

# Direct functionalization of 2-phenylquinazoline

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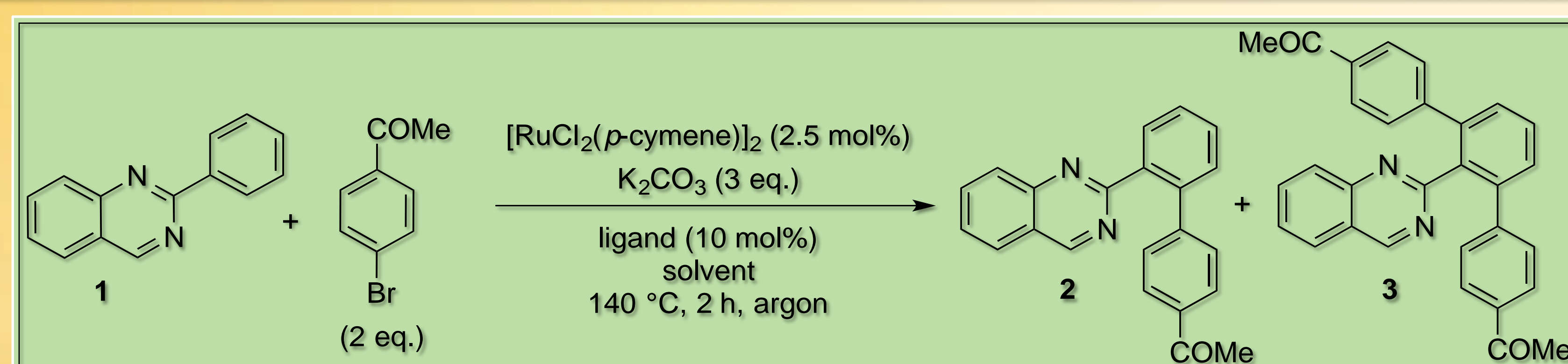
## Introduction

Catalytic transformation of carbon-hydrogen (C-H) bonds has recently become one of the most promising and powerful methods for the construction of carbon-carbon (C-C) bond frameworks.<sup>1</sup> The idea of direct functionalization of C-H bonds is appealing, because it presents an elegant way to obtain desired products by shortening and simplifying reaction pathways.

Chelation-assisted transformation has become a common and effective method for selective functionalization of *ortho* C-H bonds.

For example, quinazoline and its derivatives are well known nitrogen heterocycles often found in many alkaloids and bioactive molecules.<sup>2</sup> Their synthesis present an important value in the pharmaceutical industry.<sup>3</sup>

Therefore, the method of C-H functionalization was carried out on 2-phenylquinazoline.



**Table 1:** C-H activation of 2-phenylquinazoline in the presence of various Ru<sup>II</sup>-catalytic systems.<sup>[a]</sup> **Table 2:** C-H activation of 2-phenylquinazoline in different solvents.<sup>[a]</sup>

ligand					solvent				
example		conversion (%) <sup>[b]</sup>	mono (2) (%)	di (3) (%)	example		conversion (%) <sup>[b]</sup>	mono (2) (%)	di (3) (%)
1	NaOAc	29	25	75	1	acetonitrile	38	50	50
2	PPh <sub>3</sub>	41	71	29	2	toluene	67	38	62
3	glycolic acid	50	40	60	3	dioxane	80	8	92
4	1-naphthoic acid	69	14	86	4	tetrahydrofuran	82	7	93
5	1-phenyl-1-cyclopentanecarboxylic acid (PCCA)	96	7	93	5	2-propanol	92	5	95
6	pivalic acid	96	4	96					
7	adamantane carboxylic acid	97	3	97					

[a] Reaction conditions: **1** (0.5 mmol), 4-bromoacetophenone (1 mmol), [RuCl<sub>2</sub>(*p*-cymene)]<sub>2</sub> (0.0125 mmol), K<sub>2</sub>CO<sub>3</sub> (1.5 mmol), NaOAc (1.5 mmol), solvent (2 mL), 140 °C, 2 h, argon. [b] Conversion of **1** and the ratio of products was determined by <sup>1</sup>H NMR.

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## Conclusion

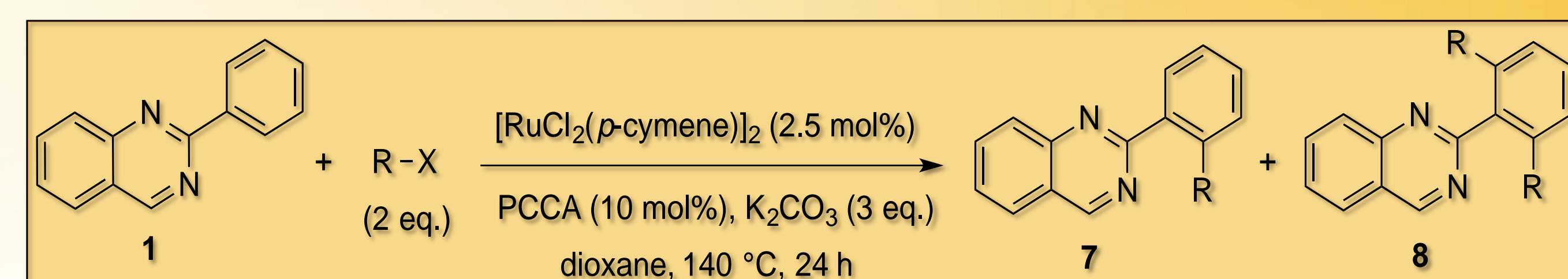
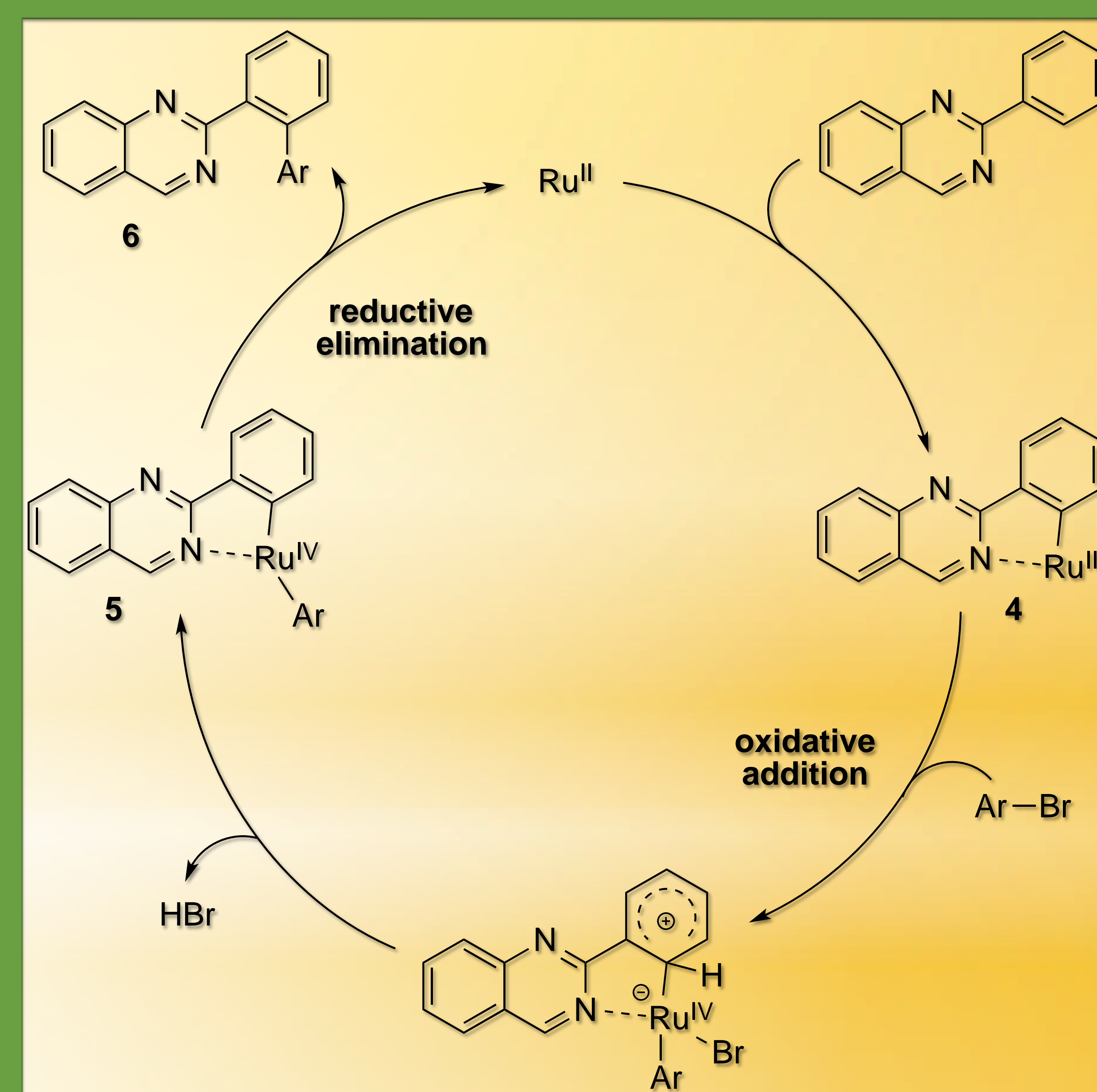
- Direct functionalization of sp<sup>2</sup>(C)-H bonds of 2-phenylquinazoline was carried out under various reaction conditions.
- In the presence of [RuCl<sub>2</sub>(*p*-cymene)]<sub>2</sub>, different ligands and potassium carbonate as the base the *ortho* positions of the quinazoline derivative **1** were successfully functionalized in different solvents.
- With PCCA as the selected ligand and dioxane as the chosen solvent, direct functionalization of 2-phenylquinazoline was carried out with several proelectrophiles which gave mono- and di-substituted products in moderate to excellent yields.

## References

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2. D'yakov, A. L.; Telezhenetskaya, M. V. Quinazoline alkaloids in nature. *Chem. Nat. Compd.* **1997**, *33*, 221-267.
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## Proposed mechanism of direct *ortho* arylation of 2-phenylquinazoline

The presumed reaction mechanism is defined as a catalytic cycle. The ruthenium(II) complex chelates on the nitrogen atom of the quinazoline derivative to form a stable ruthenacycle **4** after which the oxidative addition of the corresponding aryl halide is followed. Elimination of HBr yields the arylated ruthenacycle **5** and the reductive elimination of ruthenium affords the desired product **6**.



**Table 3:** C-H functionalization of 2-phenylquinazoline with selected proelectrophiles.<sup>[a]</sup>

R-X				
example		conversion (%) <sup>[b]</sup>	mono (7) (%)	di (8) (%)
1		17	50	50
2		29	50	50
3		94	10	90
4		96	7	93
5		96	9	91
6		90	15	85
7		29	0	100

[a] Reaction conditions: **1** (0.5 mmol), R-X (1 mmol), [RuCl<sub>2</sub>(*p*-cymene)]<sub>2</sub> (0.0125 mmol), PCCA (0.05 mmol), K<sub>2</sub>CO<sub>3</sub> (1.5 mmol), dioxane (2 mL), 140 °C, 24 h, argon. [b] Conversion of **1** and the ratio of products was determined by <sup>1</sup>H NMR.