

UČNI NAČRT PREDMETA / COURSE SYLLABUS

Predmet:	VREDNOTENJE ZNANJA
Course Title:	KNOWLEDGE ASSESSMENT

Študijski program in stopnja Study Programme and Level	Študijska smer Study Field	Letnik Academic Year	Semester Semester
MAG Kemijsko izobraževanje, 2. stopnja	/	2.	4.
USP Chemical Education, 2 nd Cycle	/	2 nd	4 th

Vrsta predmeta / Course Type:

obvezni / Mandatory

Univerzitetna koda predmeta / University Course Code:

IZO225

Predavanja Lectures	Seminar Seminar	Vaje Tutorial	Klinične vaje Work	Druge oblike študija	Samost. delo Individual Work	ECTS
15	/	30 SV	/	/	45	3

Nosilec predmeta / Lecturer:

izr. prof. dr. Jurij Reščič / Dr. Jurij Reščič, Associate Professor

Jeziki / Languages:

Predavanja / Lectures: slovenski / Slovenian

Vaje / Tutorial: slovenski / Slovenian

Pogoji za vključitev v delo oz. za opravljanje
študijskih obveznosti:

Študent oz. kandidat mora imeti predmet
opredeljen kot študijsko obveznost.

Prerequisites:

The course has to be assigned to the student.

Vsebina:

- oblike in instrumenti preverjanja in ocenjevanja znanja kemije
- pogoji izvedbe preverjanja in ocenjevanja znanja
- gradnja bank testov in testnih baterij
- osnovna statistika za vrednotenje posameznih nalog in preizkusa znanja kot celote

Content (Syllabus outline):

- forms and instruments of testing and assessment of chemistry knowledge
- providing conditions for implementing testing
- building a test bank and test batteries
- basic statistics for the analysis of test items and knowledge tests

Temeljna literatura in viri / Readings:

- Bukovec, N., Glažar, S. A. (2006). Naloge iz splošne in anorganske kemije za srednjo šolo. Ljubljana:
DZS. 139 str.
- Sagadin, J. (1999). Programska evalvacija. Sodobna pedagogika, 50 (116), 2, 196-211.
- Sagadin, J. (1991). Razprave iz pedagoške metodologije. Univerza v Ljubljani: Filozofska

fakulteta. 91- 167.

- Fraenkel, J. R. (2006). How to design and evaluate research in education. New York: McGraw-Hill. 26- 66.

Cilji in kompetence:

Spološne kompetence:

- prepoznavanje vsebine in metodike področja
- usposobljenost za preverjanje in ocenjevanje znanja in dosežkov dijakov ter oblikovanje povratnih informacij
- informacijska pismenost
- usposobljenost za pedagoško vodenje razreda in/ali skupine

Specifične kompetence:

- poznavanje specifičnih instrumentov za preverjanje in ocenjevanje kemijskega znanja
- usposobljenost za interpretacijo statističnih podatkov o dosežkih dijakov
- usposobljenost za povezovanje rezultatov preverjanja znanja učencev z učnim procesom

Objectives and Competences:

General competences:

- recognition of the thematic of the field
- ability to apply methods and instruments for testing and knowledge assessment and interpret responses
- information literacy
- ability for pedagogical guidance of the class and/or a group

Specific competences:

- ability to use specific instruments for testing and assessment of chemistry knowledge
- ability to interpret statistical data on student achievements
- ability to relate test results with the knowledge of students and the teaching process

Predvideni študijski rezultati:

Znanje in razumevanje

- pozna oblike in instrumente preverjanja in ocenjevanja znanja
- pozna mednarodne oblike preverjanja naravoslovnega znanja in je sposoben interpretirati rezultate naših dijakov s tujimi
- pozna principe gradnje in uporabe testnih bank ter testnih baterij

Uporaba

- uporabi oblike in instrumente preverjanja in ocenjevanja znanja v šoli
- zna interpretirati rezultate dijakov
- zna uporabiti principe gradnje in uporabe testnih bank ter testnih baterij

Refleksija

- zna kritično vrednotiti pisne in elektronske vire informacij
- zna kritično ovrednotiti kvaliteto svojih dosežkov v primerjavi z dosežki drugih

Prenosljive spretnosti

- predvideti vpliv rezultatov vrednotenja

Intended Learning Outcomes:

Knowledge and Comprehension

- knows forms and instruments of testing and assessment of knowledge
- knows international forms for assessment of natural sciences knowledge and is capable of interpreting the results of our students with foreign ones
- knows principles of building and applying test banks and test batteries

Application

- applies forms and instruments of testing and assessment of knowledge in school
- knows how to interpret results of high school students
- knows how to apply principles of building and applying test banks and test batteries

Analysis

- knows how to critically evaluate written and electronic sources of information
- knows how to critically evaluate the quality of his/her achievements in comparison with others

Skill-transference Ability

- predict the influence of results of the

znanja na usmerjanje pouka

knowledge assessment on conducting the class

Metode poučevanja in učenja:

- predavanja z aktivno udeležbo študentov (razlaga, diskusija, metoda postavljanja vprašanj, skupinsko in delo v parih, individualne domače naloge)
- seminarne vaje (individualna zasnova preizkusa znanja)
- individualne in skupinske konzultacije

Learning and Teaching Methods:

- lectures with active participation of students (interpretation, discussion, method of asking questions, team work and work in pairs, individual homework)
- seminar exercises (individual plan for conducting a test)
- individual and team consultations

Delež (v %) /

Načini ocenjevanja:

Weight (in %)

Assessment:

- seminar (predstavitev nalog in preizkusa znanja) 30 %	30 %	- semínar (presentation of exercises and test) 30 %
- opravljen izpit 70 %	70 %	- positively graded exam 70 %
Ocene v okviru ECTS ocenjevanja na UL (pozitivna ocena je 6 in višje).		Grades according to ECTS grading at UL (positive grade is 6 and higher).

Reference nosilca / Lecturer's references:

1. skrbik sistema Echemtest na FKKT za elektronsko preverjanje znanja
2. BONČINA, Matjaž, CERAR, Janez, GODEC, Andrej (avtor, urednik), HRIBAR, Barbara, JAMNIK, Andrej, LAH, Jurij, LAJOVIC, Andrej, LUKŠIČ, Miha, PODLIPNIK, Črtomir, PRISLAN, Iztok, **REŠČIČ, Jurij**, ŠARAC, Bojan, TOMŠIČ, Matija, VESNAVER, Gorazd. *Fizikalna kemija - praktikum*. 1. izd. Ljubljana: Fakulteta za kemijo in kemijsko tehnologijo, 2012. XXXII, 227 str., ilustr. ISBN 978-961-6756-32-7. [COBISS.SI-ID [261552640](#)]
3. BOHINC, Klemen, **REŠČIČ, Jurij**, DUFRECHE, Jean-François, LUE, Leo. Recycling of uranyl from contaminated water. *The journal of physical chemistry. B, Condensed matter, materials, surfaces, interfaces & biophysical*, ISSN 1520-6106, 2013, vol. 117, iss. 37, str. 10846-10851, doi: [10.1021/jp404822f](https://doi.org/10.1021/jp404822f). [COBISS.SI-ID [4593003](#)].
4. BONČINA, Matjaž, LAH, Jurij, **REŠČIČ, Jurij**, VLACHY, Vojko. Thermodynamics of the lysozyme-salt interaction from calorimetric titrations. *The journal of physical chemistry. B, Condensed matter, materials, surfaces, interfaces & biophysical*, ISSN 1520-6106, 2010, vol. 114, no. 12, str. 4313-4319, doi: [10.1021/jp9071845](https://doi.org/10.1021/jp9071845). [COBISS.SI-ID [33831941](#)].
5. **REŠČIČ, Jurij**, LINSE, Per. Potential of mean force between charged colloids : effect of dielectric discontinuities. *The Journal of chemical physics*, ISSN 0021-9606, 2008, vol. 129, no. 11, art. no. 114505 (9 str.), graf. prikazi. [COBISS.SI-ID [29795333](#)].