

Univerza
v Ljubljani

Fakulteta *za kemijo*
in kemijsko tehnologijo

p.p. 537, Večna pot 113
1001 Ljubljana
telefon: 01 479 80 00
faks: 01 241 91 44
dekanat@fkkt.uni-lj.si



VABILO NA PREDAVANJE
V OKVIRU DOKTORSKEGA ŠTUDIJA
KEMIJSKE ZNANOSTI / INVITATION TO THE
LECTURE WITHIN DOCTORAL PROGRAMME IN
CHEMICAL SCIENCES

Prof. Margarida Costa Gomes

*CNRS Chemistry Laboratory of the
Ecole Normale Supérieure de Lyon, France*

z naslovom / title:

**Porous ionic liquids to improve the separation
and transformation of gases**

**v sredo, 10. 5. 2023 ob 15. uri /
on Wednesday, 10. 5. 2023 at 15.00**

**v predavalnici 1 v 1. nadstropju Fakultete za kemijo in
kemijsko tehnologijo, Večna pot 113 / in lecture room 1,
1st floor at the Faculty of Chemistry and Chemical
Technology, Večna pot 113**

Vljudno vabljeni! / Kindly invited!

Abstract:

Among the alternative sorbents potentially capable of outperforming current separation technologies, in particular gas separations, ionic liquids are promising candidates. Their most attractive feature is the possibility of tuning their physical and chemical properties through proper pairing of anions and cations, which can include reactive groups, enabling the selective absorption of different gases, even at low partial pressures [i].

We will describe ionic liquid-based absorbents which are liquids with permanent porosity [ii], designed to selectively absorb different gases. The absorbents are stable suspensions of metal-organic frameworks (MOFs) in salts whose ion pairs are too voluminous to enter the solid pores [iii]. The increase in gas absorption, when compared with the pure ionic liquids, is proportional to the amount of porous solid in suspension. The thermodynamic analysis of the absorption data, as well as molecular dynamics simulations, show that the driving force for gas absorption by the porous ionic liquids is energetic as well as structural and thus is controlled by gas-solid affinity or by the porous liquid free volume [iv].

We have shown that porous ionic liquids prepared as stable suspensions of ZIF-8 in phosphonium acetate or levulinate salts can selectively absorb carbon dioxide with a capacity more than 100% higher than that of the pure MOF at 1 bar and 303 K [v]. Porous ionic liquids can also be designed to promote the reaction, at mild conditions of temperature and pressure, of the gases absorbed. Carbon dioxide can be catalytically coupled with epoxides to form cyclic carbonates in porous ionic liquids containing alkylphosphonium halides and ZIF-8. The high activity and selectivity observed at atmospheric pressure and room temperature indicate that porous ionic liquids are a promising and sustainable family of sorbents for both separating and transforming gases [vi].

References

[i] Bui *et al. Energy Environ. Sci.* **2018**, *11* (5), 1062–1176.

[ii] Giri *et al. Nature* **2015**, *527*, 216-221,

[iii] Costa Gomes *et al. Angew. Chem. Int. Ed.* **2018**, *57*, 11909-11912.

[iv] Avila *et al. Adv. Mater. Interfaces* **2021**, *8*, 2001982. J. Avila *et al Materials Adv.* **2022**, *3*, 8848-8863.

[v] Avila *et al. Angew. Chem. Int. Ed.* **2021**, *60*, 12876-12882.

[vi] Zhou *et al. Chem. Commun.* **2021**, *57*, 7922-7925.