Žnidaršič Plazl, I. Plazl. Dechlorination of Polychlorinated Phenols on Bimetallic Pd/Fe Catalyst in a Magnetically Stabilized Fluidized Bed. Chem. Eng.J., 2015, 274:50-60 2. SKUBIC, Blaž, LAKNER, Mitja, PLAZL, Igor. Microwave drying of expanded perlite insulation board. Ind. eng. chem. res.. [Print ed.], 2012, vol. 51, no. 8, str. 3314-3321. 3. BEESTON, Michael Philip, POHAR, Andrej, ELTEREN, Johannes Teun van, PLAZL, Igor, ŠLEJKOVEC, Zdenka, VEBER, Marjan, GLASS, Hylke J. Assessment of physical leaching processes of some elements in soil upon ingestion by continuous leaching and modeling. Environ. sci. technol.. [Print ed.], 2010, vol. 44, issue 16, str. 6242-6248.

OSNOVE POLIMERNEGA INŽENIRSTVA

UČNI NAČRT PREDMETA/COURSE SYLLABUS

Predmet:	Osnove polimernega inženirstva
Course title:	Principles of Polymer Engineering
Članica nosilka/UL Member:	UL FKKT

Študijski programi in stopnja	Študijska smer	Letnik	Semestri	Izbirnost
Kemijska tehnologija, prva stopnja, visokošolski strokovni	Ni členitve (študijski program)	3. letnik	poletni	obvezni

Univerzitetna koda predmeta/University course code:	0100215
Koda učne enote na članici/UL Member course code:	KTSI4

Predavanja /Lectures	Seminar /Seminar	Vaje /Tutorials	Klinične vaje /Clinical tutorials	Druge oblike študija /Other forms of study	Samostojno delo /Individual student work	ECTS
30	30	15 LV			75	5

Nosilec predmeta/Lecturer: prof. dr. Urška Šebenik

Vrsta predmeta/Course type: izbirni strokovni/Elective Professional

Jeziki/Languages:	Predavanja/Lectures:	Slovenščina
	Vaje/Tutorial:	Slovenščina

Pogoji za vključitev v delo oz. za opravljanje študijskih obveznosti: **Prerequisites:**

Študent oz. kandidat mora imeti predmet opredeljen kot študijsko obveznost.	The course has to be assigned to the student.
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Vsebina:

- Uvod v polimere;
- Polimerizacijski procesi: stopenjska polimerizacija in verižna polimerizacija;
- Kinetika in rezultat reverzibilne in ireverzibilne stopenjske polimerizacije;
- Vpliv funkcionalnosti monomerov in razmerja funkcionalnih skupin na rezultat stopenjske polimerizacije;
- Kinetika verižne polimerizacije;
- Vpliv reakcij prenosa na rezultat verižne polimerizacije;
- Verižna kopolimerizacija in njena kinetika;
- Kvantitativna analiza polimerizacijskih procesov glede na medij in njihove karakteristike: polimerizacija v raztopini, polimerizacija v masi, emulzijska polimerizacija, suspenzijska polimerizacija.

Content (Syllabus outline):

- Introduction to polymers;
- Polymerization processes: step-reaction and chain-reaction polymerization;
- Kinetics and result of reversible and non-reversible step polymerization;
- Effect of monomer functionality and functional group ratio on the result of step polymerization;
- Kinetics of chain polymerization;
- Effect of chain transfer reactions on the result of chain polymerization;
- Chain copolymerization and its kinetics;
- Quantitative analysis and characteristics of: Solution polymerization; Bulk polymerization; Emulsion polymerization; Suspension polymerization.
- Principles of batch emulsion polymerization, Harkins mechanism, kinetics;

<ul style="list-style-type: none"> - Osnove emulzijske polimerizacije v šaržnem sistemu, Harkinsonov mehanizem in kinetika; - Računski primeri; - Laboratorijske vaje (kvalitativna in kvantitativna obravnava procesov): Šaržna polimerizacija v raztopini; Kontinuirna polimerizacija v masi; Šaržna polimerizacija v suspenziji; Šaržna polimerizacija v emulziji. 	<ul style="list-style-type: none"> - Problems; - Laboratory practice (qualitative and quantitative process description): Batch polymerization in solution, Continuous polymerization in bulk; Batch suspension polymerization; Batch emulsion polymerization.
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Temeljna literatura in viri/Readings:

<ul style="list-style-type: none"> - A. Kumar in R. K. Gupta, Fundamentals of Polymers, The McGraw-Hill Companies, Inc., New York, 1998, 543 str., (50 %). - R. O. Ebewe, Polymer Science and Technology, CRC Press, Boca Raton, 2000, 463 str., (25 %). - U. Šebenik, Osnove polimernega inženirstva: Zbirka nalog, UL FKKT, Ljubljana, 2012, 41 str., (100 %). <p>Dopolnilna literatura:</p> <ul style="list-style-type: none"> - Rudin, The Elements of Polymer Science and Engineering, 2nd Ed., Academic Press, London, 1999, 483 str. - P. Rempp, E. W. Merrill, Polymer synthesis, 2nd Ed., Huthig & Wepf Verlag, Basel, 1991, 336 str.

Cilji in kompetence:

<p>Cilj predmeta je, da študentje osvojijo osnovna znanja iz področja polimernega inženirstva.</p> <p>Študentje pri predmetu pridobijo naslednje specifične kompetence:</p> <ul style="list-style-type: none"> - poznavanje načinov napovedovanja distribucije molekulskih mas; - poznavanje toplotnih prehodov, specifičnih za polimerne molekule; - poznavanje fizikalnih stanj polimerov in vpliva procesnih parametrov na fizikalna stanja; - poznavanje in kvantitativno ovrednotenje polimerizacijskih procesov; - razumevanje vpliva načina polimerizacije na lastnosti polimernega produkta. 	<h3>Objectives and competences:</h3> <p>Acquisition of basic knowledge from polymer engineering; Acquisition of knowledge about molecular weight and molecular weight distribution and methods for molecular weight distribution prediction; knowledge about thermal transitions in polymers and the ability to distinguish between different polymer physical states; Acquisition of knowledge about polymerization processes and their quantitative description; Understanding the effect of the type of polymerization and of polymerization process parameters on product properties.</p>
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Predvideni študijski rezultati:

<p>Znanje in razumevanje Študent zna kvantitativno obravnavati osnovne polimerizacijske procese in napovedovati ključne lastnosti produkta glede na vrsto in način polimerizacijskega procesa. Razume zvezo med procesnimi parametri in sintetiziranim polimerizacijskim produktom. Zna uporabiti znanja kemijske kinetike in termodinamike na področju sinteze polimerov. Uporaba Pridobljena znanja je sposoben uporabiti pri študiju kemijsko inženirskih predmetov, kot tudi pri samostojnem razvojnem in raziskovalnem delu. Sposoben je kvantitativne analize enostavnejših industrijskih polimerizacijskih procesov. Refleksija Študent je sposoben samostojno sklepati, postavljati zaključke ter uporabiti svoje znanje pri sorodnih predmetih. Znanja s področja polimernega</p>	<h3>Intended learning outcomes:</h3> <p>Knowledge and Comprehension Understanding basic principles of polymer engineering science; Ability of quantitative description of basic polymerization processes and resulting polymers; Understanding relationship between process parameters and polymer properties; Ability to employ chemical kinetics and thermodynamics to describe polymerizations. Application At other courses from chemical engineering and at individual research work; Quantitative analysis of relatively simple polymerization processes on industrial level. Analysis Ability to apply the acquired knowledge at familiar courses by critical thinking and deduction; Fundamental knowledge enables understanding similar reactive processes Skill-transference Ability</p>
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inženirstva mu omogočajo razumevanje sorodnih reakcijskih sistemov. Prenosljive spretnosti Razvita sposobnost kritičnega razmišljanja in sklepanja. Sposobnost povezovanja osnovnih znanj ter študija domače in tuje literature.	Development of the ability of critical thinking and deduction; Ability of knowledge integration and studying relevant literature from the field of polymer engineering.
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Metode poučevanja in učenja: Predavanja, seminarji, laboratorijske vaje	Learning and teaching methods: Lectures, seminars, laboratory practice
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Načini ocenjevanja:	Delež/Weight	Assessment:
Opravljenе vaje so pogoj za pristop k izpitu		Laboratory practice is a prerequisite to exam attendance
Poročila in zagovor laboratorijskih vaj.	30,00 %	Written reports and oral laboratory practice defence.
Pisni izpit.	70,00 %	Written exam.

Reference nosilca/Lecturer's references:

1. RUČIGAJ, Aleš, ALIČ, Branko, KRAJNC, Matjaž, ŠEBENIK, Urška. Curing of bisphenol A-aniline based benzoxazine using phenolic, amino and mercapto accelerators. *Express polymer letters*, ISSN 1788-618X, 2015, vol. 9, no. 7, str. 647-657, ilustr. <http://www.expresspolymlett.com/>, doi: 10.3144/expresspolymlett.2015.60. [COBISS.SI-ID 1536286915]
2. RUČIGAJ, Aleš, KRAJNC, Matjaž, ŠEBENIK, Urška. Polymerization of octamethylcyclotetrasiloxane between montmorillonite nanoplatelets initiated by surface anions. *Polymer bulletin*, ISSN 0170-0839, str. 1-16, ilustr. <http://link.springer.com/article/10.1007/s00289-015-1377-5>, doi: 10.1007/s00289-015-1377-5. [COBISS.SI-ID 1536273859]
3. RUČIGAJ, Aleš, ALIČ, Branko, KRAJNC, Matjaž, ŠEBENIK, Urška. Investigation of cure kinetics in a system with reactant evaporation : epoxidized soybean oil and maleic anhydride case study. *European Polymer Journal*, ISSN 0014-3057. [Print ed.], 2014, vol. 52, no. 1, str. 105-116, ilustr. <http://dx.doi.org/10.1016/j.eurpolymj.2014.01.009>, doi: 10.1016/j.eurpolymj.2014.01.009. [COBISS.SI-ID 1667887]
4. KAJTNA, Jernej, ŠEBENIK, Urška, KRAJNC, Matjaž. Synthesis and dynamic mechanical analysis of nanocomposite UV crosslinkable 100% solid acrylic pressure sensitive adhesives. *International journal of adhesion and adhesives*, ISSN 0143-7496. [Print ed.], 2014, vol. 49, no. 1, str. 18-25, ilustr. http://ac.els-cdn.com/S0143749613002212/1-s2.0-S0143749613002212-main.pdf?_tid=0be1f984-6c7a-11e3-b240-00000aab0f6c&acdnat=1387875899_0bee10364d8f1fce86b5ce0adea1f8d5, doi: 10.1016/j.ijadhadh.2013.12.010. [COBISS.SI-ID 1663791]
5. KAJTNA, Jernej, ALIČ, Branko, KRAJNC, Matjaž, ŠEBENIK, Urška. Influence of hydrogen bond on rheological properties of solventless UV crosslinkable pressure sensitive acrylic adhesive prepolymers. *International journal of adhesion and adhesives*, ISSN 0143-7496. [Print ed.], 2014, vol. 49, no. 1, str. 103-108, ilustr. http://ac.els-cdn.com/S0143749613002273/1-s2.0-S0143749613002273-main.pdf?_tid=1f9b954e-7921-11e3-9740-00000aacb35e&acdnat=1389267072_2b2c514569d4f86a66d77b0b12a891ad, doi: 10.1016/j.ijadhadh.2013.12.016. [COBISS.SI-ID 1664047]

OSNOVE TRAJNOSTNE KEMIJE

UČNI NAČRT PREDMETA/COURSE SYLLABUS

Predmet:	Osnove trajnostne kemije
Course title:	Fundamentals of Sustainable Chemistry
Članica nosilka/UL	UL FKKT
Member:	

Študijski programi in stopnja	Študijska smer	Letnik	Semestri	Izbirnost
Kemijska tehnologija, prva stopnja, visokošolski strokovni (od študijskega leta 2023/2024 dalje)	Ni členitve (študijski program)	1. letnik	zimski ali poletni	izbirni

Univerzitetna koda predmeta/University course code: 0643422

Predavanja /Lectures	Seminar /Seminar	Vaje /Tutorials	Klinične vaje /Clinical tutorials	Druge oblike študija /Other forms of study	Samostojno delo /Individual student work	ECTS
30	15	20 LV			60	5

Nosilec predmeta/Lecturer: doc. dr. Lev Matoh, prof. dr. Andreja Žgajnar Gotvajn

Vrsta predmeta/Course type: izbirni splošni/Elective General

Jeziki/Languages:

Predavanja/Lectures:	Slovenščina
Vaje/Tutorial:	Slovenščina

Pogoji za vključitev v delo oz. za opravljanje študijskih obveznosti: **Prerequisites:**

Študent oz. kandidat mora imeti predmet opredeljen kot študijsko obveznost.	The course has to be assigned to the student.
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Vsebina:

Osnovnih 12 principov trajnostnega razvoja in zelene kemije. Primeri uporabe principov trajnostnega razvoja v vsakdanjem življenju. Definicija in skupine onesnaževal. Viri onesnaženja. Vplivi na okolje in zdravje. Globalni in lokalni okoljski problemi. Principi trajnostne kemije pri pridelavi hrane (pesticidi, gnojila, GSO). Principi trajnostne kemije v energetiki (neobnovljivi in obnovljivi viri energije, vodik in gorivne celice, biogoriva). Principi trajnostne kemije materialov (plastika, bioplastika, kompoziti, nanomateriali). Sodobni materiali v okoljskih tehnologijah. Kemijski principi in tehnologije odstranjevanja onesnaževal iz vode, zraka in tal. Osnove recikliranja

Content (Syllabus outline):

Basic 12 principles of sustainable development and green chemistry. Examples of the application of the principles of sustainable development in everyday life. Definition of pollutants. Groups of pollutants. Sources of pollution. Effects on the environment and health. Global and local environmental problems. Principles of sustainable chemistry in food production (pesticides, fertilizers, GMOs). Principles of sustainable chemistry in energy production and usage (non-renewable and renewable energy sources, hydrogen and fuel cells, biofuels). Principles of sustainable material chemistry (plastics, bioplastics, composites, nanomaterials). Modern materials in environmental technologies. Chemical principles and technologies of removing pollutants from water, air, and soil. Basics of

<p>in ponovne uporabe. V okviru seminarja študent izbere aktualno temo in jo predstavi pred skupino. Obvezna vsebina seminarja: prikaz kemijskih osnov problema s stališča 12 principov trajnostnega razvoja, predlogi za reševanje problema. Diskusija.</p>	<p>recycling and reuse. As part of the coursework, the student chooses an up-to-date topic and presents it to the group. Mandatory content of the coursework: presentation of the chemical background of the problem based on the 12 principles of sustainable development, proposals for solution of the problem. Discussion.</p>
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Temeljna literatura in viri/Readings:

1. M. Lancaster: Green Chemistry: An Introductory Text, 3. izdaja, The Royal Society of Chemistry, 2016; izbrana poglavja.
 2. D. Drev, J. Čuvan: Okoljevarstvene tehnologije, Fit media, 2013.
- Dodatna literatura:
3. M. Knez, G. Lojen, N. Torelli: Materiali in okolje, Fit media, 2016.

Cilji in kompetence:

Cilj predmeta je študentom omogočiti osnovno razumevanje aktualnih tem, povezanih s trajnostno kemijo, razvijati zavedanje o pomenu vključevanja principov trajnostnega razvoja v vsa aplikativna področja kemije in sorodnih ved, razvijati zmožnosti za razumevanje kemijskih osnov aplikacij, razvijati sposobnosti za presojo vpliva različnih tehnologij in materialov na živo in neživo naravo.

Objectives and competences:

The objective of the course is to provide students with a basic understanding of current topics related to sustainable chemistry, to develop awareness of the importance of incorporating the principles of sustainable development into all applied fields of chemistry and related sciences, to develop the ability to understand chemical basics of applications, to develop the ability to assess the impact of various technologies and materials on nature.

Predvideni študijski rezultati:

Znanje in razumevanje

Predmet omogoči osnovno razumevanje nekaterih materialov in kemijskih procesov, povezanih z zaščito okolja. Študent spozna osnovne principe trajnostne kemije ter osnove škodljivih vplivov na človeka in okolje.

Uporaba

Študent zna uporabiti pridobljeno znanje pri vrednotenju vplivov na človeka in okolje. Na primerih uporabe v praksi se nauči vrednotiti omenjene vplive ter iskati ustrezne rešitve.

Refleksija

Študent bo na seminarjih analiziral izbrano temo, pri čemer bo uporabil principe trajnostne kemije za iskanje rešitev konkretnih problemov.

Prenosljive spretnosti

Sposobnost uporabe domačih in tujih virov literature, interpretacije in prikaza podatkov, kritična presoja. Pisanje poročil, priprava predstavitve seminarja.

Intended learning outcomes:

Knowledge and Understanding

The course provides a basic understanding of some materials and chemical processes related to protection of the environment. The student learns the basic principles of sustainable chemistry and the basics of harmful effects on humans and the environment.

Application

The student knows how to use the acquired knowledge in evaluating the impacts on people and the environment. Using practical examples, he learns to evaluate the mentioned impacts and to look for suitable solutions.

Reflection

In the seminars, the student will analyse the chosen topic, using the principles of sustainable chemistry to find solutions to concrete problems.

Skill-Transference Ability

Ability to use domestic and foreign sources of literature, interpretation and presentation of data, critical judgment. Writing reports, preparing presentations.

Metode poučevanja in učenja:

Predavanja, seminar, laboratorijske vaje.

Learning and teaching methods:

Lectures, seminar, laboratory work.

Načini ocenjevanja:**Delež/Weight****Assessment:**

Načini ocenjevanja:	Delež/Weight	Assessment:
Seminarska naloga	20,00 %	Seminar coursework
Laboratorijske vaje	10,00 %	Lab work
Pisni izpit	70,00 %	Written exam

Reference nosilca/Lecturer's references:

1. **MATOH, Lev**, ŽENER, Boštjan, KOVAČIĆ, Marin, KUŠIĆ, Hrvoje, ARČON, Iztok, LEVSTEK, Meta, LAVRENČIČ ŠTANGAR, Urška. Photocatalytic sol-gel/P25 TiO₂ coatings for water treatment: degradation of 7 selected pharmaceuticals. *Ceramics international*, 2022, 1-39.
2. **MATOH, Lev**, KOZJEK-ŠKOFIČ, Irena, ČEH, Miran, BUKOVEC, Nataša. A novel method for preparation of a platinum catalyst at low temperatures. *Journal of materials chemistry. A, Materials for energy and sustainability*, 2013, 1065-1069.
3. KUMAR, Praveen, **MATOH, Lev**, KAUR, Ramanpreet, LAVRENČIČ ŠTANGAR, Urška. Synergic effect of manganese oxide on ceria based catalyst for direct conversion of CO₂ to green fuel additive : catalyst activity and thermodynamics study. *Fuel*, Feb. 2021, vol. 285, 119083.
4. **ŽGAJNAR GOTVAJN, Andreja**, KALČIKOVÁ, Gabriela. Delamination of plastic-coated waste paper by enzymes of the white rot fungus *Dichomitus squalens*. *Journal of environmental management*, Dec. 2018, vol. 228, 165-168.
5. KORICA, Predrag, CIRMAN, Andreja, **ŽGAJNAR GOTVAJN, Andreja**. Comparison of end-of-life vehicles management in 31 European countries: a LMDI analysis. *Waste management & research*, 2022, vol. 40, 1156-1166.
6. LEKŠE, Nina, GRIESSLER BULC, Tjaša, KUKOVIČIČ, Nina, ČERNIC, Taja, **ŽGAJNAR GOTVAJN, Andreja**. Impact of preozonation on biogas potential of PVC microplastics-containing waste sludge. *Global NEST journal*, 2023, vol. 25, 13-19.

OSNOVE VEDE O MATERIALIH

UČNI NAČRT PREDMETA/COURSE SYLLABUS

Predmet:	Osnove vede o materialih
Course title:	Basics in Material Science and Engineering
Članica nosilka/UL Member:	UL FKKT

Študijski programi in stopnja	Študijska smer	Letnik	Semestri	Izbirnost
Kemijska tehnologija, prva stopnja, visokošolski strokovni	Ni členitve (študijski program)	2. letnik	zimski	obvezni

Univerzitetna koda predmeta/University course code:	0072052
Koda učne enote na članici/UL Member course code:	KT135

Predavanja /Lectures	Seminar /Seminar	Vaje /Tutorials	Klinične vaje /Clinical tutorials	Druge oblike študija /Other forms of study	Samostojno delo /Individual student work	ECTS
45	15	15 LV			75	5

Nosilec predmeta/Lecturer: prof. dr. Marjan Marinšek

Vrsta predmeta/Course type: obvezni/mandatory

Jeziki/Languages:

Predavanja/Lectures:	Slovenščina
Vaje/Tutorial:	Slovenščina

Pogoji za vključitev v delo oz. za opravljanje študijskih obveznosti:

Študent oz. kandidat mora imeti predmet opredeljen kot študijsko obveznost.

Prerequisites:

The course has to be assigned to the student.

Vsebina:

- Uvod:** Osnovne skupine materialov ter njihove značilnosti (mehanske, kemijske, optične, električne, magnetne). Kriterij za izbiro materialov. Smeri razvoja materialov.
- Zgradba snovi:** Zgradba atoma in medatomske vezi. Vpliv tipov vezi, jakost vezi in kristalne zgradbe na lastnosti materialov.
- Kristalna struktura.** Molekularne strukture. Osnove za razumevanje katere lastnosti materialov niso odvisne od njihove mikrostrukture, ter na katere lastnosti ima mikrostruktura odločilen vpliv.
- Fazna ravnotežja:** Fazno pravilo. Fazni diagrami. Kinetika fazne transformacije. Poseben poudarek na faznih ravnotežjih sistemov Fe-C, Al-Cu- Al₂O₃-SiO₂.
- Osnove difuzije v trdnem**

Content (Syllabus outline):

- Introduction:** Classification of materials and their basic properties, Choice of a material. Trends in materials' development.
- Materials structure:** The atom, Bonding in solids, Primary bonds, Secondary bonds.
- Crystal structures:** Amorphous and crystalline structure of solids, Bravais lattices and unit cells, Miller indices, Densities and packing factors, Interstitial positions and sizes
- Phase equilibria and phase diagrams:** The one-component phase diagram, phase equilibria in two-component system, Fe-C, Al-Cu- Al₂O₃-SiO₂ phase diagrams, Kinetics and microstructure of structural transformations.

<p>6. Mehanske lastnosti materialov (natezna trdnost, zlomna trdnost, žilavost, krhkost)</p> <p>7. Kovine: Lastnosti kovin ter vpliv sestave in tehnologije izdelave kovinskih materialov na njihove lastnosti. Zlitine. Mehanizmi utrjevanja kovin. Kaljenje in popuščanje jekla.</p> <p>8. Polimeri: Lastnosti polimerov ter vpliv sestave in tehnologije izdelave polimerov na njihove lastnosti. Termoplasti, duroplasti in elastomeri. Deformacija in utrjevanje polimernih materialov. Lezenje in zlom.</p> <p>9. Keramika: Značilne lastnosti keramičnih materialov. Struktura silikatne keramike. Klasična in sodobna tehnična keramika. Krhki lom in utrjevanje keramike. Funkcijska in inženirska keramika.</p> <p>10. Propad materialov: Osnove korozije. Visokotemperaturni propad. Osnove zaščite materialov.</p> <p>11. Kriteriji za izbor materialov: Možnost za njegovo obdelavo, ekonomska upravičenost izbire, stabilnost</p> <p>12. Zelene tehnologije priprave materialov: Izbrani primeri zelenih tehnologij priprave materialov.</p> <p>Seminar: Slušatelji v okviru seminarja rešujejo računске primere, ki ilustrirajo principe vpliva zgradbe snovi na lastnosti in propad materialov.</p> <p>Vaje: Pri vajah se slušatelji praktično seznanijo z bistvenimi fazami izdelave keramičnega izdelka, ter določijo njegove mikrostrukturne, termične, mehanske in električne lastnosti.</p>	<p>5. Defects in solids and diffusion: Point defects, Linear, planar and volume defects, solid state diffusion</p> <p>6. Mechanical properties: Deformation and fracture of engineering materials, brittle fracture, Fracture mechanism</p> <p>7. Metals: Metallic materials, Properties, Application, Hardening</p> <p>8. Polymers: Polymer materials, Properties, Application, Deformation mechanism, Creep</p> <p>9. Ceramics: Properties, Glass-ceramics, Silicates, Engineering ceramics</p> <p>10. Materials degradation: Basics in corrosion, High-temperature degradation</p> <p>11. Choice of materials: Possibility of materials' processing, the economic aspects of materials' choice, stability of materials.</p> <p>12. Green technologies of materials preparation: Selected examples of green technologies for materials preparation.</p> <p>Seminar work: Material science through mathematical problems</p> <p>Tutorial work: Laboratory preparation of ceramic component. Microstructure description, thermal and electrical properties determination</p>
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Temeljna literatura in viri/Readings:

- D.R. Askeland, P.P. Phule, The Science and Engineering of Materials, 5th.ed. Thomson Learning, Brooks Cole, 2006, 748 strani (50%)
- P. Glavič, Građiva, Tehniška fakulteta Maribor, 1990, 424 strani (20%)
- W,D,Jr. Callister, Materials Science and Engineering – An Introductin, 7th. ed. John Wiley & Sons, Inc., New York, 2007, 720 strani (20%)

Cilji in kompetence:

Namen predmeta je spoznavanje materialov kot sestavnih delov različnih struktur s katerimi imamo opravka v vsakdanjem življenju. Ob tem bo študent pridobil znanja o lastnosti materialov s poudarkom na kemijskih, fizikalnih in mehanskih lastnostih. Hkrati bo študent spoznal konkretne materiale (kovine, polimerne materile, keramiko in kompozitne materiale), ki se uporabljajo v industrijskih in drugih aplikacijah.

Poudarek predmeta je na razumevanju soodvisnosti načina povezovanja osnovnih gradnikov materialov ter njihovih lastnosti oziroma mikro- in makro-strukture materialov.

V okviru predmeta bo študent pridobil tudi znanja potrebna za kvalitetno napoved možnosti odpovedi gradiv na katere delujejo normalne obremenitve, ali

Objectives and competences:

Students get related with various types of materials and their basic properties (chemical, physical, mechanical).

Ability to investigate the relationship between the structure of materials at atomic or molecular scales and their macroscopic properties.

Ability to predict failure of chosen material under load or in aggressive environment.

Choice of a material in a certain application on a basis of the desired properties and relative performance.

pa so izpostavljeni zaostrenim, izrednim razmeram oziroma korozijskemu okolju. Skupaj z znanji za napoved uporabnosti gradiv bo študent pridobil tudi osnovna znanja zaščite materialov.	
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Predvideni študijski rezultati: Znanje in razumevanje Študentje spoznajo soodvisnost med zgradbo in lastnostmi materialov. V kurzu bomo podali osnovna znanja o različnih tipih materialov, njihovi pripravi ter njihovih lastnostih. Študente bomo uvajali k samostojnemu, logičnemu in kritičnemu razmišljanju o lastnostih in uporabi različnih materialov. Uporaba Študentje analizirajo znane materiale iz prakse in s tem ilustrirajo principe pridobljene pri predavanjih. Refleksija Študent pridobi znanje za smotrno analizo uporabe izbranega materiala ter možnosti, da pri njegovi uporabi zaradi izrednih pogojev pride do neželenih sprememb. Prenosljive spretnosti Splošno razumevanje področja materialov, pridobitev laboratorijskih spretnosti, seznanitev z modernimi analiznimi tehnikami, uporaba domače in tuje literature, pisna in ustna predstavitev problema, delo v skupinah.	Intended learning outcomes: Knowledge and Comprehension Understanding the relationship between the structure of materials at various scales and their macroscopic properties. Students get to know various types of materials – their properties and preparation. Students are encouraged to logical and critical thinking about various materials. Application Students analyse known various materials with respect to material science. Analysis Ability to understand materials so that right materials with the desired properties are chosen for specific application without unnecessary risk for its failure. Skill-transference Ability General understanding of materials, developed laboratory skills, modern analytical techniques, literature data collecting, data analysis and interpretation, team work.
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Metode poučevanja in učenja: Predavanja, seminar, laboratorijsko delo, poučevanje in učenje izbrane vsebine s pomočjo umetniške izkušnje	Learning and teaching methods: Lectures, seminars, tutorial work, teaching and learning of selected topics through artistic experience
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Načini ocenjevanja:	Delež/Weight	Assessment:
Pisni izpit.	40,00 %	Written exam.
Ustni izpit.	60,00 %	Oral exam.
Pisni izpit se v celoti lahko opravi z vmesnimi kolokviji in opravljeno seminarско nalogo. Pozitivna ocena 6-10, negativna ocena 1-5. Opravljene laboratorijske vaje so pogoj za opravljanje izpita.		Written exam can be accomplished also by achieving positive grades at written colloquiums during the semester and prepared individual seminar work. Pass grades from 6 to 10, fail grades from 1 to 5. Tutorial work must be done before taking an exam.

Reference nosilca/Lecturer's references: LUO, Yujia, KRALJ CIGIĆ, Irena, WEI, Quan, MARINŠEK, Marjan , STRLIČ, Matija. Material properties and durability of 19th–20th century Tibetan manuscripts. <i>Cellulose</i> . Dec. 2023, vol. 30, iss. 18, str. 11783-11795 SKALAR, Tina, ZUPAN, Klementina, MARINŠEK, Marjan . Microstructure tailoring of combustion-derived Ni-GDC and Ni-SDC composites as anode materials for intermediate temperature solid oxide fuel cells. <i>Journal of the Australian Ceramic Society</i> . Mar. 2019, vol. 55, iss. 1, str. 123-133 PECOVSKA-GJORGJEVICH, Margareta, ALEKSOVSKA, Slobotka, DIMITROVSKA-LAZOVA, Sandra, MARINŠEK, Marjan . The role of Cr/Co substitution on dielectric properties of gadolinium orthochromite. <i>Physica scripta</i> . apr. 2016, vol. 91, no. 4, art. no. 045805
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OSNOVNE OPERACIJE V KEMIJSKEM INŽENIRSTVU

UČNI NAČRT PREDMETA/COURSE SYLLABUS

Predmet:	Osnovne operacije v kemijskem inženirstvu
Course title:	Unit operations in chemical engineering
Članica nosilka/UL Member:	UL FKKT

Študijski programi in stopnja	Študijska smer	Letnik	Semestri	Izbirnost
Kemijska tehnologija, prva stopnja, visokošolski strokovni	Ni členitve (študijski program)	2. letnik	poletni	obvezni

Univerzitetna koda predmeta/University course code:	0072037
Koda učne enote na članici/UL Member course code:	KT119

Predavanja /Lectures	Seminar /Seminar	Vaje /Tutorials	Klinične vaje /Clinical tutorials	Druge oblike študija /Other forms of study	Samostojno delo /Individual student work	ECTS
45	30				75	5

Nosilec predmeta/Lecturer: prof. dr. Aleš Podgornik

Vrsta predmeta/Course type: obvezni/mandatory

Jeziki/Languages:

Predavanja/Lectures:	Slovenščina
Vaje/Tutorial:	Slovenščina

Pogoji za vključitev v delo oz. za opravljanje študijskih obveznosti: **Prerequisites:**

Študent oz. kandidat mora imeti predmet opredeljen kot študijsko obveznost.	The course has to be assigned to the student.
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Vsebina:

Uvod: Pomen kemijskega inženirstva v svetu tehnike, znanosti in gospodarstva. Domena in smeri razvoja kemijskega inženirstva. Področja aktivnosti kemijskega inženirja. **Proces in procesna shema.** Osnovne procesne spremenljivke. Osnovne faze kemijskega procesa: priprava, kemijska pretvorba, izolacija in čiščenje produkta. **Osnovne operacije.** Koncept in temeljni principi osnovnih operacij. **Mehanske in hidrodinamske operacije:** drobljenje in mletje, sejanje, mešanje, posedanje, centrifugiranje, filtracija. Osnovni principi in naprave. **Termodifuzijske operacije:** destilacija, ekstrakcija, absorpcija, adsorpcija, uparjanje, kristalizacija, sušenje. Osnovni principi in naprave. Primeri sinteze posameznih osnovnih operacij v tehnološki proces.

Content (Syllabus outline):

Introduction. The role of chemical engineering in the field of science and technology as well as economy. Directions of development of chemical engineering. Activity fields of a chemical engineer. **Process diagram.** Basic process variables. Phases in a chemical process: upstream, reaction and downstream. **Unit operations.** Concept and basic principles. **Mechanical and hydrodynamical unit operations:** grinding, screening, mixing, settling, centrifugation, filtration. Basic principles and equipment. **Thermodiffusional unit operations.** Distillation, extraction, absorption, adsorption, crystallization, evaporation, drying. Basic principles and equipment. Selected examples of integration of unit operations into whole technological process.

Temeljna literatura in viri/Readings:

- E.Ignatowitz, Kemijska tehnika (prevedel Leon Čelik), Jutro, Ljubljana, 1996, 456 str. (50%)
- J.H.Harker, J.R.Backhurst, J.F.Richardson, Chemical Engineering, Volume 2, Elsevier, 2002.

Cilji in kompetence:

Cilj predmeta je študente seznaniti z značilnostmi in koncepti kemijsko inženirske stroke. Predmetno specifične kompetence:

- študent spozna osnovno in splošno vlogo ter pomen osnovnih operacij v kemijsko tehnološkem procesu,

razume in zna pripraviti procesno shemo iz osnovnih operacij oziroma aparatov za določen tehnološki proces.

Objectives and competences:

The objective is to acquaint the student with characteristics and concepts of chemical engineering profession. Specific competences are:

- student recognizes basic and general role and significance of unit operations in a chemical process technology,

student understands and is able to prepare a process scheme containing unit operations and equipment for a particular process technology.

Predvideni študijski rezultati:

Znanje in razumevanje
Študent je po osvojitvi pojmov, zakonitosti, teorij in pojavov, ki jih podaja ta predmet, sposoben razumeti specifičnosti kemijsko inženirske stroke ter pomena osnovnih operacij v tehnološkem procesu.
Uporaba
Pridobljena znanja je sposoben uporabiti pri reševanju posameznih praktičnih primerov in problemov v industrijskih kemijsko tehnoloških procesih.
Refleksija
Uporaba splošnih znanj in osnovnih principov kemijskega inženirstva, analiza in kritično ovrednotenje tehnološkega procesa oziroma posameznega postopka in naprave.
Prenosljive spretnosti
Razvita sposobnost identifikacije in reševanja problemov, kritičnega razmišljanja in logičnega sklepanja. Sposobnost uporabe literature, zbiranja in interpretacije podatkov in njihove kritične evalvacije.

Intended learning outcomes:

Knowledge and Comprehension
After mastering notions, laws, theories and phenomenon presented by those course, student is able to understand the specifics of chemical engineering profession and the role of unit operations in the chemical process technology.
Application
Student is able to use the acquired knowledge in solving particular practical cases and problems from chemical process technology on industrial scale.
Analysis
Use of general knowledge and basic principles of chemical engineering as well as analysis and critical evaluation of chemical process technology and particular operation and equipment.
Skill-transference Ability
Developed skill to identify and solve problem, critical thinking and making logical conclusions. Ability of literature data using, data collection and interpretation as well as their critical evaluation.

Metode poučevanja in učenja:

Predavanja in seminarji, laboratorijske vaje.

Learning and teaching methods:

Lectures, seminars, laboratory exercises.

Načini ocenjevanja:

Pisni in ustni izpit Pisna seminarska naloga (pogoj za pristop k izpitu) Ocene: 6-10 (pozitivno), 1-5 (negativno).

Delež/Weight**Assessment:**

Written and oral exam Written seminar (mandatory before exam) Marks: 6-10 (positive), 1-5 (negative).

Reference nosilca/Lecturer's references:

ANDREJČIČ, Miha, PODGORNİK, Aleš. Effect of pressure drop model implemented for description of pressure drop on chromatographic monolith on estimated adsorbed layer thickness. *Chemical Engineering Science*, ISSN 0009-2509. [Print ed.], Apr. 2017, vol. 161, str. 370-381, doi: [10.1016/j.ces.2016.12.011](https://doi.org/10.1016/j.ces.2016.12.011).
GAŠPERŠIČ, Jernej, PODGORNİK, Aleš, KRAMBERGER, Petra, JARC, Marko, JANČAR, Janez, ŽORŽ, Mirjan, LENDERO KRAJNC, Nika. Separation of pegylated recombinant proteins and isoforms on CIM ion exchangers. *Journal of chromatography. B, Analytical technologies in the biomedical and life sciences*, ISSN 1570-0232, Oct. 2016, vol. 1033/1034, str. 91-96, doi: [10.1016/j.jchromb.2016.07.020](https://doi.org/10.1016/j.jchromb.2016.07.020).

ISAKARI, Yu, KISHI, Yuhi, YOSHIMOTO, Noriko, YAMAMOTO, Shuichi, PODGORNIK, Aleš.
Reaction-mediated desorption of macromolecules : novel phenomenon enabling simultaneous reaction and separation. *Biotechnology journal*, ISSN 1860-6768, 2018, vol. 13, iss. 7, str. 1-9, ilustr. doi:
[10.1002/biot.201700738](https://doi.org/10.1002/biot.201700738).

ULB
ELEKTI

POLIMERNI MATERIALI

UČNI NAČRT PREDMETA/COURSE SYLLABUS

Predmet:	Polimerni materiali
Course title:	Polymer Materials
Članica nosilka/UL Member:	UL FKKT

Študijski programi in stopnja	Študijska smer	Letnik	Semestri	Izbirnost
Kemijska tehnologija, prva stopnja, visokošolski strokovni	Ni členitve (študijski program)	2. letnik, 3. letnik		izbirni

Univerzitetna koda predmeta/University course code:	0099482
Koda učne enote na članici/UL Member course code:	KTSI1

Predavanja /Lectures	Seminar /Seminar	Vaje /Tutorials	Klinične vaje /Clinical tutorials	Druge oblike študija /Other forms of study	Samostojno delo /Individual student work	ECTS
60		15 LV			75	5

Nosilec predmeta/Lecturer: viš. pred. dr. Branko Alič

Vrsta predmeta/Course type: izbirni strokovni/Elective Professional

Jeziki/Languages:

Predavanja/Lectures:	Slovenščina
Vaje/Tutorial:	Slovenščina

Pogoji za vključitev v delo oz. za opravljanje študijskih obveznosti:

Študent oz. kandidat mora imeti predmet opredeljen kot študijsko obveznost.

Prerequisites:

The course has to be assigned to the student.

Vsebina:

- Definicije pojmov monomer, oligomer, polimer, polimerizacija, stopnja polimerizacije, ponavljajoča se enota, zamreženje, kopolimer, kopolimerizacija;
- Osnovne vrste polimernih materialov;
- Fizikalno stanje in urejenost polimernih verig: amorfno in kristalinično stanje, deformacije stanja;
- Toplotni prehodi in mehanske lastnosti;
- Osnovne vrste polimerizacij, stopenjski in verižni polimeri;
- Sintetični polimeri: plastomeri, duromeri, elastomeri;
- Polimeri za široko proizvodnjo in inženirski polimeri;
- Polimerni kompoziti, nanokompoziti, zmesi;
- Recikliranje in degradacija polimernih materialov;
- Polimeri s posebnimi lastnostmi, biopolimeri, polimeri iz obnovljivih virov;

Content (Syllabus outline):

- Definition of terms monomer, oligomer, polymer, polymerization, degree of polymerization, repeating unit, crosslinking, copolymerization, copolymer;
- Basic types of polymer materials;
- Molecular interactions, polymer crystals, amorphous bulk state;
- Heat transitions and mechanical properties;
- Basic types of polymerizations, step and chain polymers;
- Synthetic polymers: plastomers, thermosets, elastomers;
- Common polymers, polymer materials for general use and engineering polymer materials;
- Polymer composites, nanocomposites and blends;
- Recycling and degradation of polymer materials;

<ul style="list-style-type: none"> - Instrumentalne tehnike za karakterizacijo polimernih materialov; - Laboratorijske vaje: Sinteza nasičenega poliestra; zamreženje nenasičenega poliestra; Izdelava plastisola; Ekstrudiranje plastomerov. 	<ul style="list-style-type: none"> - Polymers designed for specific use, biopolymers, polymers from renewable resources; - Instrumental techniques for polymer materials characterization; - Laboratory practice: Synthesis of saturated polyester; Crosslinking of unsaturated polyester; Making plastisol; Extrusion of plastomers.
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Temeljna literatura in viri/Readings:

<ul style="list-style-type: none"> - C. E. Carraher, Jr., Polymer Chemistry: An Introduction, 4th Ed., Marcel Dekker, Inc., New York, 1996, 541 str., (60 %), - R. O. Ebewele, Polymer Science and Technology, CRC Press, Boca Raton, 2000, 471 str., (20 %). <p>Dopolnilna literatura:</p> <ul style="list-style-type: none"> - C. A. Harper, Handbook of Plastics Technologies, McGraw-Hill, New York, 2006, (loč. pag.).

Cilji in kompetence:

<p>Cilj predmeta je, da študentje osvojijo osnovna znanja o polimernih materialih in njihovih ključnih lastnostih.</p> <p>Študentje pri predmetu pridobijo naslednje specifične kompetence:</p> <ul style="list-style-type: none"> - poznavanje fizikalnih stanj in načina urejanja polimernih verig v polimernih materialih ter razumevanje vpliva na lastnosti polimernih materialov; - poznavanje osnovnih vrst polimernih materialov, njihovih specifičnosti in uporabe; - razlikovanje med osnovnimi sintetičnimi polimernimi materiali; - razlikovanje med polimernimi materiali za široko potrošnjo in inženirskimi polimernimi materiali; - poznavanje polimernih materialov s specifičnimi lastnostmi; - razumevanje pomena in prednosti polimernih zmesi, polimernih kompozitov in polimernih nanokompozitov; - poznavanje osnovnih biopolimerov; - poznavanje možnosti ter načinov recikliranja in razgradnje polimerov. 	<p>Objectives and competences:</p> <p>Acquisition of basic knowledge about polymer materials and their properties;</p> <p>Acquisition of knowledge on polymer morphology and polymer structure-property relationships, knowledge about basic polymer materials, key properties and use of basic polymer materials, distinction between basic synthetic polymer materials for common use and engineering polymer materials, polymer materials with specific properties, polymer blends and composites, polymer nanocomposites, biopolymers, polymer recycling and degradation.</p>
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Predvideni študijski rezultati:

<p>Znanje in razumevanje Študent pozna osnovne in specifične, znane polimerne materiale in njihove lastnosti ter uporabnost. Lastnosti polimernih materialov zna povezati z njihovo strukturo in fizikalnim stanjem. Razume pomen in prednosti polimernih zmesi, polimernih kompozitov in polimernih nanokompozitov. Pozna osnovne načine recikliranja polimernih materialov.</p> <p>Uporaba Znanja iz predmeta zna uporabiti pri študiju predmetov s področja polimernega inženirstva in materialov. Študent je sposoben izbirati med različnimi polimernimi materiali za izbrano aplikacijo.</p>	<p>Intended learning outcomes:</p> <p>Knowledge and Comprehension Knowledge and Comprehension: Understanding the basic principles of composition and structure of polymer materials and polymer composites, and understanding the basic principles of the relationship between polymer material properties and their composition and structure.</p> <p>Application Acquired knowledge is necessary for appropriate polymer material selection for a specific application, and for work, research and development in the field of polymer materials and polymer engineering.</p>
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<p>Refleksija Študent je sposoben pridobljeno znanje uporabiti pri sorodnih predmetih in na področjih, kjer se polimerni materiali uporabljajo. Prenosljive spretnosti Razvita sposobnost kritičnega razmišljanja in logičnega sklepanja. Sposobnost študija domače in tuje literature.</p>	<p>Analysis Ability to apply knowledge about material properties at familiar courses and/or when a polymer material selection is needed. Skill-transference Ability Ability of critical thinking and deduction; Ability of studying relevant literature from the field of polymer materials.</p>
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Metode poučevanja in učenja:

Predavanja, seminarji, laboratorijske vaje

Learning and teaching methods:

Lectures, seminars, laboratory practice.

Načini ocenjevanja:

Delež/Weight

Assessment:

Opravljene vaje so pogoj za pristop k izpitu.		Laboratory practice is a prerequisite to exam attendance
Poročila in zagovor laboratorijskih vaj.	30,00 %	Written reports and oral laboratory practice defence.
Pisni izpit.	70,00 %	Written exam.

Reference nosilca/Lecturer's references:

ŠEBENIK, Urška, KRAJNC, Matjaž, ALIČ, Branko, LAPASIN, Romano. Ageing of aqueous TEMPO-oxidized nanofibrillated cellulose dispersions : a rheological study. *Cellulose*, ISSN 0969-0239, 2018, vol. , iss. , str. 1-15, doi: [10.1007/s10570-018-2128-1](https://doi.org/10.1007/s10570-018-2128-1).

RUČIGAJ, Aleš, ALIČ, Branko, KRAJNC, Matjaž, ŠEBENIK, Urška. Investigation of cure kinetics in a system with reactant evaporation : epoxidized soybean oil and maleic anhydride case study. *European Polymer Journal*, ISSN 0014-3057. 2014, vol. 52, no. 1, str. 105-116, doi: [10.1016/j.eurpolymj.2014.01.009](https://doi.org/10.1016/j.eurpolymj.2014.01.009).

RUČIGAJ, Aleš, ALIČ, Branko, KRAJNC, Matjaž, ŠEBENIK, Urška. Curing of bisphenol A-aniline based benzoxazine using phenolic, amino and mercapto accelerators. *Express polymer letters*, ISSN 1788-618X, 2015, vol. 9, no. 7, str. 647-657, doi: [10.3144/expresspolymlett.2015.60](https://doi.org/10.3144/expresspolymlett.2015.60).

KAJTNA, Jernej, ALIČ, Branko, KRAJNC, Matjaž, ŠEBENIK, Urška. Influence of hydrogen bond on rheological properties of solventless UV crosslinkable pressure sensitive acrylic adhesive prepolymers. *International journal of adhesion and adhesives*, ISSN 0143-7496. 2014, vol. 49, no. 1, str. 103-108, doi: [10.1016/j.ijadhadh.2013.12.016](https://doi.org/10.1016/j.ijadhadh.2013.12.016). [COBISS.SI-ID [1664047](https://www.cobiss.si/id/1664047)]

ALIČ, Branko, ŠEBENIK, Urška, KRAJNC, Matjaž. Microencapsulation of butyl stearate with melamine-formaldehyde resin : effect of decreasing the pH value on the composition and thermal stability of microcapsules. *Express polymer letters*, ISSN 1788-618X, 2012, vol. 6, no. 10, str. 826-836, doi: [10.3144/expresspolymlett.2012.88](https://doi.org/10.3144/expresspolymlett.2012.88).

ALIČ, Branko, ŠEBENIK, Urška, KRAJNC, Matjaž. Differential scanning calorimetric examination of melamine-formaldehyde microcapsules containing decane. *Journal of applied polymer science*, ISSN 0021-8995. 2011, vol. 119, no. 6, str. 3687-3695, doi: [10.1002/app.33077](https://doi.org/10.1002/app.33077)

PRAKTIČNO USPOSABLJANJE

UČNI NAČRT PREDMETA/COURSE SYLLABUS

Predmet:	Praktično usposabljanje
Course title:	Industrial practice
Članica nosilka/UL Member:	UL FKKT

Študijski programi in stopnja	Študijska smer	Letnik	Semestri	Izbirnost
Kemijska tehnologija, prva stopnja, visokošolski strokovni	Ni členitve (študijski program)	3. letnik	poletni	obvezni

Univerzitetna koda predmeta/University course code:	0072053
Koda učne enote na članici/UL Member course code:	PRUSP

Predavanja /Lectures	Seminar /Seminar	Vaje /Tutorials	Klinične vaje /Clinical tutorials	Druge oblike študija /Other forms of study	Samostojno delo /Individual student work	ECTS
				150		5

Nosilec predmeta/Lecturer: prof. dr. Igor Plazl

Vrsta predmeta/Course type: obvezni/mandatory

Jeziki/Languages:	Predavanja/Lectures:	Slovenščina
	Vaje/Tutorial:	Slovenščina

Pogoji za vključitev v delo oz. za opravljanje študijskih obveznosti:

Študent oz. kandidat mora imeti predmet opredeljen kot študijsko obveznost.

Prerequisites:

The course has to be assigned to the student.

Vsebina:

Pri praksi se študenti seznanijo z zahtevnostjo in kompleksnostjo vodenja industrijskih procesov. Spoznajo, da je za uspešno in varno delo v industriji osnovni pogoj natančno poznavanje vseh faz procesa in podrobna kemijska analiza in druga karakterizacija surovin, intermediatov, procesnih tokov in končnih produktov, kot tudi celovita analiza njegovega delovanja. Uspešnost procesa je pogojena z mnogo dejavniki in za njegovo varno obratovanje je potrebno tako optimalno delovanje posameznih procesnih operacij kot tudi usklajeno delovanje sistema kot celote.

Vsebina prakse se prilagaja konkretnemu mestu kjer se opravlja. Področja na katerih študent lahko opravlja prakso so:

Content (Syllabus outline):

During practical training student meets with complexity of management of industrial processes. Student recognizes that: accurate knowledge of all stages of technological process, detailed chemical analysis and other characterization method of raw materials, intermediates and final products, process flows and entire analysis of process operation is essential for successful and safety industrial work. Success of technological process depends on many factors, among them optimal operation of separate process stages and adjustment of overall technological system are most important. Content of practical training depends on specific working environment. Fields where student can perform practical training are:
- Introducing to work of chemical engineer

<ul style="list-style-type: none"> - uvajanje v delo inženirja kemijske tehnologije, - spoznavanje s tehnološkim procesom in industrijsko proizvodnjo, - sodelovanje pri raziskovalno razvojnih nalogah in planiranju ter načrtovanju izdelkov, - nadzor proizvodnega procesa, - vhodna in izhodna kontrola kvalitete surovin in produktov, - instrumentalna analitika v raziskovalnem in kontrolnem laboratoriju, - aktivnosti v zvezi z varovanjem okolja in zagotavljanjem varnosti, - vzdrževanje aparatov, merilnih in regulacijskih sistemov. 	<ul style="list-style-type: none"> - Collaboration in technological process and industrial production - Collaboration in research projects, planning and design of final products - Supervision of technological process - Instrumental analysis in research and control labs - Activities for environment protections and to ensure working safety - Apparatus, measuring and system maintenance
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Temeljna literatura in viri/Readings:

Nabor literature bo študent dobil na mestu opravljanja prakse oziroma jo lahko dobi tudi v knjižnici UL FKKT.

Cilji in kompetence:

Namen prakse je omogočiti študentom preverjanje posredovanih teoretičnih znanj v okolju v katerem bodo delovali po zaključku študija ter jih nadgradili z znanji, ki so značilna za industrijsko tehnološko okolje in jih ni možno dobiti na šoli. Praksa poteka v povezavi študent – mentor v podjetju ali inštituciji – mentor na fakulteti.

Praktično usposabljanje uvajanja študente v praktično delo in s tem spoznavanje strokovne narave dela ter aktualnih problematik v laboratoriju, industrijski proizvodnji in drugod.

Objectives and competences:

Goal of practical training is to give opportunity that student verifies his/her theoretical knowledge, which was gained during study, under working conditions where student may operate after finishing study. Furthermore, to upgrade his/her theoretical knowledge with proficiency, specific for industrial technological practice, which he/her cannot learn during education. Practical training is supervised by industrial mentor (from establishment) and mentor at University.

Practical training introduces student to practical work, technical work profession and actual problems in industrial productions and in laboratory.

Predvideni študijski rezultati:

Znanje in razumevanje
Študent se pri opravljanju praktičnega dela usposobi za povezovanje teoretičnih in praktičnih znanj, ki jih je pridobil pri različnih predmetih med študijem z dejanskimi pogoji v praksi, tj. analiznih laboratorijih in laboratorijih za kontrolo kvalitete, industrijskih obratih. Študent spozna način reševanja posameznega problema, se seznani s tehnološko-tehničnimi parametri, se nauči strokovne komunikacije z drugim člani tima.

Uporaba
Praktično usposabljanje razvija pri študentu: sposobnost prenosa teoretičnih znanj na reševanje konkretnih problemov, predstavi sodoben pristop k reševanju inženirskih problemov, razvija sposobnost za vključevanje v skupinsko delo, sposobnost komuniciranja s sodelavci in strokovnjaki drugih disciplin, kar mu omogoča sodelovanje pri multidisciplinarnih projektih in mu razvija profesionalno etično in okoljsko odgovornost.

Refleksija

Intended learning outcomes:

Knowledge and Comprehension
By realization of practical training student knows how to connect theoretical knowledge, which was gained during education at University and practical experience present in various industrial fields (laboratories for analysis, research and quality control, industrial plants). Student finds a way to solve specific problems, learns about technological process parameters, and understands professional communication in team.

Application
Practical training develops student's skill: capability to transfer theoretical knowledge on solving concrete problems, student gains contemporary approach to solving engineering problems, student develops the ability to integrate in working team, student gains ability to communicate with fellow workers and experts from other disciplines, which enables to collaborate on multidisciplinary projects and develops ethical and environmental responsibility.

Analysis

<p>Študent je sposoben kritično analizirati in primerjati različne pristope pri reševanju problemov tako na laboratorijskem kot tudi industrijskem nivoju.</p> <p>Prenosljive spretnosti</p> <p>Usposabljanje v konkretnem delovnem okolju mu razvija sposobnost za analitično naravoslovno tehnično vrednotenje dogajanj v praksi.</p>	<p>Student is able to compare and critical examines different approaches for solving problems on level of laboratory as well as those at industrial plant.</p> <p>Skill-transference Ability</p> <p>Practical training in a certain working environment develops student's capability of science and technical evaluation of developments in practice.</p>
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Metode poučevanja in učenja:

Praksa poteka v izbranem podjetju oziroma drugi inštituciji, s katerim je vnaprej podpisana tripartitna pogodba, ki določa pogoje usposabljanja. V podjetju vodi delo študenta delovni mentor, ki mora imeti najmanj stopnjo izobrazbe SOK 7 kemijsko inženirske ali sorodne smeri.

Learning and teaching methods:

Practical training is performed in a company or other institution. Agreement with company, determining training details and conditions, has to be signed. Company supervisor should have at least SOK 7th education level of chemical engineering or similar discipline.

Načini ocenjevanja:

Študent odda dnevnik in sumarno poročilo o praksi. Potrdilo o opravljenem praktičnem usposabljanju z oceno delovnega mentorja v podjetju in fakultetnega mentorja je osnova za oblikovanje ocene. Ocenjevalna lestvica: opravljeno - neopravljeno

Delež/Weight

Assessment:

Student submits diary and summary report on practical training. The student obtains a certificate of completion practical training with an assessment of the work of the mentor in the company and assessment of the mentor at Faculty. Grading scale: Done – Failure

Reference nosilca/Lecturer's references:

STRNIŠA, Filip, BAJIĆ, Marijan, PANJAN, Peter, PLAZL, Igor, SESAY, Adama Marie, ŽNIDARŠIČ PLAZL, Polona. Characterization of an enzymatic packed-bed microreactor : experiments and modeling. Chemical engineering journal, 2018, 1-40.

MILOŽIČ, Nataša, LUBEJ, Martin, LAKNER, Mitja, ŽNIDARŠIČ PLAZL, Polona, PLAZL, Igor. Theoretical and experimental study of enzyme kinetics in a microreactor system with surface-immobilized biocatalyst. Chemical engineering journal, 2017, 313, 374-381.

PRAŠNIKAR, Anže, URBIČ, Tomaž, PLAZL, Igor. Microscale technology and biocatalytic processes. Part 1, How to minimize the gap between research and industry. Chimica oggi = Chemistry today, 2017, 35(3), 18-20.

PRAKTIKUM IZ ANORGANSKE KEMIJE

UČNI NAČRT PREDMETA/COURSE SYLLABUS

Predmet:	Praktikum iz anorganske kemije
Course title:	Practical course in inorganic chemistry
Članica nosilka/UL Member:	UL FKKT

Študijski programi in stopnja	Študijska smer	Letnik	Semestri	Izbirnost
Kemijska tehnologija, prva stopnja, visokošolski strokovni	Ni členitve (študijski program)	1. letnik	poletni	obvezni

Univerzitetna koda predmeta/University course code:	0072024
Koda učne enote na članici/UL Member course code:	KT110S

Predavanja /Lectures	Seminar /Seminar	Vaje /Tutorials	Klinične vaje /Clinical tutorials	Druge oblike študija /Other forms of study	Samostojno delo /Individual student work	ECTS
	15	60 LV			75	5

Nosilec predmeta/Lecturer: prof. dr. Franc Perdih

Vrsta predmeta/Course type: obvezni/mandatory

Jeziki/Languages:

Predavanja/Lectures:	Slovenščina
Vaje/Tutorial:	Slovenščina

Pogoji za vključitev v delo oz. za opravljanje študijskih obveznosti:

Študent oz. kandidat mora imeti predmet opredeljen kot študijsko obveznost.

Prerequisites:

The course has to be assigned to the student.

Vsebina:

Vsebina seminarjev in vaj: Študenti bodo pri predmetu sintetizirali anorganske snovi z različnimi sintezniimi tehnikami in dobljene snovi preiskali. Spoznali bodo metode karakterizacije v anorganski kemiji, prirejene predznanju študentov: uporaba rentgenske praškovne analize, termične analize in infrardeče in UV-vidne spektroskopije ter magnetne meritve. Študenti bodo sintetizirali bazični bakrov(II) sulfat, zemeljskoalkalijske oksalate hidrate, pentafluoridomanganate(III), izomere bakrovih glicinato kompleksov in do dve snovi, ki se uporabljata pri tekočem raziskovalnem delu nosilca predmeta ali njegovih sodelavcev. Pri seminarju bodo študenti dobili potrebno teoretsko osnovo in navodila za sintezo.

Content (Syllabus outline):

Syntheses and characterization of inorganic compounds. Different characterization methods used in inorganic chemistry will be studied, adapted to the knowledge level of these students: X-ray powder diffraction analysis, thermal analysis, infrared spectroscopy, UV-vis spectroscopy and measurements of magnetic susceptibility. Basic copper(II) sulphate, alkaline earth oxalates hydrates, pentafluoridomanganates(III), isomers of copper glycinato complexes and some new complexes which are subject of current research at the department of Inorganic chemistry are prepared and characterized. Theoretical background is explained at seminars.

Temeljna literatura in viri/Readings:

- S. Petriček, F. Perdih, A. Demšar: Vaje iz anorganske kemije, FKKT UL, Ljubljana, 2012, pp. 1–74.

Cilji in kompetence:

Cilj predmeta je poglobitev znanja iz anorganske kemije in pridobitev posebnih laboratorijskih veščin. Specifično, študenti z laboratorijskimi vajami nadgradijo teoretično znanje iz predmeta Anorganska kemija I s praktičnimi izkušnjami s področja sinteze in karakterizacije anorganskih snovi.

Objectives and competences:

Learning outcomes: Expanding of basic knowledge of syntheses and characterization of inorganic compounds obtained in courses of General chemistry and Inorganic Chemistry I.

Competences: Practical skills in comprehensive inorganic syntheses and characterization of inorganic compounds.

Predvideni študijski rezultati:

Znanje in razumevanje
Predmet predstavlja dopolnitev predmeta Anorganska kemija I s praktičnim delom in izkušnjami iz laboratorija. Tehnike, ki jih obvladajo študenti, bodo uporabili pri nadaljnjem študiju in pri delu.

Uporaba

Študent spozna metode sinteze spojin in določitve njihovih lastnosti. V laboratoriju dobi občutek za varno laboratorijsko delo. Nauči se uporabljati laboratorijsko opremo in instrumente. Izkušnje pri delu in občutek za snovi skupaj s teoretskim znanjem omogočajo kemiku potrebno strokovno širino.

Refleksija

Kemija je eksperimentalna veda in za diplomante tega študijskega programa je poznavanje dela v laboratoriju in kemijskem obratu posebej pomembno in hkrati motivacija pri osvajanju teoretskega znanja drugih predmetov.

Prenosljive spretnosti

Laboratorijske veščine, izkušnje in prijemi pri načrtovanju sintez so pomembni pri drugih kemijskih predmetih in pri osebnemu strokovnemu razvoju.

Intended learning outcomes:**Knowledge and Comprehension**

Subject is a supplement of a course Chemistry I with practical work and experience in the laboratory. Techniques learned in this course will be used for further studies and work.

Application

Different synthetic methods are performed and properties of the prepared materials are determined. Laboratory skills are developed also in the field of laboratory safety. Students learn how to use laboratory equipment and instruments. Experiences and sense of materials together with theoretical knowledge enables to a chemist the necessary expertise insight.

Analysis

Chemistry is an experimental science and for graduates of this study program is the chemical and laboratory knowledge particularly important and is at the same time a motivation in acquiring theoretical knowledge of other subjects.

Skill-transference Ability

Laboratory skills, experiences and the knowledge on different synthetic approaches are important in other chemistry courses and in their personal professional development.

Metode poučevanja in učenja:

Predmet se izvaja v obliki seminarjev in samostojnih laboratorijskih vaj. Na seminarju se tematiko vsake vaje umesti v širši kontekst anorganske kemije.

Learning and teaching methods:

Seminars and individual laboratory practice. At the seminar, the topic of each practical assignment is presented within the broader context of inorganic chemistry.

Načini ocenjevanja:

Pisni izpit po uspešno opravljenem praktičnem delu.

Delež/Weight**Assessment:**

Written examination after successfully passed all laboratory work.

Reference nosilca/Lecturer's references:

- **F. Perdih:** Diversity of supramolecular aggregation in copper(II) pentane-2,4-dionato compounds with methyl substituted 2-aminopyridines. *J. Coord. Chem.* 2012, 65, 1850–1591.

- **F. Perdih:** Different coordination modes and supramolecular aggregations in copper(II) pentane-2,4-dionato compounds with 2-pyridone and 3-hydroxypyridine. *Monatshefte für Chemie* 2012, 143, 1011–1017.

- **F. Perdi**, A. Perdi: Lignin selective dyes : quantum-mechanical study of their characteristics. Cellulose 2011, 18, 1139–1150.

ULFAKKA

PRAKTIKUM IZ MATERIALOV

UČNI NAČRT PREDMETA/COURSE SYLLABUS

Predmet:	Praktikum iz materialov
Course title:	Practical Course in Materials Characterisation
Članica nosilka/UL Member:	UL FKKT

Študijski programi in stopnja	Študijska smer	Letnik	Semestri	Izbirnost
Kemijska tehnologija, prva stopnja, visokošolski strokovni	Ni členitve (študijski program)	2. letnik, 3. letnik		izbirni

Univerzitetna koda predmeta/University course code:	0100223
Koda učne enote na članici/UL Member course code:	KTSI34

Predavanja /Lectures	Seminar /Seminar	Vaje /Tutorials	Klinične vaje /Clinical tutorials	Druge oblike študija /Other forms of study	Samostojno delo /Individual student work	ECTS
		25 SV + 50 LV			75	5

Nosilec predmeta/Lecturer: izt. prof. dr. Boštjan Genorio

Vrsta predmeta/Course type: izbirni strokovni/Elective Professional

Jeziki/Languages:	Predavanja/Lectures:	Slovenščina
	Vaje/Tutorial:	Slovenščina

Pogoji za vključitev v delo oz. za opravljanje študijskih obveznosti:

Študent oz. kandidat mora imeti predmet opredeljen kot študijsko obveznost.

Prerequisites:

The course has to be assigned to the student.

Vsebina:

Praktikum obsega seminarske in laboratorijske vaje. Študenti se na teoretičnem nivoju seznanijo z osnovnimi postopki in metodami preiskovanja lastnosti različnih materialov. Pri eksperimentalnem delu spoznajo aplikativno uporabo karakterizacijskih metod. Tekom dela se praktično seznanijo z mikroskopskimi tehnikami (optični mikroskop, vrstični elektronski mikroskop, mikroskop na atomsko silo, segrevalni mikroskop ...), rentgensko fotoelektronsko spektroskopijo, termo analitskimi tehnikami, karakterizacijo velikosti in porazdelitve velikosti delcev, specifične površine, porazdelitve por, merjenje mehanskih, elektrokemijskih lastnosti ... Med praktičnim delom je eksperimentalno delo podprto s sprotnim računanjem in ovrednotenjem rezultatov in njihovo kritično presojo ter posledično

Content (Syllabus outline):

The practical course includes seminar and laboratory exercises. On a theoretical level, students learn the basic procedures and methods for the investigation of material properties. In experimental work they learn the applied use of characterization methods. During the work they can learn microscopic techniques (optical microscope, scanning electron microscope, atomic force microscope, heating microscope...), x-ray photoelectron spectroscopy, thermoanalytical techniques, characterization of particle size distribution, specific surface area, pore distribution, measurement of mechanical, electrochemical properties. During the practical work, the experimental work will be supported by continuous calculation and evaluation of the results and their critical assessment and thus a discussion of the results

diskusijo dobljenih rezultatov in ustreznostjo uporabljenih metod. Delo se izvaja projektno za specifične materiale.	obtained and the appropriateness of the methods used.
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Temeljna literatura in viri/Readings:

<ul style="list-style-type: none"> - Callister, Rethwisch, Materials science and engineering, 9th edition, A John Wiley and Sons Publication, New Jersey, 2015 (20%) - Askeland, Wright, The science and engineering of Materials, 7th edition, Cengage Learning, Boston, USA, 2016 (20%) - Michio Inagaki, Feiyu Kang; Materials Science and Engineering of Carbon: Fundamentals, 2nd Edition, 2014 (10%)

Cilji in kompetence:

<p>Študenti osvojijo ter utrdijo znanje o naprednih metodah preiskovanja lastnosti in kakovosti materialov. Soočeni z realnimi problemi, spoznajo delo, možnosti uporabe in zmogljivosti posamezne metode karakterizacije. Dobljene rezultate so sposobni komentirati in aplikativno prenesti na izbrano tematiko. Pridobljene kompetence vključujejo poleg strokovne usposobljenosti tudi socialne zmožnosti sodelovanja z ostalimi kolegi in konstruktivnega reševanja problemov.</p>	<h3>Objectives and competences:</h3> <p>Students acquire or deepen knowledge of advanced methods for the investigation of properties and quality of materials. Confronted with real problems they get to know the working methods, possible applications and capabilities of the individual characterisation methods. They are able to comment on the results obtained and apply them to the chosen topic. The acquired competences include not only technical skills but also social skills for cooperation with other colleagues and constructive problem solving.</p>
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Predvideni študijski rezultati:

<p>Znanje in razumevanje Študenti pridobijo spretnosti rokovanja in delovanja metod karakterizacije. Seznanijo se s teoretičnimi osnovami metode, pripravo vzorca za karakterizacijo, pogoji meritve, obsegom dobljenih rezultatov in drugimi uporabnimi spoznanji s katerimi krepijo poznavanje tega področja.</p> <p>Uporaba Praktikum spodbuja prenos teoretičnih dejstev v uporabna znanja. Študent razvije občutek za izbor, načrtovanje in uporabo ustreznih metod na konkretnem primeru.</p> <p>Refleksija Znanja, ki so uporabna pri laboratorijskih in seminarskih vajah praktikuma so vezane na vsebine predmetov, ki obravnavajo napredne karakterizacijske metode in osnove vede o materialih.</p> <p>Interdisciplinarnost vsebine pa jih povezuje tudi z osnovnimi znanji s področij fizike in kemije.</p> <p>Prenosljive spretnosti Sposobnost uporabe analitskih tehnik karakterizacije in njihovega vrednotenja na uporabnih materialih. Zmožnost sodelovanja in strokovnega dialoga z raziskovalci na drugih področjih.</p>	<h3>Intended learning outcomes:</h3> <p>Knowledge and Comprehension Students will acquire the ability to handle and operate characterization methods. They will learn the theoretical basis of the method, how to prepare a sample for characterisation, measurement conditions, the range of results obtained and other useful knowledge that will enhance their knowledge in the field.</p> <p>Application The practical course promotes the transfer of theoretical facts into useful knowledge. The student develops a sense of selecting, planning and applying appropriate methods to a specific case.</p> <p>Analysis The knowledge usable in the laboratory and seminar exercises of the practical course is related to the contents of the subjects dealing with advanced characterization methods and the fundamentals of materials science. The interdisciplinary nature of the content also links it to fundamental knowledge from the fields of physics and chemistry.</p> <p>Skill-transference Ability Ability to apply analytical characterisation methods and evaluate them on useful materials. Ability to collaborate and engage in professional dialogue with researchers from other disciplines.</p>
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Metode poučevanja in učenja:

<p>Računski seminarji, laboratorijske vaje,</p>	<h3>Learning and teaching methods:</h3> <ul style="list-style-type: none"> - Computational seminars, - practical course,
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individualna seminarska naloga.

- individual seminar work

Načini ocenjevanja:

Delež/Weight

Assessment:

Ocena pisnega poročila	70,00 %	Written report
Ocene predstavitve sklepnih ugotovitev	30,00 %	The presentation of conclusions

Reference nosilca/Lecturer's references:

- Bobnar, J.; Lozinšek, M.; Kapun, G.; Njel, C.; Dedryvère, R.; **Genorio, B.**; Dominko, R. Fluorinated Reduced Graphene Oxide as a Protective Layer on the Metallic Lithium for Application in the High Energy Batteries. *Sci. Rep.* 2018, 8 (1), 5819. <https://doi.org/10.1038/s41598-018-23991-2>.
- Bobnar, J.; Vizintin, A.; Kapun, G.; Njel, C.; Dedryvère, R.; Dominko, R.; **Genorio, B.** A New Cell Configuration for a More Precise Electrochemical Evaluation of an Artificial Solid-Electrolyte Interphase. *Batter. Supercaps* 2020, 3, 1–10. <https://doi.org/10.1002/batt.202000255>.
- Gorgieva, S.; Osmić, A.; Hribernik, S.; Božič, M.; Svete, J.; Hacker, V.; Wolf, S.; **Genorio, B.** Efficient Chitosan/Nitrogen-Doped Reduced Graphene Oxide Composite Membranes for Direct Alkaline Ethanol Fuel Cells. *Int. J. Mol. Sci.* 2021, 22 (4). <https://doi.org/10.3390/ijms22041740>.

PRAKTIKUM IZ ORGANSKE KEMIJE

UČNI NAČRT PREDMETA/COURSE SYLLABUS

Predmet:	Praktikum iz organske kemije
Course title:	PRACTICAL COURSE IN ORGANIC CHEMISTRY
Članica nosilka/UL Member:	UL FKKT

Študijski programi in stopnja	Študijska smer	Letnik	Semestri	Izbirnost
Kemijska tehnologija, prva stopnja, visokošolski strokovni	Ni členitve (študijski program)	2. letnik	zimski	obvezni

Univerzitetna koda predmeta/University course code:	0072040
Koda učne enote na članici/UL Member course code:	KT120

Predavanja /Lectures	Seminar /Seminar	Vaje /Tutorials	Klinične vaje /Clinical tutorials	Druge oblike študija /Other forms of study	Samostojno delo /Individual student work	ECTS
		75			75	5

Nosilec predmeta/Lecturer: izt. prof. dr. Janez Cerkovnik

Vrsta predmeta/Course type: obvezni/mandatory

Jeziki/Languages:

Predavanja/Lectures:	Slovenščina
Vaje/Tutorial:	Slovenščina

Pogoji za vključitev v delo oz. za opravljanje študijskih obveznosti: **Prerequisites:**

Študent oz. kandidat mora imeti predmet opredeljen kot študijsko obveznost.	The course has to be assigned to the student.
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Vsebina:

Splošno.

- Varnost pri delu. Osebna zaščitna oprema.
- Varovanje sebe, delovnega prostora in okolja.
- Vodenje laboratorijskega dnevnika in pisanje poročil.
- Iskanje informacij v literaturi in v bazah podatkov.

Metode in tehnike.

Vaje (eksperimenti) bodo izbrane na osnovi karakterističnih sinteznih metod. Vaje bodo zajemale pripravo reagentov in kemikalij, sintezo in izolacijo produktov, ter njihovo analitiko in karakterizacijo.

Študentje bodo izvajali:

- merjenje mase in prostornine
- uporaba kemikalij in priprava raztopin,
- topila v laboratoriju (čiščenje, regeneracija, brezvodna topila)
- delo v inertni atmosferi,

Content (Syllabus outline):

General.

- Safety at work. Personal protective equipment. Personal, workspace and environmental protection.
- Lab diary and reports writing.
- Literature and databases information search.

Methods and techniques.

Exercise selection (experiments) will be based on the characteristic synthetic methods. Exercises will include a preparation of reagents and chemicals, synthesis and isolation of the products, followed by their analysis and characterization.

Students will implement:

- mass and volume measurements
- use of chemicals and preparation of solutions
- solvents in the laboratory (cleaning , regeneration , anhydrous)

<ul style="list-style-type: none"> -delo s plini (jeklenke) -kristalizacija -destilacija, - sublimacija, - ekstrakcija, - kromatografske tehnike za karakterizacijo, izolacijo in separacijo spojin - karakterizacija organskih spojin s testnimi reakcijami in z derivatizacijo - določanje fizikalnih lastnosti snovi (tališče, vrelišče itd.) 	<ul style="list-style-type: none"> - work at inert atmosphere - work with gases (cylinders) -crystallization - distillation - sublimation - extraction - chromatographic techniques, isolation and separation of the compounds - characterization of organic compounds with the test reactions and preparation of corresponding derivate - physical properties determination (melting point, boiling point, etc.).
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Temeljna literatura in viri/Readings:

- N. Gros, J. Cerar, F. Kovač, B. Kozlevčar: Praktikum iz kemije, študijsko gradivo, UL FKKT.
- G. J. Shugar, R. A. Shugar, L. Bauman, R. Shugar Bauman: Chemical Technician's Ready Reference Handbook. McGraw-Hill, 1981.
- L. M. Harwood, C. J. Moody: Experimental Organic Chemistry. Blackwell, 1989.

Cilji in kompetence:

Cilji predmeta:

Učna enota se navezuje na predmete Splošna kemija, Organska kemija . Študent z eksperimentalnim delom praktično nadgradi osnovno teoretično znanje kemije in pridobi osnovne veščine, ki so potrebne za eksperimentalno delo v kemijskem laboratoriju.

Predmetno specifične kompetence:

- varno delo v laboratoriju
- priprava in izvedba preprostih in nekaterih srednje zahtevnih eksperimentov
- izvajanje najpogostejših laboratorijskih meritev, temeljnih laboratorijskih operacij in postopkov
- izvajanje standardnih laboratorijskih tehnik za izolacijo in čiščenje spojin
- poznavanje osnov karakterizacije spojin
- dostopanje in uporaba literaturnih virov in baz podatkov

Objectives and competences:

Objectives of the subject:

The course is related to the subjects of general chemistry, organic chemistry. The basic theoretical knowledge of chemistry is upgraded and the basic skills are acquired by the practical experimental work of a student, needed for the experimental work in the chemistry laboratory.

Subject-specific competencies:

- safe work in laboratory
- preparation and implementation of simple and some moderately complex experiments
- implementation of the most common laboratory measurements of basic laboratory operations and procedures
- the application of standard laboratory techniques for the isolation and purification of compounds
- knowledge of basics of compounds characterization
- accessing and applying literature and databases

Predvideni študijski rezultati:

Znanje in razumevanje

- varno delo v laboratoriju
- osnove merjenja in pravilno izvajanje temeljnih laboratorijskih operacij in meritev
- priprava in izvedba pretvorb in eksperimentov
- ločevanje, izolacija, čiščenje in karakterizacija spojin
- pravilno načrtovanje priprave reagentov in raztopin, izbor ustreznih vrst kemikalij in pribora ter izvedba postopka
- dostopanje do literaturnih virov in baz podatkov ter njihova uporaba

Razumevanje:

- osnovni in srednje zahtevni eksperimentalni postopki in pretvorbe v kemiji
- teoretske osnove postopkov za izolacijo, čiščenje in karakterizacijo spojin.

Intended learning outcomes:

Knowledge and Comprehension

- laboratory safety
- fundamentals of measurements, measuring techniques and basic laboratory operations in accordance with good laboratory practice
- preparatory activities, transformations and experiments
- separation, isolation, purification and characterisation of substances
- preparation of solutions and reagents: planning, selecting chemicals of appropriate grade, choice of equipment, procedures
- accessing to the information sources and data bases and their use

Understanding

<p>-osnovna pravila varnega dela v laboratoriju</p> <p>Uporaba Osnovno praktično znanje kemije z razumevanjem povezav med kemijskimi področji je temeljno znanje, ki se uporablja v nadaljnjem študiju kemije in kemijske tehnologije hkrati pa je nujno potrebno vsakemu kemiku/tehnologu pri njegovem kasnejšem delu v praksi.</p> <p>Refleksija Študent bo na osnovi pridobljenega znanja sposoben izvesti preproste in srednje zahtevne meritve, eksperimente in pretvorbe v kemiji. S tem je sposoben preveriti hipoteze v praksi oziroma kritično ovrednotiti rezultate eksperimenta glede na skladnost s teoretičnimi načeli. Študent je seznanjen z dostopom do literaturnih virov in baz podatkov. Študent bo pridobil občutek za optimalno izvedbo osnovnih laboratorijskih postopkov. Študent bo pridobil kritičen odnos do priprave raztopin in se zavedal omejitev pri odmerjanju mase in prostornine ter razlik v razredih kemikalij v laboratoriju.</p> <p>Prenosljive spretnosti</p> <ul style="list-style-type: none"> - Merjenje mase in prostornine, priprava raztopin, -Pravilno izvajanje temeljnih laboratorijskih operacij -Dostopanje do literaturnih virov -Zbiranje, merjenje, analiza, razlaga in kritično vrednotenje podatkov -Identifikacija in reševanje problemov in izzivov. -Poročanje. -Kritična analiza, sinteza. 	<p>- basic experimental procedures and transformations in chemistry</p> <p>- understanding the fundamentals of procedures of isolation, purification and characterisation of substances at the basic and intermediate level</p> <p>- performance in accordance with safety precautions</p> <p>Application Fundamental chemical skills and understanding of the connections and relations between different chemical disciplines support study of prospect chemical and technological modules and are essential in professional practice.</p> <p>Analysis Student develops ability of performing measurements, experiments and transformations in chemistry at the basic and intermediate level. Student acquires skills for checking hypotheses experimentally and evaluating the conformity of the outcomes and results of experiments with theoretical expectations. Student is capable of using sources of information.</p> <p>Student develops sensitivity for adequate performance of laboratory operations. Student will be aware of the safety precautions for handling gasses. Student will develop critical attitude towards preparation of solutions and will be aware of the limitations in measuring mass and volume. Student will be aware of the distinctions between chemicals of different grades in a laboratory.</p> <p>Skill-transference Ability</p> <ul style="list-style-type: none"> - Measurement of mass and volume and preparation of solutions. - Performance of laboratory operations in accordance with good laboratory practice. - Ability of using information sources. - Ability of finding information, obtaining data by measurements, analysing, evaluating and interpreting data. - Ability of recognising and defining problems and challenges. - Reporting. - Critical analysis and synthesis of information.
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<p>Metode poučevanja in učenja: Predavanja, laboratorijske vaje.</p>	<p>Learning and teaching methods: Lectures, laboratory exercises.</p>
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<p>Načini ocenjevanja: Opravljene vaje so pogoj za pristop k izpitu. Pisni izpit, ki ga lahko nadomestita dva (sprotna) pregledna testa. Ocene: pozitivno 6-10.</p>	<p>Delež/Weight</p>	<p>Assessment: Laboratory coursework is a prerequisite for the exam. Final written exam (or two partial written tests). Positive grades 6-10</p>
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<p>Reference nosilca/Lecturer's references: 1. STRLE, Gregor, CERKOVNIK, Janez. A simple and efficient preparation of high-purity hydrogen trioxide (HOOOH). <i>Angewandte Chemie</i>, ISSN 1433-7851. [Print ed.], 2015, vol. 54, no. 34, str. 9917-9920, ilustr.</p>
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<http://onlinelibrary.wiley.com/doi/10.1002/anie.201504084/abstract>, doi: 10.1002/anie.201504084.

[COBISS.SI-ID 1536385475]

2. TUTTLE, Tell, CERKOVNIK, Janez, KOLLER, Jože, PLESNIČAR, Božo. The search for protonated dihydrogen trioxide (HOOOH) : insights from theory and experiment. *The journal of physical chemistry. A, Molecules, spectroscopy, kinetics, environment, & general theory*, ISSN 1089-5639, 2010, vol. 114, no. 30, str. 8003-8008, doi: 10.1021/jp103882e. [COBISS.SI-ID 34295813]

3. CERKOVNIK, Janez, PLESNIČAR, Božo. Recent advances in the chemistry of hydrogen trioxide (HOOOH). *Chemical reviews*, ISSN 0009-2665. [Print ed.], 2013, vol. 113, no. 10, str. 7930-7951, ilustr.

<http://pubs.acs.org/doi/ipdf/10.1021/cr300512s>, doi: 10.1021/cr300512s. [COBISS.SI-ID 1615407]

ULJEVIČ

PRAKTIKUM IZ OSNOV KEMIJSKEGA INŽENIRSTVA

UČNI NAČRT PREDMETA/COURSE SYLLABUS

Predmet:	Praktikum iz osnov kemijskega inženirstva
Course title:	Practical course in chemical engineering fundamentals
Članica nosilka/UL Member:	UL FKKT

Študijski programi in stopnja	Študijska smer	Letnik	Semestri	Izbirnost
Kemijska tehnologija, prva stopnja, visokošolski strokovni	Ni členitve (študijski program)	3. letnik	zimski	obvezni

Univerzitetna koda predmeta/University course code:	0072054
Koda učne enote na članici/UL Member course code:	KT138

Predavanja /Lectures	Seminar /Seminar	Vaje /Tutorials	Klinične vaje /Clinical tutorials	Druge oblike študija /Other forms of study	Samostojno delo /Individual student work	ECTS
		75			75	5

Nosilec predmeta/Lecturer: izt. prof. dr. Aleš Ručigaj

Vrsta predmeta/Course type: obvezni/mandatory

Jeziki/Languages:

Predavanja/Lectures:	Slovenščina
Vaje/Tutorial:	Slovenščina

Pogoji za vključitev v delo oz. za opravljanje študijskih obveznosti:

Študent oz. kandidat mora imeti predmet opredeljen kot študijsko obveznost.

Prerequisites:

The course has to be assigned to the student.

Vsebina:

Praktikum vključuje eksperimentalno in seminarsko delo, katerega teoretične osnove sodijo na naslednja temeljna področja kemijskega inženirstva:

- fluidna mehanika (mešanje);
- prenos toplote in snovi;
- kemijsko reakcijsko inženirstvo (vzbujevalno odzivna tehnika),

izbrane osnovne operacije (filtracija, destilacija, sušenje)

Content (Syllabus outline):

Practical course consists of experimental work and seminars with a theoretical background of the following basic fields of chemical engineering:

- fluid mechanics (mixing),
 - heat and mass transfer,
 - chemical reaction engineering (stimulus response technique),
- selected unit operations (filtration, distillation, drying).

Temeljna literatura in viri/Readings:

- A. Lakota, M. Linec, M. Lubej, A. Pavko, *Praktikum iz kemijskega inženirstva*, UL FKKT, Ljubljana, 2016, (80 %)

- A. Ručigaj, R. Ambrožič, M. Krajnc, Zbirka nalog: Praktikum iz osnov kemijskega inženirstva, 1. spletna izdaja, UL FKKT, Ljubljana, 2020, 119 strani (90 %)

Dopolnilna literatura / Additional literature:

Literatura, ki je navedena pri predmetih Osnove kemijskega inženirstva in Osnovne operacije v kemijskem inženirstvu.

Literature cited for the courses Chemical engineering fundamentals and Unit operations in chemical engineering

Cilji in kompetence:

Cilj predmeta je, da študentje s pomočjo laboratorijskega praktičnega dela dopolnijo predstavo in znanje o fazah proizvodnega procesa ter o vlogi kemijskega inženirstva pri tem. Tako spoznajo osnovne principe te temeljne tehniške discipline.

Študentje pri predmetu pridobijo naslednje specifične kompetence:

- vodenje kemijsko inženirskih procesov v laboratorijskem merilu;
- uporaba laboratorijskih naprav, instrumentalnih metod in programskih paketov;
- uporaba znanj fluidne mehanike;
- uporaba znanj prenosa toplote in snovi;
- uporaba osnovnih znanj kemijskega reakcijskega inženirstva;
- uporaba osnovnih znanj separacijskih procesov.

Objectives and competences:

The aim of the course is that students implement the view and theoretical knowledge about the stages of a production process and the role of chemical engineering through the laboratory experimental work. This way they become familiar with the basic principles of this fundamental technical science.

Students gain the following specific competences:

- to control chemical engineering processes at laboratory scale,
 - to use laboratory equipment, instrumentation methods and computer software;
 - to use the knowledge of fluid mechanics;
 - to use the knowledge of heat and mass transfer;
 - to use the basic knowledge chemical reaction engineering;
- to use the knowledge separation processes.

Predvideni študijski rezultati:**Znanje in razumevanje**

Študent je sposoben prepoznati vlogo in razumeti pomen poznavanja osnovnih kemijsko inženirskih znanj za načrtovanje in vodenje specifičnih kemijsko tehnoloških procesov. Študent zna identificirati ključne dejavnike pri razvoju kemijsko inženirskih procesov, produktov in opreme. Spozna osnove izvedbe in kontrole procesov v laboratorijskem merilu. Zna uporabljati instrumentalne metode in ostale analitske tehnike za nadzor procesa in karakterizacijo produkta. Zna eksperimentalne rezultate kvantitativno obravnavati v skladu z literaturo in s pomočjo sodobne programske opreme.

Uporaba

Pridobljena praktična znanja in znanje interpretiranja ter obdelovanja eksperimentalnih podatkov je sposoben uporabiti pri delu na različnih področjih kemijskega inženirstva.

Refleksija

Laboratorijske vaje se neposredno navezujejo na vsebine predmetov Osnove kemijskega inženirstva in Osnovne operacije v kemijskem inženirstvu.

Študent je sposoben uporabiti pridobljeno znanje pri reševanju praktičnih problemov iz različnih področij kemijskega inženirstva. Sposoben je povezovanja praktičnih rezultatov s teoretičnimi osnovami.

Prenosljive spretnosti

Razvita sposobnost uporabe teoretičnega znanja pri izvedbi laboratorijskih eksperimentov. Razvita sposobnost opravljanja laboratorijskih poskusov,

Intended learning outcomes:**Knowledge and Comprehension**

Student is able to recognize the role and to understand the significance of chemical engineering knowledge for specific chemical process design and control. Student knows to identify essential factors for the development of processes, products and equipment in chemical engineering. Student learns basics of process realization and control in a laboratory scale. He can use instrumental methods and other analytical techniques for the process control and product characterization. Student can qualitatively treat the results in accordance with the literature and by using contemporary computer software.

Application

Student is able to use gained practical skills and knowledge of interpreting and handling the experimental data in the various fields of chemical engineering.

Analysis

Laboratory exercises are in direct connection with the courses Chemical engineering fundamentals and Unit operations in chemical engineering.

Student is capable to use the acquired knowledge during solving of practical problems from the various fields of chemical engineering.

Student is capable to link practical results with theoretical background.

Skill-transference Ability

Developed capability of using theoretical knowledge at laboratory experiments.

vrednotenja in interpretiranja eksperimentalnih rezultatov z uporabo sodobne programske opreme na osnovi osvojenih teoretičnih znanj. Razvita sposobnost razmišljanja in logičnega sklepanja. Sposobnost zbiranja podatkov in predstavitve rezultatov. Spozna vrednosti skupinskega dela.	Developed capability of performing laboratory experiments, evaluation and interpretation of experimental results by using the contemporary computer software on the basis of accomplished theoretical knowledge. Developed capability of thinking and logical concluding. Capability of data collection and result presentation. Student learns the value of a team work.
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Metode poučevanja in učenja:

Eksperimentalno delo ki vključuje laboratorijske vaje in seminarske vaje.

Learning and teaching methods:

Experimental work including laboratory exercises and seminars.

Načini ocenjevanja:

Delež/Weight

Assessment:

Pisni in ustni izpit.	80,00 %	Written and oral exam.
Seminarska naloga (pogoj za pristop h kolokviju).	20,00 %	Written seminar (mandatory before exam).

Reference nosilca/Lecturer's references:

RUČIGAJ, Aleš, ŠTIRN, Žiga, ŠEBENIK, Urška, KRAJNC, Matjaž. Main-chain benzoxazine oligomers : effects of molecular weight on the thermal, mechanical, and viscoelastic properties. *Journal of applied polymer science*, ISSN 0021-8995. [Print ed.], 2018, vol. 135, iss. 35, str. 1-11.

RUČIGAJ, Aleš, KRAJNC, Matjaž. Kinetic modeling of a crude DERA lysate-catalyzed process in synthesis of statin intermediates. *Chemical engineering journal*, ISSN 1385-8947. [Print ed.], 2015, vol 259, no. 1, str. 11-24.

RUČIGAJ, Aleš, KRAJNC, Matjaž. Optimization of a crude DERA lysate-catalyzed process in synthesis of statin intermediates. *Organic process research & development*, ISSN 1083-6160, 2013, vol. 17, no. 5, str. 854-862.

PRAKTIKUM IZ SPLOŠNE KEMIJE

UČNI NAČRT PREDMETA/COURSE SYLLABUS

Predmet:	Praktikum iz splošne kemije
Course title:	Practical course in general chemistry
Članica nosilka/UL Member:	UL FKKT

Študijski programi in stopnja	Študijska smer	Letnik	Semestri	Izbirnost
Kemijska tehnologija, prva stopnja, visokošolski strokovni	Ni členitve (študijski program)	1. letnik	zimski	obvezni

Univerzitetna koda predmeta/University course code:	0072025
Koda učne enote na članici/UL Member course code:	KT106S

Predavanja /Lectures	Seminar /Seminar	Vaje /Tutorials	Klinične vaje /Clinical tutorials	Druge oblike študija /Other forms of study	Samostojno delo /Individual student work	ECTS
		15 SV + 60 LV			75	5

Nosilec predmeta/Lecturer: doc. dr. Nives Kitanovski

Vrsta predmeta/Course type: obvezni/mandatory

Jeziki/Languages:

Predavanja/Lectures:	Slovenščina
Vaje/Tutorial:	Slovenščina

Pogoji za vključitev v delo oz. za opravljanje študijskih obveznosti:

Študent oz. kandidat mora imeti predmet opredeljen kot študijsko obveznost.

Prerequisites:

The course has to be assigned to the student.

Vsebina:

Pri računskih vajah študenti utrjujejo poznavanje kvantitativnega obravnavanja snovi in snovnih sprememb z reševanjem računskih nalog in kvantitativnih kemijskih problemov s poudarkom na stehiometriji. Obseg in nivo pričakovanih veščin in znanja sta določena s predpisanim učbenikom (skripta).

Program laboratorijskih vaj se izvede v obliki eksperimentalnih vaj z vsebinami:

- osnove varnega dela v laboratoriju
- osnovna laboratorijska steklovina in oprema
- odmerjanje trdnih in tekočih reagentov (tehtanje, pipetiranje, itd.)
- dodajanje trdnih in tekočih reagentov
- delo pri sobni temperaturi (npr. mešanje)
- delo pri povišani temperaturi (segrevanje)

Content (Syllabus outline):

In numerical exercises students strengthen knowing of the quantitative approach of the substance and material changes with solving numerical tasks and quantitative chemistry problems, with focus on the stoichiometry. The scope and level of expected skills and knowledge are determined by the prescribed textbook (script).

Program of laboratory work is conducted in the form of laboratory work content:

- The basics of safe work in the laboratory
- Basic laboratory glassware and equipment
- Dosing of solid and liquid reagents (mass measurements, pipetting, etc.).
- Adding solid and liquid reagents
- Working at room temperature (eg, mixing)
- Work at elevated temperature (heating)

<ul style="list-style-type: none"> - delo pri nižani temperaturi (hlajenje) - kristalizacija - navadna destilacija <p>(izbor vaj se lahko spreminja) S seminarскими vajami se poudari povezanost praktičnega eksperimentalnega dela s teorijo.</p>	<ul style="list-style-type: none"> - Work at reduced temperature (cooling) - crystallization - basic distillation <p>(selection of exercises may vary) The seminar exercises emphasize the integration of practical experimental work with theory.</p>
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Temeljna literatura in viri/Readings:

- P. Šegedin, Osnove kemijskega računanja z zbirko nalog, UL Biotehniška fakulteta - Oddelek za lesarstvo, Ljubljana 1996, 149 strani.
- B. Kozlevčar, N. Kitanovski, P. Šegedin, Navodila za vaje iz splošne kemije, študijsko gradivo, UL FKKT, 2013, 27 strani.

Dodatna literatura / additional readings:

- N. Bukovec, J. Brenčič, Kemija za gimnazije 1, srednješolski učbenik, DZS, Ljubljana, 2006, 160 strani.
- J. Brenčič, F. Lazarini, Splošna in anorganska kemija, UL FKKT, Ljubljana 2004, strani 1-239.
- P. Šegedin, Zbirka izpitnih nalog iz kemije, UL Biotehniška fakulteta - Oddelek za lesarstvo, Ljubljana 1995, 65 strani.

Cilji in kompetence:

Študenti spoznajo osnove kvantitativnega obravnavanja snovi in snovnih sprememb (osnove kemijskega računanja s poudarkom na stehiometriji). Pri laboratorijskih vajah je osnovni cilj seznaniti študente z osnovnimi prijemi za delo v laboratoriju. Ob tem je poseben poudarek na osebni varnosti ter varnosti delovnega prostora in okolja.

Objectives and competences:

Students learn about the basics of quantitative treatment of materials and material changes (basic chemical calculation with focus on stoichiometry). The basic aim of laboratory exercises is to pair students with the basic approaches to laboratory work. It has a special emphasis on personal safety, safety of the working space and the environment.

Predvideni študijski rezultati:

Znanje in razumevanje
Študenti pri računskih vajah predvsem spoznajo pomen poznavanja in razumevanja osnovnih kemijskih pojmov in zakonitosti pri reševanju nalog in kvantitativnih izzivov s področja kemijskega računanja. Pri laboratorijskih vajah pridobijo osnovno praktično znanje varnega dela v kemijskem laboratoriju in spoznajo osnovna navodila o varnosti in prvi pomoči pri laboratorijskem delu.

Uporaba
Študenti se seznanijo s postopki in pristopi pri reševanju računskih nalog in problemov in jih znajo uporabiti pri njihovem reševanju. Pri izvedbi osnovnih eksperimentalnih (laboratorijskih) vaj znajo izbrati ustrezen postopek in ga na pravilen način uporabiti (izvesti).

Refleksija
Študenti so sposobni kritično ovrednotiti izvedene meritve in oceniti dobljene rezultate. Teoretične naloge in kvantitativne probleme so sposobni povezati z eksperimentalnimi nalogami in problemi, s katerimi se srečajo pri laboratorijskih vajah in se tako naučiti povezovanja in razumevanja teorije in prakse.

Prenosljive spretnosti
Pri predmetu bodo študenti pridobili laboratorijske spretnosti in se urili v reševanju kvantitativnih teoretičnih in eksperimentalnih nalog in problemov

Intended learning outcomes:

Knowledge and Comprehension
At numerical exercises students learn especially about the importance of knowing and understanding of basic chemical concepts and principles in problem solving and quantitative challenges in the field of chemical calculations. During laboratory exercises they acquire the basic practical knowledge of safe work in a chemistry lab and learn the basic instructions on safety and first aid at laboratory work.

Application
Students meet with the procedures and approach for solving numerical tasks and problems, and are able to use to tackle them. Students are able to choose the appropriate experimental procedure and use (performed) it in correct way.

Analysis
Students are able to critically evaluate the performed measurements and assess the obtained results. They are able to connect theoretical challenges and quantitative problems with the experimental tasks and problems that may arise at laboratory work, thus also learn the integration and understanding of theory and practice.

Skill-transference Ability
In this course, students gain laboratory skills and are trained in solving quantitative theoretical and experimental tasks and problems, and may learn

ter lahko pridobljene izkušnje in znanje koristno uporabil tudi pri vseh drugih kemijskih predmetih. Varnost pri delu je pomembna vrlina ne le v laboratoriju, temveč pri kakršnikoli dejavnosti.	experience and knowledge usefully applied in all other chemistry subjects. Safety at work is an important virtue not only in the laboratory but also in any activities.
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Metode poučevanja in učenja: Seminarske vaje in praktične vaje v laboratoriju. Pri seminarskih vajah poteka reševanje računskih stehiometričnih nalog in kvantitativnih kemijskih izzivov s poudarkom na skupinskem delu (sodelovalno učenje). Laboratorijske vaje so zasnovane na delu posameznika (izkustveno učenje), skupinska analiza meritev in kvalitativnih eksperimentov ter njihova razlaga s primerno teorijo.	Learning and teaching methods: Seminars and practical exercises in the laboratory. At seminars, the stoichiometric calculation problems are solving along with the quantitative chemical challenges, emphasising teamwork (collaborative learning). Laboratory exercises are based on the work of the individual (experiential learning), group analysis of measurements and qualitative experiments with their interpretation by the relevant theory.
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Načini ocenjevanja:	Delež/Weight	Assessment:
Ustni zagovor laboratorijskih vaj je pogoj za opravljanje pisnega izpita.		
Pisni izpit (lahko ga nadomestita dva pozitivno ocenjena kolokvija).		

Reference nosilca/Lecturer's references:
- KITANOVSKI, Nives , BORSAN, Nataša, KASUNIČ, Marta, FRANCETIČ, Vojmir, POPOVIČ, Jasminka, DJERDJ, Igor, ROCQUEFELTE, Xavier, REEDIJK, Jan, KOZLEVČAR, Bojan. Chromium coordination compounds with bis(3,5-dimethylpyrazol-1-yl)acetic acid or its anion. Polyhedron, 2014,70, 119-124, [COBISS.SI-ID 1666351]
- KITANOVSKI, Nives , GOLOBIČ, Amalija, ČEH, Boris. Synthesis and structural characterization of mono- and dinuclear Mo(V)-oxo-complexes containing bis(3,5-dimethylpyrazol-1-yl)acetate anion as ligand. Inorg. Chem. Commun., 2011, 14, 276-280. [COBISS.SI-ID 34680069]
- KITANOVSKI, Nives , PEVEC, Andrej, KOZLEVČAR, Bojan. Copper(II) coordination compounds with ferulic acid. Polyhedron, 2009, 28, 3642-3646, [COBISS.SI-ID 33287429]

PROCESI V INDUSTRIJSKI KEMIJI

UČNI NAČRT PREDMETA/COURSE SYLLABUS

Predmet:	Procesi v industrijski kemiji
Course title:	Processes of industrial chemistry
Članica nosilka/UL Member:	UL FKKT

Študijski programi in stopnja	Študijska smer	Letnik	Semestri	Izbirnost
Kemijska tehnologija, prva stopnja, visokošolski strokovni	Ni členitve (študijski program)	2. letnik	zimski	obvezni

Univerzitetna koda predmeta/University course code:	0072041
Koda učne enote na članici/UL Member course code:	KT118

Predavanja /Lectures	Seminar /Seminar	Vaje /Tutorials	Klinične vaje /Clinical tutorials	Druge oblike študija /Other forms of study	Samostojno delo /Individual student work	ECTS
45	30				75	5

Nosilec predmeta/Lecturer: izr. prof. dr. Boštjan Genorio, prof. dr. Urška Šebenik

Vrsta predmeta/Course type: obvezni/mandatory

Jeziki/Languages:

Predavanja/Lectures:	Slovenščina
Vaje/Tutorial:	Slovenščina

Pogoji za vključitev v delo oz. za opravljanje študijskih obveznosti: **Prerequisites:**

Študent oz. kandidat mora imeti predmet opredeljen kot študijsko obveznost.	The course has to be assigned to the student.
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Vsebina:

Katalitski postopki: vloga in delovanje katalizatorja v tehnološkem procesu, homogena in heterogena kataliza, aktivnost in druge lastnosti katalizatorja, katalizatorski strupi in deaktivacija katalizatorjev.

Razklop mineralnih surovin: Razklop mineralov, precipitacija in kristalizacija, prenasičenje, nukleacija in vpliv na lastnosti produktov kristalizacije, rast in zorenje kristalov.

Industrijski elektrokemijski procesi: Osnove elektrokemijskih procesov, prenapetost, celični separatorji in druge značilnosti elektrokemijskih reaktorjev, industrijska elektroliza in galvanotehnika.

Visokotemperaturni procesi: elektrotermični in drugi termični procesi, peči, metalurški procesi.

Content (Syllabus outline):

Catalytic processes: Role and function of catalyst in technological process; Homogeneous and heterogeneous catalysis; Activity and other properties of the catalyst; Catalyst deactivation and catalyst poisons.

Decompositions of mineral raw materials: Decompositions of minerals; Precipitation and crystallization, supersaturation, nucleation and impacts on crystalline product properties; Growth and aging of the crystals.

Industrial electrochemical processes: Fundamentals of electrochemical processes; Over voltage, current, and energy efficiency; Cell separators and other characteristics of electrochemical reactors; industrial electrolysis and electroplating.

<p>Tehnični plini: industrijski tehnični plini, osnove kriotehnike, proizvodnja, shranjevanje in distribucija plinov.</p> <p>Predelava fosilnih snovi: katalitski in nekatalitski procesi, reakcijski mehanizmi, proizvodi in čiščenje proizvodov; kemikalije iz sinteznega plina; tehnologija reakcij v plinastem stanju.</p> <p>Polimerizacijski procesi: reakcijski mehanizmi; primeri tehnologij.</p> <p>Optimizacija proizvodnih procesov: Primer optimizirane proizvodnje sladkorja.</p> <p>Kvantitativna obravnava enostavnih reakcijskih in separacijskih procesov: totalna in komponentna integralna snovna bilanca za kontrolni volumen, totalna in komponentna diferencialna snovna bilanca za kontrolni volumen; računski primeri.</p>	<p>High temperature processes: Electrothermal and other thermal processes; Furnaces; Metallurgical processes.</p> <p>Industrial gases: Industrial technical gases; principles of cryogenics; Production, storage and distribution.</p> <p>Processing of fossil raw materials: Catalytic and noncatalytic processes; Reaction mechanisms; Products and their purification; Chemicals from the synthesis gas; Technology of reactions in a gaseous phase.</p> <p>Polymerization processes: Reaction mechanisms; Examples of polymer technology.</p> <p>Optimization of production processes: Sugar production example.</p> <p>Quantitative description of simple reaction and separation processes: Total and component integral material balance for control volume; Total and component differential material balance for control volume; problems.</p>
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Temeljna literatura in viri/Readings:

- J. A. Moulijn, M. Makkee, A. van Diepen, Chemical Process Technology, Wiley, Chichester, 2001, 420 str. (40 %).
 - R. M. Felder, R. W. Rousseau, Elementary principles of chemical processes, 3. izdaja, John Wiley and Sons, New York, 2000, 606 str. (40 %)
- Dopolnilna literatura:
- Austin: Shreeves Chemical Process Industries, 6. izdaja, Mc Graw-Hill, New York, 1995, 830 str. (20 %).

Cilji in kompetence:

Slušatelji se v okviru predavanj in seminarja seznanijo z različnimi nivoji in osnovnimi značilnostmi kemijske industrijske proizvodnje, z osnovami kemijskih industrijskih procesov ter njihovimi infrastrukturnimi pogoji, možnostmi za uspešno in varno vodenje procesov.

Objectives and competences:

Students become familiar with: different levels and basic characteristics of industrial chemical production, fundamentals of chemical industrial processes and their infrastructure requirements, possibilities for successful and safe processes management.

Predvideni študijski rezultati:

Znanje in razumevanje
Študentje se naučijo analizirati industrijske procese. Spoznajo osnovne značilnosti industrijskih procesov in potrebne pogoje za sodobno industrijsko proizvodnjo.

Uporaba
Pri analizi postavitve procesov, njihovega delovanja in problemov, ki se pojavljajo v industrijskih procesih.

Refleksija
Razvija sposobnost ovrednotenja in analize kompleksnih sistemov kakršni so v kemijski in procesnih industrijah vendar lahko pridobljena znanja uporabi tudi širše.

Prenosljive spretnosti
Teoretične principe dodatno spoznava pri njihovi implementaciji v industrijsko merilo in prakso ter

Intended learning outcomes:

Knowledge and Comprehension
Students learn how to analyze industrial processes, about basic features of industrial processes and necessary conditions for modern industrial production.

Application
In analyses of process setup, process operation, and problem solving in industrial processes.

Analysis
Students develop the ability of evaluation and analysis of complex systems occurring in chemical and process industries. The acquired knowledge is widely applicable.

Skill-transference Ability
Students recognise theoretical principles in industrial scale implementations. They are able of critical evaluation and comparison between theoretical principles and practice.

kritično vrednoti skladnost med teoretičnimi načeli in praktičnim ravnanjem.	
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Metode poučevanja in učenja:	Learning and teaching methods:
Predavanja in seminar	Lectures and seminar

Načini ocenjevanja:	Delež/Weight	Assessment:
Pisni in ustni izpit. Možnost opravljanje pisnega izpita s kolokviji. Izpit je sestavljen iz ločenega preverjanja posameznih vsebin.		Written and oral exam. Written exam may be accomplished by written tests. The exam is consisted of partial assessments of the contents.

Reference nosilca/Lecturer's references:

- MOHORIČ, Ines, **ŠEBENIK, Urška**. Semibatch anionic ring-opening polymerization of octamethylcyclotetrasiloxane in emulsion. *Polymer*, ISSN 0032-3861. [Print ed.], 2011, vol. 52, no. 20, str. 4423-4428. [COBISS.SI-ID 35309317]
- **ŠEBENIK, Urška**, GOLOB, Janvit, KRAJNC, Matjaž. Comparison of properties of acrylic-polyurethane hybrid emulsions prepared by batch and semibatch processes with monomer emulsion feed. *Polymer international*, ISSN 0959-8103, 2003, vol. 52, no. 5, str. 740-748. [COBISS.SI-ID 24954117]
- RUČIGAJ, Aleš, ALIČ, Branko, KRAJNC, Matjaž, **ŠEBENIK, Urška**. Investigation of cure kinetics in a system with reactant evaporation : epoxidized soybean oil and maleic anhydride case study. *European Polymer Journal*, ISSN 0014-3057. [Print ed.], 2014, vol. 52, no. 1, str. 105-116. [COBISS.SI-ID 1667887]
- **Genorio B**, Strmcnik D, Subbaraman R, Tripkovic D, Karapetrov G, Stamenkovic V R, Pejovnik S and Marković N M 2010 Selective catalysts for the hydrogen oxidation and oxygen reduction reactions by patterning of platinum with calix [4] arene molecules *Nat. Mater.* 9 998–1003
- **Genorio B**, Lu W, Dimiev A M, Zhu Y, Raji A-R O, Novosel B, Alemany L B and Tour J M 2012 In Situ Intercalation Replacement and Selective Functionalization of Graphene Nanoribbon Stacks *ACS Nano* 6 4231–40
- Xiang C, Behabtu N, Liu Y, Chae H G, Young C C, **Genorio B**, Tsentalovich D E, Zhang C, Kosynkin D V, Lomeda J R, Hwang C-C, Kumar S, Pasquali M and Tour J M 2013 Graphene Nanoribbons as an Advanced Precursor for Making Carbon Fiber *ACS Nano* 7 1628–37
- Xiang C, Cox P J, Kukovecz A, **Genorio B**, Hashim D P, Yan Z, Peng Z, Hwang C-C, Ruan G, Samuel E L G, Sudeep P M, Konya Z, Vajtai R, Ajayan P M and Tour J M 2013 Functionalized Low Defect Graphene Nanoribbons and Polyurethane Composite Film for Improved Gas Barrier and Mechanical Performances. *ACS Nano* 7 10380–6
- Raji A-R O, Varadhachary T, Nan K, Wang T, Lin J, Ji Y, **Genorio B**, Zhu Y, Kittrell C and Tour J M 2016 Composites of Graphene Nanoribbon Stacks and Epoxy for Joule Heating and Deicing of Surfaces *ACS Appl. Mater. Interfaces* 8 3551–6

SINTEZNE METODE V ANORGANSKI KEMIJI

UČNI NAČRT PREDMETA/COURSE SYLLABUS

Predmet:	Sintezne metode v anorganski kemiji
Course title:	Methods of Synthesis in Inorganic Chemistry
Članica nosilka/UL Member:	UL FKKT

Študijski programi in stopnja	Študijska smer	Letnik	Semestri	Izbirnost
Kemijska tehnologija, prva stopnja, visokošolski strokovni	Ni členitve (študijski program)	2. letnik, 3. letnik		izbirni

Univerzitetna koda predmeta/University course code:	0100213
Koda učne enote na članici/UL Member course code:	KTSI2

Predavanja /Lectures	Seminar /Seminar	Vaje /Tutorials	Klinične vaje /Clinical tutorials	Druge oblike študija /Other forms of study	Samostojno delo /Individual student work	ECTS
	15	60 LV			75	5

Nosilec predmeta/Lecturer: doc. dr. Andrej Pevec

Vrsta predmeta/Course type: izbirni strokovni/Elective Professional

Jeziki/Languages:

Predavanja/Lectures:	Slovenščina
Vaje/Tutorial:	Slovenščina

Pogoji za vključitev v delo oz. za opravljanje študijskih obveznosti: **Prerequisites:**

Študent oz. kandidat mora imeti predmet opredeljen kot študijsko obveznost.	The course has to be assigned to the student.
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Vsebina:

Vsebina seminarjev in vaj: Študenti bodo pri predmetu sintetizirali anorganske snovi z različnimi zahtevnejšimi sintezniimi tehnikami in dobljene snovi preiskali. Pripravili bodo nekatere kovinske acetylacetonate, fluoridooksovanadate(IV), sintetizirali kalikev tris(oksalato)ferat(III) trihidrat, heksaaminkobaltov(III) klorid in silikon. Spoznali se bodo tudi s pripravo TiO₂ po sol-gel metodi. Z reakcijo v trdnem stanju bodo pripravili superprevodne spojine. Pri pripravi spojine bakrovega klorida z dimetoksietanom pa se bodo naučili dela z inertno atmosfero in vakuumom. Kot primer uporabe UV-Vis spektroskopije za določevanje kemijske kinetike bo namenjena vaja Hitrost reakcije med Cr(III) in EDTA. Spojine, ki jih

Content (Syllabus outline):

Seminars and exercises: The students synthesize inorganic compounds with various advanced synthetic techniques, and characterize the products. They prepare some metal acetylacetonates, fluoridooksovanadates(IV), potassium tris(oxalato)ferrate(III) trihydrate, hexamincobalt(III) chloride and silicone. They also learn about the preparation of TiO₂ by sol-gel method. They prepare superconducting compounds by the reaction in the solid state. In the preparation of the complex copper chloride with dimethoxyethane they learn how to work with an inert atmosphere and a vacuum. The reaction between Cr(III) and EDTA is investigated by the use of UV-Vis spectroscopy for the determination of the chemical kinetics. The compounds prepared by the students are

bodo študentje pripravljali bodo karakterizirali z razpoložljivimi metodami, kot so IR spektroskopija, UV-Vis spektroskopija, magnetne meritve, termična analiza, rentgenska praškovna analiza. Rezultate bodo zbrali in ovrednotili v poročilu. V projekte bomo kasneje zajeli nove snovi, ki jih uporablja pri tekočem raziskovalnem delu nosilec predmeta ali njegovi sodelavci. Pri seminarju bodo študenti dobili primerno teoretsko osnovo in navodila za vaje.	characterized using the available methods, such as IR spectroscopy, UV - Vis spectroscopy, magnetic measurements, thermal analysis and X-ray powder analysis. The results are collected and evaluated in the report. The new compounds will be included in the project, especially those used by lecturer or his coworkers in the current research work. During the seminar, students get adequate theoretical basis and guidelines for practice.
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Temeljna literatura in viri/Readings:

- S. Petriček, F. Perdih in A. Demšar, Vaje iz anorganske kemije, Založba FKKT UL, Ljubljana, 2012, strani 75 – 119.

Cilji in kompetence:

Cilj predmeta je študentom omogočiti poglobitev znanja metod sintez in preiskav anorganskih spojin kot nadgraditev predmeta Anorganska kemija II. Specifično, študenti pri projektno zasnovanemu predmetu spoznajo pristop k projektu od iskanja podatkov po bazah, preko laboratorijske sinteze in preiskav produktov do ovrednotenja in predstavitve rezultatov.

Objectives and competences:

The aim of this course for students is to improve knowledge about synthetic methods and characterization of inorganic compounds to upgrade course Inorganic Chemistry II. In project-based approach of this course students learn more about search data bases, laboratory synthesis techniques, characterization of the products and evaluation and presentation of results.

Predvideni študijski rezultati:

Znanje in razumevanje
Predmet predstavlja poglobitev znanja predmeta Anorganska kemija II z zahtevnejšimi laboratorijskimi tehnikami in daje študentu celovit pogled na reševanje kemijskega problema.
Uporaba
Študent spozna zahtevnejše metode sinteze spojin in določitve njihovih lastnosti. Dobi občutek za samostojno organizacijo in izvedbo dela v laboratoriju. Rezultate svojega dela je sposoben predstaviti in razložiti.
Refleksija
S spoznavanjem zahtevnejšega eksperimentalnega dela študenti spoznajo, da imajo zahtevni teoretski principi svojo praktično uporabo.
Prenosljive spretnosti
Laboratorijske veščine, izkušnje in prijemi pri načrtovanje sintez so pomembni pri drugih kemijskih predmetih in pri osebnemu strokovnemu razvoju.

Intended learning outcomes:

Knowledge and Comprehension
This course represents an advanced knowledge of the course Inorganic Chemistry II with advanced laboratory techniques and gives students a comprehensive view of solving chemical problems.
Application
Students learn advanced methods of synthesizing compounds and determining their properties. They get a sense of self-organizing and carrying out work in the laboratory. Results of their work are able to present and explain.
Analysis
By learning of advanced experimental work the students find that complex theoretical principles have their practical application.
Skill-transference Ability
Laboratory skills, experience and approaches to design synthesis are important in other chemistry courses and personal professional development.

Metode poučevanja in učenja:

Predmet se izvaja v obliki seminarjev in laboratorijskih vaj.

Learning and teaching methods:

The course takes the form of seminars and laboratory exercises.

Načini ocenjevanja:

Pisni izpit po uspešno opravljenem praktičnem delu.

Delež/Weight

Assessment:

Written exam after successful completion of practical work.

Reference nosilca/Lecturer's references:

- **Andrej Pevec**: Syntheses and Solid-State and Solution Structures of $[\text{Ba}\{(\text{C}_5\text{Me}_5)_2\text{Ti}_2\text{F}_7\}_2(\text{hmpa})]$ and $[\text{Ba}_8\text{Ti}_6\text{F}_{30}\text{I}_2(\text{C}_5\text{Me}_5)_6(\text{hmpa})_6][\text{I}_3]_2$, *Inorg. Chem.* 2004, 43, 1250-1256.
- **Andrej Pevec**, Alojz Demšar, Jiri Pinkas, Mirek Necas: Synthesis, spectroscopic and X-ray characterization of new molecular organotitanium(IV) phosphonate, *Inorganic Chemistry Communications* 2008, 11, 5-7.
- **Andrej Pevec**, Alojz Demšar: The variations in hydrogen bonding in hexafluorosilicate salts of protonated methyl substituted pyridines and tetramethylethylenediamine, *Journal of Fluorine Chemistry* 2008, 129, 707-712.
- **Andrej Pevec**, Martina Tekavec, Alojz Demšar: Cation-anion interactions involving hydrogen bonds: Syntheses and crystal structures study of hexafluorotitanate(IV) salts with pyridine and methyl substituted pyridines, *Polyhedron* 2011, 30, 549-555.

ULB FAKULTÄT

SINTEZNE TEHNIKE V ORGANSKI KEMIJI

UČNI NAČRT PREDMETA/COURSE SYLLABUS

Predmet:	Sintezne tehnike v organski kemiji
Course title:	Synthetic techniques in organic chemistry
Članica nosilka/UL Member:	UL FKKT

Študijski programi in stopnja	Študijska smer	Letnik	Semestri	Izbirnost
Kemijska tehnologija, prva stopnja, visokošolski strokovni	Ni členitve (študijski program)	2. letnik, 3. letnik		izbirni

Univerzitetna koda predmeta/University course code:	0099255
Koda učne enote na članici/UL Member course code:	KTSI6

Predavanja /Lectures	Seminar /Seminar	Vaje /Tutorials	Klinične vaje /Clinical tutorials	Druge oblike študija /Other forms of study	Samostojno delo /Individual student work	ECTS
15	15	45 LV			75	5

Nosilec predmeta/Lecturer: izt. prof. dr. Janez Cerkovnik

Vrsta predmeta/Course type: izbirni strokovni/Elective Professional

Jeziki/Languages:

Predavanja/Lectures:	Slovenščina
Vaje/Tutorial:	Slovenščina

Pogoji za vključitev v delo oz. za opravljanje študijskih obveznosti: **Prerequisites:**

Študent oz. kandidat mora imeti predmet opredeljen kot študijsko obveznost.	The course has to be assigned to the student.
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Vsebina:

1) Laboratorijske operacije

Uvod: varnost, uporaba in čiščenje steklovine ter aparatur v organskem laboratoriju, ravnanje z odpadki in nevarnimi snovmi, laboratorijski dnevnik in kemijska literatura.

2) Aparature in tehnike pri kemijskih reakcijah:

sestavljanje aparatur, uporaba inertne atmosfere, merjenje in kontrola osnovnih reakcijskih parametrov, dodajanje reagentov, koncentriranje, izolacija produktov.

3) Tehnike pri sintezi in transformaciji na nekaterih primerih:

- Izomerizacija adamantana
- Priprava cikloheksanola iz cikloheksena in cikloheksanona

Content (Syllabus outline):

Laboratory operations

Introduction: safety, using and cleaning of glassware and apparatus in organic chemistry lab, disposal of waste and hazardous chemicals, the laboratory notebook, and the chemical literature.

Apparatus and techniques in chemical reactions

Assembling the apparatus, use of an inert atmosphere, measurement and control of basic reaction parameters, addition of reagents, concentration, and isolation of products.

Techniques in synthesis and transformations using some examples

Isomerisation of adamantane, preparation of cyclohexanol from cyclohexene and cyclohexanone (oxidation), reactions of cyclohexanol (elimination, substitution,

- Reakcije cikloheksanola (eliminacija, substitucija, redukcija)
- Priprava halidov iz alkoholov (t-butil klorid iz t-butil alkohola)
- Reakcije esterifikacije (salicilna kislina, glukoza)
- Adicija diklorokarbena na alkene z uporabo katalizatorja faznega prenosa
- Grignardova reakcija: priprava alifatskih alkoholov in trifenilmetanola iz benzofenona
- Priprava amidov
- Friedel-Craftsova reakcija
- Priprava in reakcije diazonijevih soli
- Diels-Alderjeva reakcija
- Wittigova reakcija
- Priprava luminola (kemoluminiscenčna reakcija)

4) Tehnike in sinetze, ki vključujejo zaporedje reakcij (opravljanje v okviru seminarske naloge in vaj)

- Tetrafenilciklopentadienon
- Pretvorba steroidov
- 1-bromo-3-kloro-5-jodobenzen
- Sulfanilamid
- Sinteza 2,4-dinitrofenilhidrazina;
- Pretvorba steroidov (holesteril acetat – efekt tekočih kristalov);
- Iskanje spojin in sinteznih postopkov z uporabo literarnih virov in podatkovnih baz

5) Sodobne analizne tehnike v organski kemiji

Spoznavanje in uporaba sodobnih kromatografskih, spektroskopskih (UV-Vis, IR, NMR) in masno-spektrometričnih metod pri zasledovanju in kontroli organskih reakcij ter karakterizaciji produktov.

reduction), preparation of halides from alcohols (*tert*-butyl chloride from *tert*-butyl alcohol), esterification (ethyl acetate), addition of dichlorocarbene to alkene using phase transfer catalyst, the Grignard reaction (preparation of aliphatic alcohols and triphenylmethanol from benzophenone), preparation of amides, preparation and reactions of diazonium salts, Diels-Alder reaction, Wittig reaction, preparation of luminol (chemiluminescent reaction).

Experiments that use a sequence of reactions

(individual work as seminars and synthesis from the literature)

Tetraphenylcyclopentadienone, 1-bromo-3-chloro-5-jodobenzene, 2,4-dinitrophenylhydrazine, transformation of steroids (holesteryl acetate – liquid crystals), searching of compounds and synthetic methods using literature and databases.

Modern analytical techniques in organic chemistry

Understanding and application of modern chromatographic, spectroscopic (UV-Vis, IR, NMR) and mass-spectrometric methods in the pursuit and control of organic reactions and characterization of products.

Temeljna literatura in viri/Readings:

- (1) A. Ault: *Techniques and Experiments for Organic Chemistry*, University Science Books (6th ed.), 1998.
- (2) J. W. Lehman: *Student Lab Companion: Laboratory Techniques for Organic Chemistry*, Prentice Hall, 2008.

Cilji in kompetence:

Cilji predmeta:

Študent nadgradi osnovno teoretično znanje organske kemije s spoznavanjem, razumevanjem in izvajanjem nekaterih najznačilnejših in najpogosteje uporabljenih sinteznih tehnik v organski kemiji ter pridobi osnovne veščine, ki so pri tem potrebne.

Predmetno specifične kompetence:

- uporaba literarnih virov in baz podatkov
- poznavanje sinteznih tehnik v organski kemiji
- priprava in izvedba nekaterih srednje zahtevnih eksperimentov
- poznavanje sodobnih analiznih tehnik

Objectives and competences:

Objectives:

Build upon basic theoretical understanding of organic chemistry with recognizing, understanding, and performing of some most frequent synthetic and analytic techniques in organic chemistry, and gaining of some basic manual skills.

Competences:

- ability to use literature sources and databases;
- familiarity with various synthetic techniques in organic chemistry;
- ability to prepare and perform some intermediate pretentious experiments;
- knowledge of modern analytical techniques.

Predvideni študijski rezultati:

Znanje:

- samostojno obvladovanje osnovnih laboratorijskih operacij in tehnik;

Intended learning outcomes:

Knowledge:

- self-mastery of basic laboratory operations and techniques;

<p>- samostojno načrtovanje, priprava in izvedba enostavnejših kemijskih pretvorb po pravilih varnega dela v laboratoriju.</p> <p>Razumevanje teoretskih osnov enostavnejših kemijskih pretvorb in samostojno odločanje pri uporabi ustreznih laboratorijskih tehnik.</p> <p>Uporaba Študent utrdi osnovo znaje, ki ga je pridobil pri Praktikum iz kemije ter ga nadgradi z tehnikami, ki se najpogosteje uporabljajo ne samo pri izvajanju pretvorb v sintezni laboratorijih, temveč tudi v drugih kemijskih laboratorijih.</p> <p>Refleksija Študent bo na osnovi pridobljenega znanja pridobil občutek za samostojno načrtovanje, pripravo in izvedbo enostavnejših in srednje zahtevnih laboratorijskih operacij pri pretvorbah organskih spojin. Ob tem bo razumel in se zavedal nevarnosti ter tveganj uporabe posameznih tehnik pri laboratorijskem delu.</p> <p>Prenosljive spretnosti Pravilno izbiranje in izvajanje osnovnih laboratorijskih operacij ter smiselna uporaba primernih tehnik.</p> <p>Rokovanje z občutljivimi snovmi in delo v inertni atmosferi. Analiza, sinteza in poročanje o delu in dobljenih rezultatih.</p>	<p>-independent planning, preparation and execution of simple chemical transformations under the rules of safe work in the laboratory.</p> <p>Comprehension of the theoretical basis of simple chemical transformations and independent decision-making in the use of appropriate laboratory techniques.</p> <p>Application Students enhance their basic knowledge gained in Practicum in organic chemistry and upgrade the knowledge with the techniques that are most commonly used, not only in the implementation of transformations in synthetic laboratories, but also in other chemical laboratories.</p> <p>Analysis The student will gain, based on acquired knowledge, the sense of self-planning, preparation and execution of simple and moderately complex laboratory operations in the conversion of organic compounds. At the same time, he/she will understand and be aware of the dangers and risks of using specific techniques in the laboratory work.</p> <p>Skill-transference Ability Proper selection and application of basic laboratory operations and meaningful use of appropriate techniques.</p> <p>Handling of sensitive materials and work in an inert atmosphere.</p> <p>Analysis, synthesis and reporting on the work and the results obtained.</p>
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Metode poučevanja in učenja:

Predavanja, seminarske in laboratorijske vaje

Learning and teaching methods:

Lectures, seminar and laboratory exercises

Načini ocenjevanja:

Opravljenе vaje in seminar (priprava in izvedba sinteze po literaturi) in pisni izpit. 10 (odlično), 9 in 8 (prav dobro), 7 (dobro), 6 (zadostno), 5-1 (nezadostno)

Delež/Weight

Assessment:

Tutorials, seminar (preparation and execution of the synthesis from the literature) and written exam. 10 (excellent), 9 and 8 (very good) 7 (good) 6 (sufficient), 5-1 (inadequate)

Reference nosilca/Lecturer's references:

- **CERKOVNIK, Janez**, PLESNIČAR, Božo. Recent advances in the chemistry of hydrogen trioxide (HOOH). Chemical reviews, ISSN 0009-2665. [Print ed.], 2013, vol. 113, no. 10, str. 7930-7951, ilustr.<http://pubs.acs.org/doi/ipdf/10.1021/cr300512s>, doi: 10.1021/cr300512s. [COBISS.SI-ID 1615407]

- TUTTLE, Tell, **CERKOVNIK, Janez**, KOLLER, Jože, PLESNIČAR, Božo. The search for protonated dihydrogen trioxide (HOOH) : insights from theory and experiment. The journal of physical chemistry. A, Molecules, spectroscopy, kinetics, environment, & general theory, ISSN 1089-5639, 2010, vol. 114, no. 30, str. 8003-8008, doi: 10.1021/jp103882e. [COBISS.SI-ID 34295813]

- **CERKOVNIK, Janez**, PLESNIČAR, Božo, KOLLER, Jože, TUTTLE, Tell. Hydrotrioxides rather than cyclic tetraoxides (tetraoxolanes) as the primary reaction intermediates in the low-temperature ozonation of aldehydes. The case of benzaldehyde. Journal of organic chemistry, ISSN 0022-3263, 2009, vol. 74, no. 1, str. 96-101, doi: 10.1021/jo801594n. [COBISS.SI-ID 30098181]

SPLOŠNA KEMIJA

UČNI NAČRT PREDMETA/COURSE SYLLABUS

Predmet:	Splošna kemija
Course title:	General chemistry
Članica nosilka/UL Member:	UL FKKT

Študijski programi in stopnja	Študijska smer	Letnik	Semestri	Izbirnost
Kemijska tehnologija, prva stopnja, visokošolski strokovni	Ni členitve (študijski program)	1. letnik	zimski	obvezni

Univerzitetna koda predmeta/University course code:	0072027
Koda učne enote na članici/UL Member course code:	KT103S

Predavanja /Lectures	Seminar /Seminar	Vaje /Tutorials	Klinične vaje /Clinical tutorials	Druge oblike študija /Other forms of study	Samostojno delo /Individual student work	ECTS
45	15	15 SV			75	5

Nosilec predmeta/Lecturer: doc. dr. Saša Petriček

Vrsta predmeta/Course type: obvezni/mandatory

Jeziki/Languages:

Predavanja/Lectures:	Slovenščina
Vaje/Tutorial:	Slovenščina

Pogoji za vključitev v delo oz. za opravljanje študijskih obveznosti:

Študent oz. kandidat mora imeti predmet opredeljen kot študijsko obveznost.

Prerequisites:

The course has to be assigned to the student.

Vsebina:

Lastnosti in sestava snovi. Naravoslovna metoda. Osnovni kemijski zakoni in pojmi. Daltonova atomska teorija. Elementi in spojine. Stehiometrija kemijskih reakcij. Eksotermna in endotermna reakcija. Zgradba atoma - osnovni delci. Radioaktivnost in jedrska energija. Valovno-mehanski model atoma. Periodni zakon. Kemijska vez. Ionska vez - ionsko zgrajene snovi. Kovalentna vez. Geometrija molekul. Teorija o odboju valenčnih elektronskih parov. Dipolni moment molekul. Molekulske vezi. Lastnosti plinov, splošna plinska enačba. Tekočine. Viskoznost in površinska napetost. Parni tlak. Trdne snovi. Kovalentni in molekularni kristali. Kovine, kovinska vez.

Content (Syllabus outline):

- Properties and composition of matter. A scientific method. The principal chemistry laws and phenomena. Dalton atomic theory. Elements and compounds. Stoichiometry of chemical reactions. Exothermic and in endothermic reaction.
- Atom and the principal particles. Radioactivity and nuclear energy. The wave-mechanistic model of an atom. The periodic law. Chemical bond. Ionic bond – ionic compounds. Covalent bond. Molecular geometry. A valence electron pair repulsion theory. Molecular dipole moment. Molecular bonds.
- Properties of gases, ideal gas equation.
-Liquids. Viscosity and surface tension. Vapour pressure.
-Solid matter. Covalent and molecular crystals. Metals, metal bond.

<p>Raztopine. Sestava raztopin. Pojavi pri raztapljanju. Topnost, topnostni produkt. Elektroliti. Ionske reakcije.</p> <p>Zakon o vplivu koncentracij. Le Chatelier-ev princip. Hitrost in mehanizem kemijske reakcije. Homogena in heterogena kataliza.</p> <p>Voda in njene lastnosti. Brønstedova definicija kislin in baz. Protolitske reakcije. Vodikov eksponent - pH. Indikatorji. Titracija. Puferske raztopine. Redoks reakcije. Galvanski členi.</p>	<p>- Solutions (composition, dissolution, solubility). Electrolytes. Ionic reactions.</p> <p>-Principles of chemical equilibria. Le Chatelier principle.</p> <p>-Chemical kinetics and mechanism of the chemical reaction. Homogeneous and heterogeneous catalysis.</p> <p>- Water and its properties. Brønsted acid / base definition. Hydrolysis.</p> <p>- pH. Indicators. Titration. Buffer solutions.</p> <p>- Redox reactions. Galvanic cell.</p>
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Temeljna literatura in viri/Readings:

F. LAZARINI in J. V. BREŃČIČ, *Splošna in anorganska kemija*, UL FKKT, Ljubljana, 2004, 557 str. (30%)
 B. Čeh, *Splošna kemija*, UL FKKT, 2018, 389 strani (50 %)

Cilji in kompetence:

Študenti spoznajo temeljne kemijske zakonitosti in se seznanijo z osnovami kvantitativnega obravnavanja lastnosti in zgradbe snovi ter kemijskih procesov. Pridobijo pregledno osnovno znanje o zgradbi, lastnostih in vlogi značilnih skupin kemijskih elementov in njihovih najpomembnejših spojin. Predmet poteka pri predavanjih z demonstracijskimi poskusi in seminarскими ter računskimi vajami.

Objectives and competences:

Students develop fundamental chemical concepts qualitatively and quantitatively. They link composition of matter to its properties. Students practice problem solving and adopt logical approach to solving problems. They are able to think critically using basic concepts of general chemistry.

Predvideni študijski rezultati:

Znanje in razumevanje
 Učna enota se navezuje na osnovno kemijsko znanje, katerega del študenti že obravnavajo v srednji šoli. Poudarek je na kemiji, ki je splošna in se kasneje nadgradi na področjih, ki se jih natančneje obravnava v naslednjih stopnjah študija pri predmetih anorganske, organske, analize in fizikalne kemije. Študenti pri računskih vajah predvsem spoznajo pomen poznavanja in razumevanja osnovnih kemijskih pojmov in zakonitosti pri reševanju nalog in kvantitativnih problemov s področja kemijskega računanja.

Uporaba
 Znanje pridobljeno pri tem predmetu je nepogrešljivo pri nadaljnjem študiju, saj nudi osnovo in podporo vsem kemijskim področjem, s katerimi se študenti srečajo med študijem ter po njem. Študenti se seznanijo s postopki in pristopi pri reševanju računskih nalog in problemov in jih znajo uporabiti pri njihovem reševanju.

Refleksija
 Študenti so na osnovi pridobljenega znanja zmožni razumeti osnovne splošne kemijske pojme in jih uporabiti pri delu, ki ga srečajo v kemijskem laboratoriju. Sposobni so kritično ovrednotiti dobljene rezultate ter jih povezati z količinami, s katerimi se srečajo v vsakdanjem življenju. Prenosljive spretnosti

Intended learning outcomes:

Knowledge and Comprehension
 Additional depth of knowledge and understanding is built on the background obtained in previous education. Mastering core concepts of general chemistry enables successful studying of inorganic, organic, analytical and physical chemistry in higher semesters.

Skills in problem solving are sharpened by practical applications.

Application
 Knowledge of basic chemical concepts is a necessary background for a study in higher semesters. Students adopt logical approach to solving problems and learn some practical applications of general chemistry.

Analysis
 Students are able to link various fundamental concepts to solve complex problems in a chemical laboratory and evaluate results of their work.

Skill-transference Ability
 Skills in problem solving and critical thinking will be useful in almost all courses of higher semesters.

Pri predmetu si bodo študenti pridobili spretnosti za reševanje kvantitativnih teoretičnih nalog in problemov ter lahko pridobljene izkušnje in znanje koristno uporabili tudi pri vseh drugih kemijskih predmetih.	
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Metode poučevanja in učenja: Predavanja, Seminarske vaje ter vaje z računanjem Pri računskih vajah reševanje računskih nalog in kvantitativnih kemijskih problemov s poudarkom na skupinskem delu (sodelovalno učenje); možna organizacija individualne pomoči (tutorstvo).	Learning and teaching methods: Lectures, Seminar, Tutorial Lectures include some illustrative experiments; additional explanations and problem solving in seminars; stoichiometry and calculations in tutorial.
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Načini ocenjevanja:	Delež/Weight	Assessment:
Pisni izpit (50%)	50,00 %	Written exam
ustni izpit Do 10 % je možno pridobiti s sodelovanjem med semestrom (seminarji, naloge ...). Ocene 6-10 pozitivno	50,00 %	oral exam Up to 10% can be obtained through cooperation during the semester (seminars, functions, ...). Reviews 6-10 positive.

Reference nosilca/Lecturer's references: - PETRIČEK, Saša. Octahedral and tetrahedral cobalt(II) sites in cobalt chloride complexes with polyethers. Croat. chem. acta, 2011, vol. 84, no. 4, str. 515-520, doi: 10.5562/cca1747. [COBISS.SI-ID 35780869] - PETRIČEK, Saša, DEMŠAR, Alojz. Syntheses and crystal structures of manganese, nickel and zinc chloride complexes with dimethoxyethane and di(2-methoxyethyl) ether. Polyhedron. [Print ed.], 2010, vol. 29, no. 18, str. 3329-3334, doi: 10.1016/j.poly.2010.09.014. [COBISS.SI-ID 34687493] - DEMŠAR, Alojz, KOŠMRLJ, Janez, PETRIČEK, Saša. Variable-temperature nuclear magnetic resonance spectroscopy allows direct observation of carboxylate shift in zinc carboxylate complexes. Journal of the American Chemical Society, ISSN 0002-7863, 2002, vol. 124, no. 15, str. 3951-3958, graf. prikazi. [COBISS.SI-ID 24242693]
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ŠPORTNA VZGOJA

UČNI NAČRT PREDMETA/COURSE SYLLABUS

Predmet:	Športna vzgoja
Course title:	Physical education
Članica nosilka/UL Member:	UL FKKT

Študijski programi in stopnja	Študijska smer	Letnik	Semestri	Izbirnost
Kemijska tehnologija, prva stopnja, visokošolski strokovni	Ni členitve (študijski program)	1. letnik		izbirni

Univerzitetna koda predmeta/University course code:	0099481
Koda učne enote na članici/UL Member course code:	SPVZG

Predavanja /Lectures	Seminar /Seminar	Vaje /Tutorials	Klinične vaje /Clinical tutorials	Druge oblike študija /Other forms of study	Samostojno delo /Individual student work	ECTS
10	5	60 SV			75	5

Nosilec predmeta/Lecturer: viš. pred. mag. Matej Jamnik

Vrsta predmeta/Course type: obvezni/mandatory

Jeziki/Languages:

Predavanja/Lectures:	Slovenščina
Vaje/Tutorial:	Slovenščina

Pogoji za vključitev v delo oz. za opravljanje študijskih obveznosti:

Študent oz. kandidat mora imeti predmet opredeljen kot študijsko obveznost.

Prerequisites:

The course has to be assigned to the student.

Vsebina:

Predmet ŠV vključuje naslednje vsebine:
Splošno-teoretični del (osnove delovanja človekovega telesa, njegovega gibalnega, srčno-žilnega in dihalnega sistema, psihomotorične in funkcionalne sposobnosti, športno-gibalna aktivnost kot preventivna in kurativna dejavnost za ohranjanje in utrjevanje zdravja, osnove zdravega prehranjevanja in regulacije telesne teže ter drugih medicinskih vidikov športa, metode preverjanja in ugotavljanja stanja psihomotoričnih in funkcionalnih sposobnosti).
Specialno-teoretični del je vezan na izbrano športno panogo (posebnosti športne panoge in njen vpliv na človeka, tehnika, taktika in pravila, osnove telesne in tehnično taktične priprave) in se izvaja skozi praktične vaje.

Content (Syllabus outline):

Subject includes the following contents:
The general theoretical part (base functioning human body, its movement, cardiovascular and respiratory system, psychomotor, and functional abilities, sports and physical activity as a preventive and curative activity for the maintenance and reinforcement of health, the basics of healthy eating and body weight regulation, and other medical aspects of sports, methods of verification and identification and functional state of psychomotor skills).
Special theoretical part is linked to a selected sport disciplines (specialties of sport disciplines and its impact on human body, technique, tactics and rules, basic physical, technical and tactical preparation) and is carried out through practical exercises.

<p>Praktične vaje: študentje izbirajo med ponujenimi športnimi panogami v okviru ponujenih dveh modulov.</p> <p>Prvi modul: <i>Izobraževalni program</i> Vsebine izobraževalnega programa športa so športne panoge, ki se izvajajo v različnih oblikah in na več stopnjah zahtevnosti. Obsega programe učenja in izpopolnjevanja v izbranih športnih panogah (ob zahtevanih dopolnilih je lahko program izpopolnjevanja primeren tudi za usposabljanje za strokovno delo v športu), ki se izvaja kontinuirano preko obeh semestrov ali v zgoščenih oblikah.</p> <p>Drugi modul (lahko le kot dopolnilo prvemu modulu): <i>Tekmovalni program</i> Tekmovalni program obsega vsa fakultetna, univerzitetna, meduniverzitetna in mednarodna tekmovanja (evropska in svetovna univerzitetna tekmovanja ter univerziade) v izbranih športnih panogah in je sestavni del dejavnega sožitja študentov, učiteljev in drugih pripadnikov univerze z namenom sodelovanja znotraj fakultet in univerz. V ta program se vključujejo predhodno selekcionirani študenti.</p>	<p>Practical exercises: The student choose between variety of sport disciplines, that are offered in four modules. Student who selected sport programme with 5 credit point will have to carry out additional two half-day trip from the range of outdoor sport activities.</p> <p>The first module: <i>Educational programs</i> Content of educational programs are sport disciplines, to be implemented in different ways and at several levels of complexity. These consist of learning and training in the selected sport disciplines (with the required supplements, the program may be suitable for advanced training for professional work in sport) to be carried out continuously over two semesters or in concentrated forms.</p> <p>The second module (can also be used as a complement to the first module): <i>Competitive programs</i> Competitive programs include all the faculty, the university, inter-university and international competitions (the European and global academic competition, and universiade) in selected sport disciplines are an integral part of the active co-existence of students, teachers and other members of the university to cooperate within the faculties and universities.</p>
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Temeljna literatura in viri/Readings:

Splošna literatura za tiste segmente, ki so skupne vsem programom (vedenja o vplivu športne aktivnosti na zdrav način življenja in vzdrževanja psihofizičnega ravnovesja ter ohranjanja delovnih sposobnosti)/General bibliography for those segments that are common to all programs.

Obvezna literatura/ Mandatory readings:

1. Pavlin T. (2005). Zanimanje za šport je prodrlo med SLOVENCİ že v široke kroge. Ljubljana: Fakulteta za šport UL, Inštitut za šport, 143 str. (44%)
2. Društvo za zdravje srca in ožilja Slovenije (2000). Lepota gibanja tudi za zdravje (izbrana poglavja). Ljubljana: Društvo za zdravje srca in ožilja Slovenije, 336 str. (25%)
3. Rotovnik-Kozjek, N. (2004). Gibanje je življenje (izbrana poglavja). Ljubljana: Domus, 238 str. (24%)
4. Bračič M. (2006). Razvijanje moči s prostimi utežmi v košarki. Ljubljana: Fakulteta za šport UL, Inštitut za šport, 119 str. (8%)

Priporočljiva literatura/ Recommended readings:

1. Berčič, H. et al. (2001). Šport v obdobju zrelosti. Ljubljana: Fakulteta za šport UL, Inštitut za šport, 210 str.
2. Tušak M., Marinšek M., Tušak M. (2009). Družina in športnik. Ljubljana: Fakulteta za šport UL, Inštitut za šport, 224 str.
3. CINDI Slovenija (2002). Krepimo zdravje z gibanjem in zdravo prehrano (mednarodna konferenca - Radenci). Ljubljana: CINDI Slovenija, 177 str.
4. Francis, P. R. (1996). Real exercise for real people : Finding your optimum level of physical activity for a life time of healty living. Rocklin: Prima Pub,178 str.
5. Inštitut za varovanje zdravja Republike Slovenije (2000). Gibanje za zdravje (svetovni dan zdravja). Ljubljana: Inštitut za varovanje zdravja Republike Slovenije, 85 str.
6. Pokorn, D. (1988). Gorivo za zmagovalce - prehrana športnika in rekreativca. Ljubljana: Inštitut za varovanje zdravja Republike Slovenije, 153 str.
7. Ušaj, A. (1997). Kratek pregled osnov športnega treniranja. Ljubljana: Fakulteta za šport UL, Inštitut za šport, 299 str.

Specifična literatura glede na izbrane programe po posameznih športnih panogah oziroma druge programe/
The specific literature in relation to the selected programs for individual sport disciplines or other programs.

Cilji in kompetence:

Objectives and competences:

<p>Cilji:</p> <ul style="list-style-type: none"> • Ozaveščanje o vrednotah športa in preko tega vplivanje na oblikovanje pozitivnih stališč do športa in navajanje na zdrav način življenja. • Odpravljanje in preprečevanje posledic pomanjkanja gibanja oz. skrb za izboljšanje psihofizičnih sposobnosti, krepitev zdravja in ustvarjalno izrabo prostega časa. • Izpopolnjevanje znanja v izbranih športnih panogah. <p>Kompetence:</p> <ul style="list-style-type: none"> • Sposobnost samostojne skrbi za zdrav način življenja skozi športno- gibalno aktivnost. • Pripravljenost in sposobnost samostojnega vključevanja v organizirane ali neorganizirane oblike športnega udejstvovanja v novih študijskih ali delovnih okoljih. • Oblikovanje trajnega pozitivnega odnosa do športne dejavnosti in trajne skrbi za ohranjanje zdravja in delovnih sposobnosti. • Racionalno vgrajevanje športa v način življenja. • Promocija in uveljavljanje fakultete in univerze. 	<p>Objectives:</p> <ul style="list-style-type: none"> • Awareness of the values of sport and through this influence on the creation of positive attitudes towards sport and referencing to a healthy lifestyle. • Removing and preventing the consequences of the lack of movement or concern for the improvement of psychophysical abilities, health promotion and the creative use of leisure time. • Learning and training selected sport disciplines. <p>Competences:</p> <ul style="list-style-type: none"> • Ability to self-care for a healthy lifestyle through sport activities. • The willingness and ability of self-involvement in organized or unorganized forms of sport participation in the new study or work environment. • Creation of a long term positive attitude towards sport activities and a lasting concern for the preservation of health and working ability. • Racional incorporation sport in the way of life. • Promotion and enforcement of a faculty and a university.
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<p>Predvideni študijski rezultati:</p> <p>Znanje in razumevanje Osvajanje vsebin športa kot temeljev, ki omogočajo kvaliteto življenja.</p> <p>Uporaba Uporaba pridobljenega znanja za kompenzacijo vsakodnevnih stresov med študijem.</p> <p>Refleksija Uporaba pridobljenega znanja za kompenzacijo vsakodnevnih stresov v poklicu in družini.</p> <p>Prenosljive spretnosti Niso vezane le na en predmet.</p>	<p>Intended learning outcomes:</p> <p>Knowledge and Comprehension Conquering content sport as a foundation to enable the quality of life.</p> <p>Application Use the knowledge acquired to compensate for the daily stress during the study.</p> <p>Analysis Use the knowledge acquired to compensate for the daily stress in the profession and family.</p> <p>Skill-transference Ability They are not tied to just one subject</p>
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<p>Metode poučevanja in učenja:</p> <ul style="list-style-type: none"> - predavanja, - vaje, - skupinske in individualne konzultacije. 	<p>Learning and teaching methods:</p> <ul style="list-style-type: none"> - Lectures. - Tutorial - Practical training. - Group and individual consultation.
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Načini ocenjevanja:	Delež/Weight	Assessment:
Kolokvij o vplivu športne aktivnosti na zdrav način življenja in vzdrževanja psihofizičnega ravnovesja ter ohranjanja delovnih sposobnosti (20%).	20,00 %	impact of sports on healthy lifestyles and maintenance of psychophysical balance and the maintenance of work ability.
Izdelava seminarske naloge iz splošno-teoretičnih vsebin (30%)	30,00 %	Preparation of a seminar paper from general theoretical content.
Aktivna prisotnost na praktičnih vajah v izbranem vadbenem programu (75% prisotnost). Ocenjevalna lestvica: 6-10 (pozitivno), 5 (negativno).	50,00 %	Active participation in the exercises in the practical training program selected (75% presence). Assessment scale: 6-10 (positive), 5 (negative).

Reference nosilca/Lecturer's references:

Jamnik, M. (2022). How to increase sports activities of students at the University of Ljubljana after the analysis of the Bologna reform? *Physical education and sport through the centuries : journal for history, philosophy, theory and methodology*, vol. 9, iss. 2, str. 58-68

Jamnik, M. (2022). Dual career of athletes at Slovenian universities. *Zbornik radova = book of proceedings*. Lepasavić: Univerzitet u Prištini, Fakultet za sport i fizičko vaspitanje: = University of Priština, Faculty of Sport and Physical Education, str. 32-37

Jamnik, M. (2021). Differences between students who have been included in sports programs and students who have not been included in sports programs during the COVID 19 pandemic. *Zbornik radova = book of proceedings*. Lepasavić: Univerzitet u Prištini, Fakultet za sport i fizičko vaspitanje: = University of Priština, Faculty of Sport and Physical Education, str. 331-339

ULBENK

TEHNOLOGIJA PREMAZOV

UČNI NAČRT PREDMETA/COURSE SYLLABUS

Predmet:	Tehnologija premazov
Course title:	Coatings Technology
Članica nosilka/UL Member:	UL FKKT

Študijski programi in stopnja	Študijska smer	Letnik	Semestri	Izbirnost
Kemijska tehnologija, prva stopnja, visokošolski strokovni	Ni členitve (študijski program)	2. letnik, 3. letnik		izbirni

Univerzitetna koda predmeta/University course code:	0072056
Koda učne enote na članici/UL Member course code:	KTSI32

Predavanja /Lectures	Seminar /Seminar	Vaje /Tutorials	Klinične vaje /Clinical tutorials	Druge oblike študija /Other forms of study	Samostojno delo /Individual student work	ECTS
45	15	15 LV			75	5

Nosilec predmeta/Lecturer: viš. pred. dr. Branko Alič

Vrsta predmeta/Course type: izbirni strokovni/Elective Professional

Jeziki/Languages:	Predavanja/Lectures:	Slovenščina
	Vaje/Tutorial:	Slovenščina

Pogoji za vključitev v delo oz. za opravljanje študijskih obveznosti:

Študent oz. kandidat mora imeti predmet opredeljen kot študijsko obveznost.

Prerequisites:

The course has to be assigned to the student.

Vsebina:

Temeljna vsebinska področja predmeta so:

- formiranje premaznih filmov;
- tokovne lastnosti organskih premazov;
- mehanske lastnosti organskih premazov;
- stabilnost in odpornost organskih premazov na vplive okolja;
- adhezija;
- korozijska zaščita premazov,
- lateksi,
- pregled veziv in njihove lastnosti,
- topila,
- barva in pigmenti;
- pigmentne disperzije;
- načini aplikacije organskih premazov,
- defekti premaznega filma;

Content (Syllabus outline):

Formation of organic coatings, flow characteristics, mechanical properties, stability and resistance to environmental effects, adhesion, corrosion protection, latexes, solvents, colour and pigments, pigment dispersions, ways of applications of organic coatings, organic coating defects, formulations and preparations, scale-up and transfer of technology to the production.

<ul style="list-style-type: none"> - formulacije in priprava organskih premazov (izbrani primeri); - povečevalni kriteriji in prenos tehnologije v proizvodnjo. 	
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Temeljna literatura in viri/Readings:

<p>- Z. W. Wicks, F. N. Jones, S. P. Pappas, Organic Coatings: Science and Technology, 2. izdaja, John Wiley & Sons, Inc., New York, 1999, 595 str., (60 %).</p> <p>Dopolnilna literatura:</p> <p>- T. C. Patton, Paint Flow and Pigment Dispersion: A Rheological Approach to Coating and Ink Technology, 2. izdaja, John Wiley & Sons, Inc., New York, 1979, 615 str.</p>
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Cilji in kompetence:

<p>Cilj predmeta je, da študentje osvojijo različne stopnje razvoja organskih premazov.</p> <p>Študentje pri predmetu pridobijo naslednje specifične kompetence:</p> <ul style="list-style-type: none"> - poznavanje nastanka oz. formiranja premaznega filma; - poznavanje tokovnih lastnosti različnih premazov; - poznavanje ključnih lastnosti premazov in premaznih filmov z vidika njihove uporabe; - poznavanje premaznih komponent in njihove vloge; - poznavanje načina aplikacije premazov; - razumevanje izbranih primerov formulacij in priprave organskih premazov; - razumevanje povečevalnih kriterijev in prenos tehnologije v proizvodnjo. 	<p>Objectives and competences:</p> <p>Understanding the formation of coatings.</p> <p>Understanding flow characteristics of various coatings. Knowing key characteristics of coatings from the application perspective. Knowing coating components and their roles. Knowing the application of coatings. Understanding selected cases of formulation and preparation of organic coatings.</p> <p>Understanding the scale-up criteria and transfer of a technology to production.</p>
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Predvideni študijski rezultati:

<p>Znanje in razumevanje</p> <p>Študent osvoji znanja o lastnostih organskih premazov in njihovih filmov. Razume nastanek premaznega filma. Razume vpliv sestave premaza na lastnosti premaza in premaznega filma. Pozna osnovno formulacijo premaza. Pozna povečevalne kriterije za prenos tehnologije v proizvodni proces.</p> <p>Uporaba</p> <p>Pridobljena znanja je študent sposoben uporabiti pri svojem raziskovalnem delu na področju razvoja in optimizacije organskih premazov.</p> <p>Refleksija</p> <p>Študent je sposoben sintetizirati znanja s področij polimerne kemije, polimernih materialov ter produktnega inženirstva.</p> <p>Prenosljive spretnosti</p> <p>Študent je sposoben uporabljati tujo in domačo strokovno literaturo. Sposoben je samostojno sklepati, definirati problem, postavljati zaključke in problem reševati. Sposoben je zbirati in obdelovati podatke, predstaviti rezultate v pisni in ustni obliki.</p>	<p>Intended learning outcomes:</p> <p>Knowledge and Comprehension</p> <p>Understanding different levels in the development of organic coatings.</p> <p>Application</p> <p>Student is able to apply the knowledge at independent research and development work in the area of organic coatings development and optimization.</p> <p>Analysis</p> <p>Student is able of synthesis of polymer chemistry, polymer materials and product engineering scientific fields.</p> <p>Skill-transference Ability</p> <p>Ability to identify and solve problems, to collect and interpret data, to analyse results critically and to synthesize knowledge.</p>
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Metode poučevanja in učenja:

<p>Predavanja, seminarji.</p>	<p>Learning and teaching methods:</p> <p>Lectures, seminars, laboratory practice.</p>
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Načini ocenjevanja:**Delež/Weight Assessment:**

(a) Poročila za vaje in zagovor vaj (1/3 ocene) (b) Pisni in ustni izpit (2/3 ocene)
Ocene: 6-10 pozitivno

Reference nosilca/Lecturer's references:

ŠEBENIK, Urška, KRAJNC, Matjaž, ALIČ, Branko, LAPASIN, Romano. Ageing of aqueous TEMPO-oxidized nanofibrillated cellulose dispersions : a rheological study. *Cellulose*, ISSN 0969-0239, 2018, vol. , iss. , str. 1-15, doi: [10.1007/s10570-018-2128-1](https://doi.org/10.1007/s10570-018-2128-1).

RUČIGAJ, Aleš, ALIČ, Branko, KRAJNC, Matjaž, ŠEBENIK, Urška. Investigation of cure kinetics in a system with reactant evaporation : epoxidized soybean oil and maleic anhydride case study. *European Polymer Journal*, ISSN 0014-3057. 2014, vol. 52, no. 1, str. 105-116, doi: [10.1016/j.eurpolymj.2014.01.009](https://doi.org/10.1016/j.eurpolymj.2014.01.009).

RUČIGAJ, Aleš, ALIČ, Branko, KRAJNC, Matjaž, ŠEBENIK, Urška. Curing of bisphenol A-aniline based benzoxazine using phenolic, amino and mercapto accelerators. *Express polymer letters*, ISSN 1788-618X, 2015, vol. 9, no. 7, str. 647-657, doi: [10.3144/expresspolymlett.2015.60](https://doi.org/10.3144/expresspolymlett.2015.60).

KAJTNA, Jernej, ALIČ, Branko, KRAJNC, Matjaž, ŠEBENIK, Urška. Influence of hydrogen bond on rheological properties of solventless UV crosslinkable pressure sensitive acrylic adhesive prepolymers. *International journal of adhesion and adhesives*, ISSN 0143-7496. 2014, vol. 49, no. 1, str. 103-108, doi: [10.1016/j.ijadhadh.2013.12.016](https://doi.org/10.1016/j.ijadhadh.2013.12.016). [COBISS.SI-ID [1664047](https://www.cobiss.si/id/1664047)]

ALIČ, Branko, ŠEBENIK, Urška, KRAJNC, Matjaž. Microencapsulation of butyl stearate with melamine-formaldehyde resin : effect of decreasing the pH value on the composition and thermal stability of microcapsules. *Express polymer letters*, ISSN 1788-618X, 2012, vol. 6, no. 10, str. 826-836, doi: [10.3144/expresspolymlett.2012.88](https://doi.org/10.3144/expresspolymlett.2012.88).

ALIČ, Branko, ŠEBENIK, Urška, KRAJNC, Matjaž. Differential scanning calorimetric examination of melamine-formaldehyde microcapsules containing decane. *Journal of applied polymer science*, ISSN 0021-8995. 2011, vol. 119, no. 6, str. 3687-3695, doi: [10.1002/app.33077](https://doi.org/10.1002/app.33077)

VOJAŠKA DELOVANJA

UČNI NAČRT PREDMETA/COURSE SYLLABUS

Predmet:	Vojaška delovanja
Course title:	Military operations
Članica nosilka/UL	UL FKKT
Member:	

Študijski programi in stopnja	Študijska smer	Letnik	Semestri	Izbirnost
Kemijska tehnologija, prva stopnja, visokošolski strokovni (od študijskega leta 2024/2025 dalje)	Ni členitve (študijski program)	3. letnik	poletni	izbirni

Univerzitetna koda predmeta/University course code: 0643673

Predavanja /Lectures	Seminar /Seminar	Vaje /Tutorials	Klinične vaje /Clinical tutorials	Druge oblike študija /Other forms of study	Samostojno delo /Individual student work	ECTS
80	50	40 SV			100	9

Nosilec predmeta/Lecturer: Doc. dr. Andrej Androjna

Vrsta predmeta/Course type: Obvezni/Mandatory

Jeziki/Languages:

Predavanja/Lectures:	Slovenščina
Vaje/Tutorial:	Slovenščina

Pogoji za vključitev v delo oz. za opravljanje študijskih obveznosti:

Študent mora imeti predmet opredeljen kot študijsko obveznost.

Prerequisites:

The course has to be assigned to the student.

Vsebina:

- zgodovinski razvoj vojskovanja in vojaških ved;
- vojna, vojaške operacije, oboroženi spopadi;
- okolja bojevanja;
- doktrine kopenskih, zračnih in pomorskih sil ter načela delovanja združene taktične enote;
- kopensko vojskovanje; koncept bojnega delovanja voda, čete, bataljona, samostojno in znotraj enote združene taktične ravni;
- vojaške zmogljivosti in oborožitveni sistemi.

Content (Syllabus outline):

- historical development of warfare and military sciences;
- war, military operations, armed conflicts;
- warfighting environments;
- the doctrines of the land, air and naval forces and the principles of operation of the joint tactical unit;
- land warfare; the concept of combat operations of platoon, troops, battalions, independently and within a unit of the joint tactical level;
- military capabilities and armament systems.

Temeljna literatura in viri/Readings:

- Furlan, B. (2006). Bojno delovanje: skripta. Ljubljana: Poveljstvo za doktrino, razvoj, izobraževanje in usposabljanje.

2. Humar, D. (1996). Pehotna (gorska, motorizirana) četa-vod: taktični priročnik za bojevanje. Uprava za razvoj.
3. Škerbinc, M., Božič, D., Zakrajšek, P., & Petek, A. (2007). Lahki pehotni (motorizirani) bataljon: navodilo. Poveljstvo za doktrino, razvoj, izobraževanje in usposabljanje;
4. Priročnik za poveljnike oddelkov, MORS, Ljubljana, 1998;
5. Žabkar, A. (2007). Pehotna oborožitev in oprema: stanje in smeri razvoja. Defensor.

Cilji in kompetence: Pri predmetu se študenti naučijo abstraktnega razmišljanja in dobijo znanja za reševanje taktičnih nalog, v bojnem in nebojnem delovanju ter kritičnega vrednotenja in načrtovanja uporabe vojaških zmogljivosti ter oborožitvenih sistemov.	Objectives and competences: The students learn abstract thinking and gain knowledge to solve tactical tasks in combat and non-combat operations, and to critically evaluate and plan the use of military capabilities and armament systems.
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Predvideni študijski rezultati: Po uspešnem dokončanju predmeta bodo študenti znali: <ul style="list-style-type: none"> • razložiti osnovne ideje in pojme iz vsebin temeljnih vojaških ved (taktike, vojaške geografije, vojaške tehnologije) in razvoja prakse vojaškega delovanja; • analizirati in ovrednotiti okolje delovanja vojaških enot; • analizirati vojaške zmogljivosti in oborožitvene sisteme; • uporabiti metode modeliranja bojnega delovanja na taktični ravni. 	Intended learning outcomes: Upon successful completion of the course, students will be able to: <ul style="list-style-type: none"> • explain basic ideas and concepts from the contents of basic military sciences (tactics, military geography, military technologies) and the development of military practice; • analyze and evaluate the operating environment of military units; • analyze military capabilities and armament systems; • use combat modelling methods at the tactical level.
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Metode poučevanja in učenja: <ul style="list-style-type: none"> • Predavanja • Seminarji • Vaje Vaje se izvedejo po načrtu: 40 h: lab. vaje v računalnici (10 študentov/skupino).	Learning and teaching methods: <ul style="list-style-type: none"> • Lectures • Seminars • Tutorials The tutorials are carried out according to plan: 40 h: Lab. computer (10 students / group).
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Načini ocenjevanja:	Delež/Weight	Assessment:
Kolokvij (esejski tip)	20,00 %	partial exams
Seminarski nastop	30,00 %	seminars
Pisni izpit	50,00 %	written exam
Ocenjevalna lestvica v skladu z enotno lestvico Univerze v Ljubljani: 6 – 10 opravil izpit, 5 ni opravil izpita.		Grades according to the standard levels of the University of Ljubljana: 6-10 passed, 5 failed.

Reference nosilca/Lecturer's references:
<ol style="list-style-type: none"> 1. ANDROJNA Andrej, BRCKO SATLER Tanja, PAVIĆ Ivica, GREIDANUS Harm. Assessing Cyber Challenges of Maritime Navigation. <i>Journal of Maritime Science and Engineering</i>, 2020, 8 (10), 776. DOI: 10.3390/jmse8100776 2. TWRDY, Elen, ANDROJNA Andrej, PAVLIHA Marko. Proposed model of Coast Guard enhancing maritime security and safety in the Republic of Slovenia. <i>TI=Promet = Traffic & transportation</i>, 2014, Vol. 26, no. 6, str. 497-506. [COBISS.SI-ID 2559075] 3. ANDROJNA, Andrej, TWRDY, Elen. Cyber Threats to Maritime Critical Infrastructure. V: ČALETA, Denis (ur.), POWERS, James F. (ur.). <i>Cyber terrorism and extremism as threat to critical infrastructure protection</i>. E-ed. Ljubljana: Ministry of Defense, Republic of Slovenia: Institute for Corporate Security Studies; Tampa: Joint Special Operations University. 2020, str. 163-170. [COBISS.SI-ID 41510659]

4. ANDROJNA, Andrej. Slovenian contribution to "Dialogue & Cooperation for Maritime Surveillance". V: 7th Regional Seapower Symposium for the Mediterranean and Black Sea Countries. Venice: Italian Navy, 2008. [COBISS.SI-ID 2864739]

UL
ELEKTI

VOJAŠKI MENEDŽMENT, VODITELJSTVO IN VOJAŠKA ETIKA

UČNI NAČRT PREDMETA/COURSE SYLLABUS

Predmet:	Vojaški menedžment, voditeljstvo in vojaška etika
Course title:	Military management, leadership and military ethics
Članica nosilka/UL	UL FKKT
Member:	

Študijski programi in stopnja	Študijska smer	Letnik	Semestri	Izbirnost
Kemijska tehnologija, prva stopnja, visokošolski strokovni (od študijskega leta 2024/2025 dalje)	Ni členitve (študijski program)	3. letnik	poletni	izbirni

Univerzitetna koda predmeta/University course code: 0643674

Predavanja /Lectures	Seminar /Seminar	Vaje /Tutorials	Klinične vaje /Clinical tutorials	Druge oblike študija /Other forms of study	Samostojno delo /Individual student work	ECTS
60	40	40 SV			100	8

Nosilec predmeta/Lecturer: Doc. dr. Andrej Androjna

Vrsta predmeta/Course type: Obvezni/Mandatory

Jeziki/Languages:

Predavanja/Lectures:	Slovenščina
Vaje/Tutorial:	Slovenščina

Pogoji za vključitev v delo oz. za opravljanje študijskih obveznosti: **Prerequisites:**

Študent mora imeti predmet opredeljen kot študijsko obveznost.	The course has to be assigned to the student.
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Vsebina:

- koncept vojaškega voditeljstva, poveljevanja in kontrole;
- metode raziskovanja vojaške veščine in operacijskega raziskovanja;
- proces načrtovanja vojaških aktivnosti;
- štabna kultura;
- upravljanje s človeškimi viri;
- vojaški profesionalni kodeks in etika bojevnika.

Content (Syllabus outline):

- the concept of military leadership, command and control;
- research methods of military skills and operational research;
- the process of planning military activities;
- staff culture;
- human resources management;
- military professional code and warrior ethics.

Temeljna literatura in viri/Readings:

- Priročnik za usposabljanje poveljstev in enot, Mors, Ljubljana (2011);
- Kramar, M., & Cugmas, A. (2019). Vojaška didaktika (Elektronska izd., p. 1 spletni vir (1 datoteka PDF (232))). Center vojaških šol. <http://dk.mors.si/IzpisGradiva.php?id=1106>;

3. Mahmutović, J. (2011). Vojaško voditeljstvo in menedžment: gradivo za 2. letnik. Zavod IRC. http://www.zavod-irc.si/docs/Skriti_dokumenti/Vojasko_voditeljstvo_in_menedzment-Mahmutovic.pdf;

4. Unger, M. (Ed.). (2019). Pravila za štabno delo. Center vojaških šol. <http://dk.mors.si/IzpisGradiva.php?id=919>.

Cilji in kompetence:	Objectives and competences:
Pri predmetu študenti dobijo znanja iz področja metod raziskovanja vojaške veščine ter reševanja nalog voditeljstva in poveljevanja na direktni ravni. Študenti pridobijo osnovna znanja za upravljanje s človeškimi viri.	The students gain knowledge in researching military skills and solving tasks of leadership and command at the direct level. The students acquire basic knowledge of human resource management.

Predvideni študijski rezultati:	Intended learning outcomes:
<ul style="list-style-type: none"> Po uspešnem dokončanju predmeta bodo študenti znali: razložiti in pojasniti osnovne ideje in pojme koncepta poveljevanja in kontrole; opredeliti teorijo vojaškega voditeljstva; opredeliti posebnost etičnih prvin v vojski; opredeliti psihološke teorije o delovanju človeka; izdelati in uporabiti algoritme raziskovanja vojaške veščine; izdelati načrt delovanja enote z uporabo procesov in načel procesa vodenja enot; pripraviti karto in prosojnice za delo. 	<ul style="list-style-type: none"> Upon successful completion of the course, students will be able to: explain the basic ideas and concepts of command and control; define the theory of military leadership; define the specificity of ethical elements in the army; define psychological theories about human functioning; develop and use algorithms for researching military skills; create a unit operation plan using the processes and principles of the unit management process; prepare a map and slides for work.

Metode poučevanja in učenja:	Learning and teaching methods:
<ul style="list-style-type: none"> Predavanja Seminarji Vaje <p>Vaje se izvedejo po načrtu: 40 h: lab. vaje v računalnici (15 študentov/skupino).</p>	<ul style="list-style-type: none"> Lectures Seminars Tutorials <p>The tutorials are carried out according to plan: 40 h: Lab. computer (15 students / group).</p>

Načini ocenjevanja:	Delež/Weight	Assessment:
Kolokvij (esejski tip)	40,00 %	partial exams
Seminar	10,00 %	seminars
Pisni izpit	50,00 %	written exam
Ocenjevalna lestvica v skladu z enotno lestvico Univerze v Ljubljani: 6 – 10 opravi izpit, 5 ni opravi izpita.		Grades according to the standard levels of the University of Ljubljana: 6-10 passed, 5 failed.

Reference nosilca/Lecturer's references:

- ANDROJNA Andrej, BRCKO SATLER Tanja, PAVIĆ Ivica, GREIDANUS Harm. Assessing Cyber Challenges of Maritime Navigation. *Journal of Maritime Science and Engineering*, 2020, 8 (10), 776. DOI: [10.3390/jmse8100776](https://doi.org/10.3390/jmse8100776)
- TWRDY, Elen, ANDROJNA Andrej, PAVLIHA Marko. Proposed model of Coast Guard enhancing maritime security and safety in the Republic of Slovenia. *TI=Promet = Traffic & transportation*, 2014, Vol. 26, no. 6, str. 497-506. [COBISS.SI-ID 2559075]
- ANDROJNA, Andrej, TWRDY, Elen. Cyber Threats to Maritime Critical Infrastructure. V: ČALETA, Denis (ur.), POWERS, James F. (ur.). *Cyber terrorism and extremism as threat to critical infrastructure protection*. E-ed. Ljubljana: Ministry of Defense, Republic of Slovenia: Institute for Corporative Security Studies; Tampa: Joint Special Operations University. 2020, str. 163-170. [COBISS.SI-ID [41510659](https://doi.org/10.3390/jmse8100776)]

4. ANDROJNA, Andrej. Slovenian contribution to "Dialogue & Cooperation for Maritime Surveillance". V: 7th Regional Seapower Symposium for the Mediterranean and Black Sea Countries. Venice: Italian Navy, 2008. [COBISS.SI-ID 2864739]

UL
ELEKTI

VOJAŠKI OBRAMBNI SISTEMI

UČNI NAČRT PREDMETA/COURSE SYLLABUS

Predmet:	Vojaški obrambni sistemi
Course title:	Military defense systems
Članica nosilka/UL	UL FKKT
Member:	

Študijski programi in stopnja	Študijska smer	Letnik	Semestri	Izbirnost
Kemijska tehnologija, prva stopnja, visokošolski strokovni (od študijskega leta 2024/2025 dalje)	Ni členitve (študijski program)	3. letnik	poletni	izbirni

Univerzitetna koda predmeta/University course code: 0643675

Predavanja /Lectures	Seminar /Seminar	Vaje /Tutorials	Klinične vaje /Clinical tutorials	Druge oblike študija /Other forms of study	Samostojno delo /Individual student work	ECTS
40	30				80	5

Nosilec predmeta/Lecturer: Doc. dr. Andrej Androjna

Vrsta predmeta/Course type: Obvezni/Mandatory

Jeziki/Languages:

Predavanja/Lectures:	Slovenščina
Vaje/Tutorial:	

Pogoji za vključitev v delo oz. za opravljanje študijskih obveznosti:

Študent mora imeti predmet opredeljen kot študijsko obveznost.

Prerequisites:

The course has to be assigned to the student.

Vsebina:

- vojaški podsistem nacionalne varnosti Republike Slovenije;
- delovanje Slovenske vojske v sistemu kolektivne obrambe in varnosti;
- zakonitost delovanja vojaške sile v vojni, miru in na operacijah kriznega odzivanja;
- mednarodno humanitarno pravo.

Content (Syllabus outline):

- military subsystem of national security of the Republic of Slovenia;
- the operation of the Slovenian Armed Forces in the system of collective defense and security;
- the legality of the operation of military force in war, peace and crisis response operations;
- international humanitarian law.

Temeljna literatura in viri/Readings:

- Sancin, V., Švarc, D., & Ambrož, M. (2009). Mednarodno pravo oboroženih spopadov. Poveljstvo za doktrino, razvoj, izobraževanje in usposabljanje. <http://dk.mors.si/IzpisGradiva.php?id=610>

Cilji in kompetence:

Pri predmetu študenti dobijo znanja s področja organizacije in delovanja vojaškega podsistema v

Objectives and competences:

Students gain knowledge in the field of organization and operation of the military subsystem to ensure

okviru zagotavljanja nacionalne varnosti. Študenti spoznajo uporabo mednarodnega vojaškega prava in dobijo teoretične osnove za njegovo aplikacijo pri načrtovanju delovanj.	national security. Students learn about applying international military law and get the theoretical basis for its application in action planning.
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Predvideni študijski rezultati: Po uspešnem dokončanju predmeta bodo študenti znali:	Intended learning outcomes: Upon successful completion of the course, students will be able to:
<ul style="list-style-type: none"> • opredeliti in razložiti vojaški podsistem nacionalne varnosti; • opredeliti, razložiti in oceniti indikatorje vojaškega ogrožanja; • uporabiti pravila delovanja vojaške sile v vojni, miru in na operacijah kriznega odzivanja; • upoštevati določila mednarodnega vojaškega prava pri načrtovanju vojaških delovanj. 	<ul style="list-style-type: none"> • define and explain the military national security subsystem; • identify, explain and assess indicators of military threat; • apply the rules of operation of the military force in war, peace and crisis response operations; • take into account the provisions of international military law when planning military operations.

Metode poučevanja in učenja:	Learning and teaching methods:
<ul style="list-style-type: none"> • Predavanja • Seminarji 	<ul style="list-style-type: none"> • Lectures • Seminars

Načini ocenjevanja:	Delež/Weight	Assessment:
Kolokvij	40,00 %	2 partial exams
Seminar	10,00 %	seminars
Pisni izpit	50,00 %	written exam
Ocenjevalna lestvica v skladu z enotno lestvico Univerze v Ljubljani: 6 – 10 opravi izpit, 5 ni opravi izpita.		Grades according to the standard levels of the University of Ljubljana: 6-10 passed, 5 failed.

Reference nosilca/Lecturer's references:
<ol style="list-style-type: none"> 1. ANDROJNA Andrej, BRCKO SATLER Tanja, PAVIĆ Ivica, GREIDANUS Harm. Assessing Cyber Challenges of Maritime Navigation. <i>Journal of Maritime Science and Engineering</i>, 2020, 8 (10), 776. DOI: 10.3390/jmse8100776 2. TWRDY, Elen, ANDROJNA Andrej, PAVLIHA Marko. Proposed model of Coast Guard enhancing maritime security and safety in the Republic of Slovenia. <i>TI=Promet = Traffic & transportation</i>, 2014, Vol. 26, no. 6, str. 497-506. [COBISS.SI-ID 2559075] 3. ANDROJNA, Andrej, TWRDY, Elen. Cyber Threats to Maritime Critical Infrastructure. V: ČALETA, Denis (ur.), POWERS, James F. (ur.). <i>Cyber terrorism and extremism as threat to critical infrastructure protection</i>. E-ed. Ljubljana: Ministry of Defense, Republic of Slovenia: Institute for Corporate Security Studies; Tampa: Joint Special Operations University. 2020, str. 163-170. [COBISS.SI-ID 41510659] 4. ANDROJNA, Andrej. Slovenian contribution to "Dialogue & Cooperation for Maritime Surveillance". V: 7th Regional Seapower Symposium for the Mediterranean and Black Sea Countries. Venice: Italian Navy, 2008. [COBISS.SI-ID 2864739]

ZAGOTAVLJANJE KAKOVOSTI V ANALIZNEM LABORATORIJU

UČNI NAČRT PREDMETA/COURSE SYLLABUS

Predmet:	Zagotavljanje kakovosti v analiznem laboratoriju
Course title:	Quality Assurance in Analytical Laboratory
Članica nosilka/UL Member:	UL FKKT

Študijski programi in stopnja	Študijska smer	Letnik	Semestri	Izbirnost
Kemijska tehnologija, prva stopnja, visokošolski strokovni	Ni členitve (študijski program)	2. letnik, 3. letnik		izbirni

Univerzitetna koda predmeta/University course code:	0072057
Koda učne enote na članici/UL Member course code:	KTSI31

Predavanja /Lectures	Seminar /Seminar	Vaje /Tutorials	Klinične vaje /Clinical tutorials	Druge oblike študija /Other forms of study	Samostojno delo /Individual student work	ECTS
45	15	15 LV			75	5

Nosilec predmeta/Lecturer:	izr. prof. dr. Drago Kočar
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Vrsta predmeta/Course type:	izbirni strokovni/Elective Professional
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Jeziki/Languages:	Predavanja/Lectures:	Slovenščina
	Vaje/Tutorial:	Slovenščina

Pogoji za vključitev v delo oz. za opravljanje študijskih obveznosti:

Študent oz. kandidat mora imeti predmet opredeljen kot študijsko obveznost.	Prerequisites: The course has to be assigned to the student.
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Vsebina:

- Potreba po zanesljivih rezultatih (utemeljitev, družbeno-ekonomski učinek, zahteve naročnika, namen analize)
- Temeljna načela zagotavljanja in nadzora kakovosti (QA/QC, različni standardi in njihove značilnosti, najboljše prakse)
- Odvzem vzorca (definicija vzorčenja, vrste vzorcev, načrt vzorčenja, število vzorcev in velikost vzorca, obravnava vzorca in shranjevanje)
- Priprava za analizo (izbira metode, viri metod, dejavniki presoje in izbire metod, vzroki za nepravilne analize rezultate, validacija metode)
- Meritev (dobra laboratorijska praksa, kalibracija meritve, doseganje metrološke sledljivosti, nadzor kakovosti, okolje, oprema in pribor, kemikalije in potrošnji material, vzdrževanje in kalibracija opreme)

Content (Syllabus outline):

- The need for reliable results (justification, social and economic impact, customers' requirements, purpose of analysis)
- General principles of quality assurance and quality control (QA/QC, different standards and their main features, best practices)
- Sampling (sampling defined, types of samples, sampling plan, numbers and sample size, sample handling and storage)
- Preparation for analysis (selecting the method, sources of methods, factors to consider when selecting a method, reasons for incorrect analytical results, method validation)
- Making measurement (good laboratory practice, calibration of measurement, achieving metrological traceability, quality control, environment, equipment)

<ul style="list-style-type: none"> - Vrednotenje rezultatov (uporabljena statistika, nadzorni grafi, merilna negotovost) - Medlaboratorijske primerjave (sheme, organizacija, uporabljena statistika, pomen, sodelovalne študije) - Dokumentacija in upravljanje z njo (dokumentacija, mnenja in interpretacije) - Upravljanje s kakovostjo (sistem upravljanja, standardi za laboratorije, priročniki kakovosti in druga dokumentacija, nadzor, poročila upravljanja, odgovornost laboratorijskega osebja za kakovost) 	<ul style="list-style-type: none"> and glassware, chemicals and consumables, maintenance and calibration of equipment) - Data treatment (statistics, control charts, measurement uncertainty) - Proficiency testing schemes (organisation, statistics, making the most of participation in proficiency testing schemes, collaborative studies) - Documentation and its management (documentation, opinions and interpretations) - Managing quality (managing system, standards available for laboratories, quality manuals and other documentation, audits, management reviews, responsibility of laboratory staff for quality)
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Temeljna literatura in viri/Readings:

- E. Prichard, V. Barwick, Quality Assurance in Analytical Chemistry, Wiley, 2007, 316 pages.

Cilji in kompetence:

Cilj predmeta je, da se študent usposobi za delovanje v analiznem laboratoriju, ki je skladno z načeli zagotavljanja kakovosti in vodi do veljavnih analiznih rezultatov.

PREDMETNO SPECIFIČNE KOMPETENCE

- Študent razume celovitost sistema zagotavljanja kakovosti, ki temelji na pravilnem izvajanju, nadzoru in dokumentiranju vseh stopenj analiznega procesa.
- Študent razume strategije delovanja, ki vodijo do veljavnih analiznih rezultatov in zna prenesti te strategije na manj kompleksne realne primere.

Objectives and competences:

Student develops skills for quality assurance in analytical laboratory leading to valid analytical results.

COURSE SPECIFIC COMPETENCES

Student understands the quality assurance system which depends on accurate procedures, control measures and documentation in all steps of analytical process.

Student understands strategies which produce valid analytical results and is able to apply these strategies in practice.

Predvideni študijski rezultati:

Znanje in razumevanje

Študent pridobi znanja pomembna za zagotavljanje kakovosti v analiznem laboratoriju. Študent razume sistem in mehanizme zagotavljanja kakovosti.

Uporaba

Študent se usposobi za delovanje znotraj sistema zagotavljanja kakovosti v analiznem laboratoriju.

Refleksija

Študent se zaveda pomena zanesljivih analiznih rezultatov in spoštovanja načel dobre laboratorijske prakse.

Prenosljive spretnosti

Študent razvija odgovornost za zanesljivo delovanje in spoštovanje postavljenih zahtev.

Intended learning outcomes:

Knowledge and Comprehension

Student fosters knowledge which is essential for quality assurance in analytical laboratory. Student understands the quality assurance system and measures.

Application

Student develops skills for operation in the accordance with the system of quality assurance in analytical laboratory.

Analysis

Student fosters the awareness of the importance of the valid analytical results and performance in the accordance with the principles of a good laboratory practice.

Skill-transference Ability

Student develops personal responsibility for reliable performance and following the prescribed procedures.

Metode poučevanja in učenja:

Predavanja, vodeni razgovor, sodelovalno učenje, študij primerov, reševanje problemov.

Laboratorijsko projektno delo: ovrednotenje prispevkov k merilni negotovosti končnega rezultata dobljenega z analizno metodo.

Learning and teaching methods:

Lectures, guided discussions, cooperative learning, case studies, problem solving.

Laboratory project: Evaluation of the uncertainty contributions to the final result obtained by an analytical method.

Načini ocenjevanja:	Delež/Weight	Assessment:
Pisno preverjanje znanja.		Written exam

Reference nosilca/Lecturer's references:

- ŠTEFANE, Bogdan, GROŠELJ, Uroš, SVETE, Jurij, POŽGAN, Franc, **KOČAR, Drago**, BRODNIK ŽUGELJ, Helena. The influence of the quinoline moiety on direct Pd-catalyzed arylation of five-membered heterocycles. European journal of organic chemistry. 23 Jan. 2019, vol. 2019, iss. 2/3, str. 432-441, ilustr. ISSN 1434-193X. [COBISS.SI-ID 1537958339]
- ŠČAVNIČAR, Andrej, ROGELJ, Irena, **KOČAR, Drago**, KÖSE, Sevim, POMPE, Matevž. Determination of biogenic amines in cheese by ion chromatography with tandem mass spectrometry detection. Journal of AOAC International. 2018, vol. 101, no. 5, str. 1542-1547, ilustr. ISSN 1060-3271. [COBISS.SI-ID 1537748419]
- LAVRIČ, Simona, **KOČAR, Drago**, MIHELIČ, Igor, BRAYBROOK, Carl. Accurate mass determination of melamine-formaldehyde synthetic polymers after separation on preparative HPLC. Progress in organic coatings. [Print ed.]. 2015, vol. 81, no. 1, str. 27-34. ISSN 0300-9440. [COBISS.SI-ID 1536399043]

ZELENE TEHNOLOGIJE V SODOBNI DRUŽBI

UČNI NAČRT PREDMETA/COURSE SYLLABUS

Predmet:	Zelene tehnologije v sodobni družbi
Course title:	Green Technologies in Modern Society
Članica nosilka/UL	UL FKKT
Member:	

Študijski programi in stopnja	Študijska smer	Letnik	Semestri	Izbirnost
Kemijska tehnologija, prva stopnja, visokošolski strokovni	Ni členitve (študijski program)	2. letnik, 3. letnik	poletni	izbirni

Univerzitetna koda predmeta/University course code: 0643421

Predavanja /Lectures	Seminar /Seminar	Vaje /Tutorials	Klinične vaje /Clinical tutorials	Druge oblike študija /Other forms of study	Samostojno delo /Individual student work	ECTS
45	30				75	5

Nosilec predmeta/Lecturer: izr. prof. dr. Dan Podjed, izr. prof. dr. Gabriela Kalčikova

Vrsta predmeta/Course type: izbirni strokovni/Elective Professional

Jeziki/Languages:

Predavanja/Lectures:	Slovenščina
Vaje/Tutorial:	

Pogoji za vključitev v delo oz. za opravljanje študijskih obveznosti:

Prerequisites:

Študent mora imeti predmet opredeljen kot študijsko obveznost.	The course has to be assigned to the student.
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Vsebina:

Content (Syllabus outline):

- Uvod v zelene tehnologije in vplivi družbe in tehnike na razvoj zelenih tehnologij.
- Koncept zelenega prehoda v kemijskih in sorodnih tehnoloških procesih iz antropološkega vidika.
- Primerjava obstoječih konceptov in sodobnih zelenih procesov v kemijski in sorodni industriji.
- Tehnološke in procesne smernice za vpeljavo ter integracijo zelenih tehnologij
- Razumevanje prednosti in omejitev zelenega prehoda in njegova implementacija v procesih odločanja znotraj industrijskih sistemov in širše.
- Koncept prehoda od uporabe primarnih virov k sekundarnim ter obnovljivim virom v industrijskih procesih.
- Štirje koraki razvoja trajnostnih tehnologij po meri ljudi in okolja (identifikacija, analiza, interpretacija, testiranje).

- Introduction to green technologies and influence of the development of society and technology on green technologies.
- Concept of green transition in chemical and related technological processes from anthropological point of view.
- Comparison of existing and modern green processes in chemical and related industries.
- Technological and process guidelines for the introduction and integration of green technologies
- Understanding the benefits and limitations of the green transition and its implementation in decision-making processes within industrial systems and beyond.
- The concept of transitioning from the use of primary resources to secondary and renewable resources in industrial processes.
- Four steps to developing sustainable technologies for people and

<ul style="list-style-type: none"> - Pomen družbene in kulturne raznolikosti pri uvajanju zelenih tehnologij. - Družboslovne metode za spoznavanje načinov življenja, kratkoročno vplivanje na vedenje ter dolgoročno spreminjanje navad. 	<p>the environment (identification, analysis, interpretation, testing). - The importance of social and cultural diversity in the adoption of green technologies. - Social science methods for learning about lifestyles, short term influences on behavior, long term influences on habit formation.</p>
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Temeljna literatura in viri/Readings:

<ul style="list-style-type: none"> - A.R. Braden, D.J. Richards. The Greening of Industrial Ecosystems, National Academy Press, 1994. - V. Piemonte, M. De Falco, A. Basile: Sustainable Development of Chemical Engineering Innovative Technologies, Wiley and Sons, 2013. - R. Ehrlich, H. A. Geller, Renewable Energy: A First Course, CRC Press, Taylor & Francis Group, 2017 - D. Podjed idr. Four Steps for the People: People-centred Development Toolkit. Dostopno na: http://people-project.net/wp-content/uploads/2019/12/M2.4_Toolkit.pdf, 2019.

Cilji in kompetence:

<p>Študent spozna sodobne trende, ključne za zeleni prehod v kemijski in sorodnih industrijah. Seznan se z načini vključevanja sodobnih tehnologij v obstoječo ali na novo načrtovano proizvodnjo. Študent je sposoben samostojno uporabiti pridobljena znanja pri reševanju realnih izzivov zelenega prehoda v industriji. Spozna razvoj po meri ljudi in pomen odmika od siceršnje paradigme, tj. strokovnega razvoja v raziskavah in industriji. Seznan se z izdelki in storitvami, ki spodbujajo trajnostne načine življenja.</p>	<h3>Objectives and competences:</h3> <p>The student learns about recent trends that are crucial to the green transition in the chemical and related industries. He/she learns how to integrate modern technologies into existing or newly designed productions. The student is able to independently use the acquired knowledge in solving the challenges of the green transition in industry. He/she learns about development according to people needs and the importance of moving from the usual paradigm, i.e. professional development in research and industry. He/she learns about products and services that support sustainable lifestyles.</p>
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Predvideni študijski rezultati:

<h4>Znanje in razumevanje</h4> <p>Študent pridobi znanja o možnosti uporabe sodobnih tehnologij, ki so ključne pri trajnostnem razvoju in zelenem prehodu. Študent razume možnosti uporabe zelenih tehnologij za različne industrije. To mu omogoča analizo določene problematike in sintezo znanj pri njenem reševanju.</p> <h4>Uporaba</h4> <p>S pridobljenim znanjem je študent sposoben sodelovati pri vpeljavi koncepta zelene tehnologije v industrijski sistem. Sposoben je upoštevati temeljne korake razvoja po meri ljudi in okolja pri razvoju zelenih tehnologij.</p> <h4>Refleksija</h4> <p>Študent kombinira znanje, ki ga pridobi pri predavanjih ter pri pripravi projekta in s tem pridobi kompetence za ovrednotenje podatkov in prenos znanja v raziskovalni in/ali tehnološki proces. Zaveda se družbenega in okoljskega vpliva lastnih odločitev. Razume, kako rešitve, ki nastajajo v industriji, vplivajo na njegovo življenje.</p> <h4>Prenosljive spretnosti</h4> <p>Spretnost uporabe literature in teoretičnega znanja v</p>	<h3>Intended learning outcomes:</h3> <h4>Knowledge and Comprehension</h4> <p>The student acquires knowledge of the ways to use modern technologies that are critical to sustainable development and the green transition. The student understands the opportunities of green technologies for different industries. He/she is able to analyze a specific problem and apply the acquired knowledge to solve the problem.</p> <h4>Application</h4> <p>With the acquired knowledge, the student is able to participate in the implementation of the concept of green technology in the industrial system. He/she will be able to consider the basic development steps in green technologies based on human and environmental needs.</p> <h4>Analysis</h4> <p>The student combines the knowledge acquired during lectures and preparation of a project, thus acquiring skills for evaluation and knowledge transfer in the research and/or technological process. She/he is aware of the social and environmental impact of his own decisions. She/he understands how the solutions created in industry affect her/his daily life.</p>
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praktičnih aplikacijah. Sposobnost kritičnega razmišljanja. Uporaba ustnega in pisnega načina poročanje. Komunikacijske sposobnosti. Sposobnost analize, sinteze in razumevanja tehnoloških rešitev in vplivov na družbo. Etična odgovornost.	Skill-transference Ability Ability to apply literature and theoretical knowledge in practice. Ability to think critically. Application of oral and written reporting methods. Communication skills. Ability to analyse, combine and understand technological solutions and their impact on society. Ethical responsibility.
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Metode poučevanja in učenja: Predavanja, projektno delo.	Learning and teaching methods: Lectures, project.
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Načini ocenjevanja:	Delež/Weight	Assessment:
Pisni izpit	60,00 %	Written exam
Projektno delo (pogoj za pristop k izpitu)	40,00 %	Project (mandatory before exam)

Reference nosilca/Lecturer's references:

- ROZMAN, Ula, **KALČIKOVÁ, Gabriela**, MAROLT, Gregor, SKALAR, Tina, ŽGAJNAR GOTVAJN, Andreja. Potential of waste fungal biomass for lead and cadmium removal : characterization, biosorption kinetic and isotherm studies. Environmental technology & innovation. 2020.

- ŽGAJNAR GOTVAJN, Andreja, **KALČIKOVÁ, Gabriela**. Delamination of plastic-coated waste paper by enzymes of the white rot fungus Dichomitus squalens. Journal of environmental management. 2018.

- **KALČIKOVÁ, Gabriela**, ŽGAJNAR GOTVAJN, Andreja. From dumping to sustainable landfilling : the concept of aerobic landfills. V: JACKSON, Carla H. Landfills and recycling centers : processing systems, impact on the environment and adverse health effects. New York: Nova Science Publishers, 2015.

- ZAVRATNIK, Veronika, **PODJED, Dan**, TRILAR, Jure, HLEBEC, Nina, KOS, Andrej, STOJMENOVA DUH, Emilija. Sustainable and community-centred development of smart cities and villages. Sustainability. 2020, št. 10, 3961, str. 1-17.

- **PODJED, Dan**. Razvoj etnografsko utemeljene tehnološke rešitve. Glasnik Slovenskega etnološkega društva. 2019, 59, št. 1, str. 39-48.

- **PODJED, Dan**. Razvoj rešitev za ravnanje z odpadki po meri ljudi. V: PODJED, Dan (ur.), et al. Nevidno življenje odpadkov. 1. izd. Ljubljana: Založba ZRC, 2022. Str. 125-142.