

UČNI NAČRT PREDMETA / COURSE SYLLABUS

Predmet:	SODOBNI ANORGANSKI MATERIALI IN KATALIZATORJI
Course Title:	MODERN INORGANIC MATERIALS AND CATALYSTS

Študijski program in stopnja Study Programme and Level	Študijska smer Study Field	Letnik Academic Year	Semester Semester
MAG Kemija, 2. stopnja	/	1.	1.
USP Chemistry, 2 nd Cycle	/	1 st	1 st

Vrsta predmeta / Course Type: izbirni strokovni / Elective Professional

Univerzitetna koda predmeta / University Course Code: K2I03

Predavanja Lectures	Seminar Seminar	Vaje Tutorial	Klinične vaje Work	Druge oblike študija	Samost. delo Individual Work	ECTS
15	15	45 LV	/	/	75	5

Nosilec predmeta / Lecturer: doc. dr. Romana Cerc Korošec, prof. dr. Anton Meden / Dr. Romana Cerc Korošec, Assistant Professor; Dr. Anton Meden, Full Professor

Jeziki / Languages: slovenski / Slovenian
Predavanja / Lectures: slovenski / Slovenian
Vaje / Tutorial: slovenski / Slovenian

Pogoji za vključitev v delo oz. za opravljanje študijskih obveznosti: Študent oz. kandidat mora imeti predmet opredeljen kot študijsko obveznost.

Prerequisites: The course has to be assigned to the student.

Vsebina:

Kemija materialov:

- uvod: različni tipi materialov in njihova uporaba;
- vplivi kemijske sestave, molekularne in kristalne (ali amorfne) strukture ter mikrostrukture materialov na njihovo funkcijo, polimorfizem;
- primeri različnih oblik materialov: monoliti, volumenski materiali, monodisperzni delci, tanki filmi, nano delci;
- kemijske osnove in pregled nekaterih postopkov priprave različnih materialov - na primer: reakcije v trdnem, hidrotermalna sinteza, sol-gel metoda, sinteza z uporabo

Content (Syllabus outline):

Materials chemistry:

- introduction: different types of materials and their application;
- the influence of chemical composition, molecular and crystal (or amorphous) structure and microstructure of materials on their function, polymorphism;
- examples of different forms of materials: monoliths, bulk materials, monodispersed particles, thin films, nano-particles.
- chemical basis and overview of some preparation procedures of different materials – for instance: solid-state reactions, hydrothermal synthesis, sol-gel method, microwave synthesis,

mikrovalov, zgorevalna sinteza, koprecipitacija, priprava monodisperznih delcev, tankih filmov, nanodelcev;

- električne, magnetne in optične lastnosti materialov;
- osnove heterogene katalize (mehanizem in kinetika reakcij na površini).

Pregled različnih vrst materialov, njihove priprave, strukture in lastnosti:

- molekule v trdninah, samo-urejanje ob kristalizaciji, vpliv oblike molekul na kristalno strukturo;
- anorganski, in kovinsko – organski polimeri ter kompoziti;
- stekla in keramike, fazni diagrami;
- oksidi in nitridi, polprevodniki (dopirani);
- kovine in zlitine, napake v kristalih in njihove posledice;
- tekoči kristali;

Primeri materialov in njihove uporabe:

- porozni zeolitni in zeolitom podobni katalizatorji;
- materiali za shranjevanje in proizvodnjo energije – komponente gorivnih celic, baterij in akumulatorjev, trdni ionski prevodniki in fotoelektrokemijske celice;
- fotokatalizatorji;
- luminiscentni materiali;
- kromogeni materiali (elektrokromni, gasokromni, fotokromni, termokromni).
- različni nanomateriali.

Vaje: Skupinska izvedba projekta priprave in karakterizacije materiala po podatkih iz literature.

combustion synthesis, co-precipitation, preparation of uniform particles, thin films, nanoparticles;

- electrical, magnetic and optical properties of materials;
- the basis of heterogeneous catalysis (mechanism and kinetics of surface reactions).

Overview of different materials, their preparation, structure and properties:

- molecules in solids, self-assembling during crystallization, the influence of molecule's shape on crystal structure;
- inorganic and metal-organic polymers and composites;
- glasses and ceramic materials, phase diagrams;
- oxides and nitrides, semiconductors (doped);
- metals and alloys, crystal defects and their consequences;
- liquid crystals;

Case-studies of materials and their application:

- porous zeolites and zeolite-like catalysts;
- materials for energy storage and its production – components of fuel cells, batteries, solid ionic conductors, photoelectrochemical cells,
- photocatalysts,
- luminescent materials;
- chromogenic materials (electrochromic, gasochromic, photochromic, thermochromic)
- various nanomaterials.

Practical lab: Student team performs project from preparation to characterization of materials according to data obtained from the literature.

Temeljna literatura in viri / Readings:

- Introduction to Materials Chemistry, H.A. Allcock, Wiley, 2008, 460 strani (30%).
 - Inorganic Materials Chemistry, M.T. Weller, Oxford University Press, 2005, 92 strani (50%).
 - Synthesis of Inorganic Materials, U. Schubert and N. Hüsing, Wiley, 2. Izdaja, 2005, 409 strani (25 %).
- The Basis and Application of Heterogeneous Catalysis, M. Bowker, Oxford University Press, 1998, 90 strani (70%).

Cilji in kompetence:

Cilji: Poglobljeno spoznavanje določenih tipov sodobnih anorganskih materialov in katalizatorjev, sinteznih tehnik in raznovrstnih metod za njihovo karakterizacijo.

Kompetence: Izpeljava celotnega projekta: sinteza in karakterizacije tako katalizatorja kot anorganskega materiala, vključno z literaturnim pregledom in načrtovanjem.

Objectives and Competences:

Objectives: Acquiring in-depth knowledge of certain modern inorganic materials, preparation techniques and characterization methods.

Competences: Students acquire competences to carry out a project involving synthesis and characterization, both for the catalyst or inorganic material, including a literature survey and planning.

Predvideni študijski rezultati:Znanje in razumevanje

Podrobno poznavanje izbranih vrst anorganskih materialov ter katalizatorjev in lasnosti, ki so povezane s strukturo, morfologijo in obliko posameznega materiala. Poznavanje različnih preparativnih tehnik za njihovo pripravo in raznovrstnih metod karakterizacije. Povezava teoretičnega in praktičnega znanja z izvedbo dveh različnih projektnih nalog: sinteza in karakterizacija novega materiala ter sinteza in karakterizacija katalizatorja.

Uporaba

Pridobljeno znanje zna povezati in s pomočjo dosegljive literature načrtovati modificirane sintezne postopke priprave novega materiala oz. katalizatorja. Poznavanje metod karakterizacije, njihovo komplementarnost in dopolnjevanje vodijo v razumevanje vsebine znanstvenih člankov s odročja materialov.

Refleksija

Študent razume in utrdi razumevanje povezave med lastnostmi in sestavo ter zgradbo snovi na teoretičnem nivoju, na dveh konkretnih, samostojno izvedenih primerih pa je sposoben novo znanje uporabiti v praksi, kar vodi v samostojno načrtovanje in karakterizacijo novih materialov oz. katalizatorjev.

Prenosljive spretnosti

Študent zna uporabljati zbirke podatkov,

Intended Learning Outcomes:Knowledge and Comprehension

In-depth knowledge of selected types of inorganic materials and catalysts and their properties, which originate from the structure, morphology and shape of the material. Knowledge of different preparation techniques and characterization methods. Connection between theoretical and practical knowledge through performing two different projects: synthesis and characterisation of a new material and synthesis and characterisation of a catalyst.

Application

Integration of acquired knowledge and planning modifying synthesis routes for the preparation of a new material or catalyst with the aid of available literature. The knowledge of characterisation methods and the ways they complement one another leads to the understanding of the contents of research articles in the field of materials chemistry.

Analysis

The student solidifies the understanding of the connections between properties, composition and structure of material on a theoretical level. Acquired knowledge is then transferred to the practical level and applied to two specific cases. This enables independent planning and characterisation of new materials and catalysts.

Skill-transference Ability

The student is able to use databases, chooses

izbrati najbolj primerne sintezne postopke za pripravo novega materiala oz. katalizatorja in ve, katere metode karakterizacije bo potreboval. Ker vseh metod karakterizacije ni mogoče individualni izvesti, se sooči s »timskim« delom. Rezultate karakterizacije zna kritično ovrednotiti in z njihovo pomočjo optimirati sintezni postopek, da bo dobljeni material izboljšal. Pisno in ustno zna poročati o rezultatih projektne naloge in jih komentirati.

the most appropriate synthesis routes for the preparation of a new material or catalyst and knows which characterisation methods are required. Since not all characterisation methods can be performed individually, group work is required. The student is able to critically evaluate the results obtained from characterisation and uses them as a basis for the optimisation of the synthesis procedures in order to obtain materials with the best properties. The student can report and comment on the results of the project in both written and oral form.

Metode poučevanja in učenja:

- predavanja
- seminar
- izpeljava projektne naloge

Learning and Teaching Methods:

- Lectures
- Seminars
- Execution of one project

Načini ocenjevanja:

Pogoj za pristop k izpitu je oddano poročilo o projektu (zaključene vaje).
Pisni izpit.
Ocene 6-10: pozitivno, ocene 1-5: negativno.

Delež (v %) /

Weight (in %)

Assessment:

Submitted project report (completed practice) is mandatory before the exam.
Written exam.
Marks 6-10: pass, marks 1-5: fail.

Reference nosilca / Lecturer's references:

- R. Cerc Korošec, P. Bukovec, B. Pihlar, A. Šurca Vuk, B. Orel, G. Dražić: Preparation and structural investigations of electrochromic nanosized NiO_x films made via the sol-gel route. - *Solid state ion.* **2003**, *165*, 191-200.
- J. Moškon, R. Dominko, M. Gaberšček, R. Cerc Korošec, J. Jamnik. Citrate-derived carbon nanocoatings for poorly conducting cathode. - *J. Electrochem. Soc.* **2006**, *153*, A1805-A1811.
- P. Umek, R. Cerc Korošec, B. Jančar, R. Dominko, D. Arčon: The Influence of the Reaction Temperature on the Morphology of Sodium Titanate 1D Nanostructures and a Study of Their Thermal Stability. - *J. nanosci. nanotechnol.* **2007**, *9*, 3502-3508.